

# Enhanced desulfurization performance of hybrid membranes using embedded hierarchical porous SBA-15

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## Electronic Supplementary Material

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## S1 Nitrogen adsorption-desorption isotherms

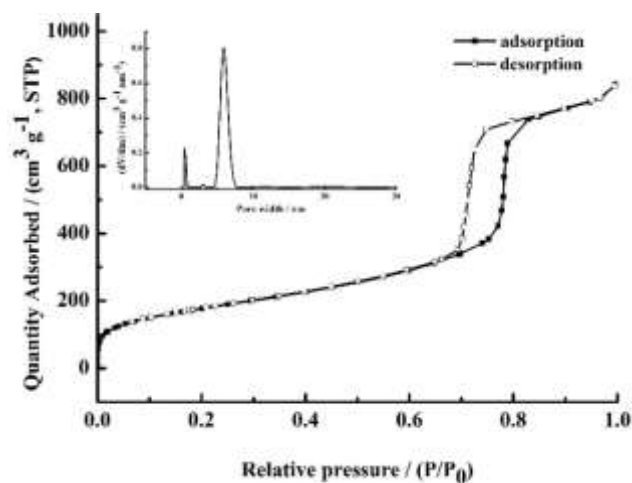


Fig.S1. Nitrogen adsorption-desorption isotherms of SBA-15 (Inset indicates the pore size distribution of SBA-15)

## S2 DSC study of membranes

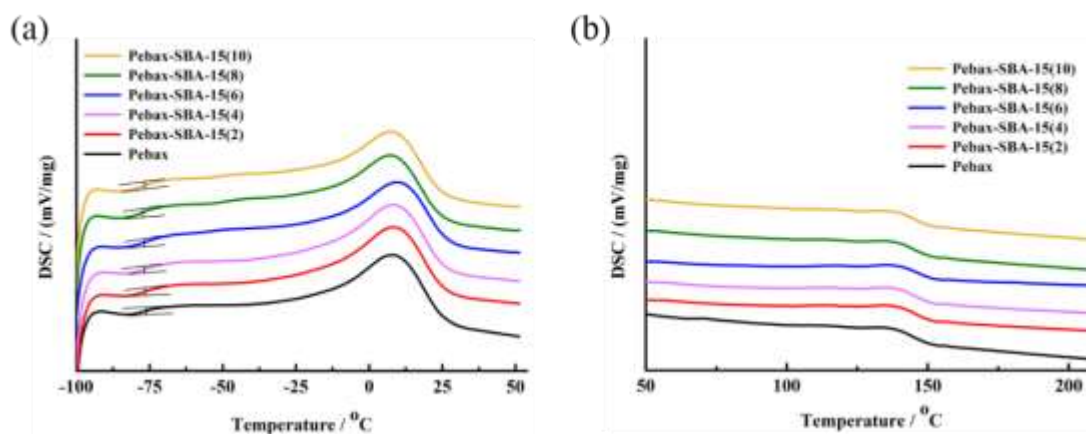


Fig.S2. DSC curves of pristine Pebax membrane and Pebax-SBA-15(X) membranes: (a) low-temperature zone; (b)

high-temperature zone

**Table S1**  $T_g$  and  $T_m$  of pristine Pebax membrane and Pebax-SBA-15(*X*) membranes.

Membrane	$T_g$ <sup>a)</sup> / °C	$T_{m1}$ <sup>b)</sup> / °C	$T_{m2}$ <sup>c)</sup> / °C
<b>Pebax</b>	-75.79	7.8	133.4
<b>Pebax-SBA-15(2)</b>	-75.83	8.5	133.9
<b>Pebax-SBA-15(4)</b>	-76.91	8.6	133.9
<b>Pebax-SBA-15(6)</b>	-77.04	9.5	134.0
<b>Pebax-SBA-15(8)</b>	-77.87	7.2	134.8
<b>Pebax-SBA-15(10)</b>	-77.88	7.2	134.0

a)  $T_g$ : glass-transition temperature; b)  $T_{m1}$ : melting point of PTMO segments; c)  $T_{m2}$ : melting point of PA12 segments

### S3 Free volume characteristics of membranes

**Table S2** Free volume characteristics of pristine Pebax membrane and Pebax-SBA-15(*X*) membranes.

Membrane	$I_3$ <sup>a)</sup> / %	$r_3$ <sup>b)</sup> / nm	$f_{app}$ <sup>c)</sup>
<b>Pebax</b>	21.90	0.3414	3.6484
<b>Pebax-SBA-15(2)</b>	22.11	0.3416	3.6737
<b>Pebax-SBA-15(6)</b>	22.32	0.3428	3.7643
<b>Pebax-SBA-15(10)</b>	22.54	0.3402	3.7156

(a)  $I_3$ : intensity of ortho-positronium; b)  $r_3$ : the free volume cavities; c)  $f_{app}$ : the free volume

## S4 Comparison of the membrane separation performance

**Table S3** Comparison of the membrane separation performance in this study with previous desulfurization membranes in literatures.

Membrane	Temperature /K	Sulfur content in feed/ppm	Thickness/ $\mu\text{m}$	Permeation flux/(kg/(m <sup>2</sup> h))	Enrichment factor	Reference
Pebax-SBA-15(6)/PSf	333	500	2	22.07	6.76	This study
Pure Pebax	333	500	2	16.99	4.49	This study
Pebax-Cu <sup>+</sup> Fe <sup>2+</sup> @CNS-5/PSf	333	500	2	13.42	7.11	[1]
Pebax-MoS <sub>2</sub> -4/PSf	333	500	1.8	11.42	9.11	[2]
Pebax-Ag <sup>+</sup> @SNW-1-9/PSf	333	500	2	16.35	6.80	[3]
PDMS-MIL101(Cr)-6/PVDF	313	500	15	5.2	5.6	[4]
PDMS-NH <sub>3</sub> -(TMOS)16/PSf	303	500	/	7.36	4.98	[5]
Pebax/PVDF	313	1280	11	3.8	4.0	[6]
PDMS-dopamine/Cu/PSf	303	500	16	7.42	4.81	[7]
PDMS-DATi(2.2.2.2)/PSf	313	500	18	6.61	4.80	[8]
Ethyl Cellulose-C <sub>60</sub>	348	300	/	2.32	4.72	[9]
PEG-CuY	383	1190	/	3.19	2.95	[10]
PDMS-Ag <sup>+</sup> -SiO <sub>2</sub> /PES	306	500	14	7.76	4.3	[11]
PES/Pluronic F127	303	500	200	3.1	3.5	[12]
PDMS-Ni <sup>2+</sup> Y/PS	303	500	/	3.26	4.84	[13]
PDMS-DAAg/5-5.0/PSf	313	500	16~25	8.22	5.03	[14]
PDMS-Ag <sup>+</sup> /TiO <sub>2</sub> (0.01)-5.0/PSf	313	500	/	4.14	8.56	[15]
PDMS-GNS(0.2)/PVDF	313	500	20	6.22	3.58	[16]
PDMS-CuBTC-8/PVDF	313	500	35	6.47	5.20	[17]
PDMS-AgY/PAN	323	/	15	8.15	3.50	[18]
PEG/PU	383	1200	/	2.5	4.03	[19]
Cl-PBPP	353	400	12.4	1.38	5.6	[20]
Pebax-CuBTC/PSf	343	500	2	16.45	4.04	[21]
Pebax-Ag-PDA/GNS-6/PSf	313	500	6	4.42	8.76	[22]
Pebax-Ag-PDA/GNS-6/PSf	333	500	6	14	6.48	[22]
Pebax-HPSiNT(2)/PSf	333	500	1.8	19.0	5.6	[23]

### Abbreviation note:

**CNS:** Carbon nitride nanosheets      **SNW-1:** N-rich Schiff based porous organic

framework

**MIL101(Cr):** Materials of Institute Lavoisier Frameworks 101

**TMOS:** Tetramethyl orthosilicate      **PVDF:** Polyvinylidene difluoride

**DATi:** Dopamine/titanium(IV) bis-(ammonium lactate) dihydroxide

**C<sub>60</sub>:** Fullerene      **PEG:** Poly(ethylene glycol)      **PES:** Poly(ether sulfone)

**PS:** Polysulfone      **DAAg:** Dopamine-silver nanoparticles      **GNS:** Graphene nanosheets

**CuBTC:** Metal-organic framework from Hong Kong University of Science and Technology

**PAN:** Polyacrylonitrile      **PU:** Polyurethane

**Cl-PBPP:** Chlorine incompletely substituted poly [bis (phenoxy) phosphazene]

**PDA:** Polydopamine      **HPSiNT:** Hierarchical porous organosilica nanotubes

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