

Electronic Supplementary Material

Solid-conversion synthesis of three-dimensionally ordered mesoporous ZSM-5 catalysts for the methanol-to-propylene reaction

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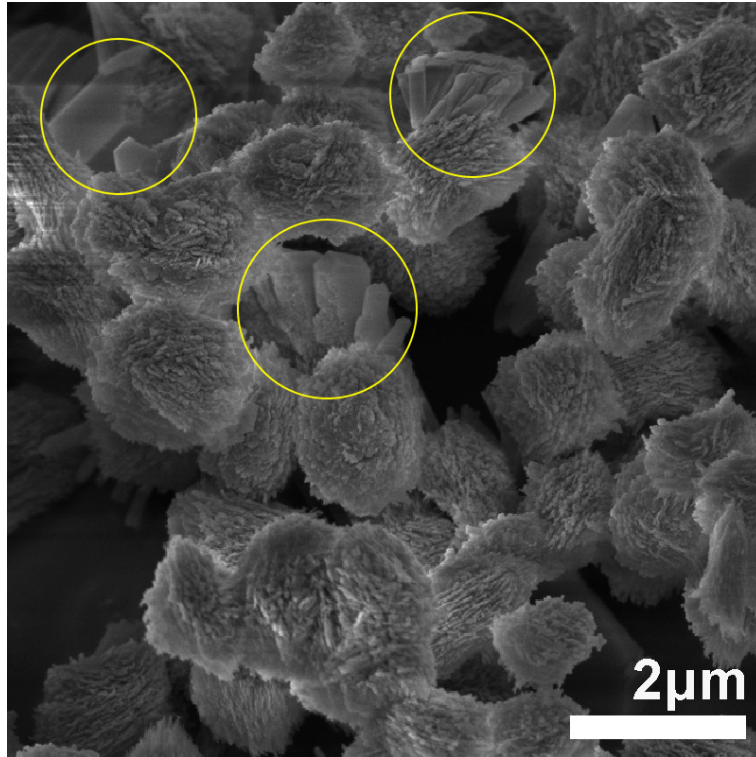


Figure S1. SEM image of 3DZ5_S/C.

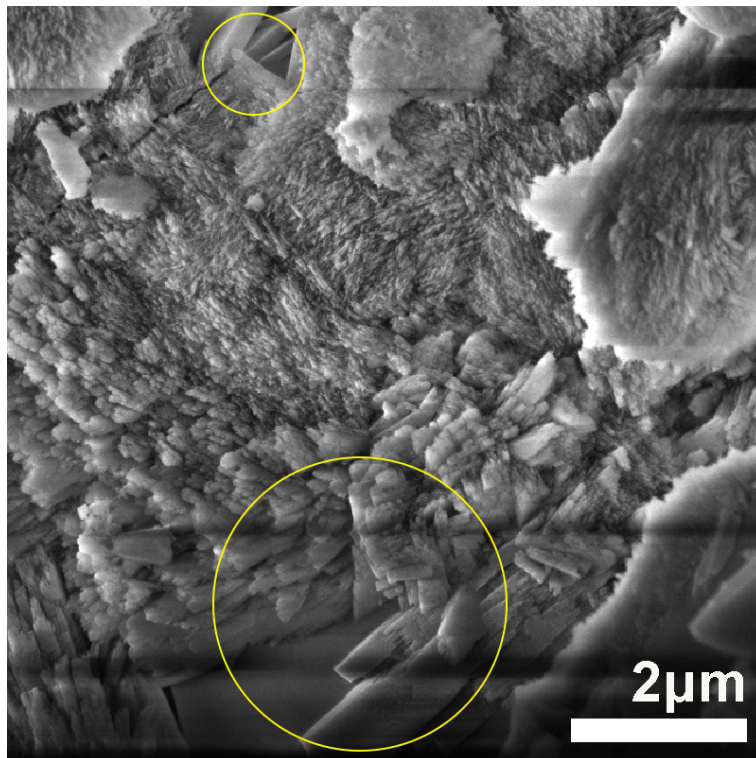


Figure S2. SEM image of 3DZ5_S.

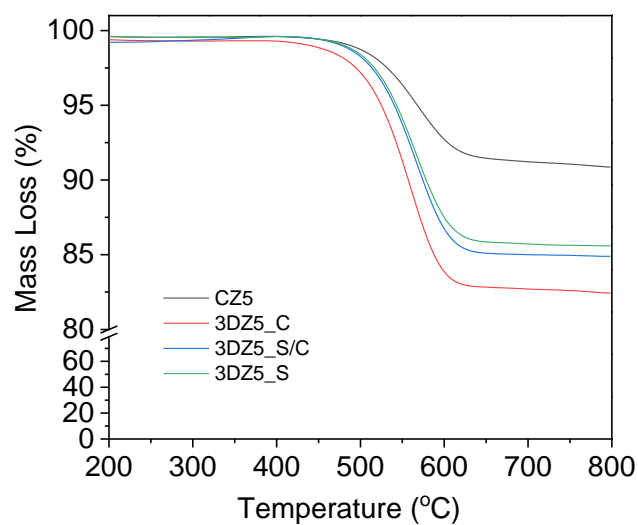


Figure S3. Thermogravimetric analysis curves of used catalysts after MTP reaction.

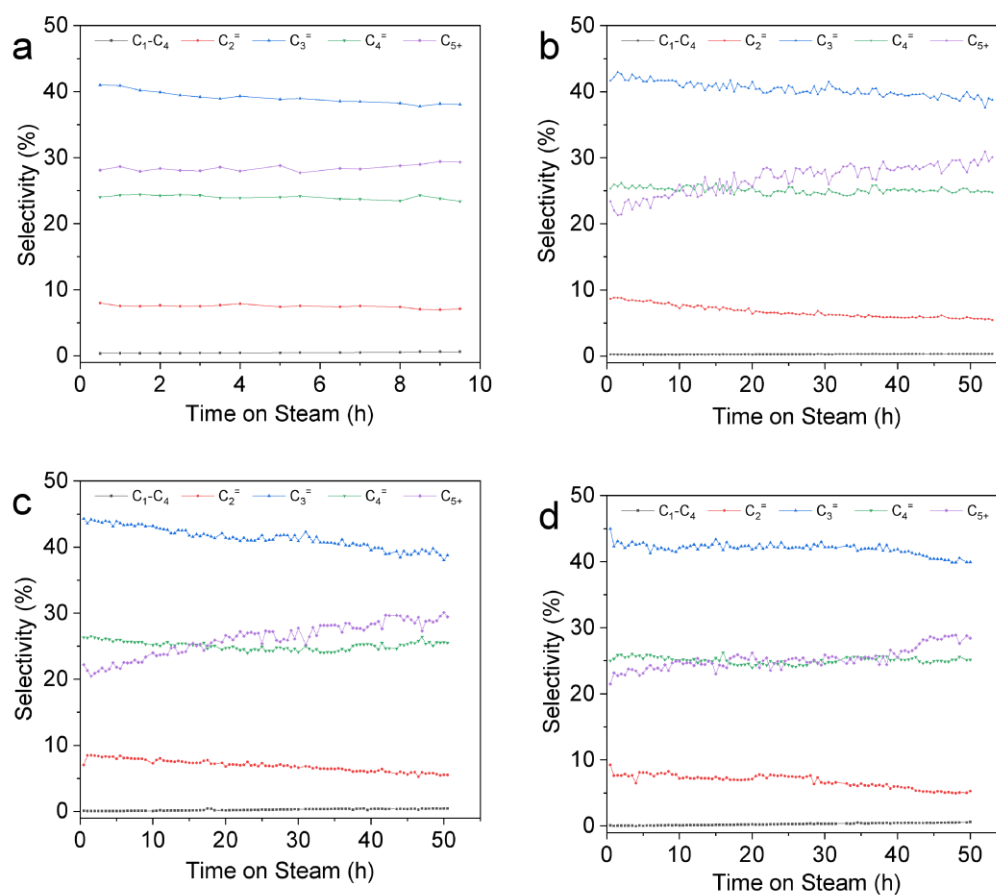


Figure S4. Product distributions versus time on stream over various synthesized 3DZ5 and the commercial CZ5 catalysts. Reaction conditions: $T = 723 \text{ K}$, $P = 1 \text{ atm}$, $\text{WHSV} = 1.7 \text{ h}^{-1}$.

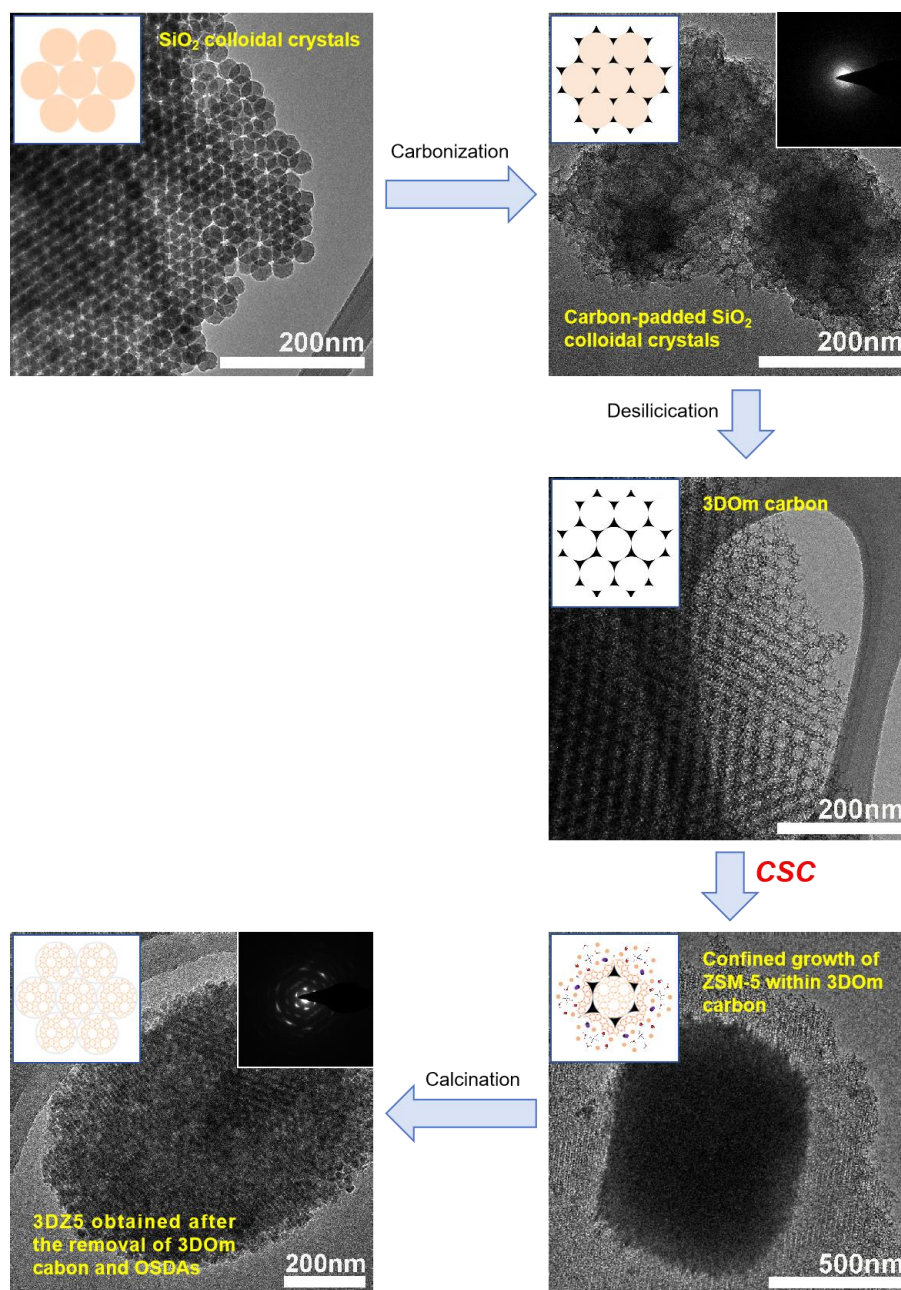


Figure S5. TEM observations of initial SiO₂ colloidal crystals, intermediate products, and final 3DZ5 zeolites by the CSC approach (Route I).

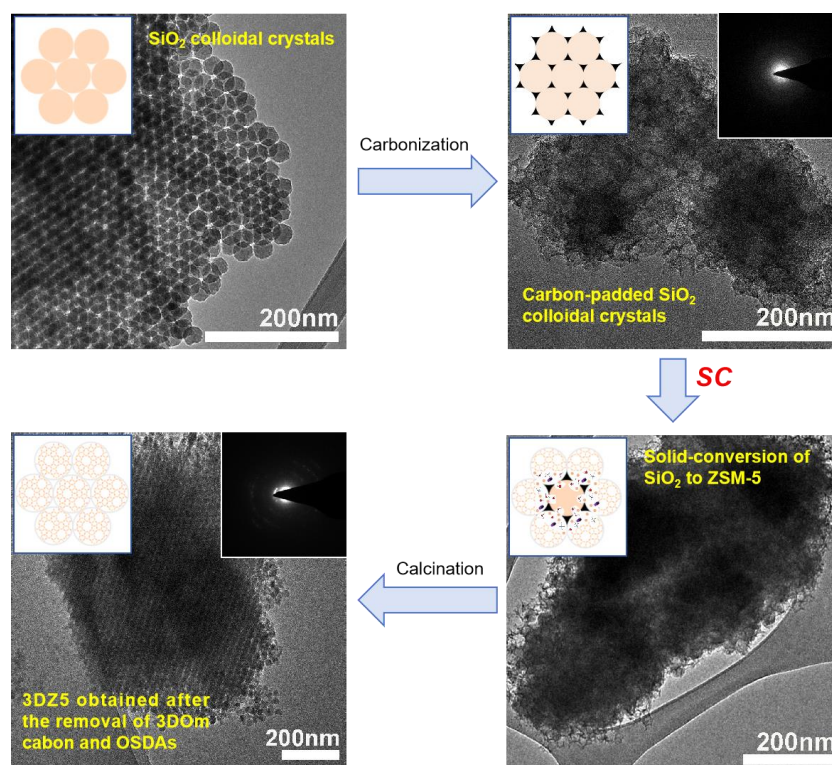


Figure S6. TEM observations of initial SiO_2 colloidal crystals, intermediate products, and final 3DZ5 zeolites by the modified solid-conversion synthesis of 3DZ5 from the carbon-padded SiO_2 colloidal crystals (Route II).

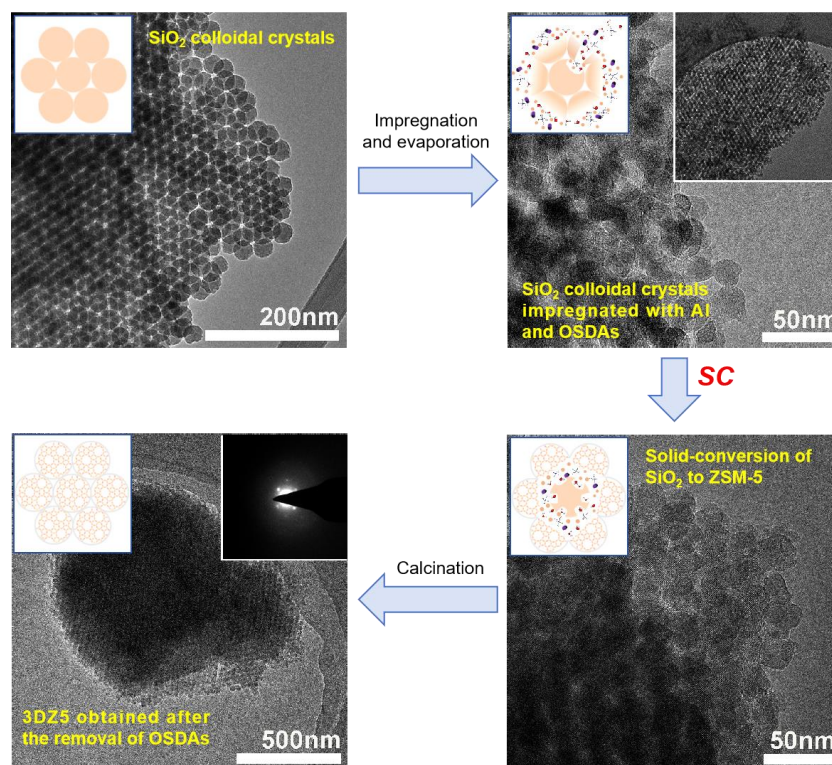


Figure S7. TEM observations of initial SiO_2 colloidal crystals and the the in-situ zeolitization process (Route III).

Table S1. Coke contents and coke formation rates (R_{coke}) of various synthesized 3DZ5 and the commercial CZ5 catalyst in

MTP reaction		
Sample	Coke Content [mg . gcat ⁻¹]	R_{coke} [mg . gcat ⁻¹ . h ⁻¹]
CZ5	81	2.02
3DZ5_C	164	1.49
3DZ5_S/C	145	1.61
3DZ5_S	138	1.73

Table S2. Product properties and synthesis efficiency of various synthesis approaches

Sample	Crystallinity ^{a)} [%]	Mesoporosity ^{b)} [%]	Acidity ^{c)} [%]	Time-efficiency ^{d)} [%]	Productivity ^{e)} [%]
3DZ5_C	72.3	34.1	56.3	35.0	82.2
3DZ5_S/C	83.2	32.6	56.2	56.0	90.5
3DZ5_S	87.3	29.7	58.0	82.4	88.8

a) Calculated according to the diffraction peaks based on the standard test method ASTM D5758-01 (2015) and the sample of CZ5 was used as a benchmark of $RC = 100\%$; b) Defined as the fraction of external surface area ($S_{\text{ext}}/S_{\text{BET}}$); c) Defined as the fraction of strong acid sites ($N_{\text{strong}}/N_{\text{total}}$); d) Defined as the deviation from the time of SiO₂ colloidal crystals preparation ($T_{\text{SiO}_2}/T_{\text{total}}$); e) Defined as the synthesis yield from raw material.