

## Electronic Supplementary Material

### *In-situ* hydrophobic environment triggering reactive fluorescence probe to real-time monitor mitochondrial DNA damage

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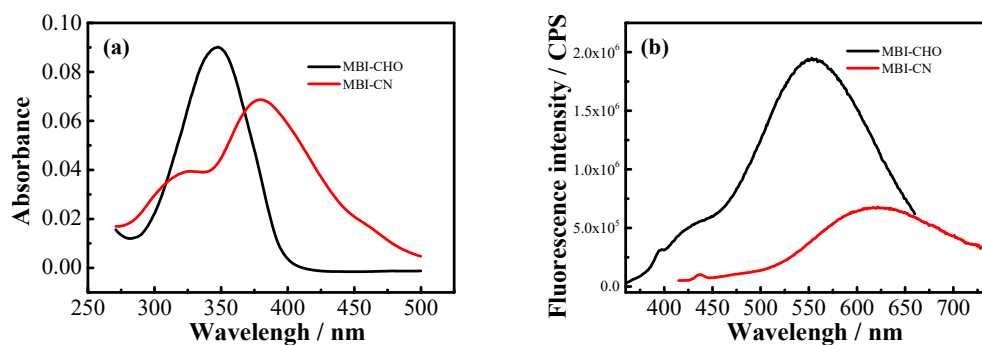
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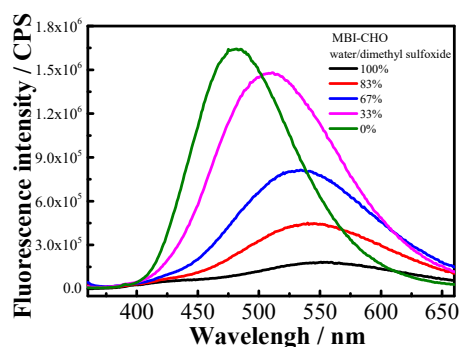
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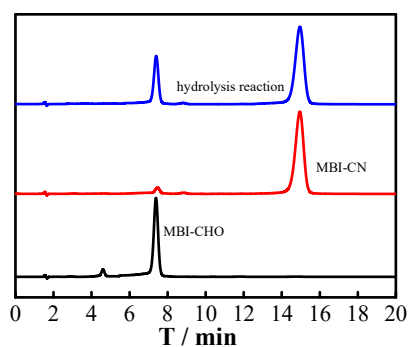
**Fig. S1** Absorption and emission of MBI-CN and MBI-CHO in water.



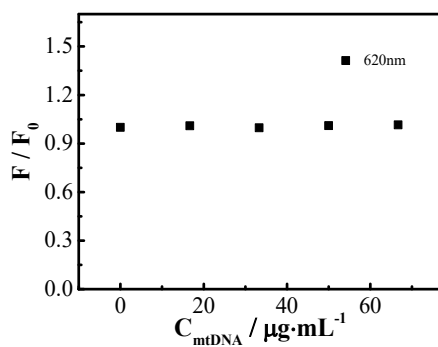
**Fig. S2** The fluorescence spectra of MBI-CHO in different water and dimethyl sulfoxide mixed solutions with hydrophilicity and hydrophobicity.

Mechanism of hydrolysis of probe MBI-CN

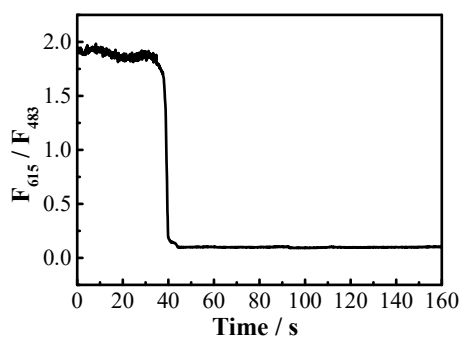
The hydrolysis of MBI-CN to MBI-CHO was verified by High Performance Liquid Chromatography (HPLC). According to the results of chromatogram, the retention time of probe MBI-CN was 14.95 min, and the retention time of MBI-CHO was 7.39 min. When the probe MBI-CN was placed in a hydrophobic environment, the results of chromatogram showed that a new chromatogram peak appeared at 7.40 min. This is basically consistent with the retention time of the chromatogram peak of MBI-CHO, which proves that the probe MBI-CN can undergo a hydrolysis reaction in a hydrophobic environment to generate MBI-CHO.



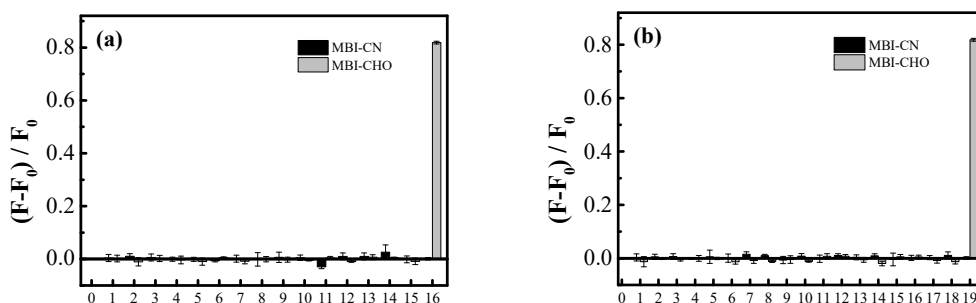
**Fig. S3** The hydrolysis of MBI-CN to MBI-CHO in hydrophobic environment was verified by HPLC. (mobile phase: water/methanol (1:1, v/v), velocity of flow: 1.5 mL/min, ultraviolet detection wavelength: 380 nm).



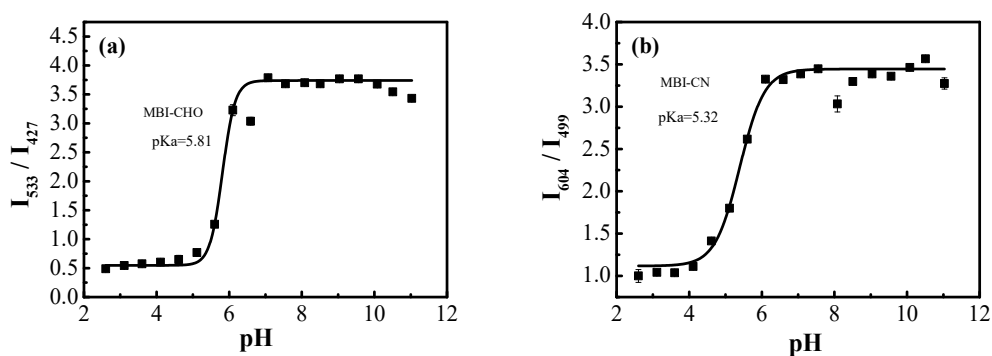
**Fig. S4** Fluorescence response after adding mtDNA to MBI-CN in water. ( $F_0$ : the initial fluorescence intensity of probes MBI-CN;  $F$ : the fluorescence intensity after adding mtDNA).



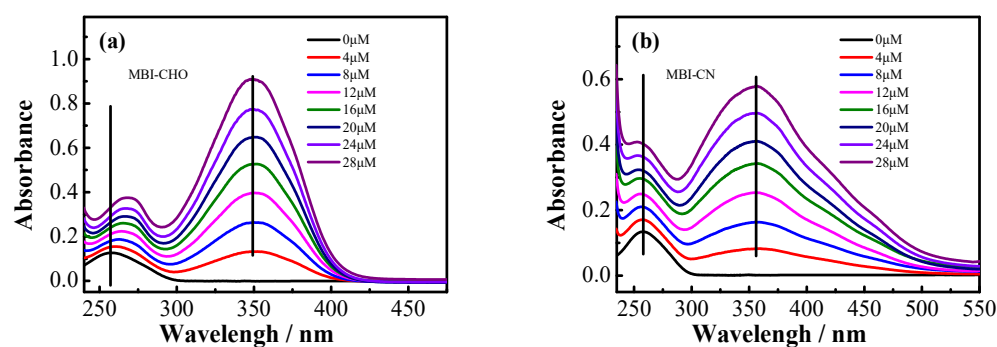
**Fig. S5** Hydrolysis kinetics of MBI-CN.



**Fig. S6** Determination of MBI-CHO interference with related active small molecules in organisms. (a) Ion interference determination. 0 Blank; 1 SnCl<sub>2</sub>; 2 CaCl<sub>2</sub>; 3 MnCl<sub>2</sub>; 4 CoCl<sub>2</sub>; 5 CaSO<sub>4</sub>; 6 HgCl<sub>2</sub>; 7 Zn(NO<sub>3</sub>)<sub>2</sub>·6H<sub>2</sub>O; 8 FeCl<sub>2</sub>·7H<sub>2</sub>O; 9 CaCl<sub>2</sub>·6H<sub>2</sub>O; 10 FeCl<sub>3</sub>·6H<sub>2</sub>O; 11 Al(NO<sub>3</sub>)<sub>3</sub>; 12 MgSO<sub>4</sub>; 13 Li<sub>2</sub>CO<sub>3</sub>; 14 Na<sub>2</sub>CO<sub>3</sub>; 15 dipotassium hydrogen phosphate (20 μmol/L); 16 mtDNA (100 μg/mL). (b) Determination of amino acid interference. 0 Blank; 1 glycyl-DL-phenylalanine; 2 lysine; 3 DL-threonine; 4 cystine; 5 glutamine; 6 glutamic acid; 7 D-methionine; 8 L-cysteine; 9 glycine; 10 DL-leucine; 11 L-glutamic acid; 12 arginine; 13 hypoxanthine; 14 L-aspartic acid; 15 valine; 16 DL-Homocysteine; 17 Serine; 18 Glutathione (20 μmol/L); 19 mtDNA (100 μg/mL). (F<sub>0</sub>: the initial fluorescence intensity of probes MBI-CN or MBI-CHO; F: the fluorescence intensity after adding interference).



**Fig. S7** Fluorescence signal changes of MBI-CHO and MBI-CN at different pH.



**Fig. S8** In the Tris-HCl (0.05 mol, pH = 7.4), the concentration of fixed mtDNA (5 μg/mL) increased the concentration of the molecule (0-28 μmol/L). The interaction of

MBI-CHO and MBI-CN with mtDNA were tested by ultraviolet absorption spectroscopy

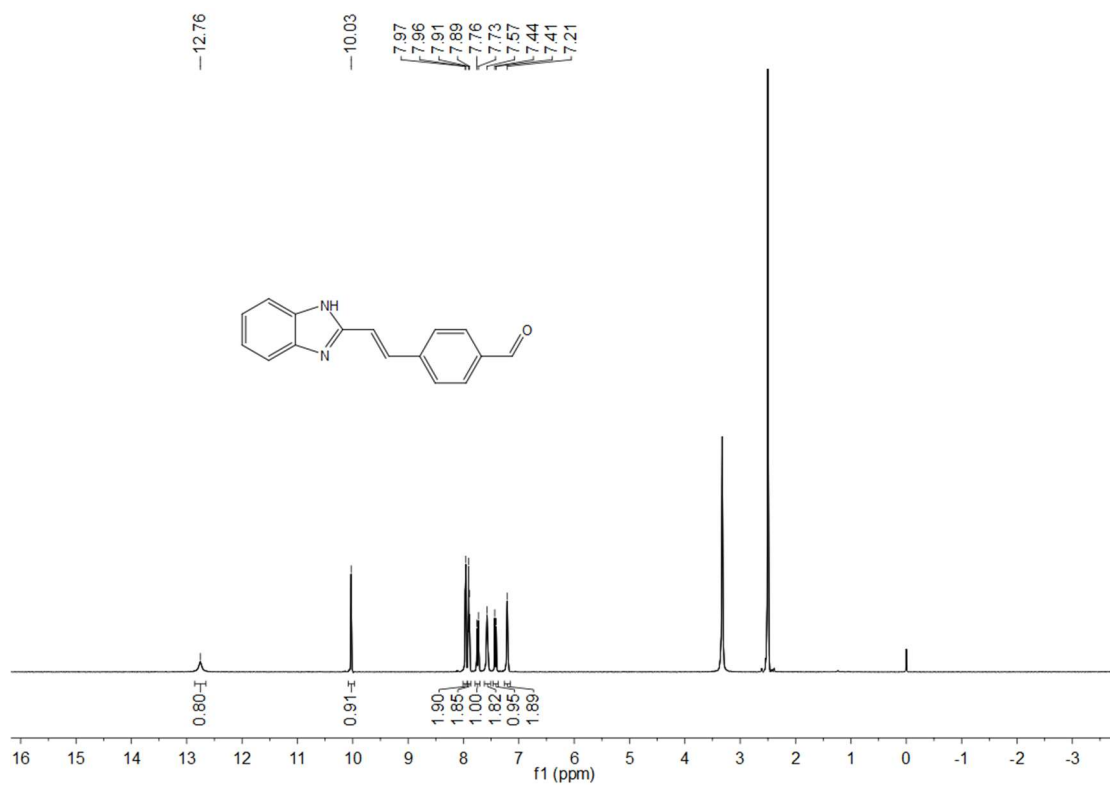


Fig. S9  $^1\text{H}$  NMR of compound MBI-CHO

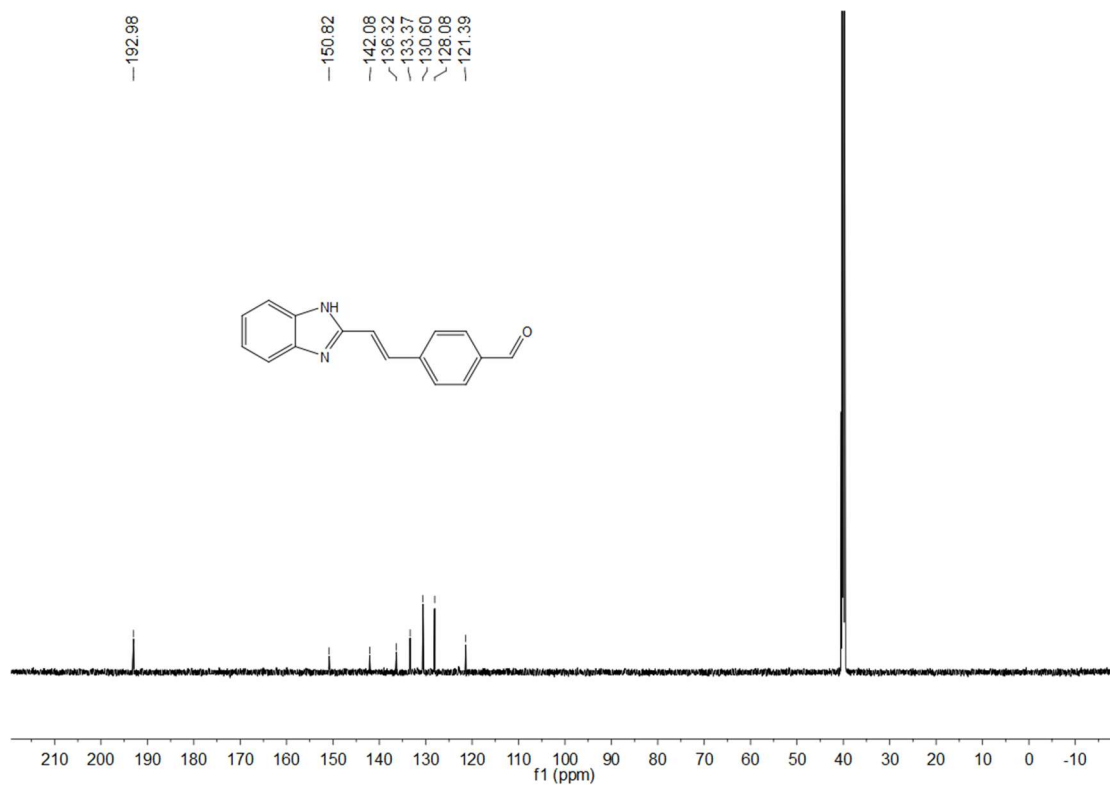
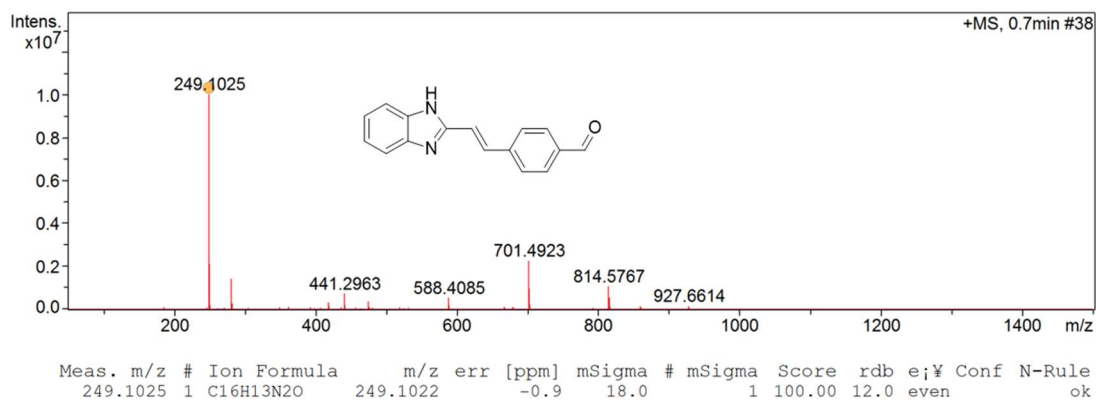
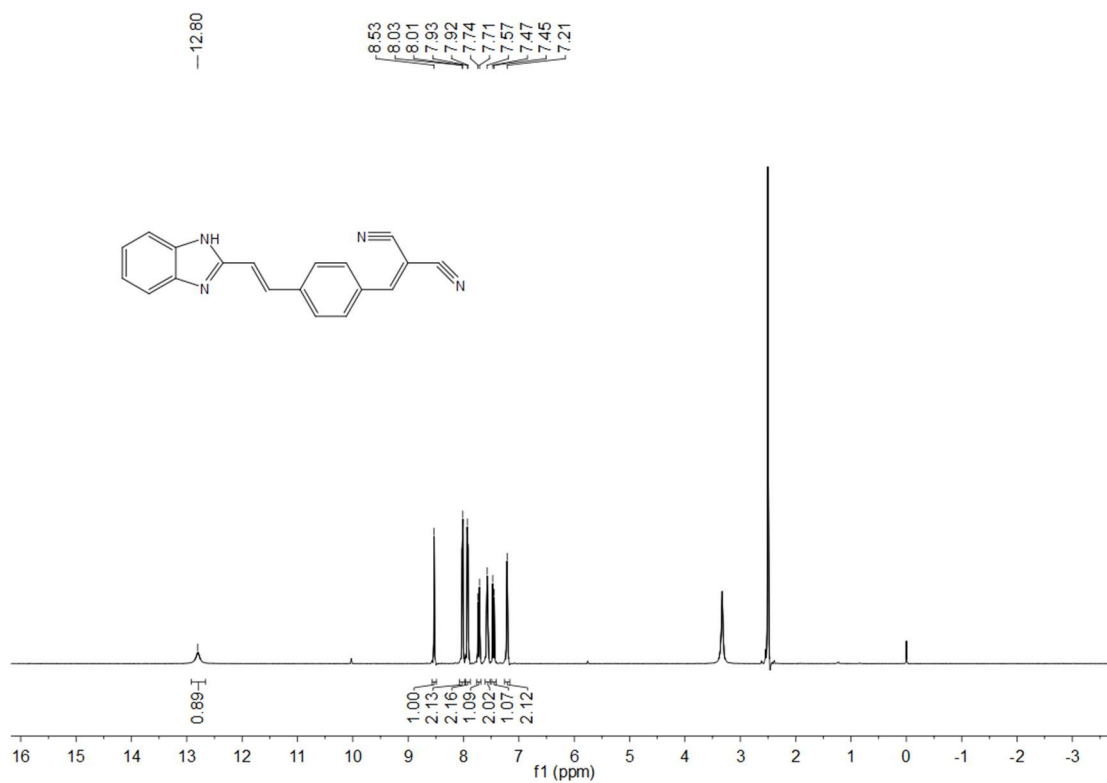


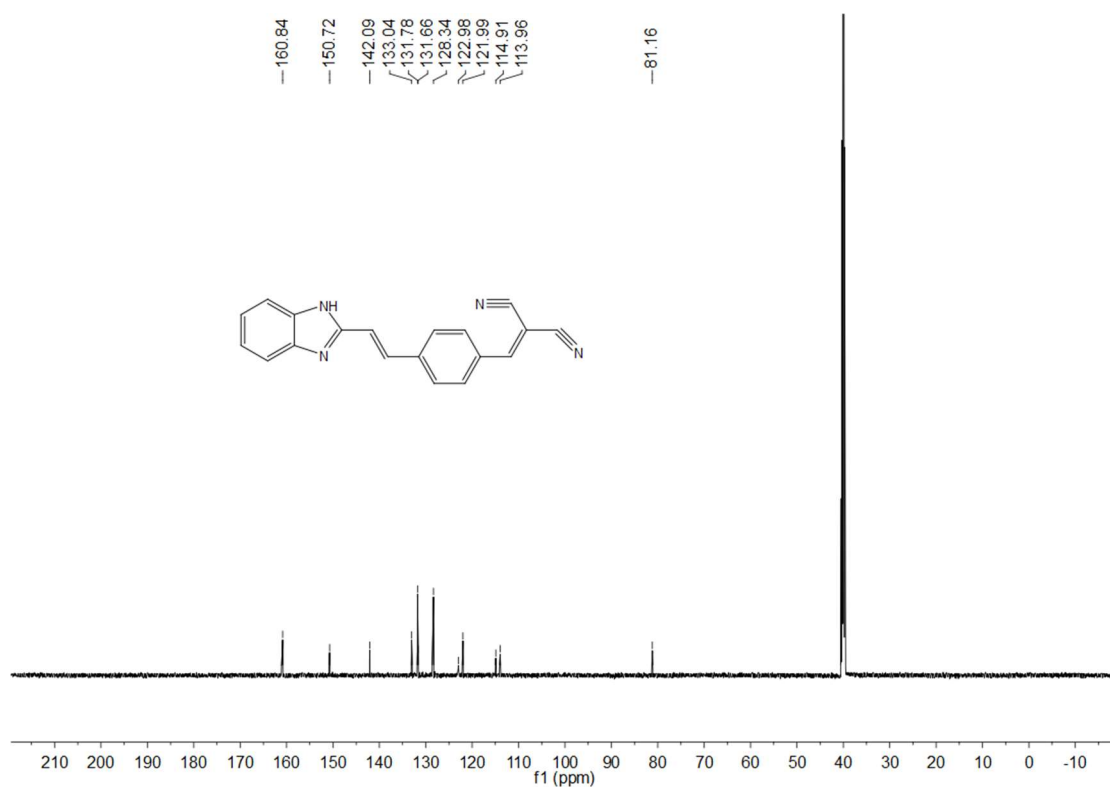
Fig. S10  $^{13}\text{C}$  NMR of compound MBI-CHO



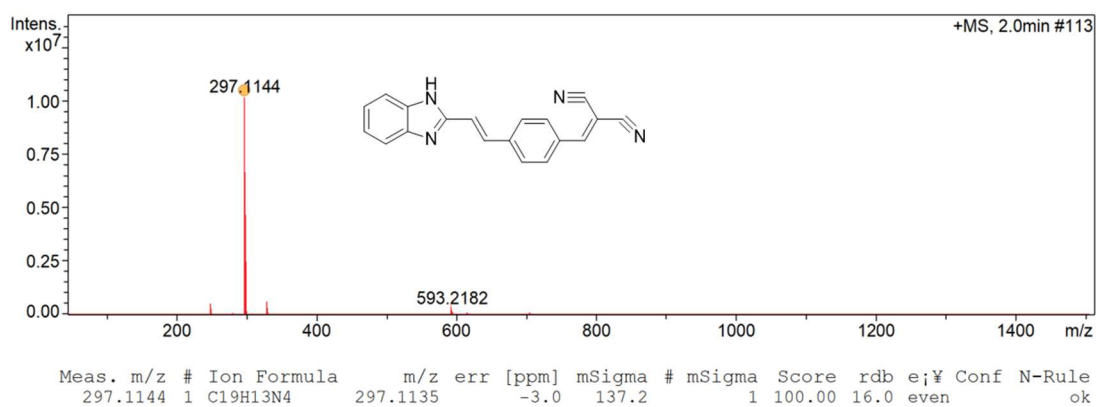
**Fig. S11** Mass spectroscopy of compound MBI-CHO



**Fig. S12** <sup>1</sup>H NMR of compound MBI-CN



**Fig. S13**  $^{13}\text{C}$  NMR of compound MBI-CN



**Fig. S14** Mass spectroscopy of compound MBI-CN