

state [7] and specific surface area with high reaction ability down to monolayer limit [8, 9], thus booming in the potential applications of electronic, optoelectronic, magnetic, catalytic conversion and storage devices [4, 10–20]. Different from MoS₂ and ReS₂ material families with weak interlayer interaction, layered hafnium chalcogenides (HfS₂ and HfSe₂) share much stronger van der Waals force, resulting in the extreme difficulty for the exfoliation and bottom-up synthesis of ultrathin 2D layers [21]. Meantime, layer dependent moderate bandgap from bulk to monolayer of 1.13–1.18 eV for HfSe₂, is similar with traditional silicon semiconductor for transistor channel application [22]. The theoretical band structure calculation also has predicted HfSe₂ with high phonon-limited mobility of 3500 cm²·V⁻¹·s⁻¹, which is nearly 10 times higher than that of the extensively studied MoS₂ (340 cm²·V⁻¹·s⁻¹) [23]. The electronic transport prediction for 2D HfSe₂ field effect transistors demonstrated the record on-state current density of 5000 μA·μm⁻¹ (about 650 times that of MoS₂) [24]. Moreover, high dielectric constant HfO₂ is a compatible oxide of 2D HfSe₂, which could be achieved by the slow oxidation process similar to the Si/SiO₂ and Bi₂O₂Se/Bi₂SeO₅ interface [1, 22, 25]. Therefore, scalable synthesis of layered 2D HfSe₂ and its heterostructure should be highly desirable to realize low-power electronic and optoelectronic devices [24].

Most recently, ultrathin exfoliated HfSe₂ nanoflakes from chemical vapor transport synthesized single crystals demonstrated the promising transistor performance (on/off ratio > 10⁶; on current, 30 μA/μm) due to the interfacial trap free native oxides [22]. Although 2D HfSe₂ nanoflakes and their heterostructures have been confirmed with excellent electronic device performance, scalable synthesis of large area 2D HfSe₂ nanoflakes and their heterostructures were still rarely reported [22, 26–29]. Low pressure chemical vapor deposition with the utilization of the selenolate complexes [Cp₂Hf(SeR)₂] showed the synthesis of randomly nucleated flower-like nanostructure films [30]. Two-dimensional HfSe₂ nanostructures and MoSe₂/HfSe₂ van der Waals heterostructures were synthesized by molecular beam epitaxy (MBE), also providing the obvious evidence of an ordered Se adlayer by high-resolution scanning tunneling microscopy characterization [31, 32]. Polycrystalline wafer-scale 2D HfSe₂ thin films were achieved by the MBE with the solid supply of Hf and Se, which were also applied to memristor crossbar array for energy-efficient neural network hardware [33]. Randomly vertical oriented HfS₂ nanosheets, polycrystalline small-sized HfSe₂ samples, and HfS_{2(1-x)}Se_{2x} alloys were grown by the highly volatile HfCl₄ precursors, which are still far from the requirement of electronic devices application in terms of domain size and the lateral flakes geometry [34–36]. Therefore, the stable synthesis of high-quality 2D single-crystal HfSe₂ nanosheets on the diverse

substrates laterally is highly desirable, which also could be utilized as scalable oxidation templates for HfO₂–HfSe₂ heterostructures, avoiding the nucleation problem of dielectric layer growth on dangling bond free surface [37].

Herein, we report a remote molten salt (NaCl) vapor-assisted stable volatilization of high melting point hafnium-based precursor (HfO₂) for the CVD synthesis of high-quality single-crystal HfSe₂ nanosheets and thin films on various substrates. The size (6–40 μm), thickness (~4.5 nm), and shape (hexagonal to circular) of HfSe₂ nanosheets can be precisely tuned by varying the growth parameters. High quality HfO₂–HfSe₂ heterostructures were formed by the slow natural oxidation, which has been confirmed with clear lattice incorporation of oxygen atoms and the formation of surface amorphous oxide layers. Further, we investigated the natural oxidation process of HfSe₂ in air and explored the memristor application of HfO₂–HfSe₂ heterostructures. These findings provide a new idea for the scalable synthesis of two-dimensional hafnium-based chalcogenides laterally and demonstrate that the HfO₂–HfSe₂ heterostructures hold great promise in the field of information storage devices.

2 Experimental section

2.1 Preparation of HfSe₂ nanosheets

Two-dimensional HfSe₂ nanosheets were synthesized on Mica, Si/SiO₂ substrates using remote molten salt (NaCl)-assisted volatilization of a high melting point hafnium-based precursor (HfO₂) in an atmospheric pressure chemical vapor deposition system. Specifically, the high melting point hafnium-based precursor (HfO₂, Aladdin, ≥99.9%) was placed in a quartz boat at a high temperature in the center of a tube furnace. The substrate is reversely placed over the precursor. The NaCl (Aladdin, ≥99.5%) powder was placed upstream of the tube furnace at approximately 780 °C. The Se (Aladdin, ≥99.5%) powder was similarly placed upstream of the tube furnace and heated using an external heating band set at 260–340 °C. Before growth, the quartz tube was purged using high-purity argon (Ar), and then the furnace temperature was increased to 780–960 °C and kept constant for 10 min under the argon flow rate of 30 sccm. The specific growth schematic is shown in Fig. 1(a).

2.2 Sample characterization

The optical microscope (OM, OLYMPUS BX53M) has been used to obtain the morphology of the HfSe₂ nanosheets. The Raman spectrometer (Raman, Renishaw inVia Reflex) with a 532 nm laser as the excitation source was used to collect spectra of HfSe₂

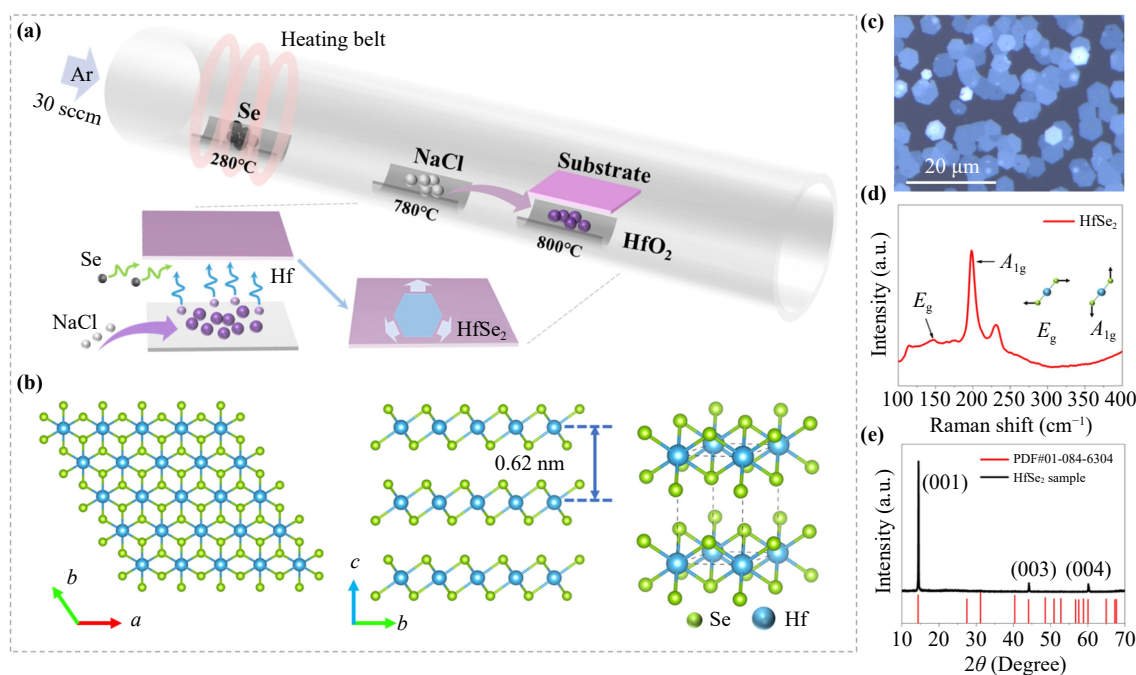


Fig. 1 (a) Schematic diagram of remote alkali halide vapor assisted stable sublimation of high melting point hafnium dioxide for controlled synthesis of 2D HfSe₂ nanosheets. The NaCl and HfO₂ powder were arranged at the separated temperature of 780 °C, and 800 °C, respectively. (b) Schematic atomic structures of 1T-HfSe₂ along the ab and bc crystal planes, also for its unit cell. (c) Typical optical image of the synthesized HfSe₂ nanosheets on the mica substrate. (d) Raman spectra and (e) XRD pattern of the synthesized HfSe₂ nanosheets.

nanosheets to determine molecular vibration bonds of the sample. The crystal structures were characterized by X-ray diffraction (XRD, PERSEE XD-3). The thickness of the sample was determined using an atomic force microscope and the surface potential of the samples was tested using Scanning Kelvin Probe Microscopy (OXFORD MFP-3D Origin). Transmission electron microscopy (TEM, FEI Tecnai F20) is used to analyze the crystal structure and elemental composition of the HfSe₂ nanosheets.

2.3 Device fabrication and property measurement

The device is fabricated via an Ultraviolet Maskless Lithography machine (TuoTuo Technology (Suzhou) Co., Ltd.). The naturally oxidized 2D HfSe₂ (25 °C, 40% humidity, 48 h) were used to fabricate the devices. The transfer platform was used to transfer the sheet to a specific location. Physical vapor deposition equipment (VNANO VZZ-300S) was used to deposit Cr/Au as electrodes. The memristor properties of the devices were tested by a semiconductor analyzer (PDA FS-Pro) connected with a probe station at room temperature.

3 Results and discussion

To achieve high-quality two-dimensional (2D) HfSe₂

nanosheets as the heterostructure template, we propose a remote alkali halide vapor assisted chemical vapor deposition [Fig. 1(a)]. Different from the normal growth using the precursor mixtures, our design strategy separated the location of HfO₂ and NaCl to avoid the precursor supply shortage or erupt [38]. The alkali halide powder was placed around its melting temperature of 780 °C, separately arranging the HfO₂ powers (800–940 °C), which could continuously provide the vapor phase of alkali halide molecule to react with the HfO₂ surface for stable sublimation. The freshly exfoliated mica or SiO₂/Si substrate was settled on the top of HfO₂ powders, in which the reaction species were transported to the substrate surface and formed the nuclei. Thus, the nuclei could epitaxially grow towards in-plane direction to form 2D HfSe₂ under the continuous mild supply, avoiding the small sizes or the vertical nanosheets formation [36, 39–41]. Figure 1(b) shows the schematic crystal structure of 1T phase HfSe₂ (−4.550 eV), which has a lower formation energy than 2H-HfSe₂ (−4.045 eV). In details, 1T-HfSe₂ belongs to the hexagonal crystal system with space group of *P3m1*, showing the lattice parameters of $a = b = 3.744 \text{ \AA}$, $c = 6.155 \text{ \AA}$, $\alpha = \beta = 90^\circ$, $\gamma = 120^\circ$ [27]. Monolayer 1T-HfSe₂ consists of two layers of Se atoms sandwich a layer of Hf atoms with interlayer covalent bonding, in which the Hf atoms are located at the center of an octahedron composed of Se atoms, with a spacing distance of 0.62 nm [Fig. 1(b)] [42].

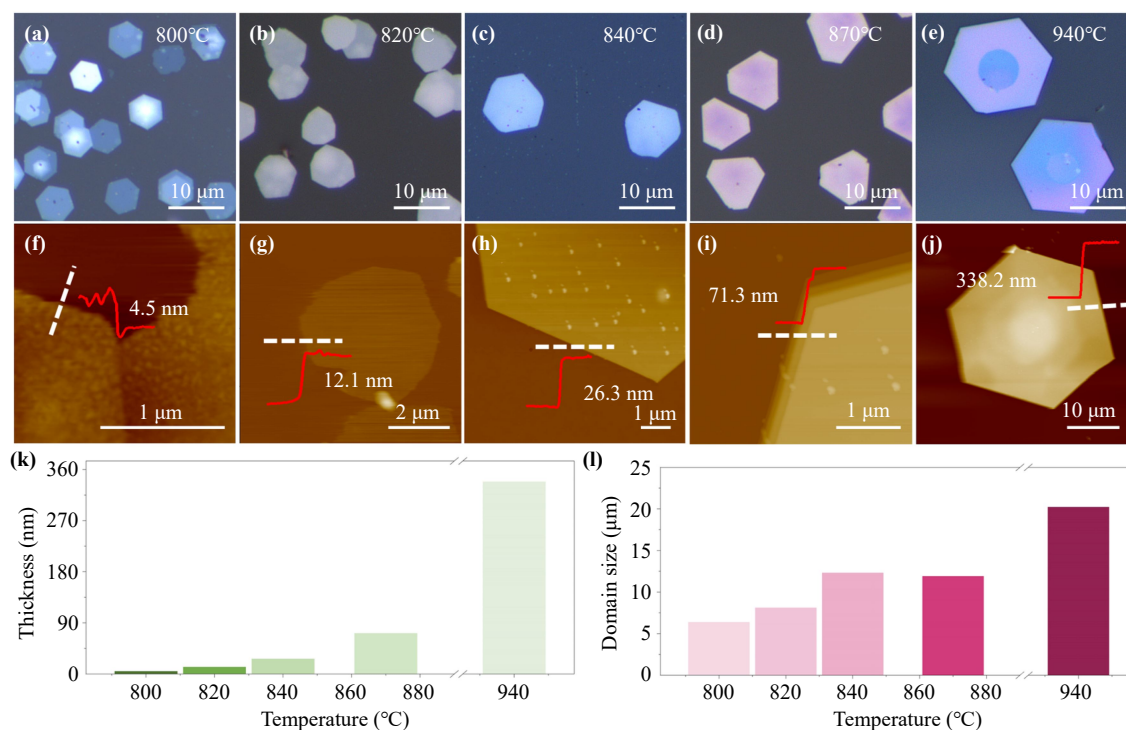


Fig. 2 (a–e) Optical images of HfSe₂ nanosheets grown on mica substrates at different growth temperatures: 800 °C, 820 °C, 840 °C, 870 °C, and 940 °C, respectively, under the argon flow rate of 30 sccm, Se temperature of 280 °C and growth time of 10 min. (f–j) Corresponding representative AFM images. (k) Statistical graph of the thickness of HfSe₂ nanosheets at different growth temperatures. (l) Statistical graphs of HfSe₂ nanosheet domain sizes at different growth temperatures.

Figure 1(c) shows an OM image of 2D single crystalline HfSe₂ nanosheets synthesized on a mica substrate, which are obtained by the separated precursor strategy. The HfSe₂ nanosheets exhibit a regular hexagonal geometry with uniform optical contrast and an average domain size of about 6.4 μm. The typical growth results of 2D single crystalline HfSe₂ nanosheets are better than the previously reported polycrystalline HfSe₂ thin films or vertical small size nanosheets (HfS₂) [34, 41], which are due to the high nucleation sites from high volatilization of low melting point HfCl₄. Similarly, the normal precursor mixtures could provide erupt chemical supply, resulting in high nucleation sites for randomly vertical 2D HfSe₂ [Fig. S1 of the Electronic Supplementary Materials (ESM)] [34, 39]. Therefore, such a design enables NaCl vapor to continuously interact with HfO₂, promotes the stable volatilization of the metal precursor, and achieves a smooth and continuous supply of reactants, thus facilitating the synthesis of high-quality lateral epitaxy of 2D HfSe₂ nanosheets [43]. Raman spectra of the prepared samples [Fig. 1(d)] were obtained using a 532 nm excitation laser for the identification of HfSe₂ nanosheets. The in-plane E_g mode of 146.7 cm⁻¹ and the out-of-plane A_{1g} mode of 198.6 cm⁻¹ were observed, both of which are in good agreement with previously reported results [31, 41]. Three clear and sharp diffraction peaks at $2\theta = 14.4^\circ$, 44.2° , and 60.1° that correspond to the (001),

(003), and (004) crystal planes of HfSe₂ were shown, which matched well with the standard PDF card of the HfSe₂ crystals (PDF#01-084-6304), indicating that the grown HfSe₂ samples have a 1T-phase structure with specific growth direction [Fig. 1(e)] [33]. Therefore, high-quality 2D HfSe₂ thin nanosheets were achieved by the remote alkali halide vapor assisted chemical vapor deposition.

The thickness, domain size, and surface morphology of 2D material have a great influence on its physical properties such as band gap and carrier mobility [11, 36, 44]. Therefore, it is crucial to realize the precise control for achieving large-size and ultrathin 2D single-crystal. Herein, the temperature dependence of growth procedures has been systematically carried out to tune the thickness, domain size, and surface morphology. Figures 2(a)–(e) show the optical images of hexagonal HfSe₂ nanosheets grown on mica substrates at different temperatures, in which other growth conditions were kept as argon flow rate of 30 sccm, Se temperature of 280 °C and NaCl temperature of 780 °C. Two obvious growth results trends were observed according to the above optical images analysis. First, the optical contrast of the HfSe₂ nanosheets changes from light blue to white and finally to silver-pink with the increased growth temperature, indicating that the thickness of the nanosheets is gradually getting much thicker. This is reflected more clearly in

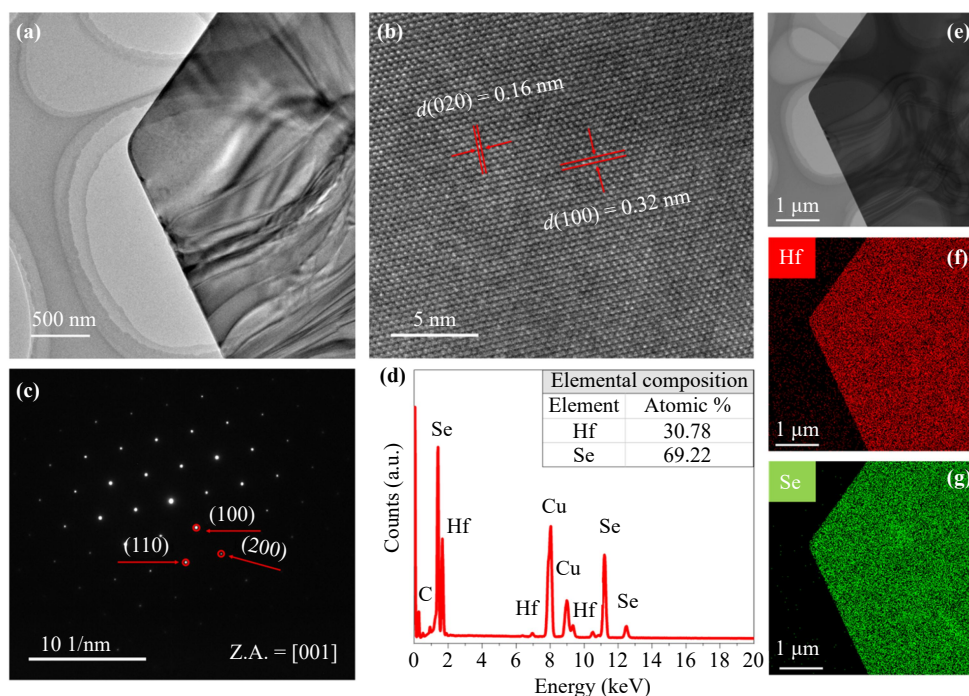


Fig. 3 (a) Low-resolution TEM image and (b) High-resolution TEM image of 2D HfSe₂ nanosheet. (c) Corresponding selected-area electron diffraction pattern (SAED). (d) EDS elemental analysis spectra of 2D HfSe₂ nanosheet and the inset shows the stoichiometric ratio of Hf and Se. (e) Dark-field TEM image of 2D HfSe₂ nanosheet. (f, g) EDS mapping images of Hf and Se, respectively.

the corresponding AFM images [Figs. 2(f–j)] and the thickness statistics histogram [Fig. 2(k)]. The thickness of HfSe₂ nanosheets increased from 4.5 to 338.2 nm when the growth temperature was increased from 800 to 940 °C. It is worth noting that 4.5 nm is not the thinnest sample we could obtain, due to the easily oxidizable property of HfSe₂ making the samples less than 4.5 nm thick are difficult to measure (Fig. S2 of the ESM). Secondly, the average domain size of HfSe₂ nanosheets increases gradually with the increased growth temperature. As shown in Fig. 2(l), the average domain size of HfSe₂ nanosheets increased from 6.3 to 20.2 μm when the temperature was increased from 800 to 940 °C. At lower growth temperatures, the low diffusivity of active reactive precursors prefers to bond with the active edge of two-dimensional HfSe₂ sheet at the substrate surface, which tends to produce thin two-dimensional HfSe₂ nanosheets. As the growth temperature increases, the concentration of active reactants is elevated and the migration rate of precursors on the substrate surface is enhanced, which promote both in-plane and out-of-plane growth of 2D nanosheets, resulting in large and thick materials with lower surface energy. 2D HfSe₂ nanosheets grown on SiO₂/Si substrates (Fig. S3 of the ESM) exhibit the similar temperature dependent thickness and domain size regularity. The HfSe₂ morphology changes from hexagonal to circular nanosheets and large area thin films were also realized

by the tuned ratio of Hf to Se precursor (Fig. S4 of the ESM), and the decreased gas flow rate, respectively (Fig. S5 of the ESM). Thus, the systematic growth results of 2D HfSe₂ were carefully explored and shown with high-quality samples under the wide range of synthesis conditions [41, 45].

In order to further investigate the microstructure, elemental composition, and crystallinity of 2D HfSe₂ nanosheets, the measurement of transmission electron microscopy was carried out. Vertically grown HfSe₂ was pressed directly onto a TEM copper grid for the clean surface quality (Fig. S6 of the ESM). A low-resolution TEM image of a HfSe₂ nanosheet exhibited a regular hexagonal corner [Fig. 3(a)]. Figure 3(b) shows the corresponding high-resolution TEM image with the clear lattice stripes showing the excellent crystallinity, in which the lattice spacing of 0.32 and 0.16 nm can be indexed to the (100) and (020) crystal planes of HfSe₂, respectively [22]. The selected-area electron diffraction pattern of the HfSe₂ nanosheet was shown with regular hexagonal pattern in Fig. 3(c), in which single crystalline diffraction patterns are consistent with the simulated electron diffraction and other electron diffraction from different regions (Fig. S7 of the ESM). Figures 3(f) and (g) show the corresponding elemental mappings of Hf and Se for the samples in Fig. 3(e), in which the Hf and Se elements are uniformly distributed throughout the nanosheets. The elemental analysis spectra of HfSe₂

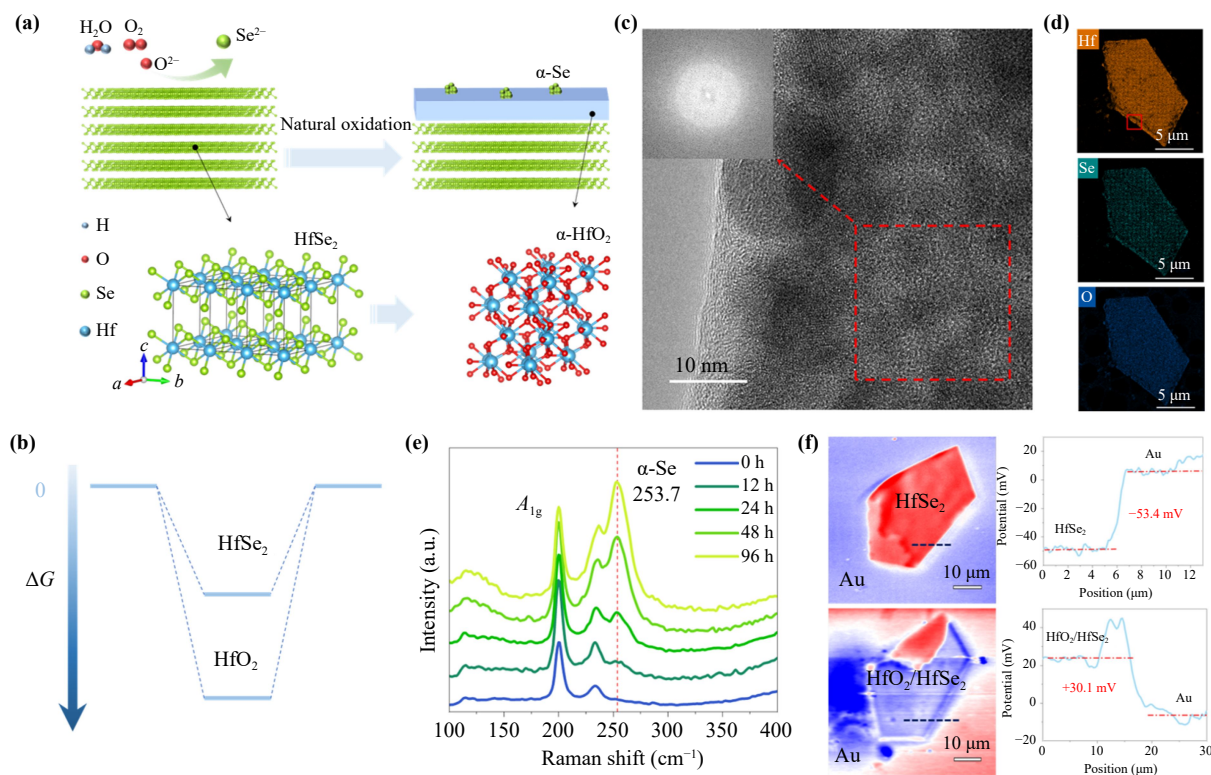


Fig. 4 (a) Schematic of natural oxidation of 2D HfSe₂ nanosheets in air, forming Se clusters on the surface of HfO₂-HfSe₂ heterostructure under the H₂O and O₂ atmosphere. (b) Schematic of the relative magnitudes of standard molar Gibbs free formation energies of HfSe₂ and HfO₂. (c) High-resolution TEM image of HfSe₂ nanosheets after 48 h oxidation in air, and the inset image is the corresponding FFT of amorphous HfO₂ structure. (d) Corresponding EDS elemental mapping for Hf, Se, and O, respectively. (e) Raman spectra of HfSe₂ nanosheets naturally oxidized in air for different oxidation times. (f) Surface potentials comparison of 2D HfSe₂ nanosheets before and after the oxidation.

nanosheets [Fig. 3(d)] indicates that the stoichiometric ratio of Hf to Se is around 1:2.2, which is close to the theoretical chemical ratio.

Thin Hf-based oxide nanostructure or ultrathin film has attracted much interest in recent years, which is highly desirable to develop scalable synthesis methods rather than only relying on atomic layer deposition, etc. [46] The nonlayered crystal structure and high melting temperature make it difficult to synthesize in large scale 2D forms and get high crystal quality with the cheap semiconductor procedure [47]. Herein, high-quality 2D HfSe₂ nanosheets with low Gibbs free formation energy were proposed as the ultrathin template, oxidating into the nonlayered HfO₂-HfSe₂ heterostructure, which could be much easier to implement compared to stable 2D HfS₂ [48]. Figure 4(a) illustrates the schematic natural oxidation of 2D HfSe₂ in air, where the unstable 2D HfSe₂ is exposed to the moist air. The microelectrochemical charge transfer may occur between the material basal plane and the water redox pairs, thus leading to surface oxidation of 2D HfSe₂, where the selenium clusters segregate onto the surface of the material in the form of amorphous selenium. The active oxygen atoms could replace the selenium atoms and generate Hf-O chemical

bonds, eventually forming the HfO₂-HfSe₂ heterostructure or entire HfO₂ amorphous structures. Figure 4(b) shows the relative standard molar Gibbs free energies of HfSe₂ and HfO₂, which are much lower for HfO₂ (-1162.4 kJ/mol) than for HfSe₂, which confirms the thermodynamical possibility of the transformation of HfSe₂ into HfO₂. An air-exposed of ~40 nm HfSe₂ nanosheet was used to record Raman spectra at different oxidation time. The optical contrast of HfSe₂ nanosheet changes from light blue to dark brown with the increase of exposure time, and there are obvious dark spots appearing on the surface of 2D HfSe₂ (Fig. S8 of the ESM). In the time dependent Raman spectrum, it can be seen that after 12 h of natural oxidation, in addition to the representative peaks A_{1g} of HfSe₂, a new active peak at 253.7 cm⁻¹ appeared [Fig. 4(e)]. This Raman characteristic peak can be indexed as amorphous Se and the peak intensity increases with time. The above results fully confirm that HfSe₂ has amorphous Se production during the oxidation process with the maintained Hf-Se chemical bonds. High-resolution TEM images of oxidized HfSe₂ nanosheets with corresponding EDS elemental analyses are shown in Figs. 4(c) and (d), respectively. The high-resolution TEM image of oxidized HfSe₂

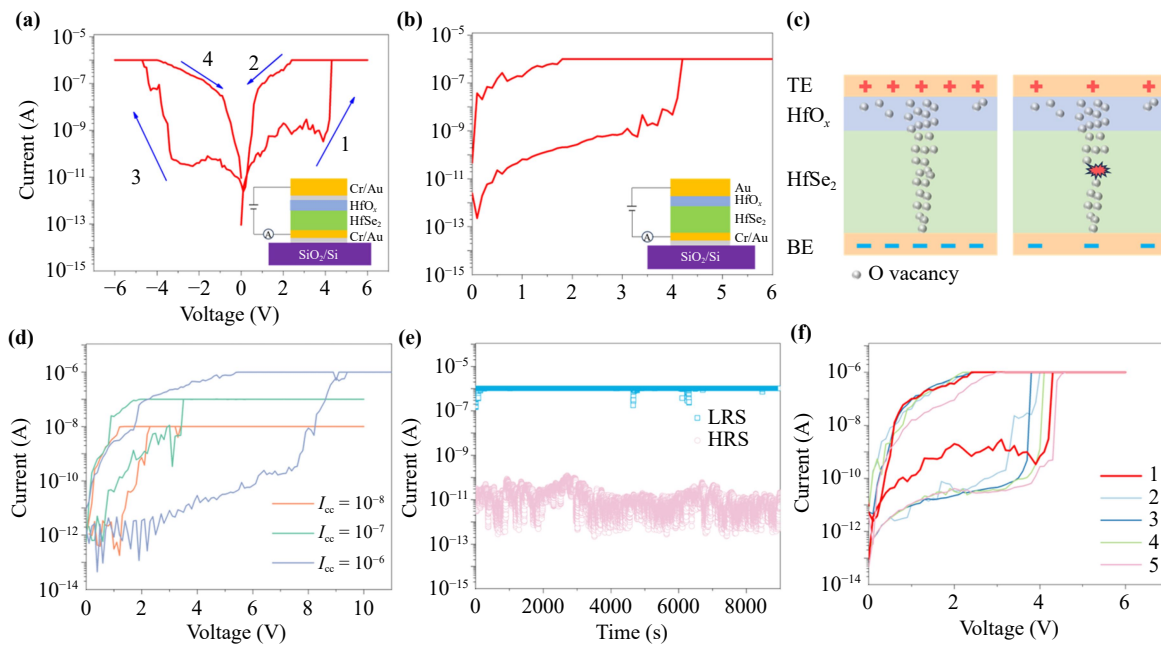


Fig. 5 (a) Typical I - V switching curve of the HfO_2 - HfSe_2 heterostructure memristor with Cr/Au top electrode and Cr/Au bottom electrodes. The inset picture shows the schematic device structure. (b) Typical I - V switching curve of the HfO_2 - HfSe_2 heterostructure memristor with inert Au top electrode and Au bottom electrodes. The inset picture shows the schematic device structure. (c) Schematic mechanism diagram of oxygen vacancies dominated resistance change in the HfO_2 - HfSe_2 heterostructure. (d) I - V switching curves of the HfO_2 - HfSe_2 memristor under different limiting currents. (e) Holding characteristics of the high and low resistance states of the HfO_2 - HfSe_2 memristor. (f) The multiple cycling stability measurement of the HfO_2 - HfSe_2 memristor devices.

nanosheets showed the clear disordered structures with amorphous diffraction rings, indicating the formation of amorphous HfO_2 surface layers [Fig. 4(c)]. The relatively uniform Hf and O elemental mapping and the discrete elemental distribution of weak Se intensities also indicate the chemical oxidation induced by the surface reaction. The unusual FFT diffraction spots of other different regions reveal other specific orientation formation of HfSe_2 , which may arise the volumetric expansion and lattice strain in the heterostructure (Fig. S9 of the ESM). The surface potential of 2D HfSe_2 nanosheets before and after oxidation was measured by the kelvin probe force microscopy, in which the samples were transferred on the gold substrates shown with small roughness change [Fig. 4(f) and Fig. S10 of the ESM]. The surface potential image of HfO_2 layers demonstrated the uniform potential distribution, indicating the slow oxidation kinetics process [Fig. 4(f)]. The surface potential of HfO_2 is higher by 80.5 mV compared to the fresh grown HfSe_2 , indicating the lowering Fermi level and the decreased electron density [49]. The above results clearly confirm the formation of high-quality HfO_2 - HfSe_2 heterostructures with the controllable oxidation process.

The vertical memristor devices were constructed by the layer-by-layer transfer and fabrication, using the natural oxidized HfO_2 - HfSe_2 heterostructure as the active layers [Fig. 5(a)]. A layer of Cr/Au (4/8 nm) was

made on the SiO_2/Si surface as the bottom electrode, and the naturally oxidized HfSe_2 nanosheets were transferred to the bottom electrode surface. Finally, the top electrode was deposited a layer of Cr/Au (10/70 nm). The characteristic I - V switching curves for this typical memristor were measured by the cyclic voltage scanning in the range 0–6 V with 0.1 V step [Fig. 5(a)]. As the applied positive voltage increases, the device switches from the high-resistance state to the low-resistance state at about 4.1 V. When the voltage drops to 2.3 V, the device spontaneously and gradually switches from the low-resistance state to the high-resistance state. It is possible the surface HfO_2 layer helps reduce lower resistance states of the memristor to obtain a high switching ratio. The symmetrical characteristic I - V relation was also observed for our devices. The CVD derived HfO_2 - HfSe_2 heterostructure memristor has demonstrated a switching ratio of up to 10^5 . In addition, the device could spontaneously and gradually switch from a low-resistance state to a high-resistance state when the external electric field decreases, indicating that the device is volatile, and shown as a threshold switching devices. In order to understand the working mechanism of HfO_2 - HfSe_2 memristors, the inert metal of Au layers was used as the top and bottom electrodes [Fig. 5(b)]. A significant facile amnesic behavior was observed in several devices [Fig. 5(b) and Fig. S11 of the ESM], in which the immobile

gold atoms inferred oxygen vacancy filament is the main reason of resistance conductivity transition. When a forward bias is applied to the device, the enriched oxygen vacancies in the amorphous HfO₂ layer could gradually move to the bottom electrode. The oxygen vacancies are continuously enriched and migrated, and eventually form a conducting channel across the dielectric layer and show a low resistance state. When the forward voltage decreased, the oxygen vacancy conducting filaments could rupture at the most fragile part, bringing the device back to the high-resistance state [Fig. 5(c)].

The HfO₂-HfSe₂ memristor was operated and measured under different limiting currents of 10⁻⁸, 10⁻⁷ and 10⁻⁶ A, where the device maintained a switching ratio of over 10³ with a clear memristor windows even the current is as low as 10 nA [Fig. 5(d)]. High device performance at low current density helps to reduce the power consumption of the memristor device and attenuates the adverse effects of high current density on device's lifetime. The hold-up characteristics in the high-resistance and low-resistance states at bias voltages of 3.2 V were measured [Fig. 5(e)]. The measured results showed that the high/low resistive states are virtually unchanged over the experimental time scale of 9000 seconds, indicating that the device is virtually unaffected by read disturbances. The multiple scanning performance of this HfO₂-HfSe₂ memristor is shown in Fig. 5(f), demonstrating the voltages window with small fluctuation within two small ranges of 3.8–4.1 V and 2–2.9 V, which indicates the good cycling stability. The relatively high performance and scalable fabrication of HfO₂-HfSe₂ heterostructure devices clearly confirm our excellent stable synthesis strategy, which also reveal the oxygen vacancies dominated working mechanism of the memristor.

4 Conclusion

In conclusion, we have successfully and controllably synthesized high-quality 1T-HfSe₂ nanosheets on mica and SiO₂/Si substrates using stable molten salt-assisted volatilization of high melting point hafnium-based precursors by chemical vapor deposition. The lateral-size, thickness, and micro-morphology can be precisely adjusted by changing the growth parameters. The high-quality HfO₂-HfSe₂ heterostructures were achieved by the easy natural oxidation, in which the selenium clusters segregated onto the surface forming amorphous selenium and oxygen atoms incorporated into HfSe₂ lattice replacing the Se atoms. The fabricated HfO₂-HfSe₂ heterostructures demonstrated excellent performance of a high switching ratio of 10 [5], and long retention time of over 9000 s under low current density. Our work provides a new synthesis strategy for epitaxy growth of lateral Hf-based layered materials and its heterostructure for high performance scalable electronic devices.

Declarations The authors declare that they have no competing interests and there are no conflicts.

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