

high stability.

Atomically precise nanoclusters offer new opportunities for exploring excellent properties and can serve as building blocks with tunable electronic structures for assembling materials and devices [14]. Among them, “superatomic clusters” (namely, “superatoms”) possess outstanding stability and are promising for self-assembling [15, 16]. Indeed, some superatomic cluster assembled nanomaterials have already been synthesized. For instance, Re_6Se_8 clusters can be assembled into a 2D hierarchical semiconductor via strong interaction by substitutionally labile Cl atoms, whose transport properties are tunable from semiconducting to metallic and to superconducting upon n -type doping through Cl dissociation [17–19]. C_{60} together with $\text{Co}_6\text{Se}_8/\text{Cr}_6\text{Te}_8/\text{Ni}_6\text{Te}_8$ clusters were patterned into three-dimensional crystalline array with CdI_2 or NaCl type [20]. On the theoretical side, Guo *et al.* [21] proposed a strategy to construct stable 2D lattices using superatomic tetrahedral clusters (i.e., B_4 , Al_4 , Ga_4 , In_4 , Cr_4 , Mo_4 , and Te_4) covalently linked by oxygen atoms. Du *et al.* [22] examined 27 dimers of endohedral cage clusters and predicted that M@X_{16} ($\text{M} = \text{Ti}, \text{Zr}, \text{or Hf}; \text{X} = \text{Si}, \text{Ge}, \text{or Sn}$) are potential zero-dimensional vdW Lego blocks. Xing *et al.* [23, 24] presented a joint experiment-theory investigation of a rationally designed layered superatomic crystal of $\text{Au}_6\text{Te}_{12}\text{Se}_8$ cubes stacked by noncovalent intercubic quasibonds, finding a sequential-emerged anisotropic triple-cube charge density wave and polarized metallic states below 120 K. Zhao *et al.* [25] designed a series of fullerene-based 1D chains, namely $\text{U}_2\text{C}@C_{80}\text{-M}$ ($\text{M} = \text{Cr}, \text{Mn}, \text{Mo}, \text{and Ru}$) 1D chains, with both ferroelectric and ferromagnetic properties.

ReNCl_4 cluster with nearly ideal C_{4v} symmetry has been synthesized in a facile synthesis from perrhenate, sodium azide and HCl gas [26–28]. Inspired by the aforementioned cluster assemblies with extraordinary properties, herein we design 1D nanowires of ReNX_4 ($\text{X} = \text{F}, \text{Cl}, \text{Br}, \text{and I}$) by cluster self-assembling and extend them to 1D nanowires of MF_5 ($\text{M} = \text{V}, \text{Nb}, \text{and Ta}$) by isoelectronic substitution. The cluster-assembling rule takes into account both the closed electronic shells for ReNX_4 and MF_5 clusters as well as the binding strength between metal and N/F atoms. The molecular orbitals and superatomic characters of these clusters with 40 valence electrons were identified for the first time. Owing to second-order Jahn–Teller effect, these assembled nanowires intrinsically hold superior ferroelectricity with Curie temperature above 300 K, thereby paving a new route for the design of intrinsic 1D ferroelectrics from self-assembling.

2 Computational methods

Density functional theory (DFT) calculations were

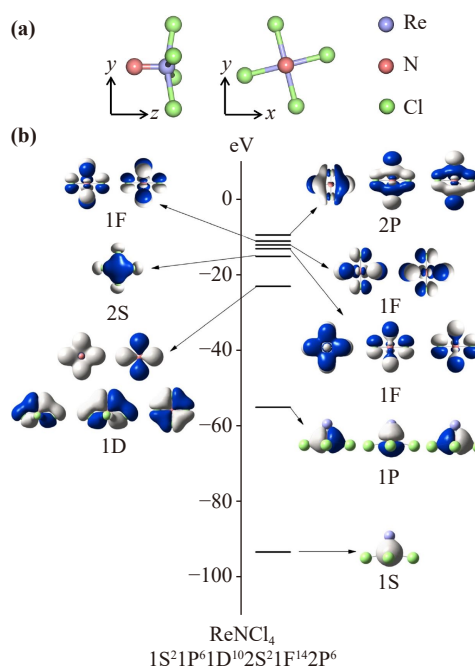


Fig. 1 (a) Atomic structure of ReNCl_4 cluster. (b) Molecular orbitals referred to the vacuum energy level as zero and a classification of orbital pattern into superatomic 1S, 1P, 1D, 2S, 1F and 2P states for ReNCl_4 cluster. Blue and red colors denote positive and negative phases of the wavefunction, respectively. The isosurface value is ± 0.05 a.u. The Re, N and Cl atoms are shown in purple, pink and green colors, respectively.

performed using the Vienna *ab initio* simulation package (VASP) [29], with planewave basis set (energy cutoff of 500 eV), projector augmented wave potentials [30], and Perdew–Burke–Ernzerhof (PBE) exchange–correlation functional [31]. Based on equilibrium structures, a hybrid HSE06 functional [32] was further used to compute the electronic band structures. The convergence criteria for total energy and force were set to 10^{-7} eV and $0.01 \text{ eV}\cdot\text{\AA}^{-1}$, respectively. Uniform \mathbf{k} -point meshes with spacing of $\sim 0.015 \text{ \AA}^{-1}$ were adopted to sample the Brillouin zones. Phonon dispersion analysis was performed using the Phonopy code [33] based on the density functional perturbation theory. The frontier molecular orbitals and natural population of individual ReNX_4 and MF_5 clusters were analyzed using the PBE functional accompanied with 6-311+G(d) [34] and SDD [35] basis sets, implemented in Gaussian16 suite [36].

3 Results and discussion

We firstly investigate the atomic structure and electronic properties of an individual ReNCl_4 cluster. As shown in Fig. 1(a), four equivalent Cl^- atoms and one N^{3-} atom are bonded with one Re^{7+} atom to form a stable N-Re-Cl_4 unit with C_{4v} symmetry, with N–Re and

Table 1 Lattice parameters a , off-center displacement (d_2-d_1 and d_3) of metal along the axial z direction, formation energy E_f , band gaps E_g , potential barrier of E_B , spontaneous polarization P_{FE} , Curie temperature T_C , piezoelectric response e_{11} and pyroelectric response p .

Materials	ReNF ₄	ReNCl ₄	ReNBr ₄	ReNI ₄	VF ₅	NbF ₅	TaF ₅
a (Å)	4.10	4.08	4.04	3.88	4.63	4.23	4.11
d_2-d_1 (Å)	0.74	0.70	0.63	0.41	0.82	0.40	0.20
d_3 (Å)	0.32	0.36	0.35	0.28	0.43	0.34	0.21
E_f (eV)	-0.28	-0.22	-0.28	-0.43	-0.10	-0.12	-0.18
E_g (eV)	4.25 (d)	2.25	1.43	--	4.94 (d)	6.97	6.60
E_B (eV)	0.74	0.81	0.85	--	0.50	0.10	0.09
P_{FE} (10^{-10} C/m)	52.03	25.64	28.58	--	55.5	24.64	36.96
T_C (K)	224	280	321	--	95	86	72
e_{11} (10^{-10} C/m)	30.29	22.86	22.21	--	35.47	31.47	61.28
p (10^{-10} C/(K·m ²))	0.06	0.02	0.02	--	0.13	0.07	0.12

Re–Cl bond length of 1.67 and 2.30 Å, respectively, and Cl–Re–N angle of 81.24°. According to the valence electronic configurations of Re ($5d^56s^2$), N ($2s^22p^3$) and Cl ($3s^22p^5$) atoms, each ReNCl₄ cluster carry totally 40 valence electrons and achieves a closed electronic shell of $1S^21P^61D^{10}2S^21F^{14}2P^6$, which is evidently demonstrated in Fig. 1(b). For the first time, the superatomic characteristics [37] for valence molecular orbitals of ReNCl₄ cluster are identified, suggesting its great potential as a building block to assemble nanomaterial.

According to the geometric feature of ReNCl₄ cluster, Re atom is prone to bond with another N atom in adjacent ReNCl₄ cluster to form distorted corner-sharing ReN₂Cl₄ octahedra without tilt. In this manner, ReNCl₄ units can be self-assembled to construct a ReNCl₄ wire along the z axial direction with P_4 symmetry and 1D lattice parameter of 4.08 Å. Here, each N³⁻ ion is bonded in a distorted linear geometry to two Re⁷⁺ atoms and each Cl⁻ ion is bonded to one Re⁷⁺ atom with bond length of 2.31 Å, resulting in invariable valence states of Re, N and Cl atoms as confirmed via the analysis of on-site Bader charge [38] (Re: $-2.1e$; N: $0.9e$; Cl: $0.3e$). There are one shorter (1.69 Å) and one longer (2.30 Å) Re–N bond with bond overlap population [39] of 0.93 and 0.11, corresponding to nature of strong and weak covalent bond respectively.

To verify the energetic stability of ReNCl₄ nanowire, we calculate its formation energy ΔH defined as $\Delta H = E_{\text{wire}} - E_{\text{cluster}}$, where E_{wire} and E_{cluster} are the energy for one unit of 1D nanowire and an individual cluster, respectively. ReNCl₄ nanowire shows negative ΔH of -0.22 eV per formula (Table 1), indicating that its formation is exothermic. The kinetic stability of ReNCl₄ nanowire is also confirmed by the phonon dispersions in Fig. 2(b).

In light of the stability of self-assembled ReNCl₄ nanowire, Cl atoms can be substituted by other halogens ($X = F, Br$ and I) and a series of hypohetic 1D ReNX₄ compounds might possess analogous structure as ReNCl₄

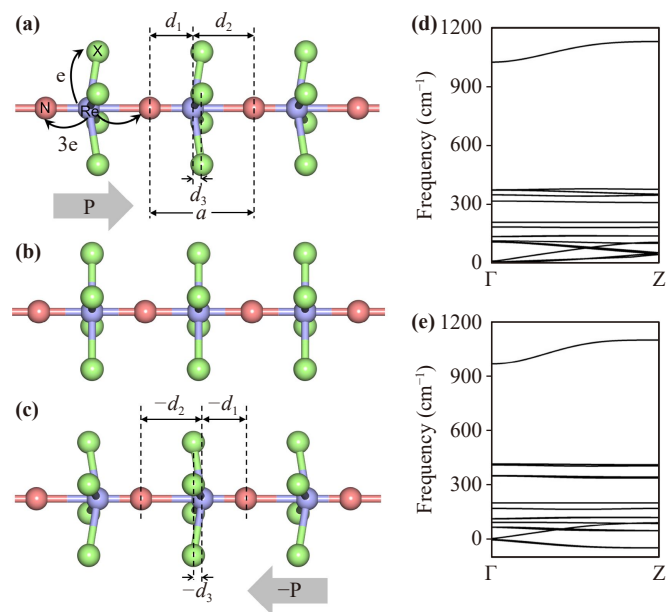


Fig. 2 Atomic structures of cluster-assembled nanowires with (a) FE, (b) PE and (c) $-FE$ state, respectively. The Re (or V, Nb and Ta), N (or F) and Cl (or F and Br) atoms are shown in purple, pink and green colors, respectively. d_1 and d_2 represents the bond length of metal-linked atoms, and d_3 represents displacements of metal and halogen atoms along (001) chain direction. Phonon spectrum of (d) FE and (e) PE phase of ReNCl₄ nanowire, respectively.

nanowire. More importantly, all these ReNX₄ clusters exhibit superatomic characters. Additionally, isoelectronic substitution is another effective strategy to design isostructural materials. Note that MX₅ ($M = V, Nb$ and Ta ; $X = Cl, F, Br$ and I) hold same number of valence electrons with that of ReNCl₄ ($40e$), resulting in similar geometric structure for both types of clusters, that is, MX₅ cluster also displays X–M–X₄ arrangement like ReNCl₄ cluster. Therefore, 1D distorted MX₅ nanowires can be also constructed by self-assembling of MX₅ clusters (see Fig. 2). The energetic and dynamical stabilities of

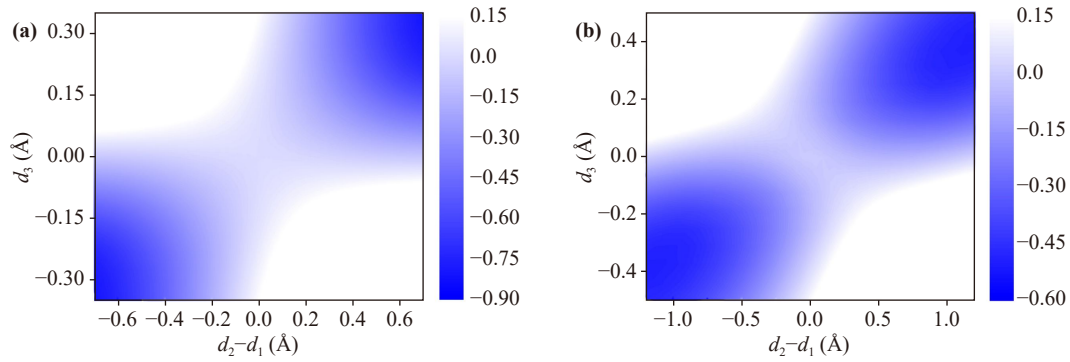


Fig. 3 The energy contour plot (in eV) of 1D (a) ReNCl_4 and (b) VF_5 unit cell as a function of polar displacements of the metal ions. The energy of the PE phase is set to zero.

these predicted nanowires are systematically assessed by their formation energies and phonon dispersions [Fig. S1 of the Electronic Supplementary Materials (ESM)]. Eventually, we screen out ReNF_4 , ReNCl_4 , ReNBr_4 , ReNI_4 , VF_5 , NbF_5 and TaF_5 nanowires with exothermic formation energies ranging from -0.01 to -0.43 eV per formula, suggesting experimental feasibility for the synthesis of these 1D structures via cluster self-assembling. These results also support that self-assembling superatomic clusters is beneficial for thermodynamic stability. Specifically, the lattice parameters of ReNX_4 nanowires in the range of 3.88 – 4.10 Å are smaller than 4.11 – 4.63 Å for MF_5 nanowires (Table 1). In addition, the lattice constant of ReNX_4 (MF_5) nanowire decreases with increasing atomic number of X (M) element. In all situations, metal ions show an off-center displacement along z axial direction by 0.20 – 0.82 Å. The electronic band structures of ReNX_4 and MF_5 nanowires are depicted in Fig. S2 of the ESM. Expect ReNI_4 , all assembled nanowires are semiconductors with moderate or wide band gaps, which allows switching of the polarization a ferroelectric by applying an external field.

Due to non-centrosymmetry of ReNX_4 and MF_5 nanowires, there is spontaneous polarization along z axial direction. In order to explore the ferroelectric feature of these 1D nanowires, the centrosymmetric phase (P_4/m symmetry) without displacement between metal and halide atoms along z axis is firstly identified as the paraelectric (PE) configuration (Fig. 2). As a representative, Fig. 2(c) displays the phonon dispersion of 1D PE ReNCl_4 . Here, three imaginary branches including acoustic and optical phonon modes are throughout the entire Brillouin zone, indicating the geometric instability of PE configuration and a tendency of phase transition from ferroelectric state to a paraelectric state below Curie temperature (T_C). In addition, PE phase has higher energy than that of FE phase by 0.09 – 0.85 eV per formula. Taking ReNCl_4 and VF_5 nanowires as examples, the scanned free-energy contour is plotted with different values of d_2-d_1 and d_3 in Fig. 3 (d_1 and d_2 represent the bond length of metal-linked

atoms, and d_3 represents displacements of metal and halogen atoms along z chain direction). FE and $-$ FE phases are equivalent states with lowest energy, separated by the nonpolar PE state ($d_2 - d_1 = d_3 = 0$) located at the diagonal direction. These indicate the higher stability of FE and $-$ FE phases compared with PE phase. The energy maps of other nanowires are presented in Fig. S3 of the ESM, showing similar phenomena.

The central symmetry of 1D FE ReNX_4 and MF_5 nanowires is eliminated due to the cooperative displacements between metal and halogen atoms, resulting in an evident spontaneous polarization P_{FE} along the chain direction. According to Berry phase through the modern theory of polarization [40], the calculated P_{FE} of 1D FE ReNX_4 and MF_5 nanowires are $(25.64$ – $52.03) \times 10^{-10}$ C/m and $(24.64$ – $55.50) \times 10^{-10}$ C/m, respectively, which are comparable and even larger than the values of many FE materials, such as transition-metal dichalcogenides [41], group-IV monochalcogenides [42, 43], bulk BaTiO_3 [44, 45] and lead-zirconate-titanate (PZT) [46, 47].

We further explore the origin of ferroelectricity in these nanowires. 1D ReNX_4 and MF_5 are semiconductors with band gaps in range of 1.43 – 1.25 eV and 4.94 – 6.60 eV, respectively, and display similar features of electronic structure (Fig. S2 of the ESM). Note that ReNF_4 and VF_5 possess direct band gaps. Taking 1D FE ReNCl_4 nanowire as representative, the projected density of states for each element are displayed in Fig. 4. The hybridization between $\text{Re}-d_{xz}/d_{yz}$, $\text{N}-p_z$ and $\text{Cl}-p_z$ near the Fermi level is originated from $\text{Re}-\text{Cl}$ and $\text{Re}-\text{N}$ bonding. Here, the cooperation of empty d orbitals for Re^{5+} and p orbitals for N^{3-} along the polar direction regulates the existence of FE states in 1D ReNCl_4 , resulting in the second-order Jahn–Teller effect aroused by d^0 -configuration [48, 49], which is similar with the well-known FE perovskite oxides [44, 50]. Analogously, the ferroelectricity of VF_5 nanowire also results from d^0 principle generated by V^{5+} empty d orbitals. Furthermore, more remarkable $pd\pi$ interaction of $\text{Re}-d_{xz}/d_{yz}/\text{N}-p_{x/y}$ than $pd\sigma$ interaction signifies the strong ferroelectric

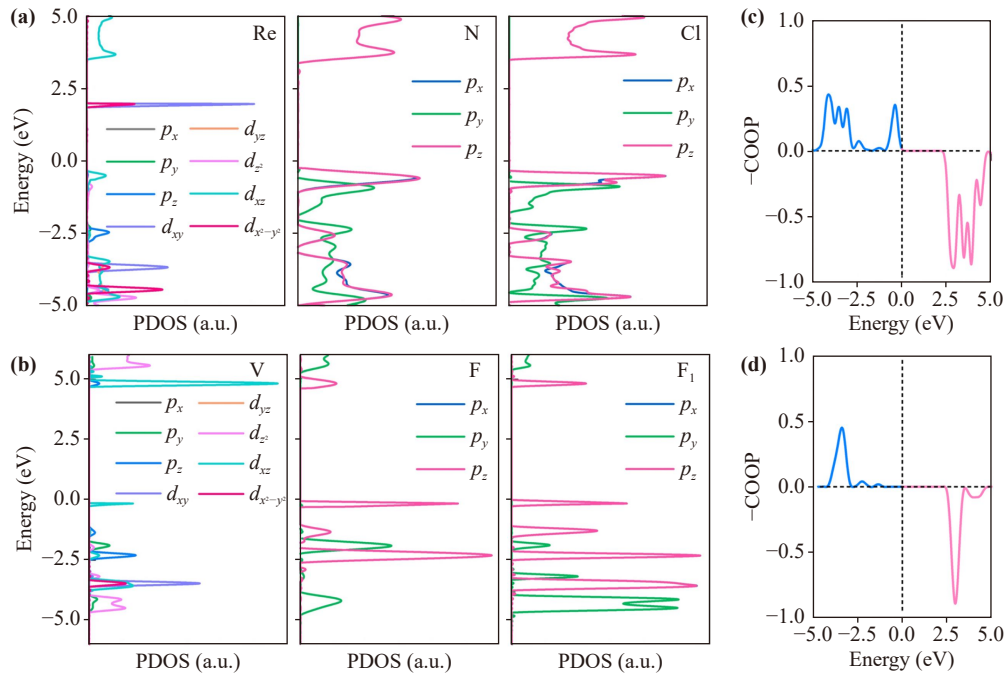


Fig. 4 The PDOS of (a) ReNCl_4 and (b) VF_5 . Here, F is the atom linking two clusters and four F_1 atoms are located at xy plane. $-\text{COOP}$ curves for (c) ReNCl_4 and (d) VF_5 .

effect, which is also appeared in BaTiO_3 and PbTiO_3 [44]. Figures 4(c, d) and Fig. S4 of the ESM present the $-\text{COOP}$ (Crystal Orbital Overlap Population) integrals for Re–N and M–F bonds, where a positive $-\text{COOP}$ denotes favorable bonding interaction and a negative $-\text{COOP}$ indicates unfavorable antibonding interaction. The results reveal that the occupied states are bonding states, while the unoccupied states belong to antibonding ones, further confirming that the FE states is stabilized by Re–N and M–F bonding.

Within the Landau–Ginzburg theory [51], the free energy and polarization should satisfy $E = \sum_i \left[\frac{\alpha}{2}(P_i)^2 + \frac{\beta}{2}(P_i)^4 \right] + \frac{\gamma}{2} \sum_{i,j} (P_i - P_j)^2$. The first two terms are related to the energy contribution to polarization in each unit cell, which can be described by anharmonic double-well shape of the potential profile. α and β parameters are obtained by fitting the double-well potentials in Fig. 5 and Fig. S5 of the ESM. The third term reflects the nearest-neighbor dipole–dipole coupling within the nanowire, and coefficient γ is fitted using the mean-field theory within nearest-neighbor approximation, which can be plotted as a parabola shape in second-order approximation shown in Fig. 5(b) and Fig. S6 of the ESM. The energy barrier E_B for the transition from ferroelectric to paraelectric phases of ReNX_4 and MF_5 nanowires can be determined as 0.74–0.85 eV and 0.09–0.50 eV, respectively. It is noteworthy that E_B is one order of magnitude larger than that of other frequently used ferroelectrics such as BaTiO_3 and PbTiO_3 [44], suggesting high Curie temperature for

these ferroelectric nanowires.

Ab initio molecular dynamic (AIMD) simulations [52] with $1 \times 1 \times 8$ supercells are also performed to evaluate the Curie temperature of these nanowires. Figure 5(c) and Fig. S7 of the ESM show P_{FE} as a function of temperature. Taking ReNCl_4 and VF_5 nanowires as representatives, the static P_{FE} values of 25.64 C/m and 55.50 C/m are maintained at finite temperature up to about 300 and 100 K, respectively. Then the Curie temperature for transition of FE to PE phases can be fitted as the temperature-dependent $P(T) = a/(1 + e^{-k(T-T_C)})$, where a and k are constants. Importantly, T_C value is usually related to the energy barrier E_B and the dipole–dipole coupling between the adjacent unit cells, that is, T_C increases with increasing E_B . Here, ReNX_4 nanowires with larger E_B results in higher T_C value (224–321 K) compared with MF_5 nanowires (72–95 K), which also can be obtained from the pyroelectric response defined as $p = -(dP/dT)$. Moreover, the piezoelectric response e_{11} is dependent on a linear fitting of 1D polarization changes with respect to the uniaxial strains ε_{11} along the z direction. A linear relation between polarization change and the strain within $\pm 1\%$ for ReNX_4 and MF_5 nanowires has been demonstrated in Fig. 5(d) and Fig. S8 of the ESM. The obtained e_{11} is as high as 61×10^{-10} C/m, which is an order of magnitude larger than that of MoS_2 monolayer (3.64×10^{-10} C/m) [53]. The active pyroelectric and piezoelectric responses in these nanowires suggest their broad applications in sensors and energy conversion devices. Particularly, the giant pyroelectric coefficients of ReNX_4 and MF_5

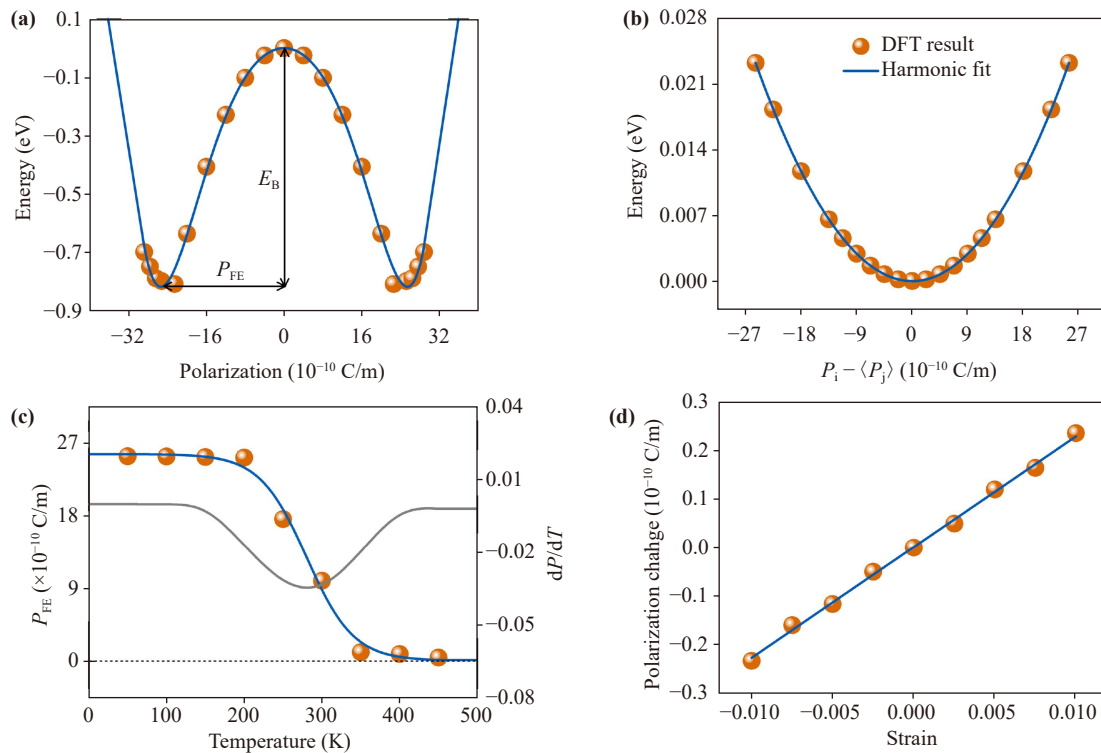


Fig. 5 (a) Double-well potential versus polarization for ReNCl₄ nanowire. E_B is the potential barrier. (b) The dipole-dipole interaction energy for ReNCl₄ nanowires calculated by the mean-field method. (c) Temperature dependence of polarization obtained from *ab initio* molecular dynamics (orange balls). Blue line is a sigmoid fit to MD results and grey line is the pyroelectric response (dP/dT) for ReNCl₄ nanowire. (d) Piezoelectric response (relaxed-ion e_{11} coefficient) for ReNCl₄ nanowire from DFT calculations. The polarization and strain are all along the x direction. The corresponding relaxed-ion piezoelectric coefficient e_{11} can be obtained from slope of the line.

nanowires in range of $(0.02\text{--}0.13)\times 10^{-10}$ C/(K·m²) (Table 1) are larger than those of most ferroelectric crystals, for instance, 2.0 , 8.3 , and 1.8×10^{-4} C/(K·m²) for BaTiO₃, LiNbO₃ and LiTaO₃, respectively [54].

4 Conclusion

In summary, we have designed a series of ReNX₄ (X = F, Cl, Br and I) and MF₅ (M = V, Nb and Ta) nanowires by self-assembling superatomic clusters, based on the criteria of superatomic orbital characters within the shell model and orbital interaction between adjacent clusters. Self-assembling superatomic clusters can strengthen the stability of these 1D structures and achieve ferroelectric effect. These cluster-self-assembled 1D materials are all dynamically stable and possess a moderate or wide band gap of 1.43–6.60 eV (expect for metallic ReNl₄ nanowire). Notably, ReNX₄ and MF₅ nanowires exhibit sizable spontaneous polarization and low switching barrier, and ReNX₄ nanowires hold high T_C above room temperature that is essential for ferroelectric devices. The extraordinary ferroelectric properties in these assembled materials are derived from an electronic mechanism, i.e., the second-order Jahn–Teller effect with

d^0 -configuration transition-metal ions in the materials. These theoretical results open unprecedented opportunities for the discovery of new ferroelectric materials and point to a feasible route for the exploration of exotic 1D physics from cluster assemblies.

Declarations The authors declare that they have no competing interests and there are no conflicts.

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