

Supporting Materials

Table S1 Structural parameters for FeTiO_x and FeCe_aTiO_x catalysts

Samples	Specific surface area (m ² /g)	Pore volume (cm ³ /g)
FeTiO _x	119.8	0.21
FeCe _{0.1} TiO _x	146.4	0.22
FeCe _{0.2} TiO _x	175.4	0.21
FeCe _{0.6} TiO _x	91.8	0.10
FeCeTiO _x	35.9	0.08
FeCe _{1.6} TiO _x	62.1	0.08

Table S2 Semi-quantitative analysis of surface atomic concentration (% in molar ratio) over FeCe_{0.2}TiO_x, FeTiO_x catalysts and Fe₂O₃, CeO₂ reference samples derived from XPS data

Samples	Fe (%)	Ce (%)	Ti (%)	O (%)	Ce ³⁺ /(Ce ³⁺ + Ce ⁴⁺) (%)	O _a /(O _β + O _a) (%)
FeCe _{0.2} TiO _x	44.4	1.4	2.3	51.9	20.0	42.8
FeTiO _x	45.0	–	2.7	52.3	–	38.5
Fe ₂ O ₃	49.0	–	–	51.0	–	34.1
CeO ₂	–	35.7	–	64.3	19.8	29.0

Table S3 Peak deconvolution results of the H₂-TPR profiles for Fe₂O₃, FeTiO_x and FeCe_{0.2}TiO_x

Samples	H ₂ consumption (mmol H ₂ /g) ^{a)}	Area ratio (%) ^{b)}		
		Black	Red	Blue
Fe ₂ O ₃	18.9	12.8 (298+323+338)	21.2 (432+504)	66.0 (560+626+697+757)
FeTiO _x	9.4	18.1 (293+305+319)	16.0 (431)	65.9 (513+585+743+825+865)
FeCe _{0.2} TiO _x	7.5	25.3 (304+315)	10.4 (438)	64.3 (511+598+694+815+885)

Notes: a) The calculated H₂ consumption amount and area ratios of sub-bands derived from the deconvoluted H₂-TPR profiles. b) The values in the parentheses represented the temperature positions of H₂ consumption peaks

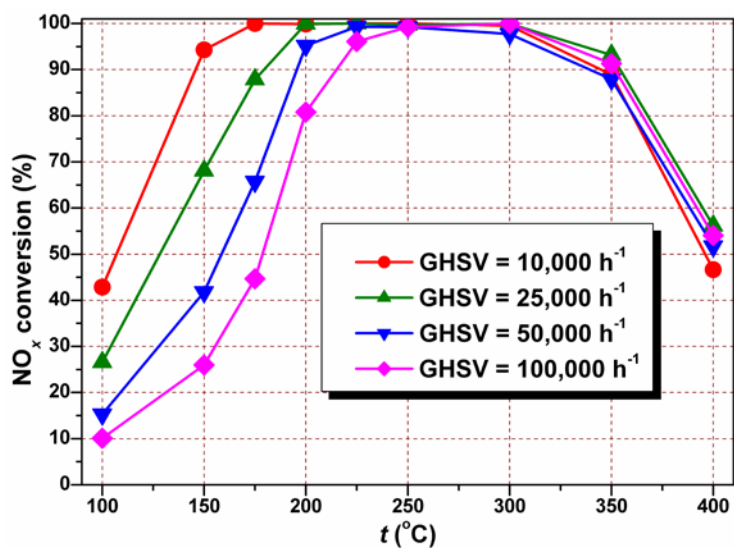


Fig. S1 The influences of GHSV on the NO_x conversion over FeCe_{0.2}TiO_x catalyst

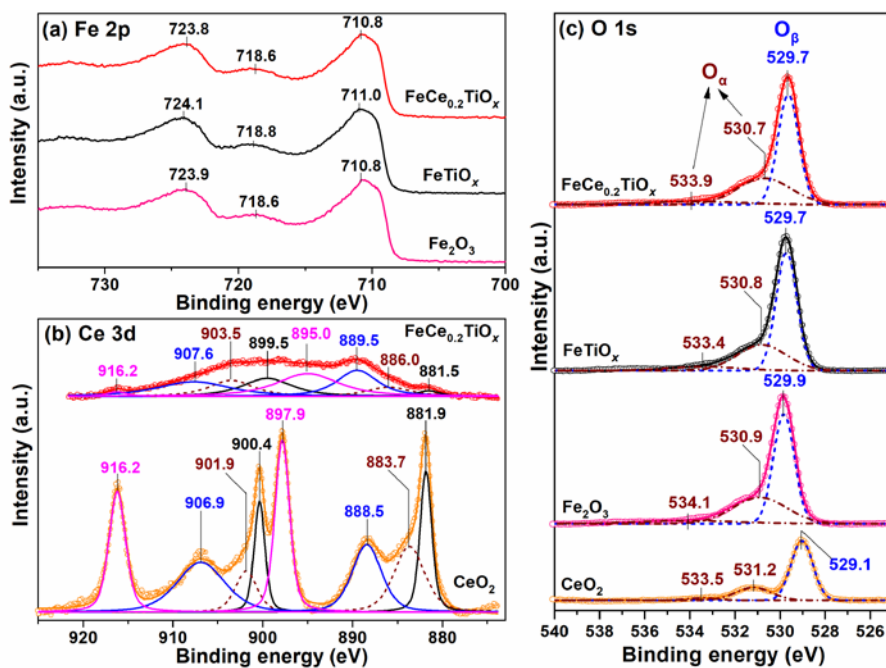


Fig. S2 XPS results of (a) Fe 2p, (b) Ce 3d, and (c) O 1s in FeTiO_x, FeCe_aTiO_x catalysts together with Fe₂O₃ and CeO₂ reference samples

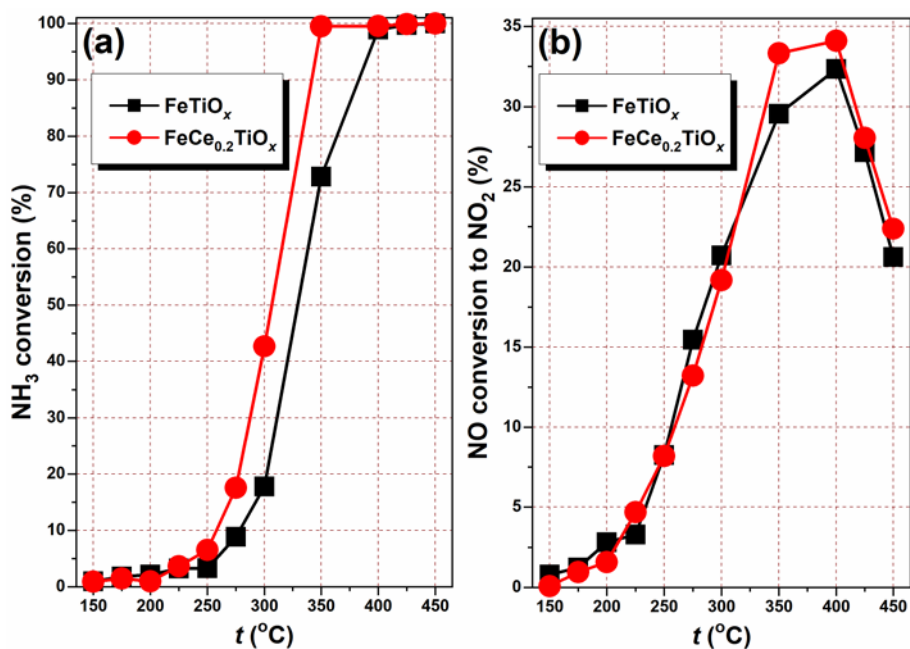


Fig. S3 (a) NH₃ conversion and (b) NO conversion to NO₂ as a function of the reaction temperature in NH₃ oxidation reaction and NO oxidation reaction over FeTiO_x and FeCe_{0.2}TiO_x catalysts

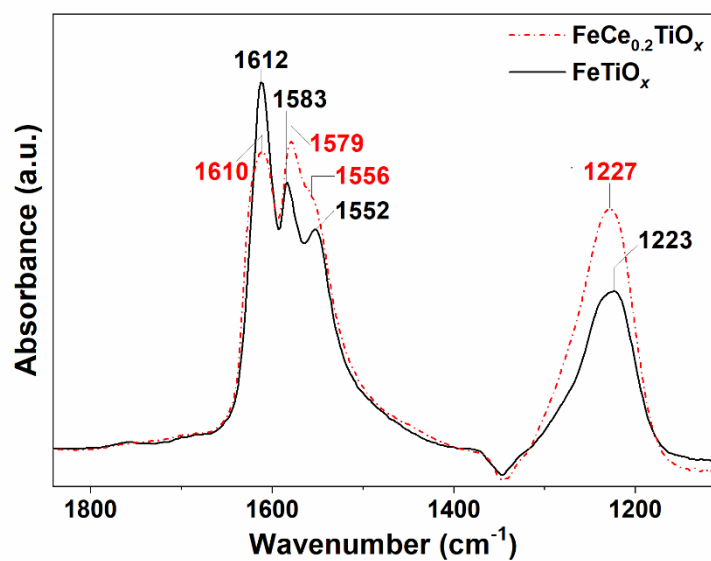


Fig. S4 *In situ* DRIFTS of NO + O₂ adsorption on FeTiO_x and FeCe_{0.2}TiO_x at 150°C collected in N₂ flow