

MoS₂ is the most optimistic TMDCs because of its surplus and innocuous and easily feasible synthesis [10, 11]. Structurally, MoS₂ is a typical hexagonal layered compound with strong in-plane covalent and weak out-of-plane van der Waals (vdW) interaction, resulting in the absence of dangling bonds on the surface [12]. When cleaved from bulk to monolayer (1L), energy band structure of MoS₂ transforms from indirect to direct due to the quantum confinement effect [6]. Field-effect transistors (FETs) based on monolayer MoS₂ have been demonstrated to exhibit a high on/off current ratio 10⁸ and a high mobility of 200 cm²·V⁻¹·s⁻¹ at room temperature [13]. These intriguing properties give it a wide range of potential applications in transistors [13–15], memories [16, 17], flexible electronics [18, 19], photodetectors [20–23], energy storage and conversion and hydrogen evolution reaction [24].

In the past few decades, wide bandgap semiconductors (WBSs), such as GaN and SiC, have drawn considerable attention because of their outstanding properties, and widely used in the emerging electronics and optoelectronics. GaN possesses wide bandgap (3.4 eV), high room-temperature electron mobility (1300 cm²·V⁻¹·s⁻¹), and large short-wavelength absorption coefficient, making it an important semiconductor for applications in light emitting diodes (LED), FETs, solar cells, lasers, and photodetectors [25–29]. SiC exists as over 170 polytypes with the 4H polytype is most commonly used in high-temperature and high-frequency electronic devices due to its high carrier mobility (900 cm²·V⁻¹·s⁻¹), large band gap (3.26 eV) and low dopant ionization energy [30].

Beyond separate studies of 3D WBSs or 2D TMDCs, it is favourable to integrate 2D/3D TMDCs/WBSs heterostructures, which benefits for improving the performance of devices, such as LEDs, photodetectors, sensors and energy storage devices, and further providing the material platform of novel physical phenomena [25]. Due to the great light absorption of multilayer MoS₂, Goel et al. have successfully fabricated the MoS₂/GaN UV photodetector with the detectivity of 10¹¹ Jones and the external spectral responsivity of 10³ A·W⁻¹ [31]. The MoS₂/GaN heterostructure device fabricated by Moun et al. demonstrated a detectivity of 10¹⁴ Jones and an ultra-high optical responsivity of 10⁵ A·W⁻¹ under a 405 nm laser radiation [32]. Because of its suitable band gap, SiC has always been an important material for UV photodetectors [33]. Then a MoS₂/4H-SiC photodetector operating in both the UV and visible regions were fabricated, which exhibits a responsivity of 5.7 A·W⁻¹ under UV irradiation and a noise equivalent power (NEP) of 10⁻¹³–10⁻¹⁵ W·Hz^{-1/2} [20]. Based on the dual-photogating effect, MoS₂/SiC photodetectors were demonstrated to have a ultra-high responsivity of 1.6 × 10³ A·W⁻¹ and 1 × 10⁴ A·W⁻¹ in VIS and UV irradiation, respectively [21]. The MoS₂/WSe₂ heterojunction photodetector

fabricated on n-doped 4H-SiC also shows the excellent performance with the maximum responsivity of 7.17 A·W⁻¹, the corresponding maximum EQE of 1.67 × 10³ %, and detectivity of 5.51 × 10¹¹ Jones with a gain of 10³ [34].

In the aspects of photocatalyst, gas sensor and other applications, the MoSSe/SiC and WSSe/SiC heterostructures were certified to be a high-efficient photocatalyst due to their specific electronic and optical properties based on the density functional theory (DFT) calculations [35]. And the MoS₂/GaN photoanode could enhance the photocurrent density to 5.2 mA·cm⁻², which is 2.6 times higher than that of GaN photoanode [36]. As compared to the MoS₂ or GaN gas sensors, the sensitivity can be significantly enhanced (20 times) when integrating the MoS₂/GaN heterostructure [37, 38]. The MoS₂/GaN heterojunction nanowires could be used in memory device, which possesses the good retention characteristics of 3.4 × 10³ s under a low switching voltage [39]. Although above experimental and theoretical works have confirmed the enormous advantage of the integrated TMDCs/WBSs, the detailed study and analysis of the interfacial properties, especially multiple interfacial interactions modulated energy band alignment, of these heterostructures are still quite limited.

In this study, monolayer (1L), bilayer (2L) and quadlayer (4L) MoS₂ were deposited on 4H-SiC (0001) substrates by reactive sputtering deposition. The interfacial properties of MoS₂/SiC heterostructures were studied by combining the first-principles calculations and X-ray photoelectron spectroscopy. Experimental (theoretical) VBO increases from 1.49 (1.46) to 2.19 (2.36) eV, with increasing MoS₂ layers up to 4L. Theoretical calculations further reveal that a strong interlayer interaction was demonstrated at 1L MoS₂/SiC interface and Fermi level pinning and totally surface passivation were realized for the 4H-SiC (0001) Si surface. 1L MoS₂/SiC interface exhibits type I band alignment with the asymmetric CBO and VBO. For 2L and 4L MoS₂/SiC, Fermi level was just pinning at the lower MoS₂ monolayer. They exhibit the type II band alignments and the enlarged CBOs and VBOs. High efficiency of charge separation will emerge due to the asymmetric band alignment and built-in electric field for all the MoS₂/SiC interfaces, and enhance the device performance significantly.

2 Experimental and theoretical methods

The 4H-SiC substrates were prepared using a four-step cleaning process. It was ultra-sonicated in acetone, ethanol and DI water for 10 mins each at 45 °C to remove particle and organic contaminants. After which, dilute HF (1:1) was used to etch the surface to remove any SiO₂ oxidation. MoS₂ layers were grown via reactive magnetron sputtering whereby a molybdenum target

was sputtered using argon in a chamber with ambient sulfur. The partial pressure of sulfur vapour was controlled at 1.2×10^{-5} mbar through the heating of sulfur powder in a separate compartment. One of the substrates was left bare while the rest were sputtered with a DC power source at 10 W, 3.2×10^{-4} mbar and 700 °C for 5 mins, 9 mins, 12 mins and 2 hours (1L, 2L, 4L and bulk MoS₂).

Thermo Scientific DXR microscope with a 514.5 nm laser were used to take the Raman and photoluminescence (PL) spectra of the as-grown heterojunction samples. Atomic force microscopy (AFM) images were obtained by using the Bruker Dimension Icon system and the tapping mode. High-resolution transmission electron microscopy (TEM) images were taken from samples grown directly on 8 nm SiO₂ support membranes. High resolution X-ray photoelectron spectroscopy (XPS) spectra were taken on a VG ESCALAB 220i-XL XPS system. The binding energy scale was calibrated with pure Au, Ag, and Cu standard samples by setting the Au 4f_{7/2}, Ag 3d_{5/2}, and Cu 2p_{3/2} at binding energies of 83.98 eV, 368.26 eV, and 932.67 eV, respectively. C 1s signal from the sample surface was used to correct the core-level binding energy.

The first-principles calculations were performed by using the Vienna ab initio simulation package (VASP) [40]. The exchange-correlation energy is described by the Perdew-Burke-Ernzerhof (PBE) exchange-correlation functional [41]. The semi-empirical correction scheme of Grimme (DFT-D3) were utilized to correct the effect of vdW interactions in multilayer TMDCs system [42].

Furthermore, Heyd-Scuseria-Ernzerhof (HSE06) was used to simulate all the density of states (DOSs), because of its accuracy for obtaining the bandgap of the calculated system [43]. We also considered the dipole correction in the calculation to eliminate possible artificial dipole interactions [44]. The charge transfer was calculated by using the Bader analysis [45].

3 Results and discussion

Based on the above sputtering parameters, we could obtain the extremely low growth rate of MoS₂ layers. The successful deposition of monolayer and few-layer MoS₂ on different substrates have been achieved by precisely controlling the growing time, which has also been demonstrated by our previous works [46–50]. Raman spectroscopy was further used to characterize the number of layers in our samples. It has been shown in previous studies that the difference in Raman shift between the in-plane (E_{2g}^1) and out-of-plane (A_{1g}) characteristic vibration modes increases with the number of layers of MoS₂ and hence can be used to accurately determine the number of layers within 1 to 5 MoS₂ layers [51–53]. Figure 1(a) shows the E_{2g}^1 and A_{1g} vibration modes of 1L, 2L, 4L and bulk MoS₂/SiC samples. A Raman shift difference of 19.5, 21.7, 23.9 and 25.5 cm⁻¹ was observed from the samples, which corresponds to 1L, 2L, 4L and bulk MoS₂ respectively.

AFM images of SiC substrate before etching, after etching, and 1L MoS₂ grown on SiC substrate were

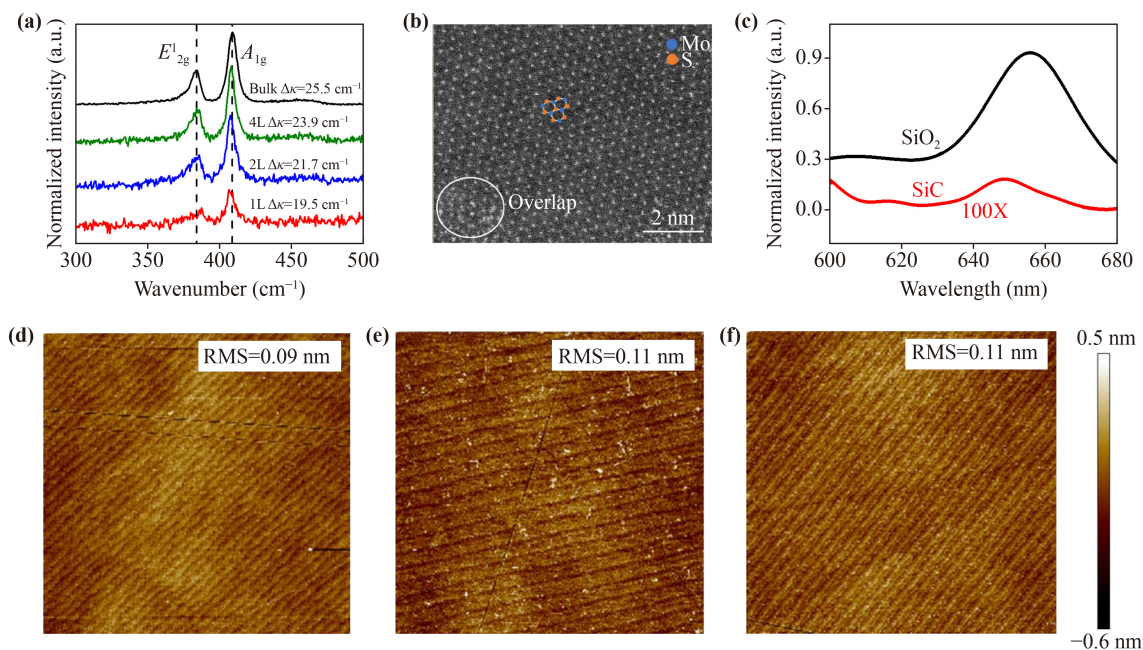


Fig. 1 (a) Raman spectra of 1L, 2L, 4L and bulk MoS₂/SiC, (b) ADF-STEM image of 1L MoS₂, (c) PL spectra of 1L MoS₂ grown on SiC against 100 nm SiO₂/Si. AFM images (5 μm × 5 μm) of (d) SiC substrate before etching, (e) SiC substrate after etching, (f) 1L MoS₂ grown on SiC substrate.

taken to determine the uniformity of the samples, as shown in Figs. 1(d)–(f). The step-terrace structure on the surface is due to different etching rates of the atomic layers in the 4H-SiC crystal structure during the material processing [54]. The dilute HF wet etching process only resulted in a slight increase in RMS value from 0.09 nm to 0.11 nm and the deposition of MoS₂ did not result in an increase in surface roughness showing good monolayer uniformity.

Figure 1(b) exhibits annular dark field STEM (ADF-STEM) image of monolayer MoS₂ with alternating Mo and S columns arranging into hexagonal structure, which can be identified to be the 2H-MoS₂ polytype. Meanwhile, moiré patterns (overlap on the lower left) are also observed. Therefore, TEM results confirm the high quality 2D MoS₂. Figure 1(c) shows the PL spectra of 1L MoS₂ grown on SiC and SiO₂ as a comparison. The PL intensity of the 1L MoS₂ on SiC was about 500× weaker in comparison to 1L MoS₂ grown on a 100 nm SiO₂/Si substrate. Though a tiny PL peaks were observed at 649 nm from the 1L MoS₂/SiC sample, the quenching of PL intensity forebodes a strong interaction between the as-grown MoS₂ and SiC substrate, which agrees well with the following theoretical observations. And no PL peaks were observed from 2L, 4L and bulk samples.

Surface chemistry was probed with XPS measurements. The fitting rule adopted in this work is that the full width at half maximum (FWHM) of both components are comparable and the intensity ratios follow the expect quantum mechanically predicted ratio when XPS doublet have narrow separations. Figure 2 shows the Si 2p scan for SiC, 1L, 2L and 4L MoS₂/SiC. The SiC Si 2p peak was resolvable into 2p_{1/2} and 2p_{3/2} peaks due to the asymmetry of the 2p peak and was done to enhance the accuracy of the subsequent band alignment calculations. The Si 2p_{3/2} peak for all samples except bulk MoS₂/SiC was aligned to the literature value of 100.4 eV to account for charging effects [55]. The spectra for the bare SiC showed no signs of atmospheric oxidation

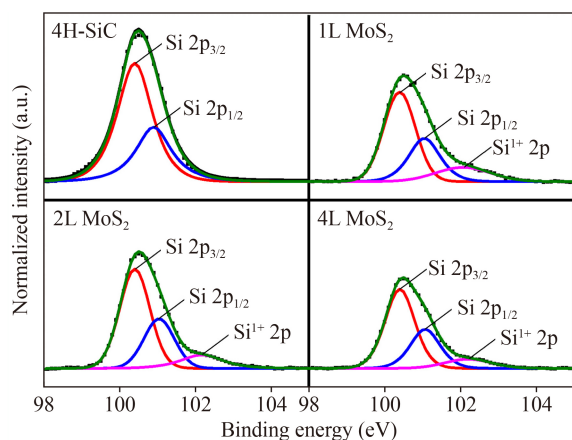


Fig. 2 XPS Si 2p spectra of SiC, 1L, 2L and 4L MoS₂/SiC.

which supports that the etching process removed any surface oxidation on the SiC surface. A convoluted Si¹⁺ peak can be observed at a binding energy of 102.07, 102.16 and 102.21 eV for the 1L, 2L and 4L MoS₂ grown samples. This is because the XPS characterization was done ex-situ and the MoS₂ grown samples were left in air for a longer time for deposition whereas the bare SiC substrate was characterized immediately after etching. Nonetheless, there was only a small amount of Si¹⁺ present.

Figure 3 shows the Mo 3d and S 2p scan for 1L, 2L, 4L and bulk MoS₂. The Mo⁴⁺ 3d_{5/2} and 3d_{3/2} peaks were located at 228.69, 228.59, 228.60 and 231.83, 231.78, 231.76 eV for 1L, 2L and 4L MoS₂/SiC. Whereas the S²⁻ 2p_{3/2} and 2p_{1/2} peaks were located at 161.50, 161.43, 161.42 and 162.69, 162.62, 162.62 eV respectively. This agrees well with the current literature values [56–59]. Due to the absence of the SiC peak in bulk MoS₂, the spectrum was aligned to the Mo 3d_{5/2} of 4L MoS₂ for presentation purposes. Due to incomplete sulfurization, small amounts of metallic Mo was detected on 2L (3.5 at %), 4L (3.0 at %) and bulk (2.0 at %) MoS₂.

To derive the band alignment of the MoS₂/SiC heterostructures, we adopted a direct method suggested by Santoni *et al.* [60], which requires to align the XPS spectra of SiC, 1L, 2L and 4L MoS₂ to a common reference. The alignment has already been done to the Si 2p_{3/2} peak at 100.4 eV. The Si 2p_{3/2} peak in SiC was chosen to be the reference as it can be detected in all the required samples in sufficient quantity. The valence band maximum (VBM) was then derived by the intersection of the linear regressions of the leading edge of valence band and the baseline of its spectra, as shown in Fig. 4. The VBM of bare SiC was identified to be 1.26 eV. Interestingly, the valence band spectra of 1L, 2L and 4L MoS₂/SiC exhibit two leading edges with different slope. According the following theoretical density of states (DOS), the red linear regression of the leading edge comes from the Mo 4d₂ orbital, while the

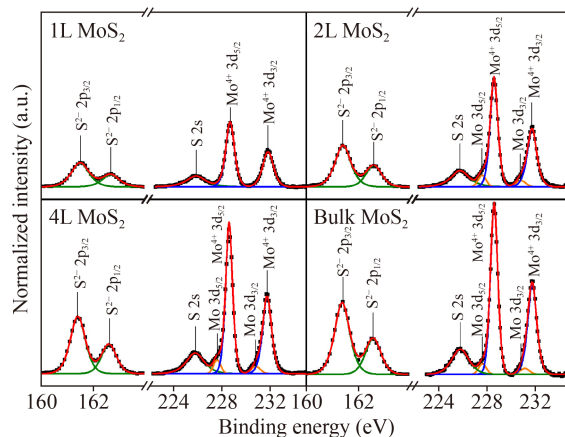


Fig. 3 XPS Mo 3d and S 2p spectra of 1L, 2L, 4L and bulk MoS₂/SiC.

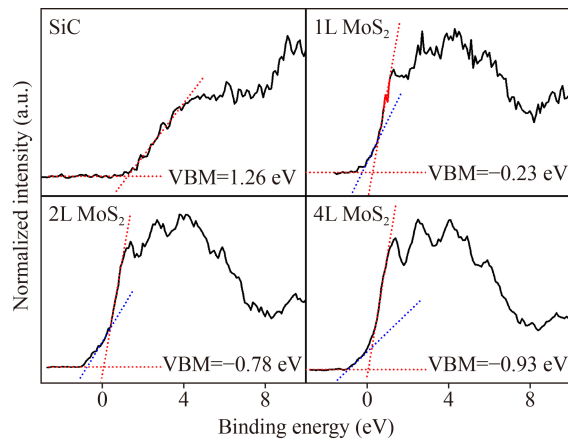


Fig. 4 Valence band spectra for SiC, 1L, 2L and 4L MoS₂/SiC.

blue one originates from the Mo $4d_{(x^2-y^2)}$ orbital. It is the Mo $4d_{(x^2-y^2)}$ orbital that contributes the VBM of 1L MoS₂. Then, the VBM of the 1L, 2L and 4L MoS₂/SiC could be identified to be -0.23 , -0.78 , and -0.93 eV, respectively. Further considering the VBM of SiC substrate of 1.26 eV, the valence band offsets (VBOs) could be obtained to be 1.49 , 2.04 , and 2.19 eV for the 1L, 2L and 4L MoS₂/SiC, respectively. Therefore, the VBOs increase from 1.49 eV up to 2.19 eV with increasing MoS₂ layers up to 4L.

To understand above experimental observations and illuminate the interfacial interactions and electronic properties in the MoS₂/4H-SiC heterostructures, the first-principles calculations were performed by using the Vienna ab initio simulation package (VASP). Figure 5 shows the optimized atomic structure of 1L MoS₂ and the HSE06 total DOSs of 1L MoS₂ and partial DOSs of Mo and S atom. The lattice constants of 3.19 Å and the HSE06 bandgap of 2.10 eV are obtained for 1L MoS₂. These basic parameters are in good agreement with the previous theoretical and experimental results [12, 61]. The conduction band minimum (CBM) and VBM

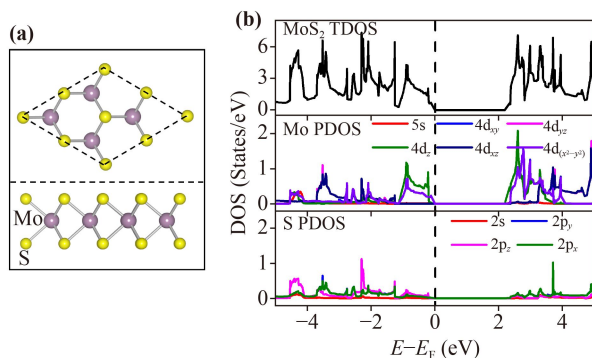


Fig. 5 (a) Top and side view of the optimized atomic structure of 1L MoS₂. (b) The HSE06 total DOSs of 1L MoS₂ and partial DOSs of Mo and S atom. The Fermi level is shifted to 0 eV.

mainly come from the Mo atom. Careful observation shows that the CBM mainly comes from the Mo $4d_{z^2}$ orbital. The valence band mainly originates from the $4d_{(x^2-y^2)}$ and $4d_{z^2}$ orbitals, while it's the $4d_{(x^2-y^2)}$ orbital that contributes the VBM of 1L MoS₂. The different orbital origins of the valence band lead to the formation of two different slope distributions in the DOS of MoS₂ at the valence band. This is consistent well with the above XPS valence band spectra of MoS₂, as exhibited in Fig. 4, and is beneficial to the accurate identification of the VBM of 1L, 2L and 4L MoS₂ in XPS valence bands.

The lattice constant of bulk 4H-SiC was obtained to be 3.09 Å, consistent with the reported value [62, 63]. The HSE06 bandgap is 3.49 eV, as shown in Fig. 6(b), which is only 7% larger than the reported bandgap of 3.26 eV. The lattice constant of the substrate 4H-SiC was slightly expanded in the model constructed to make it consistent with the lattice constants of MoS₂, due to the weak vdW interlayer interaction in multilayer MoS₂, which will be constructed in our calculations. The bandgap of the strained bulk 4H-SiC is 3.10 eV, as shown in Fig. 6(b), which is only about 5% smaller than the reported bandgap of 3.26 eV due to the slight tensile strain. Therefore, the model based on the strained 4H-SiC can effectively ensure the accurate results of the band alignment.

The 4H-SiC (0001) surface is usually formed by a Si or C atomic layer due to the polarity. Previous researches have demonstrated that the Si-face is energy stabilized surface as compared to the C-face [64]. Then

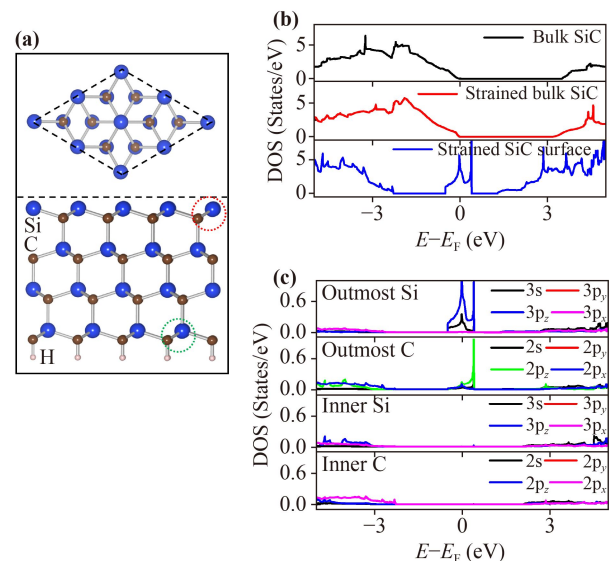


Fig. 6 (a) Top and side view of the strained atomic structure of 4H-SiC (0001) Si surface. (b) The HSE06 total DOSs of bulk, strained bulk 4H-SiC and strained 4H-SiC (0001) Si surface. (c) Partial DOSs of the outmost and inner Si and C atoms as indicated by the red and green circles in (a), respectively. The Fermi level is shifted to 0 eV.

the geometrical and electronic structures of 4H-SiC (0001) Si surface under slight tensile strain were adopted and calculated in this study. Figure 6(a) shows the strained atomic structure of 4H-SiC (0001) Si surface. Due to the Si dangling bond formed on the surface, the HSE06 total DOSs, as shown in Fig. 6(b), exhibit the strong surface states in the middle of the bandgap. And the Fermi level just passes through the surface states, indicating the Fermi level pinning, which is in agreement with the case of 2H-SiC (0001) Si surface [59]. Figure 6(c) shows the partial DOSs of the outmost and inner Si and C atoms as indicated by the red and green circles in Fig. 6(a), respectively. It is found that the mid-gap surface states mainly come from the outmost unsaturated Si atoms due to the broken Si-C bonds at the surface. It seems like that the outmost Si and C atomic layers contribute greatly to these surface states, while almost no surface states were observed in the bandgap for the inner Si and C atomic layers. Therefore, the inner Si and C atoms keep the electronic properties of bulk 4H-SiC.

Furthermore, 1L MoS₂/4H-SiC configuration was first constructed by a MoS₂ monolayer (containing 1 Mo, 2 S atomic layers) and 8 atomic layers of the 4H-SiC(0001) Si surface (containing 4 Si and 4 C atomic layers). Considering the lattice matching and hexagonal symmetry, six representational high-symmetric 1L MoS₂/4H-SiC configurations were constructed as shown in Figs. 7(a)–(f). The binding energies of these configurations were calculated by using the following equation:

$$E_b = E_{total} - E_{SiC} - E_{MoS_2}, \quad (1)$$

where E_{total} , E_{SiC} and E_{MoS_2} are the total energies of MoS₂/4H-SiC system, isolated 4H-SiC and MoS₂,

respectively. Among above six possible configurations, the preferred atomic structure of 1L MoS₂/4H-SiC, as shown in Figs. 7(c) and (f), has the lowest binding energy of -0.96 eV. In this energy favorable configuration, S atom locates on the top of the Si atom of the 4H-SiC (0001) Si surface, while Mo atom is aligned with the C atom below. As shown in Fig. 7(f), a distinct Si-S chemical bond is formed at the 1L MoS₂/4H-SiC interface. Moreover, the Si-S bond length is calculated to be 2.20 Å. This value is shorter than the interlayer distance of 2.5 – 3.5 Å in the vdW systems and similar with the Si-S bond length of 2.0 Å in other covalent bonding system [65, 66]. Above large binding energy and formation of a chemical Si-S bond indicate a significantly strong interlayer interaction at the 1L MoS₂/4H-SiC interface, agreement well with the above experimental observation of the quenching of PL.

To identify the interface properties, especially the band alignment, atomic DOSs of 1L MoS₂/SiC interface were calculated and aligned from bottom to up according to the atomic order, as shown in Fig. 8(a). As compared to that of bare 4H-SiC (0001) Si surface, the original mid-gap surface states vanish when 1L MoS₂ layer adsorbed on the surface. That is, the surface states originated from the Si dangling bond was passivated by the S atom by forming the chemical Si-S bond. Though the absence of the dangling bond of the MoS₂ surface, the existing of lone pair at the S surface accounts for the surface passivation of the SiC substrate, according to our previous calculations [67]. Then the Fermi level shifts up and passes through the bottom of conduction band, indicating the Fermi level pinning at the interface states of the 1L MoS₂/4H-SiC interface.

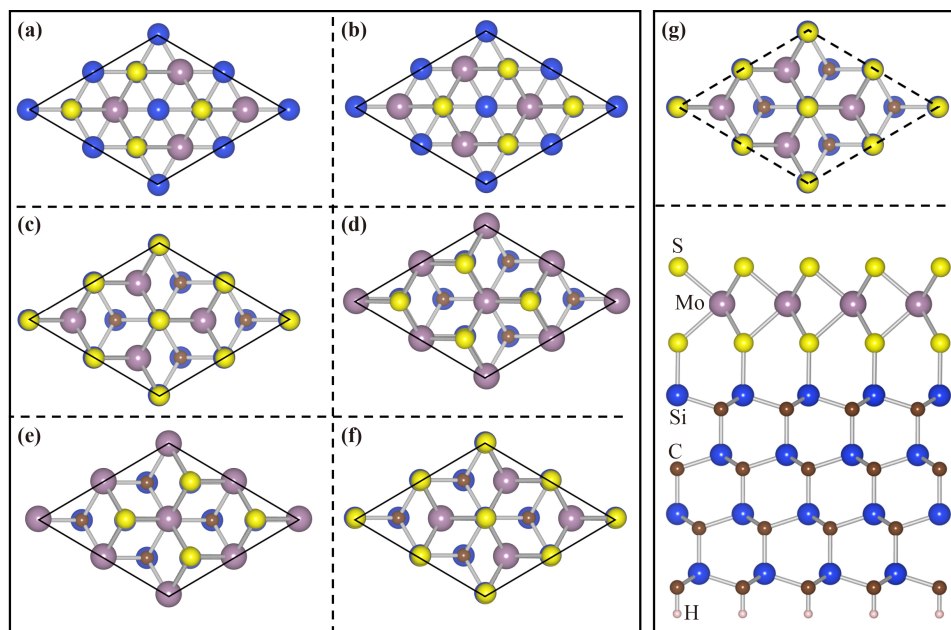


Fig. 7 (a–f) Top views of six representational high-symmetric 1L MoS₂/4H-SiC configurations. (g) Top and side view of the preferred atomic structure of 1L MoS₂/4H-SiC interface.

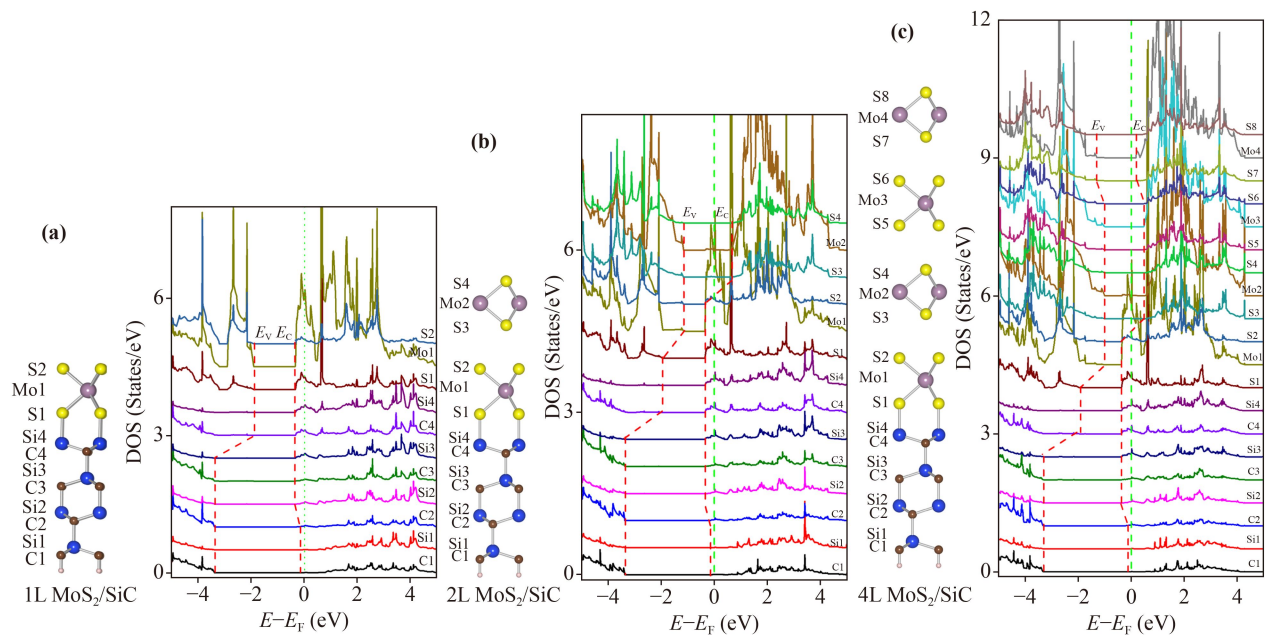


Fig. 8 (a) Atomic DOSs of 1L MoS₂/SiC, (b) 2L MoS₂/SiC, and (c) 4L MoS₂/SiC interfaces were calculated and aligned from bottom to up according to the atomic order illustrated in the left inset. The green dotted line indicates the Fermi level at 0 eV. The red dashed lines indicate the VBM and CBM of the different atomic layer of the MoS₂/SiC interface.

According to the real space atomic DOSs alignments, we could directly identify the VBM and CBM of each atomic layer and further obtained the band alignment at the 1L MoS₂/SiC interface, as indicated by the red dashed lines in Fig. 8. Atomic DOSs of bottom C1, Si1 and C2 exhibit the inner SiC electronic properties with the bandgap of about 3.23 eV. The CBM and VBM lie at the -0.13 and -3.36 eV, respectively. Looking up from Si2 to S2, the CBM shifts a little down to -0.33 eV. However, the VBM shifts largely up to -1.90 eV from C4 to S2. Then the bandgap shrinks from 3.23 eV to 3.03 eV, further to 1.57 eV from bottom to up in the 1L MoS₂/SiC heterostructure. And the CBM and VBM exhibits special asymmetric evolution, which would play important role in the carrier distribution. The new bandgap of 1L MoS₂ (1.57 eV) in 1L MoS₂/SiC heterostructure is much smaller than that of bare MoS₂ (2.10 eV), further indicating a strong interaction between MoS₂ and SiC. The interaction at the MoS₂/SiC interface is totally different from the previous weak interaction among the WS₂/HfO₂, MoS₂/HfO₂ and WS₂/SiO₂ interfaces [68–70]. The CBM and VBM distributions demonstrate the formation of type I band alignment at the 1L MoS₂/SiC interface. However, due to the special asymmetric evolution, the CBM locates at the SiC and MoS₂ region, while the VBM only lies at the MoS₂ region, different from the traditional type I band alignment.

Moreover, the charge transfer between 1L MoS₂ and SiC substrate was identified by using the Bader analysis [45]. The calculation results show that about 0.96e transfer from the SiC substrate to the MoS₂ layer. Since

that, an electric field forms and it directs from SiC to MoS₂. The built-in electric field is significantly important for the high performance of the electronic and optoelectronic devices based on the MoS₂/SiC heterostructure [20]. Combined with the built-in electric field and the special band alignment, the photogenerated holes tend to accumulate to the MoS₂ layer, while electrons tend to accumulate to the SiC layer. The efficiency of separation of photogenerated carriers is benefiting for enhancing the performance of the photodetectors and photocatalytic devices.

In order to understand the evolution of VBOs as the MoS₂ thickness increasing, atomic DOSs of 2L and 4L MoS₂/SiC interfaces were further calculated and aligned from bottom to up according to the atomic order, as shown in Figs. 8(b) and (c). According to our group's works by using the X-ray diffraction (XRD) and TEM [46, 47], the as-grown multilayer MoS₂ on different substrates tend to exhibit the 2H-phase. And our and others previous first-principles calculations [12, 71] also demonstrated that the 2H-phase MoS₂ generates the lowest formation energy in the multilayer stacking systems. Therefore, 2H-phase MoS₂ were mainly constructed to clarify the issue of band alignment evolution in this work. For 2L MoS₂/SiC interface, the Fermi level also passes through the interface states locating at the region between the lower MoS₂ monolayer and SiC substrate, similar with the case at 1L MoS₂/SiC interface. However for the upper MoS₂ layer, almost no interface states were observed in the bandgap and the Fermi level just passes through the center of the band gap. Therefore, Fermi level was just pinning at the interface states

in the lower 1L MoS₂/SiC heterostructure. The interaction between the upper and lower MoS₂ layer keeps the weak vdW interaction. That is, the lower MoS₂ monolayer can effectively passivate the dangling bonds of the SiC surface and obtain a new weak vdW surface.

We could also identify the VBM and CBM of each atomic layer and obtain the band alignment at the 2L MoS₂/SiC interface, as indicated by the red dashed lines in Fig. 8(b). Similar with 1L MoS₂/SiC, atomic DOSs of bottom C1, Si1 and C2 also exhibit the inner SiC electronic properties with the bandgap of about 3.23 eV. The CBM and VBM also lie at the -0.13 and -3.36 eV, respectively. Looking up from Si2 to S2, the CBM firstly shifts a little down to -0.33 eV. Then, the CBM shifts largely up to 0.67 eV from S3 to S4, further indicating the weak vdW interaction between the upper and lower MoS₂ layer. For the VBM, it firstly shifts largely up to -1.95 eV from C4 to S1. Then, the VBM continues to shift up to -1.14 eV from Mo1 to S4 due to the interlayer orbital coupling interaction in bilayer MoS₂. The interlayer orbital coupling tends to split and shift the VBM to the higher energy region, which has been demonstrated in the previous theoretical calculations of multilayer MoS₂ [12]. Then the bandgap changes from 3.23 eV to 3.03 eV, 1.62 eV, 0.81 eV and 1.81 eV from bottom to up in the 2L MoS₂/SiC heterostructure. Therefore, the CBM and VBM exhibit more complicated asymmetric evolution, as compared with that of 1L MoS₂/SiC. The CBM locates at the interface from Si2 to S2, while the VBM locates at the MoS₂ layer from Mo1 to S4. On the whole, the complicated CBM and VBM evolution demonstrates the type II band alignment of the 2L MoS₂/SiC heterostructure. Compared with 1L MoS₂/SiC, 2L MoS₂/SiC possess thicker layers, smaller bandgap distribution, and type II band alignment, which would more effectively improve the light absorption, separate the photogenerated carriers and enhance the performance of optoelectronic devices.

For 4L MoS₂/SiC interface, the interface states also locate at the region from Si2 to S2, and Fermi level is pinned at them, similar with the cases in the 1L and 2L MoS₂/SiC interfaces. Except the lowest MoS₂ monolayer, the upper MoS₂ layers show almost no interface states in the bandgap and the Fermi level just passes through the center of the bandgap. Therefore, Fermi level was just pinning at the interface states in the lowest 1L MoS₂/SiC heterostructure. The interaction between the upper MoS₂ layers and lowest MoS₂ monolayer keeps the weak vdW interaction, similar with the case of 2L MoS₂/SiC interface.

The evolution of CBM and VBM of 4L MoS₂/SiC interface, as indicated by the red dashed lines in Fig. 8(c), exhibits the similar trend as that of 2L MoS₂/SiC interface. Atomic DOSs of bottom C1, Si1 and C2 also exhibit the inner SiC electronic properties with the bandgap of about 3.23 eV. The CBM and VBM also lie

at the -0.13 and -3.36 eV, respectively. Looking up from Si2 to S2, the CBM firstly shifts a little down to -0.37 eV. Then, the CBM shifts largely up to 0.48 eV from S3 to S6, indicating the weak vdW interaction between the upper and lowest MoS₂ layer. And from S7 to S8, the CBM shifts a little down to 0.19 eV. For the VBM, it firstly shifts largely up to -1.90 eV from C4 to S1. Then, the VBM continues to shift up to -1.00 eV from Mo1 to S6 due to the interlayer orbital coupling interaction in quadlayer MoS₂. Finally, the VBM shifts a little down to -1.29 eV. Then the bandgap changes from 3.23 eV to 2.99 eV, 1.53 eV, 0.63 eV, 1.48 eV, and 1.48 eV from bottom to up in the 4L MoS₂/SiC heterostructure. The CBM locates at the interface of lowest MoS₂ and SiC substrate from Si2 to S2, while the VBM locates at the MoS₂ layer from Mo1 to S6. The 4L MoS₂/SiC heterostructure also exhibits the type II band alignment. It will have similar advantages to 2L MoS₂/SiC in enhancing electronic and optoelectronic device performance.

Figures 9 (a)–(c) exhibits the theoretical energy band alignments at 1L, 2L, and 4L MoS₂/SiC interfaces. The 1L MoS₂/SiC interface exhibits the type I band alignment with the CBO of 0.20 eV and VBO of 1.46 eV. The large VBO and small CBO are attributed to the strong interaction between 1L MoS₂ and SiC substrate and the Fermi level pinning at the interface states. Due to the asymmetric CBM and VBM evolution, it is different from the traditional type I band alignment. Further due to the built-in electric field, the photogenerated electrons accumulate at SiC region, while the photogenerated holes accumulate at MoS₂ region.

For the 2L (4L) MoS₂/SiC interface, they exhibit the type II band alignments and the CBO and VBO are 0.80 (0.61) and 2.22 (2.36) eV, respectively. The enlarged CBOs and VBOs, as compared to that of 1L MoS₂/SiC, are attributed to the weak vdW interaction between the lower MoS₂ monolayer and upper MoS₂ layers and the strong interlayer orbital coupling. Then the photogenerated electrons and holes tend to accumulate to the SiC and MoS₂ layer, respectively, due to type II band alignment and built-in electric field. The high efficiency of charge separation in these heterostructures will enhance the device performance significantly. Finally, both the experimental and theoretical (2H-phase stacking) VBOs of the 1L, 2L and 4L MoS₂/SiC interfaces were plotted in Fig. 9(d). As the MoS₂ thickness increasing from 1L to 4L, the experimental VBO increases from 1.49 eV to 2.04 eV and 2.19 eV. Theoretically, the VBO (2H-phase stacking) increases from 1.46 eV to 2.22 eV and 2.36 eV. Therefore, the experimental and theoretical (2H-phase stacking) VBOs are in good agreement with each other, confirming the validity of the interface properties demonstrated in this study.

Since the metastable 3R-phase stacking MoS₂ were also observed in experiments, we further calculated the atomic DOSs of the 3R-phase stacking 2L and 4L MoS₂/

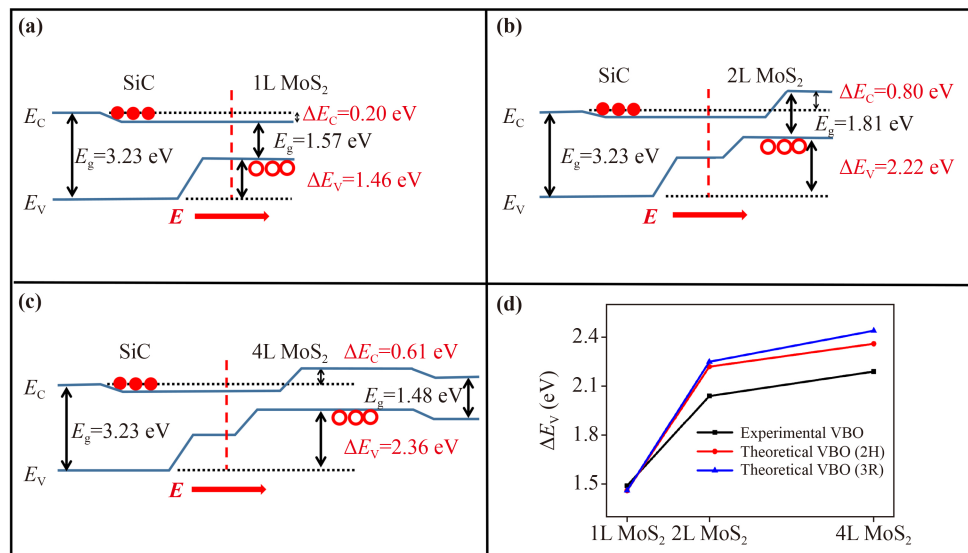


Fig. 9 Theoretical energy band alignments at (a) 1L, (b) 2L, and (c) 4L MoS₂/SiC interfaces. The red dashed line indicates the real space boundary between the SiC and MoS₂ layers. The VBO and CBO were indicated by ΔE_V and ΔE_C . The built-in electric field and photogenerated electrons and holes were denoted by the red arrow, red solid and hollow circles, respectively. (d) Comparison of the experimental and theoretical VBOs (2H-phase and 3R-phase stacking) of the 1L, 2L and 4L MoS₂/SiC interfaces.

SiC heterostructures and the theoretical VBOs (3R-phase stacking) of the 2L and 4L MoS₂/SiC interfaces were also exhibited in Fig. 9(d) for comparison. The VBOs of 3R-phase stacking structure increase from 1.46 eV to 2.25 eV and 2.44 eV. As compared to that of 2H-phase, the increased theoretical VBO of 3R-phase stacking is attributed to the small shrunken bandgap and enlarged VBM of the 3R-phase MoS₂, which is agreement with the previous study [71]. Therefore, both the theoretical 2H- and 3R-phase VBOs are in good agreement with each other, and the 2H- and 3R-phase stacking structures do not influence significantly the interfacial properties, especially the band alignment, of multilayer MoS₂/SiC heterostructures.

4 Conclusions

In summary, the interfacial properties of 2D/3D MoS₂/4H-SiC heterostructures were studied by XPS and first-principles calculations. It is found that the VBO increases from 1.49 eV to 2.19 eV with increasing MoS₂ layers up to 4L, consistent with the first-principles calculations. For 1L MoS₂/SiC, a strong interlayer interaction between the 1L MoS₂ and SiC sheet was demonstrated by the formation of a chemical Si-S bond (2.20 Å) and a large binding energy (−0.96 eV/unit cell). Then Fermi level pinning at the interface states and a passivation of 4H-SiC (0001) Si surface were realized. About 0.96e per unit cell transferring from SiC to MoS₂ monolayer indicates an electric field forms, and it directs from SiC to MoS₂. Then, the 1L MoS₂/SiC interface exhibits the type I band alignment with the CBO of 0.20 eV and

VBO of 1.46 eV. The large VBO and small CBO are attributed to the strong interaction between 1L MoS₂ and SiC substrate and the Fermi level pinning at the interface states. For 2L and 4L MoS₂/SiC, Fermi level was just pinning at the lower MoS₂ monolayer. The interaction between the upper MoS₂ layers and lower MoS₂ monolayer keeps the weak vdW interaction. Then, they exhibit the type II band alignments and the CBO and VBO are 0.80 (0.61) and 2.22 (2.36) eV, respectively. The enlarged CBOs and VBOs are attributed to the weak vdW interaction between the lower MoS₂ monolayer and upper MoS₂ layers and the strong interlayer orbital coupling. Despite of the different band alignment type, the photogenerated electrons and holes tend to accumulate to the SiC and MoS₂ layer, respectively, due to the asymmetric band alignment and built-in electric field in all the 1L, 2L and 4L MoS₂/SiC interfaces. The high efficiency of charge separation in these heterostructures will enhance the device performance significantly. The multiple interfacial interactions discussed in this study provide a new modulated perspective for the next-generation electronics and optoelectronics based on the 2D/3D semiconductors heterojunctions.

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