

# Electronic Supplementary Material

## CO<sub>2</sub> capture using membrane contactors: a systematic literature review

Sanaa Hafeez\*<sup>1</sup>, Tayeba Safdar\*<sup>1</sup>, Elena Pallari<sup>2</sup>, George Manos<sup>3</sup>, Elsa Aristodemou<sup>1</sup>, Zhien Zhang<sup>4</sup>,

S. M. Al-Salem<sup>5</sup>, Achilleas Constantinou (✉)<sup>1,6</sup>

1 Division of Chemical & Petroleum Engineering, School of Engineering, London South Bank University, London SE1 0AA, UK

2 Medical Research Council Clinical Trials Unit, University College London, London WC1V 6LJ, UK

3 Department of Chemical Engineering, University College London, London WC1E 7JE, UK

4 William G. Lowrie Department of Chemical and Biomolecular Engineering, The Ohio State University, Ohio 43210, USA

5 Environment & Life Sciences Research Centre, Kuwait Institute for Scientific Research, Safat 13109, Kuwait

6 Department of Chemical Engineering Cyprus University of Technology, 3036 Limassol, Cyprus

E-mail: [constaa8@lsbu.ac.uk](mailto:constaa8@lsbu.ac.uk)

\* These authors contributed equally to the work.

## Appendix A ‘Carbon Dioxide Main Characteristics Post Membrane Type Reactor Removal’

Table 1. The table depicts the studies included in the systematic review, with the extracted data on membrane material, reactor type, flow configuration, solvent, wetting, average calculated flux, gas flow rate, liquid flow rate and percentage of CO<sub>2</sub> in the feed stream.

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Niwa, Ohya [1]	porous ceramic	ceramic membranes	concurrent	3 M HCL	-	9E-06	-	-	50
Saha and Chakma [2]	polymeric membrane.	-	counter current flow	3 M DEA	-	3.14E-12	-	-	-
Langevin, Pinoche [3]	-	facilitated transport membrane	counter current flow	Monoethanolamine (MEA)	-	1.75E-16	0.03 - 0.33	0.5	41
Tokuda, Fujisawa [4]	cardo polyimide	hollow fibre membrane	co and counter current flow	N-Methyl-2-pyrrolidon (NMP), N,N-Dimethylacetamide (DMAc)	-	3.19E-06	-	-	-
Li and Teo [5]	silicone rubber or polyethersulphone	hollow fibre membrane	co and counter current flow	NaOH 2.5 M	-	1.32E-04	-	-	4
Suzuki, Tanaka [6]	Poly (ethylene oxide)-containing polyimide	composite hollow fibre membranes	concurrent flow	p-chlorophenol	-	1.97E-04	-	-	-
Chen, Kovvali [7]	glycerol	based immobilized liquid membranes (ilms)	parallel flow	glycine-Na-glycerol solution (2.5M)	-	9.66E-05	-	-	-
Lee, Noble [8]	potassium carbonate	hollow fibre membrane	parallel flow	-	-	-	-	9.6 - 100.2	-
Mizukami, Takaba [9]	-	zeolite membranes	-	-	-	-	-	-	-
Vu, Koros [10]	carbon molecular sieve (CMS) hollow fibre membranes	-	concurrent flow	-	-	1.52E-04	-	-	10
Xu, Zhang [11]	PVDF-PVA hydrogel	hydrogel membranes	concurrent flow	-	-	-	-	-	0.6 - 1.3
Zhang, Wang [12]	-	polymeric membranes	-	-	-	4.69E-06	-	-	35
Kapantaidakis, Koops [13]	polyimide and polyethersulfone blends	hollow fibre membrane	counter current flow	1-methyl-2-pyrrolidone (NMP)	-	7.32E-05	160	-	55

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Kim, Park [14]	poly (ether sulfone) (PES) membrane	-	-	2-ethoxyethanol	-	-	-	-	-
Jordal, Bredesen [15]	Pd membranes	-	counter current and cross flow	-	-	-	-	-	-
Kim and Lee [16]	-	hollow fibre membrane	-	-	-	-	-	-	38
Li, Falconer [17]	SAPO-34	-	-	H3PO4 (85 wt.% aqueous solution)	-	1.33E-04	60	-	50
Moon, Ahn [18]	tetrapropylammonium bromide (TPABr)-templating silica sols	silica/alumina composite membrane	-	TEOS: EtOH:H2O: HCl of 1.0 : 3.8 : 5.0 : 0.0053	-	6.00E-03	-	-	-
Teramoto, Kitada [19]	-	facillated transport membrane	concurrent flow	monoethanolami/ne (MEA), diethanolamine (DEA) and 2- amino-2- methyl-1- propanol	-	4.4E-02	100	25-150	-
Wang, Li [20]	-	hollow fibre membrane contactor	counter current flow	2-amino-2-methyl-1- propanol (AMP), diethanolamine (DEA) and methyldiethanoamine (MDEA)	-	2.12E-05	-	-	-
Johannessen and Jordal [21]	-	catalytic membrane reactor	counter-current flow	-	-	-	-	-	-
Jordal, Bolland [22]	Pd membrane	-	counter current flow	-	-	-	-	-	-
Li, Alvarado [23]	SAPO-34	-	parallel flow	tetraethylammonium hydroxide (TEAOH)	-	2.02E-04	-	-	50
Li, Martinek [24]	SAPO-34membranes	-	cross-flow	-	-	6.48E-04	150	-	-
Liu, Chakma [25]	poly(ether block amide) (PEBA)	hollow fibre membrane	counter current flow	1-butanol	-	4.20E-05	-	-	62
Qin, Chung [26]	6FDA-Durene/1,3-phenylenediamine (mPDA) copolyimide	hollow fibre membranes	-	6 FDA-Durene/mPDA (PI), N-methyl-pyrrolidone (NMP) and tetrahydrofuran (THF) with a weight ratio of 20:50:30	-	2.44E-04	-	0.1	-
Wang, Zhang [27]	polypropylene	hollow fibre membrane	concurrent flow	2 MDEA	-	3.60E-06	-	-	20
Gong, Wang [28]	-	hollow fibre membrane	counter current flow	methyldiethanolamine/m onoethanolamine (MDEA/MEA) aqueous solutions and distilled water	-	1.65E-05	157	20-50	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Ismail and Yaacob [29]	asymmetric polysulfone	hollow fibre membranes	concurrent flow	N,N-dimethylacetamide	-	2.38E-06	-	-	-
Okabe, Nakamura [30]	polyolefin	-	cross-flow	-	-	-	-	-	-
Zhang, Wang [31]	-	hollow fibre membrane	concurrent flow	diethanolamine (DEA) solution 2M	-	1.77E-06	-	-	20
Francisco, Chakma [32]	poly(vinyl alcohol) matrix	-	concurrent flow	monoethanolamine (MEA), 2-amino-2-methyl-1-propanol (AMP), diethanolamine (DEA) and N-methyldiethanolamine (MDEA)	-	1.1E-01	-	-	-
Himeno, Tomita [33]	deca-dodecasil 3R (DD3R)	hydrophobic ddr-type zeolite membrane	concurrent flow	-	-	4.32E-04	-	-	50
Kelman, Lin [34]	poly(ethylene oxide) [PEO] material	polymeric membranes	concurrent flow	1-hydroxycyclohexylphenyl ketone (HCPK)	-	5.50E-04	-	-	45
Kwon, Kim [35]	zeolite membrane : silica/aluminophosphate(SAPO-4)	-	-	-	-	3.60E-09	-	-	-
Ohta, Takaba [36]	zeolite-like porous membranes	-	counter current flow	-	-	-	-	-	-
Sakamoto, Nagata [37]	mesoporous silica membranes	-	concurrent flow	NaOH	-	4.95E-04	-	-	20
Shim, Lee [38]	-	hollow fibre membrane	-	-	-	-	-	-	-
Sridhar, Suryamurali [39]	poly(ether-block-amide) (PEBAX-1657)	-	continuous flow	2,4-toluylene diisocyanate (TDI)	-	3.70E-06	-	-	-
Wang, Zhu [40]	-	-	-	-	-	-	-	-	-
Xiao, Feng [41]	trimesoyl chloride	crosslinked chitosan membranes	-	sodium hydroxide aqueous solution (1 mol/L)	-	7.86E-05	-	-	-
Yan, Fang [42]	polypropylene (PP)	hollow fibre membrane	counter current flow	monoethanolamine (MEA) and methyldiethanolamine (MDEA)	-	2.66E-06	-	-	-
Yegani, Hirozawa [43]	2,3-diaminopropionic acid (DAPA)	-	counter current flow	-	-	5.5E-01	2.24E-02	-	3.65

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Al-Marzouqi, El-Naas [44]	polypropylene (PP)	hollow fibre membrane reactor	concurrent flow	monoethanol amine (MEA) and sodium hydroxide (NaOH) solutions	-	-	200-600	200-600	10
Al-Marzouqi, El-Naas [45]	PTFE	hollow fibre membrane (hfm) contactor	concurrent flow	Mathematical Modelling Exercise	-	3.00E-03	200-500	200-500	10
Kai, Kouketsu [46]	poly(amidoamine) (PAMAM) dendrimer composite membrane	-	concurrent flow	-	-	2.42E-05	-	-	5
Keshavarz, Fathikalajahi [47]	polypropylene and polytetrafluoroethylene (PTFE)	hollow fibre membrane	counter current flow	diethanolamine (DEA)	-	3.90E-06	-	-	20
Kosuri and Koros [48]	Torlon®, a polyamide-imide polymer	-	-	-	-	1.22E-05	-	-	90
Li, Falconer [49]	silicoaluminophosphate (SAPO) membranes	-	cross flow	NaOH	-	1.34E-03	-	-	50
Low, Xiao [50]	copoly(4,4'-diphenylene oxide/1,5-naphthalene-2,2'-bis(3,4-dicarboxylphenyl)hexafluoropropanediimide (6FDA-ODA/NDA) d	polymeric membranes	concurrent flow	dimethylformamide (DMF)	-	5.14E-06	-	-	-
Modigell, Schumacher [51]	-	polymeric membranes	counter current flow	-	-	-	-	-	-
Paul, Ghoshal [52]	-	flat sheet membrane	concurrent flow	monoethanolamine(MEA),diethanolamine(DEA),N-methyldiethanolamine (MDEA) and 2-amino-2-methyl-1-propanol (AMP)	-	8.40E-05	-	-	-
Yan, Fang [53]	polypropylene (PP)	polymeric hollow fibre membranes	concurrent flow	-	-	-	-	-	-
Zhang, Wang [54]	polypropylene (PP) and polyvinylidene fluoride (PVDF)	hollow fibre membrane	concurrent flow	diethanolamine (DEA)	-	1.50E-03	-	-	-
Zhao, Cao [55]	cross-linker, poly(propylene glycol) block poly(ethylene	mixed matrix membranes (mmms)	-	tetrahydrofuran (THF)	-	2.46E-05	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
	glycol) block poly(propylene glycol) diamine (PPG/PEG/PPGDA)								
Faiz and Al-Marzouqi [56]	polyvinylidene fluoride (PVDF)	hollow fibre membrane (hfm) contactors	counter current flow	MEA (0.005 - 2M)	-	1.41E-06	200-500	200-600	25 – 90
Ji, Cao [57]	-	hollow fibre composite membrane	concurrent flow	-	-	1.22E-05	-	0.7	-
Kai, Kazama [58]	-	cesium-incorporated carbon membranes	-	1-methyl-2-pyrrolidone (NMP)	-	1.34E-07	100-20	-	5
Kosuri and Koros [59]	poly(tetrafluoroethylene)	-	concurrent flow	-	-	1.72E-17	-	-	30
Li, Rui [60]	samarium-doped ceria (SDC)-Li/Na/K <sub>2</sub> CO <sub>3</sub> (43.5/31.5/25 mol%) and SDC-Li/Na <sub>2</sub> CO <sub>3</sub>	ionic-conducting ceramic/carbonate composite membrane	concurrent flow	-	-	-	-	-	-
Lu, Zheng [61]	-	amino acid membrane contractor	counter current flow	-	-	-	-	-	-
Nistor, Shishatskiy [62]	3-glycidoxypropyl trimethoxysilane (GPTMS)	-	concurrent flow	tetrahydrofuran (THF)	-	-	-	-	-
Piroonlerkgul, Laosiripojana [63]	-	-	concurrent flow	-	-	-	-	-	40
Rezvani, Huang [64]	-	-	-	-	-	-	-	-	-
Safari, Ghanizadeh [65]	glassy polymeric: 6FDA-2,6-DAT membrane	-	cross-flow	-	-	-	-	-	5-20
Sandru, Kim [66]	polyvinyl amine (PVAm) casted on two polysulfone (PSf)	fixed-site-carrier (fsc) membranes	-	-	-	1.13E-01	-	-	10
Simons, Nijmeijer [67]	porous PP	hollow fibre membrane	cross-flow	MEA	-	7.84E-07	10	160	-
Watanabe, Keskin [68]	Cu(hfipbb)(H <sub>2</sub> hfipbb) 0.5 (H <sub>2</sub> hfipbb = 4,40-	metal organic framework	-	-	-	-	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
	(hexafluoroisopropylidene) bis(benzoic acid)								
Xing and Ho [69]	crosslinked polyvinylalcohol (PVA)/polyethyleneglycol (PEG) blend membrane	-	-	-	-	1.67E-04	60-30	-	-
Yang, Wang [70]	-	-	concurrent flow	-	-	-	-	-	-
Yave, Car [71]	poly(trimethylene terephthalate)-block-poly(ethylene oxide (PTT-b-PEO) copolymers as CO <sub>2</sub> -philic membrane	-	-	5wt.% of trifluoroacetic acid	-	5.77E-07	-	-	-
Ahmad, Lau [72]	-	-	-	-	-	-	-	-	-
Cong and Yu [73]	aminosilane-modified poly(ethylene glycol) (PEG), PEG-block-poly(propylene glycol) (PPG)-block-PEG (PEPEG), and PPG-block-PEG-block-PPG (PPEPG) diacrylate polymer membrane	-	-	triethylamine, THF	-	1.95E-04	-	-	-
Constantinou and Gavriilidis [74]	Metal (nickel) microstructure mesh	Flat sheet	co-current flow	2M NaOH	+	1.20E-10	177-354	1.28-2.56	20
El-Naas, Al-Marzouqi [75]	polypropylene hollow fibre membrane	hollow fibre membrane contactor	parallel flow	MEA (0.01 M)	-	-	-	-	10
Franz and Scherer [76]	-	single membrane module	co and counter current flow	-	-	-	-	-	61-67
Hudiono, Carlisle [77]	SAPO-34	ionic liquids (poly(rtll), rtll and zeolite membranes	-	divinylbenzene	-	-	-	-	-
Kumar, Yuan [78]	S. platensis UTEX LB 2340	hollow fibre membrane	continuous flow			-	-	-	2 – 15
Li and Fan [79]	SAPO-34 membrane	-	-	tetraethylammonium hydroxide (TEAOH)		9.94E-04	-	-	50
Li, Carreon [80]	SAPO-34 membrane	-	-	dipropylamine		4.14E-04	-	-	50

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Li and Chung [81]	Pebax	□ mixed matrix membranes (mmms)	-	N-methyl-2-pyrrolidone (NMP)	-	2.22E-04	-	-	-
Liu, Hu [82]	ZIF-69 membranes	-	counter current flow	-	-	2.34E-05	100	-	50
Lu, Ji [83]	polypropylene (PP)	hollow fibre membrane	counter current flow	amino acid salt	-	1.38E-05	8.33E-04	30-120	-
Mansourizadeh and Ismail [84]	polyvinylidene fluoride (PVDF)	polymeric hollow fibre membranes	concurrent flow	1-methyl-2-pyrrolidone (NMP)	-	2.85E-06	-	200	-
Mansourizadeh, Ismail [85]	porous polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter current flow	NaOH (1 M)	-	1.65E-04	200	1.5	-
Mansourizadeh, Ismail [86]	porous polyvinylidene fluoride (PVDF)	polymeric hollow fibre membranes	counter current flow	1-methyl-2-pyrrolidone (NMP)	-	1.95E-06	-	320	20
Marzouk, Al-Marzouqi [87]	polytetrafluoroethylene (PTFE)	hollow fibre membrane	counter current flow	5 M aqueous solutions of MEA, DEA and TETA	-	2.10E-04	200	100	9.5
Park, Han [88]	poly(1-methylsilyl-1-propyne) (PTMSP)	-	-	N-methyl-2-pyrrolidone (NMP)	-	1.02E-04	-	-	-
Reijerkerk, Knoef [89]	poly(ethylene glycol) and poly(dimethyl siloxane)	-	continuous flow	ethanol	-	1.27E-04	40-25	40-25	50
Sandru, Haukebo [90]	poly(phenylene oxide) (PPO) and polysulfone (PSf)	hollow fibre membrane	counter current flow	-	-	7.33E-03	-	-	10
Scholes, Smith [91]	metallic, polymeric and organic membranes	-	-	-	-	2E-02	-	-	-
Simons, Nijmeijer [92]	amino Acid	membrane contractors	counter current flow	monoethanolamine=Mon oethanolamine (MEA) (0.5–3M)	-	5.04E-03	-	160	20
Tiscornia, Kumakiri [93]	microporous titanosilicate ETS-10 membranes	-	-	10 NaOH:4 KF:1 TiO <sub>2</sub> :10 SiO <sub>2</sub> : 675 H <sub>2</sub> O	-	3.06E-05	60	-	50
Yave, Car [94]	poly(ethylene oxide)-poly(butylene terephthalate) multiblock copolymer	polymeric membranes	-	-	-	1.3E-01	-	-	-
Yave, Szymczyk [95]	segmented poly(butylene terephthalate)-poly(ethylene oxide)	thin film composite membranes	-	-	-	3E-02	-	-	20

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
	(PBT-PEO) block copolymers,								
Yu, Wang [96]	polymerization of 3,3-Diamino-N-methylpropylamine (DNMDAm) and trimesoylchloride (TMC) on the polysulfone (PS)	thin film composite membranes	co and counter current flow	3,3-Diamino-N-methylpropylamine (DNMDAm)	-	3.82E-05	-	-	20
Zhang, Xiao [97]	4,40-(hexafluoroisopropylidene) diphthalic anhydride (6FDA) and 4,40-oxydianiline (ODA)	polymeric membranes	-	N-methyl-pyrrolidone	-	3.1E-02	-	1	-
An, Swenson [98]	zeolite membrane: natural clinoptilolite material	-	concurrent flow	-	-	2.70E-05	-	-	-
Banihashemi, Pakizeh [99]	Silicalite-1 and ZSM-5 zeolites membrane	-	-	n-hexane	-	4.77E-05	-	-	-
Boributh, Assabumrungrat [100]	-	hollow fibre membrane contactor	counter current flow	-	+	-	5.70E-03	3.42 - 9.45	100
Chen, Qiu [101]	polyimide, 6 FDA-DAM:DABA.	hollow fibre membrane	asymmetric	THF and ethanol	-	1.18E-04	-	-	50
Chew, Ahmad [102]	H-SAPO-34 zeolite membrane	-	-	tetraethylammonium hydroxide	-	2.23E-04	-	-	-
Couling, Prakash [103]	-	-	counter current flow	-	-	-	-	-	-
Erucar and Keskin [104]	IRMOF-1/Matrimid and CuBTC/Matrimid	metal-organic frameworks (mofs)	-	-	-	-	-	-	10 - 50
Eslami, Mousavi [105]	polytetrafluoroethylene (PTFE) hollow fibre membrane	-	counter current flow	potassium glycinate (PG) aqueous solution	-	5.83E-08	-	-	14
Favre and Pierre [106]	nylon-SiO2 membrane	biocatalytic hybrid membranes	counter current flow	aqueous carbonic anhydrase solution (1M)	-	1.27E-05	3140	-	10
Iarikov, Hacarlioglu [107]		ionic liquids (rtils) membranes	-	1-butyl-4-methylpyridinium tetrafluoroborate or methyltrioctylammonium	-	6.12E-06	-	-	20-90

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
				bis(trifluoromethylsulfonyl)imid					
Jin, Han [108]	hydroxylcontaining polyimide	hollow fibre membrane	co and counter current	-	-	-	-	-	-
Jindaratamee, Shimoyama [109]	polypropylene (PP)	liquid membranes of amine liquids	concurrent flow	amine/glycol mixture - Monoethanolamine (MEA)	-	4.98E-10	-	1000	-
Khaisri, deMontigny [110]	polytetrafluoroethylene (PTFE)	hollow fibre membrane	-	Monoethanolamine (MEA)	-	-	10-18.6	0.5 - 3.5	-
Khan, Li [111]	sulfonated aromatic poly(ether ketone) (S-PEEK)	polymeric membranes	-	3,3',5,5'-tetramethyl diphenyl-4,4'-diol	-	3.09E-05	1000	-	100
Khan, Li [112]	poly(ether etherketone) (S-PEEK) blend membranes	-	-	DMA	-	4.12E-05	-	-	-
Kim, Baek [113]	1-ethyl-3-methylimidazolium-bis(trifluoromethylsulfonyl) imide ([emim][Tf2N]) and polyvinylidene fluoride (PVDF)	polymeric hollow fibre membranes	-	40 vol% methanol	-	1.86E-04	-	-	-
Kim, Uddin [114]	polyvinylamine (PVAm)	polymer composite membranes	-	-	-	3.33E-09	-	-	10
Kumbharkar, Liu [115]	polybenzimidazole (PBI) based asymmetric	hollow fibre membrane	concurrent flow	(DMAc, >99.5%)	-	6.92E-08	0.5	-	-
Kwisnek, Heinz [116]	UV-cured PEG-diacrylate	polymer membranes	-	2,2-dimethoxy-2-phenylacetophenone (DMPA)	-	2.78E-05	-	-	-
Lee, Kim [117]	Pd-Cu membrane	hybrid membranes	concurrent flow	-	-	-	-	-	5-15
Li, Pramoda [118]	1-vinyl-3-butylimidazolium bis(trifluoromethylsulfonyl)imide ([vbim][Tf2N])	ionic liquid (rtil)	-	-	-	1.07E-03	-	-	50
Li, Remias [119]	(NF/RO) membrane-ammonia	zeolite membranes	concurrent flow	-	-	-	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Lotrič, Sekavčnik [120]	-	membrane reactor	counter current flow	-	-	-	-	-	5.2
Mansourizadeh and Ismail [121]	porous polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter current flow	1-methyl-2-pyrrolidone (NMP)	-	8.70E-11	1-5	60	-
Mansourizadeh and Ismail [122]	polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter current flow	1-methyl-2-pyrrolidone (NMP)	-	4.68E-06	-	-	-
Martin, Dijkstra [123]	-	membrane reformer	counter current flow	-	-	-	-	-	1.8
Nguyen, Lasseuguette [124]	thin skin (Teflon AF®, PTMSP) coated on a porous support (PP)	hollow fibre membrane	concurrent flow	monoethanolamine (MEA)	-	2.85E-04	-	-	30
Ostwal, Singh [125]	3-aminopropyltriethoxysilane	inorganic membranes	-	-	-	1.78E-08	-	-	-
Peters, Hussain [126]	PVAm/PVA (polyvinyl amine and polyvinyl alcohol)	-	counter current flow	diethanolamine (DEA)	-	2.3E-05	1180*	-	-
Reijerkerk, Jordana [127]	hydrophilic, poly(ethylene oxide) membranes	-	-	-	-	2.30E-05	-	-	14
Reijerkerk, Wessling [128]	poly(ethylene glycol)	-	-	ethanol	-	5.40E-04	-	-	-
Sanaeepur, Amooghin [129]	polymeric membranes	-	-	-	-	1.15E-06	-	-	-
Sandström, Sjöberg [130]	MFI film on a graded alumina support	zeolite membrane	continuous flow	ethanol	-	9E-02	-	-	-
Sohrabi, Marjani [131]	-	hollow-fibre membrane contactor	parallel flow	MEA, DEA, MDEA, AMP and K <sub>2</sub> CO <sub>3</sub> (10wt%)	-	4.10E-08	-	-	10
Spadaccini, Mukerjee [132]	PMDA-ODA	-	-	-	-	-	-	-	-
Tuinier, Hamers [133]	-	-	-	-	-	-	-	-	-
Venna and Carreon [134]	SAPO-34 seeds and membranes	-	cross-flow	-	-	2.94E-04	-	100	-
Wade, Lee [135]	carbonate electrolyte membrane	-	counter current flow	organic solvents	+	7.92E-06	100	100	-
Xia, Liu [136]	poly(ethylene glycol) (PEG)	-	-	water/ethanol (30/70 wt %)	-	2.16E-03	-	-	-
Xing and Ho [137]	polyvinylalcohol-polysiloxane/fumed silica	mixed matrix membranes (mmms)	-	-	-	3.95E-03	30	163	20



Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Li, Paul [152]	1-vinyl-3-ethylimidazolium dicyanamide ([veim][dca]), 1-vinyl-3-butylimidazolium dicyanamide ([vbim][dca]) and 1-vinyl-3-heptylimidazolium dicyanamide ([vhim][dca])	ionic liquids (rtils) membranes	concurrent flow	-	-	1.25E-04	-	-	50
Lively, Dose [153]	6FDA-based polyimide — 6FDA-DAM:DABA(4:1)	hollow fibre membrane	counter current flow	3,5-diaminobenzoic acid (DABA)	-	8.29E-05	740	-	20
Madhusoodana, Patil [154]	-	ceramic membranes	-	-	-	-	-	-	15.4
Marzouk, Al-Marzouqi [155]	expanded poly(tetrafluoroethylene) (ePTFE) and poly(tetrafluoroethylene-co-perfluorinated alkylvinyl ether) (PFA)	hollow fibre membrane	concurrent flow	2.0 M NaOH	-	1.08E-05	100-600	10-25	5
Merkel, Wei [156]	-	-	counter current flow	-	-	4.91E-04	-	-	40-20
Modarresi, Soltanieh [157]	polysulfone gas-separation membranes	-	-	N-methyl-2-pyrrolidone	-	5.53E-09	-	-	-
Naim, Ismail [158]	microporous polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter current flow	dimethylacetamide (DMAc)	-	9.66E-05	-	60	-
Naim, Ismail [159]	microporous polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter current flow	N-methyl-2-pyrrolidone (NMP)	-	2.42E-04	-	110	-
Rahbari-Sisakht, Ismail [160]	polysulfone (PSf)	hollow fibre membrane	concurrent flow	1-methyl-2-pyrrolidone	-	1.05E-06	200	50-200	-
Rahbari-Sisakht, Ismail [161]	modified polyvinylidene fluoride (PVDF)	hollow fibre membrane	cross-flow	1-methyl-2-pyrrolidone (NMP)	-	4.62E-06	100	300	-
Rahbari-Sisakht, Ismail [162]	modified Polysulfone (PSf)	polymeric hollow fibre	concurrent flow	1-methyl-2-pyrrolidone (NMP)	-	3.48E-06	100	50-300	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
		membranes							
Rongwong, Boributh [163]	-	capillary membrane	counter current flow	monoethanolamine (MEA) 0.5 M		3.60E-06	500	-	20-40
Scholes, Simioni [164]	-	-	concurrent flow	30 wt.% monoethanolamine (MEA)	-	-	-	-	-
Shirazian, Marjani [165]	-	hollow-fibre membrane contactor	parallel flow	amines aqueous solutions (MEA & MDEA)		4.01E-08	50	50	50 – 100
Smart, Vente [166]	255 EtOH:4 TEOS:1 Co(NO <sub>3</sub> ) <sub>2</sub> .6H <sub>2</sub> O:9H <sub>2</sub> O <sub>2</sub> : 40H <sub>2</sub> O	-	-	-		1.65E-04	-	1.7-122	50
Uddin and Hägg [167]	PVAm/PVA blend composite membrane	-	concurrent flow	-	-	2E-02	-	-	2 - 10
Yilmaz and Keskin [168]	ZIF-8 + poly(1,4-phenylene ether-ether-sulfone) (PPEES)	mixed matrix membranes	-	-	-	-	-	-	-
Zhang, Hu [169]	polybenzimidazole (PBI) and zeolitic imidazolate-framework-7 (ZIF-7)	mixed matrix membranes (mmms)	-	-	-	-	-	-	-
Zhao and Ho [170]	poly-N-isopropylallylamine	fixed site carrier	counter-current flow	monoethanolamine (MEA) and other amine solvents	-	2.65E-03	60-30	0.03	-
Bae and Long [171]	nanocrystals of M <sub>2</sub> (dobdc) (M = Mg, Ni, Zn; dobdc <sup>4-</sup> = 1,4-dioxido-2,5-benzenedicarboxylate)	-	-	-		1.02E-03	-	-	-
Cao, Tao [172]	CAU-1-NH <sub>2</sub> and poly(methyl methacrylate) polymer membrane	metal organic structures	counter current flow	chloroform solvent		-	-	-	-
Choi, Park [173]	Pd-Cu membrane	-	-	-		-	-	-	-
Duan, Taniguchi [174]	hybrid membrane of poly(amidoamine) (PAMAM) dendrimer/cross-linked poly(vinyl alcohol) (PVA) membrane	flat sheet membrane	-	-	-	3.38E-11	10	-	80

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Ghadiri, Marjani [175]	-	polymeric membrane contactor	-	monoethanolamine (MEA) aqueous solution	-	3.00E-06	-	-	-
Ghasem, Al-Marzouqi [176]	polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter current flow	-	-	3.06E-06	-	-	10
Hao, Li [177]	zeolite imidazolate framework-8 (ZIF-8) nanoparticles	-	-	N,N0-methylenebis(acrylamide)	-	9.55E-14	-	-	50
Hassanlouei, Pelalak [178]	-	hollow-fibre membrane	parallel flow	MEA, MDEA and potassium glycinate	-	9.30E-03	200-900	200-900	10
Hu, Dong [179]	poly(2,6-dimethyl-1,4-phenylene oxide) (PPE) membrane	-	-	-	-	2.36E-05	-	-	22
Kai, Taniguchi [180]	poly(amidoamine) Dendrimer/polyme	flat sheet membrane	concurrent flow	-	-	4.53E-10	100	-	80
Kanehashi, Kishida [181]	1-butyl-3-methylimidazolium bis(trifluoromethylsulfonyl)imide ([BMIM][Tf2N])	polymeric and ionic liquid composite membranes	-	1-butyl-3-methylimidazolium bis(trifluoromethylsulfonyl)imide ([BMIM][Tf2N])	-	-	60	60	-
Kangas, Sandström [182]	-	zeolite membranes	-	-	-	1E-02	-	-	-
Khan, Klaysom [183]	polysulfone acrylate	mixed matrix membranes (mmms)	concurrent flow	Amino functional groups	-	1.84E-05	-	-	-
Kim, Abouelnasr [184]	-	iza zeolite structures	-	-	-	1.01E-03	-	-	50
Kim, Ingole [185]	PEBAX/PEI	composite hollow fibre membranes	counter current flow	n-butanol	-	8.41E-06	-	-	-
Kim, Vrålstad [186]	polyvinylamine (PVAm) Fixed-Site-Carrier (FSC) composite membrane	-	-	0.1 M KOH	-	1.5E-01	-	-	4
Koutsonikolas, Kaldis [187]	polymeric, ceramic and metallic	-	-	-	-	-	-	-	-
Lee, Lee [188]	Pd-Au composite	Pd-Au composite membrane	-	aluminium nitrate	-	-	-	-	40
Li, Wang [189]	diethylene glycol bis(3-aminopropyl) ether (DGBAmE) and diaminopolyethylene glycol (DAmPEG)	thin film composite (tfc) membranes	-	PDMS solution	-	2.64E-04	-	-	15

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Lin, Chen [190]	mesoporous Fluorocarbon-Modified Silica Aerogel	SiO <sub>2</sub> eorgel membrane	-	-	-	-	-	-	-
Sainan, Gongping [191]	polydimethylsiloxane (PDMS) and poly (ethylene glycol) diacrylate (PEGDA)	polymeric composite membranes	-	-	-	2.75E-06	-	-	25-90
Mehdipour, Karami [192]	-	hollow fibre membrane	concurrent flow	-	-	1.13E-06	75	-	-
Moghadassi, Rajabi [193]	carbon nano tubes (MWCNTs) as nanofillers and poly ethylene glycol (PEG) as second polymer was prepared by solution casting method. Both raw-MWCNTs (R-MWCNTs) and functionalized carboxyle-MWCNTs (C-MWCNTs)	mixed matrix membranes	concurrent flow	-	-	1.80E-05	-	-	-
Mondal and Mandal [194]	crosslinked poly(vinyl alcohol)/poly (allylamine)/2-amino-2-hydroxymethyl-1,3-propanediol/polysulfone	polymeric membranes	counter current flow	n,n-dimethylacetamide (DMAc)	-	2.27E-04	28	30	20
Nagumo, Iwata [195]	-	membrane separation technology	concurrent flow	-	-	-	-	-	-
Naim and Ismail [196]	microporous polyvinylidene fluoride (PVDF)	hollow fibre membrane contractor	counter current flow	dimethylacetamide (DMAc)	-	1.35E-06	-	2	-
Nasir, Mukhtar [197]	PES-DEA and PES-MDEA	-	-	monoethanolamine (MEA), diethanolamine (DEA), and methyl-diethanolamine (MDEA)	-	6.80E-05	-	-	-
Peydayesh, Asarehpour [198]	SAPO-34 zeolite loaded Matrimid@5218	mixed matrix membranes (mmms)	concurrent flow	1-Methyl-2-pyrrolidone	-	3.71E-05	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Qiao, Wang [199]	polyvinylamine (PVAm)	composite membrane	concurrent flow	-	-	7E-02	-	-	-
Rahbari-Sisakht, Ismail [200]	modified polyvinylidene fluoride (PVDF)	hollow fibre membrane	-	1-Methyl-2-pyrrolidone (NMP)	-	3.24E-05	-	300	-
Rahman, Filiz [201]	PEBAX	-	-	ethanol/water (70/30 wt%)	-	2.41E-05	-	-	-
Razavi, Razavi [202]	-	nanoporous membrane contactor	counter current flow	2-amino-2-methyl-1-propanol and piper-azine	-	-	200	100-400	-
Ryi, Lee [203]	Pd-based composite	pd-based composite membrane	concurrent flow	-	-	-	-	-	40
Sandru, Kim [204]	polymeric polyvinylamine	polymeric membranes	concurrent flow	-	-	-	-	-	-
Shao, Dal-Cin [205]	-	-	-	-	-	-	-	-	5-10
Shen, Yu [206]	carboxymethyl chitosan and polyethylenimine	facilitated transport membranes	concurrent flow	-	-	1E-02	-	-	-
Stanislawski, Holmes [207]	-	-	-	-	-	-	-	-	-
Tomé, Patinha [208]	-	ionic liquid membranes (silms)	-	1-ethyl-3-methylimidazolium bis (trifluoromethylsulfonyl) imide ([C2mim][NTf2])	-	8.55E-07	-	-	-
Tseng, Itta [209]	SBA-15/CMS	polyphenylene oxide (ppo)-derived composite cms membranes	-	-	-	3.53E-04	50	-	-
Tziaila, Veziri [210]	zeolitic imidazolate framework ZIF-69 membranes	-	-	N,N-dimethylformamide	-	5.58E-08	-	-	44
Wang, Zhang [211]	polypropylene (PP) and polyvinylidene fluoride (PVDF)	hollow fibre membrane contactor	cross-flow	MEA solution (1 mol L <sup>-1</sup> )	-	3E-02	2667	145	15
Wang, Wang [212]	3,5-diaminobenzoate on a cross-linked polydimethylsiloxane coating polysulfone membrane	-	concurrent flow	diethylene glycol bis(3-aminopropyl) ether	-	1.29E-03	-	-	15

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Wang, Fang [213]	-	hollow fibre membrane	cross-flow	monoethanolamine (MEA) 20 - 30wt%	-	-	-	-	50
Wang, Fang [214]	hydrophobic microporous polypropylene (PP)	membrane vacuum regeneration (mvr)	concurrent flow	triethylenetetramine (TETA) and N-methyldiethanolamine (MDEA)	-	-	2000	20	-
Wang, Fang [215]	polypropylene (PP)	hollow-fibre membrane	cross-flow	monoethanolamine (MEA) (3.2M)	-	3.69E-12	6667	100	20
Yin, Chu [216]	porous alumina	thin zeolite t/carbon composite membranes	-	TMAOH (Tetramethylammonium hydroxide)	-	3.60E-05	100	-	-
Yoshimune and Haraya [217]	sulfonated poly(phenylene oxide) (SPPO)	hollow fibre membrane	concurrent flow	-	-	8.40E-06	-	-	50
Zhang, Zou [218]	yttria stabilized bismuth oxide (YSB) membranes	-	concurrent flow	-	-	7E-01	-	100	-
Zhu, Chai [219]	triazine-framework-based membranes	-	-	NaOH	-	1.04E-07	-	-	-
Ahmadpour, Shamsabadi [220]	poly (amide e 6 e b e ethylene oxide) (Pebax MH 1657) copolymer	-	concurrent flow	DMF	-	-	-	-	10-50
Amrei, Memardoost [221]	PTFE	hollow fibre membrane reactor	counter current flow	Monoethanolamine (MEA) aqueous solution 0.5 - 1M	10%	2.37E-05	1000	25	15
Chen, Thong [222]	Pebax/poly(dimethylsiloxane)/polyacrylonitrile (Pebax/PDMS/PAN)	composite hollow fibre membrane	counter current flow	ionic liquid (RTIL), 1-ethyl-3-methylimidazolium tetrafluoroborate [emim][BF <sub>4</sub> ]	-	1.94E-04	3	1.5	50
Constantinou, Barrass [223]	PTFE	flat membrane	concurrent	2M NaOH	-	8.00E-03	354	2.56	20
Deng and Hägg [224]	carbon nanotube (CNT) reinforced polyvinyl amine/polyvinyl alcohol (PVAm/PVA) blend	nanocomposite membrane	-	hydrophilic PVAm/PVA blended solution	-	3E-02	-	-	-
Ghasem, Al-Marsouqi [225]	polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter current flow	sodium hydroxide	-	5.40E-06	-	-	-
He, Kim [226]	polyvinylamine (PVAm)/polyvinylalcohol (PVA)	hybrid fixed-site-carrier (fsc) membranes	-	5 wt% PVAm aqueous solution	-	1.3E-01	50	-	10-50

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Huang, Chen [227]	Zn(BDC) (TED)0.5 (BDC¼benzenedicarboxylate, TED ¼ triethylenediamine)	metal-organic frameworks (mofs)	-	N,N-dimethylformamide	-	1.59E-03	-	-	-
Hussain, Nasir [228]	polyimides, polyetherurethaneurea, polyphenylene oxide, cellulose acetate and polyamide ether	-	concurrent and cross flow	-	-	-	-	-	9.5-2.9
Junaidi, Khoo [229]	SAPO-34	zeolite membranes	-	ethanol or isopropanol	-	3.96E-04	-	-	-
Jusoh, Lau [230]	-	-	concurrent flow	-	-	-	-	-	25
Kimball, Al-Azki [231]	PTFE fibre	hollow fibre membrane	counter current flow	30% Monoethanolamine (MEA)	-	-	-	-	12-15
Kosinov, Auffret [232]	High-silica (gel Si/Al ¼ 100) SSZ-13	zeolite membranes	-	-	-	1.08E-03	200	200	-
Kundu, Chakma [233]	-	hybrid membranes	cross-flow	Monoethanolamine (MEA)	-	-	-	-	70-100
Li, Wang [234]	-	fixed carrier membranes	-	3,3'diamino-N-methyldipropylamine (DNMDAm)	-	4.20E-04	-	-	-
Lin, He [235]	polyester	thin film composite membranes	cross-flow	-	-	4.20E-04	-	-	1-40
Lin, Ko [236]	Sol-gel preparation of polymethylsilsesquioxane aerogel	-	cross-flow	ethanol	-	8E-02	-	-	-
Lindqvist, Roussanaly [237]	-	polymeric membranes	-	monoethanolamine (MEA)	-	-	-	-	-
Lu, Lu [238]	-	ionic liquid membranes (silms)	co and counter current flow	1-butyl-3-methylimidazolium tetrafluoroborate ([bmim][BF4]) and 1-(3-aminopropyl)-3-methylimidazolium tetrafluoroborate ([apmim][BF4])	-	2.11E-06	200	100	-
Ma and Koros [239]	ester-crosslinked	hollow fibre membranes	counter current flow	DABA	-	2.77E-04	-	-	50
Ma, Qiao [240]	methylcarbamate (MC)	polymeric membrane	-	-	-	1.73E-03	-	-	10-40
Maghsoudi and	α-alumina	zeolite	concurrent flow	N, N, N-trimethyl-1-	-	3.75E-05	-	-	2.13

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Soltanieh [241]		membranes		adamantylammonium hydroxide					
Makhloufi, Lasseguette [242]	Teflon (AF2400) and commercial (TPX)	hollow fibre membrane	counter current flow	monoethanolamine (MEA)	-	3.23E-09	-	-	-
Mansourizadeh, Aslmahdavi [243]	hydrophobic polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter current flow	dimethylacetamide (DMAc)	-	1.62E-08	100	40-300	-
Mansourizadeh and Pouranfard [244]	microporous polyvinylidene fluorid	polymeric hollow fibre membranes	counter current flow	1-methyl-2-pyrrolidone (NMP)	-	2.70E-07	-	110	-
Masoumi, Keshavarz [245]	4-diethylamino-2-butanol (DEAB) in a amino alcohol	hollow fibre membrane contactor	cross-flow	monoethanolamine (MEA), diethanolamine (DEA), methyl-diethanolamine (MDEA) and DEAB solutions	-	5.40E-05	-	-	20
Mohshim, Mukhtar [246]	polyethersulfone-SAPO-34	mixed matrix membranes (mmms) - flat sheet membrane	-	N-methyl-pyrroline (NMP)	-	1.31E-03	-	-	-
Mondal and Mandal [247]	poly(vinyl alcohol) (PVA)/polyvinylpyrrolidone (PVP)	-	-	KOH	-	7.96E-04	30	0.02 - 0.075	20
[248]	poly (vinyl alcohol) (PVA)/polyvinylpyrrolidone (PVP) blend membrane	-	-	tetraethylenepentamine	-	3.79E-04	-	0.02-0.075	-
Nabian, Ghoreyshi [249]	polysulfone (PSF)	flat sheet membrane	counter current flow	dimethylformamide(DMF)	-	4.20E-07	3-12	9.7	-
Nafisi and Hägg [250]	PEBAX-2533	mixed matrix membrane	-	ethanol	-	1.29E-03	-	-	10
Patel, Kim [251]	amphiphilic PCZ-r-PEG	-	-	THF	-	2.07E-05	-	-	-
Pedram, Omidkhah [252]	diethanolamine-impregnated cross-linked polyvinylalcohol/glutaraldehyde	-	counter current flow	Diethanolamine (DEA)	-	1.01E-02	-	-	15
Rabiee, Soltanieh [253]	poly(ether-b-amide6) (Pebax1657)/ glycerol triacetate (GTA) gel membrane	-	cross-flow	glycerol triacetate	-	2.11E-03	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Rahbari-Sisakht, Rana [254]	polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter current flow	-	-	1.26E-05	50-200	200	-
Rahim, Ghasem [255]	polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter current flow	monoethanolamine (MEA), diethanolamine (DEA), and 2-amino-2-methyl-1-propanol (AMP)	-	4.20E-04	600	10-50	-
Rezaei, Ismail [256]	PVDF/montmorillonite	hollow fibre membrane	-	0.02M sodium hydroxide (NaOH)	-	1.13E-05	150	25-200	-
Rodenas, van Dalen [257]	NH 2 -MIL-53(Al)	mixed matrix membranes (mmms)	-	-	-	6.93E-06	-	-	50
Rodenas, van Dalen [258]	polysulfone (PSF) and polyimide (PI) polymers	mixed matrix membranes (mmms)	-	N,N-dimethylformamide	-	1.31E-06	-	-	50
Roh, Kim [259]	mesoporous TiO <sub>2</sub> hollow nanospheres (f-MTHS)	matrix membranes (mmms)	-	-	-	1.83E-05	-	-	-
Ryi, Lee [260]	nickel membrane	-	-	-	-	-	-	-	0-20
Salih, Yi [261]	Polyether	thin film composite (tfc) membrane	concurrent flow	polyetheramine (PEA) and Trimesoyl Chloride (TMC)	-	1.91E-04	400-450	-	20 – 10
Scholes, Ho [262]	poly(benzoxazole) (PBO) membrane	-	cross-flow	3,30-dihydroxy-4,40-diamino-biphenyl (HAB)	-	5.13E-04	-	-	-
Scholes, Ribeiro [263]	polymeric membranes	-	-	-	-	-	-	-	12
Shi [264]	SAPO-34 zeolite membranes	-	-	-	-	1.93E-03	-	-	50
Taniguchi and Fujikawa [265]	poly(amidoamine) (PAMAM)	-	counter current flow	N,N-dimethylethylenediamine	-	2.27E-10	-	-	-
Thompson, Vaughn [266]	zeolitic imidazolate framework (ZIF)	mixed matrix membranes (mmms)	counter current flow	methanol	-	1.04E-05	6	-	-
Tseng, Chang [267]	homogeneous PPO membrane	-	-	N-methyl-2 pyrrolidone (NMP)	-	5.93E-05	50	-	-
Wang, Li [268]	poly(ether block amide) (Pebax 1657) composite membrane	-	-	-	-	1.51E-05	-	-	-
Wang, Liu [269]	Pebax-PEG-MWCNT	-	-	ethanol/water mixture (70/30wt%)	-	1.49E-03	-	-	20 - 30
Wang, Fang [270]	polypropylene (PP) and Polyvinylidene fluoride (PVDF)	-	-	20 wt% MEA	-	-	-	1.5	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Wickramanayake, Hopkinson [271]	fabricated utilizing Matrimids and Torlons	hollow fibre membrane	-	N-methyl-2-pyrrolidone (NMP)	-	6.73E-06	3.2	60-90	-
Wu, Maurin [272]	porous Zr-carboxylate	metal organic framework	-	-	-	-	-	-	10-15
Wu, Wang [273]	alumina-supported AlPO-18	-	-	1.0 Al <sub>2</sub> O <sub>3</sub> :3.16 P <sub>2</sub> O <sub>5</sub> : 6.32 tetraethyl ammonium hydroxide (TEAOH): 186 H <sub>2</sub> O	-	1.12E-04	-	-	-
Xin, Wu [274]	TiO <sub>2</sub>	mixed matrix membranes (mmms)	counter current flow	N,N-Dimethyl acetamide (DMAc)	-	3.27E-04	20	-	10 - 30
Xiong, Gu [275]	hydroxide-exchange membrane	-	concurrent flow	KOH	-	-	-	-	10
Yan, He [276]	amino acid salt, potassium L-argininate (PA)	membrane contractor	counter current flow	monoethanolamine (MEA)	-	-	-	-	-
Yan, Zhang [277]	-	hollow fibre membrane	counter current flow	MDEA/PZEA [methyl-diethanolamine/2-(1-piperazinyl)-ethylamine] 0.5mol/L	-	1E-01	-	-	14
Yin, Wang [278]	amino-functionalized MOF membrane: tubular CAU-1 membrane	metal organic structures	-	-	-	8.04E-04	-	-	1-9
Zaidiza, Billaud [279]	-	hollow fibre membrane	counter current flow	MEA	-	-	-	-	15
Zhang and Wang [280]	porous polyetherimide (PEI)	hollow fibre membranes	concurrent flow	2M sodium taurinate	-	9.00E-06	-	-	-
Zhang, Gong [281]	Al <sub>2</sub> O <sub>3</sub>	silver carbonate membranes	counter current flow	-	-	-	-	-	-
Zhang, Gong [282]	silver-carbonate	silver carbonate membranes	concurrent flow	-	-	-	-	-	-
Zhang, Seames [283]	PETE	membrane contractor	-	MEA and NaOH	-	-	120	120	-
Zhang, Yan [284]	-	gas liquid hollow fibre membrane contactor (hfmc)	counter current flow	MDEA/PZEA 0.3:0.7 mol/L	-	-	-	-	14

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Zhang, Yan [285]	polypropylene (PP)	hollow fibre membrane contactor (hfmc)	counter current flow	monoethanolamine (MEA)	-	-	-	-	-
Zhao, Ren [286]	Poly(amide-6-b-ethylene oxide)/SAPO-34	mixed matrix membranes (mmms)	-	-	-	1.15E-04	-	-	-
Zhao, Jung [287]	polyvinylalcohol	multiwalled carbon nanotubes (mwnts)	-	-	-	2.59E-03	30	-	-
Zhou, Tran [288]	Perfluorocyclobutyl polymer	thin film composite membranes (tfc)	-	methanol	-	4.82E-03	-	-	-
Zhou, Korelskiy [289]	Zeolite SAPO-34	zeolite membranes	-	-	-	3E-02	-	-	50
Azizi and Mousavi [290]	polyurethane membrane	-	-	-	-	1.64E-04	-	-	-
Carapellucci, Giordano [291]	Polyactive® PEBAX®/PDMS-PEG TR-PBI	one-dimensional hollow-fibre model	-	monoethanolamine (MEA)	-	-	-	-	-
Dong, Sun [292]	PEBA2533	metallo-supramolecular polymers membrane	-	-	-	-	-	-	-
Farjami, Moghadassi [293]	polyvinylidene fluoride (PVDF)	hollow fibre membrane contactor	counter current flow	distilled water	-	7.50E-05	-	-	-
Fazaeli, Razavi [294]	-	hollow-fibre membrane	-	tetramethylammonium glycinate ([N1111][Gly])	-	-	175 - 475	100	15
Gilassi and Rahmanian [295]	polytetrafluoroethylene (PTFE)	flat sheet membrane	cross-flow	-	-	-	-	-	-
Goyal, Suman [296]	micro-porous Polypropylene (PP)	hollow-fiber membrane module (hfmm)	concurrent flow	diethanolamine (DEA)	-	3.51E-06	-	-	-
Hwang, Chi [297]	polystyrene (PS) core and ZIF-8 shell	mixed matrix membranes	-	MeOH	-	1.25E-04	-	-	-
Kammakakam, Nam [298]	6FDA-durene polyimide (PI)	-	-	1,4-Dibromobutane	-	1.92E-04	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Khalilinejad, Sanaeepur [299]	poly (ether-6-block amide) (Pebax-1657) on a polyvinylchloride (PVC)	thin film composite membranes (tfcs)	-	-	-	1.24E-05	-	-	-
Kim, Jeon [300]	poly(dimethylsiloxane)-g-poly(oxyethylene methacrylate) (PDMS-g-POEM)	-	-	EtOH	-	4.55E-05	-	-	-
Konruang, Sirijarukul [301]	polysulfone (PSF) membranes	-	counter current flow	-	-	3.90E-02	-	-	-
Lee, Magnone [302]	fluoroalkylsilanized Al <sub>2</sub> O <sub>3</sub>	hollow fibre membrane	concurrent flow	-	-	2.6E-01	10-50	20-50	20
Lee, Choi [303]	PSf	hollow fibre membrane	concurrent flow	-	-	7.24E-05	-	-	-
Li, Wang [304]	hydrophilic membrane amino group-rich polyvinylamine (PVAm)	-	-	-	+	1.12E-03	-	960	-
Li, Wang [305]	poly(diallyldimethylammonium carbonate-co-vinylamine) (P(DAD-MACA-co-VAm))	-	concurrent flow	DADMAC	-	4.24E-03	20	-	15
Li, Cheng [306]	CNTs and GO	mixed matrix membranes (mmms)	-	DMF	-	1.53E-05	-	-	10-30
Li, Jiang [307]	polyethylenimine-functionalized graphene oxide nano-sheets (PEG-PEI-GO)	mixed matrix membranes (mmms)	-	-	-	9.95E-04	-	-	-
Li, Ma [308]	-	composite matrix membranes	-	70 wt% ethanol and 30 wt% water	-	2.44E-03	-	-	-
Li, Li [309]	calcium lignosulfonate (CaLS)	anionic surfactant-doped pebax membrane	-	ethanol	-	5.03E-09	-	-	10-30
Li, Xin [310]	-	polymer electrolyte membrane	concurrent flow	-	-	9.05E-04	-	-	10
Liao, Wang [311]	polyvinylamine (PVAm) fixed carrier membrane	-	concurrent flow	-	-	3.66E-04	-	-	15

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Lin, Ge [312]	NH2-MIL-101(Al)	mixed matrix membranes (mmms)	-	N,N-dimethylacetamide (DMAc)	-	4.07E-03	-	-	-
Lin, Chang [313]	hydrophobic fluorocarbon-modified silica aerogel	-	counter current flow	-	-	3E-02	-	-	-
Lock, Lau [314]	-	hollow fibre membrane module	co and counter current	-	-	-	-	-	10-60
Loloei, Omidkhah [315]	polyethylene glycol (PEG 200) as a CO <sub>2</sub> -philic polymer membrane	mixed-matrix membrane (mmm)	concurrent flow	-	-	2.35E-05	-	-	-
Mahmoudi, Asghari [316]	polymer/liquid/solid (PEBA/PEG/nanozeolite X)	mixed matrix membranes (mmms)	concurrent flow	ethanol	-	1.53E-04	-	-	-
Mondal, Barooah [317]	25wt% poly(allylamine) and 15wt% 2-amino-2-hydroxymethyl-1,3-propanediol	polymeric membranes	counter current flow	water	-	1.04E-05	0.05	0.03	-
Mulukutla, Chau [318]	polyamidoamine (PAMAM)	hollow fibre membrane	cross-flow	monoethanolamine (MEA)	-	-	-	-	-
Nabian, Ghoreyshi [319]	polysulfone (PSF)	flat sheet membrane	counter current flow	N,N-dimethylformamide (DMF)	-	1.92E-07	1000	3-12.5	-
Nasir, Mukhtar [320]	polyethersulfone (PES) and PES/carbon molecular sieve (CMS)	mixed matrix membranes	-	N-Methyl-2-pyrrolidone (NMP)	-	4.91E-05	-	-	-
Nasir, Mukhtar [321]	polyethersulfone (PES) membrane	-	counter current flow	diethanolamine (DEA)	-	4.96E-05	-	-	-
Nordin, Racha [322]	zeolitic imidazole framework 8 (ZIF-8)	metal-organic frameworks (mofs)	-	tetrahydrofuran (THF) and methanol	-	2.03E-05	-	-	-
Nwogu, Kajama [323]	-	inorganic membranes	concurrent flow	-	-	-	-	-	-
Park, Lee [324]	poly(ethylene glycol) behenyl ether methacrylate (PEGBEM) and poly(oxyethylene methacrylate) (POEM)	-	-	2-methylpropionitrile	-	1.17E-04	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Park, Lee [325]	Poly(methoxy(polyethyleneoxy)propyl-co-methacryloxypropyl)silsesquioxane membranes	-	-	-	-	2.47E-03	-	-	50
Qiao, Wang [326]	amines (SMAs) (ethanediamine (EDA), piperazine (PIP), monoethanolamine (MEA) and diethanol amine (DEA))	-	-	monoethanolamine (MEA)	-	1E-02	960 - 30	-	-
Rabiee, Alsadat [327]	poly(amide-12-b-ethyleneoxide)(Pebax1074)/SAPO-34	mixed matrix membranes (mmms)	cross-flow	1-Butanol	-	1.80E-12	-	-	-
Rahim, Ghasem [328]	PVDF	hollow fibre membrane	counter current flow	MEA and NaOH	-	3.90E-04	10-100	10 - 40	10
Rezaei, Ismail [329]	1 montmorillonite and Cloisite 15A	mixed matrix membranes (mmms)	concurrent flow	0.02M sodium hydroxide (NaOH)	-	6.00E-06	140	20-350	-
Sadoogh, Mansourizadeh [330]	PVDF	hollow fibre membrane	-	1 methyl-2-pyrrolidone (NMP) (1 M)	-	6.75E-05	-	-	-
Scofield, Gurr [331]	poly(dimethylsiloxane)-b-poly(ethylene glycol) copolymer	thin film composite membranes (tfc)	-	n-butanol	-	7.04E-04	-	1	-
Seoane, Coronas [332]	-	mixed matrix membranes (mmms)	cross and counter current flow	-	-	-	-	-	-
Shen, Liu [333]	-	-	-	-	-	-	-	-	-
Shin, Hwang [334]	Pd-based membrane	-	counter current flow	-	-	-	-	-	40
Skorek-Osikowska, Bartela [335]	-	-	-	-	-	-	-	-	-
Sorribas, Comesaña-Gándara [336]	ETS-10	mixed matrix membranes	-	-	-	2.10E-05	-	-	-
Sun, Wen [337]	-	nanoporous graphene	-	-	-	-	-	-	-
Taniguchi, Kai	Poly(amidoamine)	-	-	-	-	1.34E-17	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
[338]	(PAMAM)								
Tong, Zhang [339]	ceramic-carbonate dual-phase	ceramic membranes	concurrent flow	-	-	-	15-50	120	-
Vakharia, Ramasubramanian [340]	crosslinked polyvinylalcohol (PVA) matrix	polymeric facilitated transport membranes	counter current flow	-	-	-	-	-	20-31
Wang, Sun [341]	faujasite/polyethersulfone composite	-	-	Tetramethylammonium hydroxide (25% aqueous)	-	3.6E-11	-	-	-
Wang, Mundstock [342]	Amine-modified Mg-MOF-74/CPO-27-Mg membrane	-	-	N-dimethylformamide	-	5.43E-06	50	50	-
Wang, Tian [343]	Poly(ethylene glycol)-containing polymeric sub microspheres (PEGSS) membranes	-	-	N,N-dimethylformamide (DMF) and ethanol	-	3.12E-05	20	-	-
Xin, Gao [344]	graphene oxide (GO-DA-Cys) nanosheets into a sulfonated poly(ether ketone) (SPEEK) polymer	-	-	2-(3,4-dihydroxyphenyl)-ethylamine (dopamine)	-	1.90E-04	-	-	-
Xin, Li [345]	sulfonated poly(ether ketone)	mixed matrix membranes (mmms)	-	N-dimethylformamide	-	2.66E-04	20	-	10 - 30
Xing, Peters [346]	-	molten carbonate dual phase membranes	counter current flow	-	-	-	-	-	-
Zaidiza, Belaissaoui [347]	polypropylene (PP), polyvinylidene fluoride (PVDF) and polytetrafluoroethylene (PTFE)	hollow fibre membrane	counter current flow	30wt% Monoethanolamine (MEA)	-	8E-02	-	-	-
Zhang, Qu [348]	Polypropylene	membrane contractor	counter current flow	Monoethanolamine (MEA)	-	-	-	-	-
Zhang, Qu [349]	polypropylene	hollow fibre membranes	-	-	-	-	-	-	14
Zheng, Hu [350]	SSZ-13 membranes - all silica and zeolite	-	-	N,N,N, trimethyl-1-adamantammonium hydroxide and tetraethylammonium hydroxide	-	2.40E-04	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Zhou, Wang [351]	SSZ-13	-	-	N,N,N-trimethyl-1-adamantammonium	-	2.28E-04	150 - 400	-	-
Zhou, Luo [352]	Pebax	mmms	-	ethanol/water (70/30 wt%)	-	6.03E-05	50	20	-
Adewole and Ahmad [353]	polymeric membranes: 6FDA-DAM:DABA 2: 1	-	cross flow	-	-	-	-	-	23.58 – 50
Amooghini, Omidkhan [354]	nano-porous sodium zeolite-Y	mixed matrix membranes	-	-	-	5.93E-06	-	-	10
Arias, Mussati [355]	-	-	cross and counter flow	-	-	2.11E-04	-	-	13
Hosseinzadeh Beiragh, Omidkhan [356]	nanoporous ZSM5	mixed matrix membrane	-	water/ethanol (30/70 wt. %)	-	1.54E-04	-	-	-
Chang and Kang [357]	poly(vinylidene fluoride-co-hexafluoropropylene) (PVDF-HFP)	polymeric membrane	-	tetrafluoroboric acid solution	-	-	-	-	-
Chen and Ho [358]	polyvinylamine (PVAm)/piperazine glycinate (PG) membrane	-	counter current flow	-	-	2.29E-05	-	-	-
Dai, Bai [359]	thin film composite (TFC) membrane	-	-	Ethanol (96 wt%), toluene and n-hexane	-	1.71E-06	-	-	10
Dong, Hou [360]	hyperbranched polyethylenimine (HPEI), and trimesoyl chloride (TMC) coating on a polysulfone membrane	fixed carrier composite membrane	-	-	-	1.95E-06	0.5	-	10
Dong, Zhang [361]	porous graphene-based macroscopic membrane	reduced graphene nanosheets	-	70 wt% ethanol and 30 wt% water	-	3.22E-05	-	-	-
Dong, Zhang [362]	montmorillonite (MMT) functionalized with poly(ethylene glycol) methyl ether (PEG) and aminosilane coupling agents in a PEBA membrane	-	-	2 wt % in n-butanol	-	9.01E-05	-	-	-
Dong, Liu [363]	MIL-68(Al) (MIL5Material of Institute Lavoisier)	mixed matrix membrane (mmm)	-	DMF	-	5.71E-05	50	-	50

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Ebrahimi, Mollaiy-Berneti [364]	mixed matrix membranes (MMMs) using polyethersulfone (PES)	-	concurrent flow	-	-	-	-	-	-
Fan, Xie [365]	[Choline][Pro]/polyethylene glycol 200 (PEG200)	ionic liquid membranes (silms)	concurrent flow	~45 wt% in methanol	-	3.87E-04	-	-	10 – 14
Gilassi and Rahmanian [366]	-	hollow fibre membrane	-	monoethanolamine (MEA)	-	3.02E-04	-	-	20
Jeon, Kim [367]	poly(dimethylsiloxane)-graft-poly(4-vinylpyridine) (PDMS-g-P4VP)	-	-	-	-	3.32E-07	-	-	-
Jomekian, Behbahani [368]	ZIF-8/Pebax 1657/PES	hollow fibre membrane	-	Dimethylformamide (99.99%, DMF)	-	3.04E-04	-	-	-
Jusoh, Lau [369]	-	-	-	-	-	-	-	-	25
Kang, Peng [370]	poly(ether imide) (Ultem) or polybenzimidazole (PBI)	mixed matrix membranes (mmms)	-	-	-	-	-	-	-
Karamouz, Maghsoudi [371]	PEBAX-1074. Membranes	-	concurrent flow	Dimethylformamide (DMF)	-	8.32E-04	-	-	-
Karimi, Korelskiy [372]	acetate-functionalized silica	silica membranes	-	-	-	1E-02	-	-	-
Kertik, Khan [373]	polyimide (PI) Matrimid® and ZIF-8, ZIF-7 and NH <sub>2</sub> -MIL-53(Al)	mixed matrix membranes (mmms)	-	chloroform and methanol	-	-	-	-	-
Kim, Choi [374]	poly(2-ethyl-2-oxazoline) (POZ) and semicrystalline poly(amide-6-b-ethylene oxide) (PEBAXVR MH 1657)	mixed matrix membranes (mmms)	-	ethanol	-	1.61E-06	-	-	-
Kim, Fu [375]	polyethylene glycol(PEG) matrix	mixed matrix membrane	-	-	-	2.41E-06	0.6-6.3	-	30
Kim, Fu [376]	polyethylene glycol (PEG)-based ultra-thin film composite	mixed matrix membranes (utfc-mmms)	-	tris(2-aminoethyl amine)	-	1.49E-03	-	-	-
Kim, Park [377]	polypropylene (PP)	polymeric hollow fibre membranes	counter current flow	-	-	2.33E-06	150-300	200–800	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Kim, Chi [378]	poly(1-trimethylgermyl-1-propyne) (PTMSP)	mixed matrix membrane (mmm)	-	THF	-	3E-02	-	-	-
Koolivand, Sharif [379]	graphene-oxide nanosheet	mixed matrix membranes (mmms)	-	tetrahydrofuran (THF) and N-methyl-2-pyrrolidone (NMP) solvents	-	5.07E-03	-	-	-
Korelskiy, Grahn [380]	sub-micron (ca. 0.5 mm) b-oriented MFI zeolite	zeolite membranes	-	-	-	4E-02	-	-	-
Krea, Roizard [381]	copolyetherimide polymers (PEI)	mixed matrix membranes	-	DMAC	-	1.28E-04	-	-	-
Lee, Jung [382]	PEDOT-PSS embedded comb copolymer	polymeric membranes	-	N,N-dimethylformamide (DMF)	-	7.55E-06	-	-	-
Li, Zheng [383]	graphene-based	mof	concurrent flow	-	-	-	-	-	-
Li, Zhang [384]	-	graphene oxide membranes	-	-	-	-	-	-	-
Liao, Wang [385]	poly(diallyldimethylammonium carbonate) (PDDACA)	fixed carrier membranes	-	ethanol	-	1.49E-03	-	-	-
Lin and Kuo [386]	mesoporous bis(trimethoxysilyl)hexane (BTMSH)/tetraethylorthosilicate (TEOS)-based hybrid silica aerogel	-	cross-flow	ethanol	-	8E-02	-	-	-
Liu, Liu [387]	polyvinylidene Fluoride (PVDF)	ionic liquid membranes (silms)	concurrent flow	bis((trifluoromethyl)sulfonyl)imide ([BMIM][NTf <sub>2</sub> ])	-	3.59E-03	-	-	-
Lu, Khan [388]	polysulfone (PSf)	ionic liquids (ils) membrane	-	-	-	7.14E-04	-	-	50
Moradi, Chenar [389]	PDMS coated TFC-RO membranes	thin film composite (tfc) membrane	concurrent flow	n-hexane	-	-	-	-	-
Mosleh, Mozdianfard [390]	polyethersulfone (PES)	-	concurrent flow	DMAc	-	5.23E-05	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Mubashir, Yeong [391]	tetrapropylammonium bromide (TPABr)-templating silica sols	-	concurrent flow	-	-	1.48E-03	-	-	10
Nematollahi, Dehaghani [392]	alumina nano particles (Al <sub>2</sub> O <sub>3</sub> ) and poly(4-methyl-1-pentyne)	mixed matrix membranes (mmms)	-	-	-	4.01E-04	-	-	-
Nematollahi, Dehaghani [393]	Al <sub>2</sub> O <sub>3</sub> nanoparticles	mixed matrix membranes (mmms)	-	-	-	5.51E-03	-	-	-
Nguyen, Gong [394]	Ca-A zeolite	mixed matrix membranes	-	dichloromethane (DCM)	-	4.62E-05	-	-	50
Otani, Zhang [395]	tetrabutylphosphonium 2-cyanopyrrolide ([P4444][2-CNpyrr])	facilitated transport membrane	-	-	-	-	-	-	-
Pohlmann, Bram [396]	PolyActiveTM	thin film multilayer composite membrane	counter current flow	-	-	-	-	-	14.5
Qin, Lv [397]	SPEEK/amino acid salt membranes	-	concurrent flow	N,N-Dimethyl acetamide (DMAc)	-	1.19E-04	-	-	-
Rafiq, Deng [398]	cross-linked poly (ethylene oxide) (PEO)	facilitated transport membranes	-	-	-	7.11E-05	-	-	10-20
Rahmani, Kazemi [399]	organic matrices (polymer matrices)	mixed matrix membranes (mmms)	cross flow	chloroform	+	1.63E-05	0.01	-	-
Razavi, Shirazian [400]	-	hollow fibre membrane contractors	counter current flow	DEA	-	-	-	-	-
Rui, James [401]	1 (IRMOF-1)	metal-organic framework (mof) membranes	-	-	-	6.2E-01	-	-	-
Saedi, Seidi [402]	amino-cellulose (AC)	-	cross-flow	N,N-dimethyl aminopropylamin	+	7.19E-06	-	-	30
Saeed and Deng [403]	PVA-mimic enzyme	carbon nanotube composite membrane	-	NaOH	+	9.24E-05	-	-	-
Sanaeepur, Kargari [404]	nano-porous sodium zeolite Y (NaY)	mixed matrix membranes (mmms)	concurrent flow	-	-	4.07E-06	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Sánchez-Laínez, Zornoza [405]	d ZIF-11 (nZIF-11 and ZIF-11, respectively)	nano and micro sized polymeric mixed matrix membrane	-	DMAc solvent	-	-	-	-	-
Sánchez-Laínez, Zornoza [406]	ZIF-8	mixed matrix membranes (mmms)	-	MeOH	-	-	-	-	-
Scofield, Gurr [407]	poly(ethylene glycol)-block-poly(pentafluoropropyl acrylate) diblock copolymers	-	-	n-butanol	-	1.69E-03	-	1	-
Shen, Liu [408]	graphene oxide (GO)-polyether block amide (PEBA)	mixed matrix membranes	-	-	-	6.63E-05	-	-	-
Shen, Zhang [409]	UiO-66-polyether block amide	mixed matrix membranes	-	N,N-dimethylformamide (DMF)	-	7.84E-05	-	-	-
Shen, Wang [410]	MoS <sub>2</sub>	mixed matrix membranes (mmms)	-	-	-	3.16E-05	-	-	-
Solimando, Lherbier [411]	Pebax	-	-	N,N-dimethylformamide (DMF)	-	3.90E-04	-	-	-
Sumer and Keskin [412]	polymeric membranes (Ultem, Matrimid, 6FDA-DAM, and modified PDMS)	mixed matrix membranes (mmms)	-	-	-	-	-	-	-
Sun, Srivastava [413]	faujasitic (FAU)	zeolite membrane	-	tetramethylammonium hydroxide solution (TMAOH, SACHEM)	-	3.44E-04	-	-	20
Waheed, Mushtaq [414]	polysulfone	mixed matrix membranes (mmms)	concurrent flow	4-aminophenazone (4-AMP)	-	1.71E-04	-	-	-
Wang, Li [415]	-	polymeric membranes	concurrent flow	-	-	6.51E-04	-	-	-
Wang, Yang [416]	-	nanoporous graphene membranes	-	-	-	-	-	-	15
Wong, Goh [417]	-	nanocomposite membrane	cross-flow	-	-	-	-	-	-
Woo, Dong [418]	poly(benzoxazole-co-imide)	hollow fibre membranes	cross-flow	hydroxyl polyimide-co-polyimide (HD5)	-	3.38E-04	6000	-	14 - 80 vol%

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Wu, Zhao [419]	nanoporous polyethersulfone (PES) membranes	-	counter current flow	PES/N-methyl-2-pyrrolidone (NMP)/2-methoxyethanol (2-ME)	-	1.00E-04	60-30	-	-
Xiang, Pan [420]	poly(ether-block-amide)	mixed matrix membranes (mmms)	concurrent flow	water/ethanol (30/70 wt %)	-	1.13E-04	-	-	-
Xin, Zhang [421]	poly(ether-block-amide) (Pebax 1657)	mixed matrix membranes (mmms)	-	tris(hydroxymethyl) amino methane (Tris)	-	6.49E-05	20	-	-
Xin, Zhang [422]	sulfonated poly(ether ketone) (SPEEK)	mixed matrix membranes (mmms)	-	DMF	-	2.56E-04	-	-	10 - 30
Zaidiza, Wilson [423]	-	hydrophobic microporous hollow fibre membrane	counter current flow	-	-	2E-02	-	-	14-24.2
Zhang, Guo [424]	N-isopropylacrylamide hydrogel (NIPAM-CNTs)	mixed matrix membranes (mmms)	concurrent flow	N,N,N',N''-pentamethyldiethylenetriamine	-	6.33E-04	-	-	-
Zhang, Li [425]	Polypropylene	hollow fibre membrane	counter current flow	monoethanolamine (MEA)	-	3.00E-14	8333	400	12
Zhang, Tong [426]	M microstructure of an ALD-ZrO <sub>2</sub>	silver carbonate membranes	cross-flow	-	-	-	-	-	-
Zhong, Bu [427]	aluminophosphate (AIPO)-17 and silicoaluminophosphate (SAPO)-17	-	-	-	-	5.76E-04	-	8000	-
Alavi, Kargari [428]	polydimethylsiloxane/zeolite 4A (PDMS/zeolite 4A)	mixed matrix membranes	-	N,N-dimethylformamide (DMF) and hexane	-	2.10E-03	-	-	-
Azizi, Arzani [429]	polymeric nanocomposite membranes (PNMs)	-	cross-flow	dimethylacetamide (DMA), Dimethylformamide (DMF) and 1-Butanol	-	7.90E-05	-	-	-
Azizi, Mohammadi [430]	polymeric nanocomposite membranes (PNMs)	-	-	-	-	3.98E-04	-	-	20
Azizi, Mohammadi [431]	poly(ether-block-amide) (Pebax 1074, Pebax 1657 and Pebax 2533)	nanocomposite membranes - modified multi-walled carbon nanotubes	cross-flow	-	-	1.26E-04	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
		(mwcnts)							
Benito, Sánchez - Laínez [432]	poly[1-(trimethylsilyl)-1-propyne] (PTMSP)	-	-	-	-	4.46E-05	-	-	10
Brunetti, Cersosimo [433]	polybenzoxazole-co-imide (TR-PBOI)	mixed matrix membrane	-	N-Methyl-2-pyrrolidone (NMP)	-	1.02E-04	-	-	-
Cheng, Wang [434]	copper 1,4-benzenedicarboxylate nanosheets, CuBDCs	mixed matrix membranes, metal organic structure	-	-	-	8.18E-05	-	-	-
Galaleldin, Mannan [435]	titanium dioxide (TiO <sub>2</sub> ) nanoparticles: 5 % loading of TiO <sub>2</sub>	mixed matrix membranes	-	-	-	5.25E-06	-	-	-
Ghadiri, Marjani [436]	2-amino-2-methyl-1-propanol (AMP)	hollow-fibre contactor	counter current flow	-	-	-	170	10-130	-
Hosseini and Mansourizadeh [437]	polypropylene	hollow fibre membrane contactor (hfmc)	cross-flow	monoethanolamine (MEA)	-	-	-	-	-
Hu, Cheng [438]	SAPO 34	ionic liquids membranes	concurrent flow	ethanol/water solvent (70/30, weight ratio)	-	8.22E-05	-	-	50
Hussain [439]2017)	PVAm (polyvinylamine) and PVA (polyvinyl alcohol) blended membrane layer on a polysulfone support	-	-	-	-	-	-	-	9.5-2.9
Ilyas, Muhammad [440]	ionic liquids membrane	-	cross-flow	3-aminopropyl) trimethoxysilane and acetic acid	-	5.43E-05	-	-	-
Isfahani, Sadeghi [441]	Poly (ethylene glycol) (PEG)-based polyurethane (PU) membranes	-	-	-	-	1.26E-04	-	-	-
Jin, Huang [442]	gas-liquid membrane contractor (GLMC): Poly(vinylidene fluoride) (PVDF)	hollow fibre membrane	cross-flow	water, monoethanolamine (MEA, primary amine), potassium carbonate (K <sub>2</sub> CO <sub>3</sub> , inorganic salt), potassium hydroxide (KOH, inorganic salt),	-	1.17E-06	250	50-140	40

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
				and potassium sarcosine (PS, organic salt)					
Jo, An [443]	1,3-cyclohexanebis-methylamine (CHMA) and trimesoyl chloride(TMC)	thin film composite (tfc) hollow fibre membrane	concurrent flow	N-hexane	-	6.03E-06	-	-	70
Jung, Park [444]	copolymer comprising of poly(ethylene-alt-maleic anhydride) (PEMA) main chains and poly(propylene glycol) (PPG) side chains	-	-	butanol (BuOH)	-	1.79E-05	-	-	-
Jusoh, Yeong [445]	(ZIF-8) nanofiller and 6FDA-durene polymer	mixed matrix membranes (mmms)	-	triethylamine (TEA)	-	6.57E-04	-	-	90
Kang, Chan [446]	poly(vinylidene fluoride) (PVDF)	hollow fibre membrane	cross-flow	methyldiethanolamine (aMDEA)	-	4.80E-06	4500	100	70
Karousos, Labropoulos [447]	1-alkyl-3-methylimidazolium	ionic liquid membranes (silms)	concurrent flow	-	-	8.10E-07	-	-	1-50
Karunakaran, Villalobos [448]	ultrathin graphene oxide (GO)	ionic liquid ultrathin composite membranes	-	1-ethyl-3-methylimidazolium acetate ([EMIM][Ac]) and 1-ethyl-3-methylimidazolium tetrafluoroborate ([EMIM][BF <sub>4</sub> ])	-	7.44E-06	-	-	-
Kgaphola, Sigalas [449]	nanocomposite SAPO-34 membranes	-	-	-	-	1.38E-05	-	-	-
Khakpay, Rahmani [450]	nanoporous graphene (NPG) and graphene oxide (NPGO)	-	-	-	-	1.11E-06	-	-	-
Khalilinejad, Kargari [451]	hydrophilic/hydrophobic silica nanoparticles (0-10 wt%)	mixed matrix membranes	-	ethanol and water	-	1.28E-04	-	-	-
Khosravi, Omidkhah [452]	CuBTC and NH <sub>2</sub> -CuBTC	mixed matrix membranes (mmms)	-	ethanol	-	2.52E-10	-	-	-
Kim, Park [453]	poly(glycidyl methacrylate-g-poly(propylene glycol))-co-	-	-	ethanol/water mixed solvent (70/30 wt %)	-	1.01E-04	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
	poly(oxyethylene methacrylate) (PGP-POEM) combcopolymer								
Kim, Kim [454]	-	hollow fibre membrane	counter current flow	-	-	8.49E-04	-	2600 - 4000	14
Kline, Weidman [455]	crosslinked poly(ethylene oxide) (PEO) membranes	-	-	3.4 mL DMAc	-	6.28E-04	-	-	-
Lai, Yeong [456]	(ZIF)-8	zeolite membranes	-	methanol	-	1.66E-05	30	200	36
Lai, Yeong [457]	zeolitic imidazolate framework-8 (ZIF-8) membrane	-	-	-	-	4.80E-05	-	-	15,50,90
Lee, Park [458]	zeolitic imidazolate framework-8 (ZIF-8)	-	-	-	-	9.83E-05	-	-	50
Lee, Lee [459]	poly (benzoxazole-co-imide) (XTR-PBOI)	hollow fibre membrane	concurrent flow	-	-	1.25E-04	130-140	-	14
Lee, Binns [460]	-	hollow fibre membrane	counter current flow	-	-	2.88E-04	-	-	14
Li, Ding [461]	porous graphene (PG)	-	-	sodium hydroxide (NaOH)	-	2.26E-07	-	-	-
Li, Pyrzynski [462]	PEEK	hollow fibre membrane	counter current flow	methyl-diethanolamine (aMDEA)	-	-	330000-775000	3400-5900	6.4-16.6
Liu, Zhou [463]	[TESPMIM][BF4]	ionic liquids (rtils)	cross-flow	tetraethylammonium hydroxide	-	1.20E-04	150-400	-	-
Liu, Li [464]	poly(amide-b-ethylene oxide) (Pebax MH 1657)	mixed-matrix membrane (mmm)	concurrent flow	polydopamine	-	9.05E-05	30	30	10
Lu, Ge [465]	Potassium glycinate (PGLY)	ionic liquid membranes (silms)	counter current flow	monoethanolamine glycinate ([MEA][GLY])	-	3E-02	500	200	-
Mahdavi, Azizi [466]	Pebax 1657/PEG1000/ $\gamma$ -Al <sub>2</sub> O <sub>3</sub> membrane	-	cross-flow	DMF	-	8.30E-05	-	-	-
Mannan, Mohshim [467]	polyethersulfone (PES)	ionic liquid membranes	-	1-methyl-2-pyrrolidone EMPLURA1 (NMP)	-	4.57E-06	-	-	-
Marti, Wickramanayake [468]	polymeric porous Hollow Fiber membrane	-	-	-	-	6.92E-06	-	0.23 - 013	20
Martín-Gil, López [469]	ETS-10, TS-1 having Si/Ti=100 and TS-1 using Si/Ti=25	mixed matrix membranes	-	N-methyl-2-pyrrolidinone, NM	-	1.17E-05	-	-	50

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Mirfendereski and Mohammadi [470]	polypropylene porous	hollow fibre membrane	counter current flow	methyl-diethanolamine (MDEA) (0.5mol/L)	-	-	360000	11.7	4-12
Monteiro, Nabais [471]	-	metal organic framework	-	-	-	-	-	-	-
Morris, Jacques [472]	PIM-1	mofs	-	-	-	4.65E-03	-	-	-
Nordin, Ismail [473]	modified ZIF-8	mixed matrix membranes	-	N,N-methylpyrrolidone (NMP), tetrahydrofuran (THF) and ethanol	-	3.22E-05	-	-	-
Park, Lee [474]	MgCO <sub>3</sub> -crystal-	mixed matrix membranes (mmms)	-	ethanol	-	6.29E-06	-	-	-
Peng, Wang [475]	graphene oxide (GO) nanosheets	-	-	dopamine (DA) solution	-	5.54E-05	-	-	-
Prasad and Mandal [476]	polymeric (CS-TEPA) membrane	-	concurrent flow	-	-	7.6E-01	0.05	0.03	20
Qu, Li [477]	nitrogen-rich membranes, g-C <sub>3</sub> N <sub>4</sub> , g-C <sub>3</sub> N <sub>3</sub> , C <sub>2</sub> N-h <sub>2</sub> D, g-C <sub>12</sub> N <sub>8</sub> , and p-BN	-	-	-	-	-	-	-	-
Quan, Li [478]	cross-linked poly (ethylene oxide) (PEO)	mixed matrix membranes (mmms)	-	-	-	5.03E-10	-	-	-
Rahmawati, Nurkhamidah [479]	hydrophobic polypropylene	polymeric hollow fibre membranes	cross-flow	-	-	1.48E-05	400	400	-
Ranjbaran, Kamio [480]	ionic liquid (IL)-based gel membrane	-	concurrent flow	N,N'-methylenebis(acrylamide) (MBAA)	-	2.94E-03	100	-	50
Ur Rehman, Rafiq [481]	triethanolamine formate (TEAF) and triethanolamine acetate (TEAA)	ionic liquid membranes	-	triethanolamine	-	7.52E-05	200	-	-
Ricci, Minelli [482]	polyimide membrane (Matrimid®)	polymeric membranes	-	5(6)-amino-1-(4'-aminophenyl)-1,3,-trimethyl indane	-	-	-	-	-
Rudaini, Naim [483]	PVDF polymer	hollow fibre membrane	counter current flow	monoethanolamine (MEA)	-	4.07E-04	200	50	-
Saidi [484]	polytetrafluoroethylene (PTFE), polypropylene (PP) and polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter-current flow	diethanolamine (DEA)	-	-	150	100	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Saidi [485]	4-diethylamino-2-butanol (DEAB)	hollow fibre membrane	cross-flow	MEA, DEA, TEA, MDEA	-	2.26E-07	-	-	20
Selyanchyn and Fujikawa [486]	MTR Polaris™	-	counter current flow	monoethanolamine (MEA)	-	1.1E-01	-	-	-
Shafie, Man [487]	polycarbonate-Silica	-	-	3-aminopropyl trimethoxysilane and N(2 aminoethyl) 3-amino-propyltrimethoxysilane	-	1.33E-03	-	-	-
Shamsabadi, Seidi [488]	TiO <sub>2</sub> nanoparticles	polymer matrix membrane	-	3-aminopropyl-diethoxymethylsilane	-	4.85E-05	-	-	-
Shin, Chi [489]	PVC-POEM/ZIF-8	mixed matrix membranes	-	1,1,4,7,10,10hexamethyl triethylene tetramine (HMTETA, 99%)	-	6.48E-06	-	-	-
Song, Liu [490]	-	hybrid cryogenic membrane process	concurrent flow	-	-	-	-	-	15-84
Song, Qiu [491]	ZIF, ZIF-8, pore size, ~ hybrid polysulfone (PSF) and (ZIF, ZIF-8	-	-	-	-	2.73E-05	-	-	-
Taniguchi, Kinugasa [492]	olydimethylsiloxane (PDMS)	thin film composite membranes	-	methanol	-	1.52E-03	0.8	-	-
Taniguchi, Wada [493]	poly(amidoamine)s (PAMAMs) incorporated into a cross-linked poly(ethylene glycol)	-	concurrent flow	monoethanolamine (MEA)	-	7.65E-05	100	100	40
Tong and Ho [494]	poly(N-vinylamine)s, namely poly(N-methyl-N-vinylamine), poly(N-isopropyl-N-vinylamine), and poly(N-tertiary-butyl-N-vinylamine)	-	counter current flow	-	-	1.39E-03	60 and 30	-	20
Tseng, Chuang [495]	mesopore SBA-15	mixed matrix membranes (mmms)	-	HCL solution	-	2.18E-08	-	-	-
Turi, Ho [496]	-	-	counter current flow	-	-	-	-	-	-

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Usman, Dai [497]	1-Butyl-3-methylimidazolium Tricyanomethanide ([Bmim][TCM])	tubular membrane contractor: hybrid technology	counter current flow	-	-	4.47E-06	56	20	45
Wang, Li [498]	stacked graphene oxide (GO) membranes	-	counter current flow	-	-	-	-	-	-
Wang, Xie [499]	graphene oxide membranes	-	concurrent flow	-	-	7.06E-05	-	-	50
Wang, Ren [500]	NH <sub>2</sub> -UiO-66	mixed matrix membranes	-	DMF	-	1.69E-03	-	-	30 – 70
Xiang, Sheng [501]	amino-functionalized zeolitic imidazolate framework ZIF-7 (ZIF-7-NH <sub>2</sub> )	zeolite membranes	-	-	-	3.02E-06	-	-	50
Yoon, Kim [502]	1-Butyl-3-methylimidazolium tetrafluoroborate/zinc oxide	ionic liquid membranes	concurrent flow	-	-	-	-	-	-
Zainab, Iqbal [503]	polyamide 6/carbon nanotube (PA/CNT) nano-fibre/net	polymeric membranes	-	ethanol or isopropanol	-	-	200	-	-
Zhang, Zhang [504]	polyvinyl acetate (PVAc)	poly(ionic liquid) membranes	-	tetrahydrofuran (THF)	-	8.70E-06	-	-	-
Zhang, Wang [505]	Polyimide(PI)	hollow fibre membranes	-	-	-	1.36E-04	-	6667	12
Zhang, Tu [506]	four diamine-monocarboxylate-based protic	ionic liquids (pils)	-	N, N, N',N'-Tetramethyl-1,3-propanediamine	-	4.23E-05	-	-	-
Zhang, Wang [507]	montmorillonite (MMT) and Mg-Al hydrotalcite (HT)	-	-	sodium hydroxide (NaOH)	-	2.23E-06	-	-	-
Zhao, Ren [508]	Pebaxs membranes /CNTs/GTA	mixed matrix membranes (mmms)	concurrent flow	ethanol/water (70/30 wt%)	-	3.52E-04	-	-	-
Zhao, Sang [509]	d diacetylenic linkages with heteroatoms hydrogen, fluorine, and oxygen (GDY X, X = H, F, and O)	-	-	-	-	6.09E-05	-	-	50
Zhou, Tien [510]	graphene oxide	hollow fibre membranes	-	-	-	-	-	-	15

Authors	Membrane material	Reactor type	Flow configuration	Solvent (with molarity)	Wetting (+ = hydrophilic; - = hydrophobic)	Average flux (mol/ min cm <sup>2</sup> )	Gas flow rate (ml/min)	Liquid flow rate (ml/min)	Feed vol % CO <sub>2</sub>
Zhu, Swihart [511]	poly[2,20-(m-phenylene)-5,50-bisbenzimidazole] (PBI)	-	counter current flow	n N,N-dimethylacetamide (DMAc)	-	2.03E-07	-	-	50
Ahmad, Navarro [512]	MOF based Nanoparticles of CO <sub>2</sub> -philic UiO-66 (Zr-BDC)	metal organic structures	-	Poly(amic) PPA solution	-	3.08E-05	-	-	50
Baghban and Azar [513]		hollow fibre membrane	-	-	-	-	-	-	10-50
Constantinou, Barrass [514]	Nickel	flat membrane	concurrent	NaOH	+	3.00E-03	230	1.66	20
Fu, Li [515]	polyetheramine (PEA) and methyl-containing polyisophthalamides (MPA)	copolymer membranes	-	polyetheramine (PEA) and methyl-containing polyisophthalamides (MPA)	-	2.53E-04	-	-	10
Hu, Cheng [516]	polydimethylsiloxane (PDMS) membranes	-	-	-	-	7.44E-04	-	-	40
Hu, Cheng [517]	polyacrylonitrile polydimethylsiloxane (PDMS)	hollow fibre-supported membrane	-	3-aminopropyltriethoxysilane	-	9.81E-04	-	-	-
Ko [518]	-	hollow fibre membrane	counter current flow	-	-	-	-	-	10
Ovalle-Encinia, Pfeiffer [519]	Ce 0.85 Sm 0.15 O <sub>2</sub> -Sm 0.6S r 0.4 Al 0.3Fe 0.7O <sub>3</sub> composite	ceramic membranes	counter current flow	-	-	3.60E-10	60	60	15
Pang, Gong [520]	polyvinylidene fluoride (PVDF)	hollow fibre membrane	counter current flow	2-Pyrrolidone (NMP)	-	7.86E-06	130	70	19
Ramli, Hashim [521]	polypropylene (PP)	ionic liquids (ils) membranes	concurrent flow	1-Ethyl-3-methylimidazolium bis(trifluoromethylsulfon yl) imide [emim] [NTf <sub>2</sub> ]	-	-	-	-	-
Russo, Prpich [522]	polydimethylsiloxane (PDMS) membrane	-	counter current flow	-	-	1.06E-03	-	5000 - 10000	10
Wang, Kang [523]	polytetrafluoroethylene (PTFE)	hollow fibre membrane	counter current flow	-	-	-	1.75	0.2-0.8	20-50
Yu, Kanezashi [524]	amine-silica membranes	-	-	HCl and EtOH	-	4.50E-03	50	-	-
Zhang, Peng [525]	(SP-Zn <sup>2+</sup> )	-	-	anhydrous ethanol (C <sub>2</sub> H <sub>5</sub> OH)	-	1.69E-04	30	-	-

\* Flow in kmol/hr

## References

1. Niwa M, Ohya H, Tanaka Y, Yoshikawa N, Matsumoto K, Negishi Y. Separation of gaseous mixtures of CO<sub>2</sub> and CH<sub>4</sub> using a composite microporous glass membrane on ceramic tubing. *Journal of membrane science*, 1988, 39(3): 301-314.
2. Saha S Chakma A. Separation of CO<sub>2</sub> from gas mixtures with liquid membranes. *Energy Conversion and Management*, 1992, 33(5-8): 413-420.
3. Langevin D, Pinoche M, Se E, Me M, Roux R. CO<sub>2</sub> facilitated transport through functionalized cation-exchange membranes. *Journal of membrane science*, 1993, 82(1-2): 51-63.
4. Tokuda Y, Fujisawa E, Okabayashi N, Matsumiya N, Takagi K, Mano H, Haraya K, Sato M. Development of hollow fiber membranes for CO<sub>2</sub> separation. *Energy conversion and management*, 1997, 38: S111-S116.
5. Li K Teo W. Use of permeation and absorption methods for CO<sub>2</sub> removal in hollow fibre membrane modules. *Separation and Purification Technology*, 1998, 13(1): 79-88.
6. Suzuki H, Tanaka K, Kita H, Okamoto K, Hoshino H, Yoshinaga T, Kusuki Y. Preparation of composite hollow fiber membranes of poly (ethylene oxide)-containing polyimide and their CO<sub>2</sub>/N<sub>2</sub> separation properties. *Journal of Membrane Science*, 1998, 146(1): 31-37.
7. Chen H, Kovvali A, Sirkar K. Selective CO<sub>2</sub> Separation from CO<sub>2</sub>- N<sub>2</sub> Mixtures by Immobilized Glycine-Na- Glycerol Membranes. *Industrial & engineering chemistry research*, 2000, 39(7): 2447-2458.
8. Lee Y, Noble R D, Yeom B-Y, Park Y-I, Lee K-H. Analysis of CO<sub>2</sub> removal by hollow fiber membrane contactors. *Journal of Membrane Science*, 2001, 194(1): 57-67.
9. Mizukami K, Takaba H, Kobayashi Y, Oumi Y, Belosludov R V, Takami S, Kubo M, Miyamoto A. Molecular dynamics calculations of CO<sub>2</sub>/N<sub>2</sub> mixture through the NaY type zeolite membrane. *Journal of Membrane Science*, 2001, 188(1): 21-28.
10. Vu D Q, Koros W J, Miller S J. High pressure CO<sub>2</sub>/CH<sub>4</sub> separation using carbon molecular sieve hollow fiber membranes. *Industrial & engineering chemistry research*, 2002, 41(3): 367-380.
11. Xu L, Zhang L, Chen H. Study on CO<sub>2</sub> removal in air by hydrogel membranes. *Desalination*, 2002, 148(1-3): 309-313.
12. Zhang Y, Wang Z, Wang S. Synthesis and characteristics of novel fixed carrier membrane for CO<sub>2</sub> separation. *Chemistry letters*, 2002, 31(4): 430-431.
13. Kapantaidakis G, Koops G, Wessling M, Kaldis S, Sakellariopoulos G. CO<sub>2</sub> Plasticization of polyethersulfone/polyimide gas - separation membranes. *AIChE journal*, 2003, 49(7): 1702-1711.
14. Kim K-J, Park S-H, So W-W, Ahn D-J, Moon S-J. CO<sub>2</sub> separation performances of composite membranes of 6FDA-based polyimides with a polar group. *Journal of membrane science*, 2003, 211(1): 41-49.
15. Jordal K, Bredesen R, Kvamsdal H, Bolland O. Integration of H<sub>2</sub>-separating membrane technology in gas turbine processes for CO<sub>2</sub> capture. *Energy*, 2004, 29(9-10): 1269-1278.
16. Kim H Lee D. Simulation study for CO<sub>2</sub> separation process by using hollow fiber membrane. Ajou University, 2004.
17. Li S, Falconer J L, Noble R D. SAPO-34 membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of Membrane Science*, 2004, 241(1): 121-135.

18. Moon J-H, Ahn H, Hyun S-H, Lee C-H. Separation characteristics of tetrapropylammoniumbromide templating silica/alumina composite membrane in CO<sub>2</sub>/N<sub>2</sub>, CO<sub>2</sub>/H<sub>2</sub> and CH<sub>4</sub>/H<sub>2</sub> systems. *Korean Journal of Chemical Engineering*, 2004, 21(2): 477-487.
19. Teramoto M, Kitada S, Ohnishi N, Matsuyama H, Matsumiya N. Separation and concentration of CO<sub>2</sub> by capillary-type facilitated transport membrane module with permeation of carrier solution. *Journal of Membrane Science*, 2004, 234(1-2): 83-94.
20. Wang R, Li D, Liang D. Modeling of CO<sub>2</sub> capture by three typical amine solutions in hollow fiber membrane contactors. *Chemical Engineering and Processing: Process Intensification*, 2004, 43(7): 849-856.
21. Johannessen E, Jordal K. Study of a H<sub>2</sub> separating membrane reactor for methane steam reforming at conditions relevant for power processes with CO<sub>2</sub> capture. *Energy Conversion and Management*, 2005, 46(7-8): 1059-1071.
22. Jordal K, Bolland O, Möller B F, Torisson T. Optimization with genetic algorithms of a gas turbine cycle with H<sub>2</sub>-separating membrane reactor for CO<sub>2</sub> capture. *International journal of green energy*, 2005, 2(2): 167-180.
23. Li S, Alvarado G, Noble R D, Falconer J L. Effects of impurities on CO<sub>2</sub>/CH<sub>4</sub> separations through SAPO-34 membranes. *Journal of Membrane Science*, 2005, 251(1-2): 59-66.
24. Li S, Martinek J G, Falconer J L, Noble R D, Gardner T Q. High-pressure CO<sub>2</sub>/CH<sub>4</sub> separation using SAPO-34 membranes. *Industrial & engineering chemistry research*, 2005, 44(9): 3220-3228.
25. Liu L, Chakma A, Feng X. CO<sub>2</sub>/N<sub>2</sub> separation by poly (ether block amide) thin film hollow fiber composite membranes. *Industrial & engineering chemistry research*, 2005, 44(17): 6874-6882.
26. Qin J-J, Chung T-S, Cao C, Vora R. Effect of temperature on intrinsic permeation properties of 6FDA-Durene/1, 3-phenylenediamine (mPDA) copolyimide and fabrication of its hollow fiber membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of membrane science*, 2005, 250(1-2): 95-103.
27. Wang R, Zhang H, Feron P, Liang D. Influence of membrane wetting on CO<sub>2</sub> capture in microporous hollow fiber membrane contactors. *Separation and Purification Technology*, 2005, 46(1-2): 33-40.
28. Gong Y, Wang Z, Wang S. Experiments and simulation of CO<sub>2</sub> removal by mixed amines in a hollow fiber membrane module. *Chemical Engineering and Processing: Process Intensification*, 2006, 45(8): 652-660.
29. Ismail A, Yaacob N. Performance of treated and untreated asymmetric polysulfone hollow fiber membrane in series and cascade module configurations for CO<sub>2</sub>/CH<sub>4</sub> gas separation system. *Journal of membrane science*, 2006, 275(1-2): 151-165.
30. Okabe K, Nakamura M, Mano H, Teramoto M, Yamada K. Separation and recovery of CO<sub>2</sub> by membrane/absorption hybrid method. in *Proceedings of the Eighth International Conference on Greenhouse Gas Control Technologies*. 2006. Elsevier.
31. Zhang H-Y, Wang R, Liang D T, Tay J H. Modeling and experimental study of CO<sub>2</sub> absorption in a hollow fiber membrane contactor. *Journal of membrane science*, 2006, 279(1-2): 301-310.

32. Francisco G J, Chakma A, Feng X. Membranes comprising of alkanolamines incorporated into poly (vinyl alcohol) matrix for CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of Membrane Science*, 2007, 303(1-2): 54-63.
33. Himeno S, Tomita T, Suzuki K, Nakayama K, Yajima K, Yoshida S. Synthesis and permeation properties of a DDR-type zeolite membrane for separation of CO<sub>2</sub>/CH<sub>4</sub> gaseous mixtures. *Industrial & Engineering Chemistry Research*, 2007, 46(21): 6989-6997.
34. Kelman S, Lin H, Sanders E S, Freeman B D. CO<sub>2</sub>/C<sub>2</sub>H<sub>6</sub> separation using solubility selective membranes. *Journal of Membrane Science*, 2007, 305(1-2): 57-68.
35. Kwon W T, Kim S R, Kim E B, Bae S Y, Kim Y. H<sub>2</sub>/CO<sub>2</sub> Gas Separation Characteristic of Zeolite Membrane at High Temperature. in *Advanced Materials Research*. 2007. Trans Tech Publ.
36. Ohta Y, Takaba H, Nakao S-i. A combinatorial dynamic Monte Carlo approach to finding a suitable zeolite membrane structure for CO<sub>2</sub>/N<sub>2</sub> separation. *Microporous and mesoporous materials*, 2007, 101(1-2): 319-323.
37. Sakamoto Y, Nagata K, Yogo K, Yamada K. Preparation and CO<sub>2</sub> separation properties of amine-modified mesoporous silica membranes. *Microporous and Mesoporous Materials*, 2007, 101(1-2): 303-311.
38. Shim H M, Lee J S, Wang H Y, Choi S H, Kim J H, Kim H T. Modeling and economic analysis of CO<sub>2</sub> separation process with hollow fiber membrane modules. *Korean Journal of Chemical Engineering*, 2007, 24(3): 537-541.
39. Sridhar S, Suryamurali R, Smitha B, Aminabhavi T. Development of crosslinked poly (ether-block-amide) membrane for CO<sub>2</sub>/CH<sub>4</sub> separation. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 2007, 297(1-3): 267-274.
40. Wang B, Zhu D c, Zhan M c, Liu W, Chen C s. Combustion of coal - derived CO with membrane - supplied oxygen enabling CO<sub>2</sub> capture. *AIChE journal*, 2007, 53(9): 2481-2484.
41. Xiao S, Feng X, Huang R Y. Trimesoyl chloride crosslinked chitosan membranes for CO<sub>2</sub>/N<sub>2</sub> separation and pervaporation dehydration of isopropanol. *Journal of Membrane Science*, 2007, 306(1-2): 36-46.
42. Yan S-p, Fang M-X, Zhang W-F, Wang S-Y, Xu Z-K, Luo Z-Y, Cen K-F. Experimental study on the separation of CO<sub>2</sub> from flue gas using hollow fiber membrane contactors without wetting. *Fuel Processing Technology*, 2007, 88(5): 501-511.
43. Yegani R, Hirozawa H, Teramoto M, Himei H, Okada O, Takigawa T, Ohmura N, Matsumiya N, Matsuyama H. Selective separation of CO<sub>2</sub> by using novel facilitated transport membrane at elevated temperatures and pressures. *Journal of Membrane Science*, 2007, 291(1-2): 157-164.
44. Al-Marzouqi M, El-Naas M, Marzouk S, Abdullatif N. Modeling of chemical absorption of CO<sub>2</sub> in membrane contactors. *Separation and Purification Technology*, 2008, 62(3): 499-506.
45. Al-Marzouqi M H, El-Naas M H, Marzouk S A, Al-Zarooni M A, Abdullatif N, Faiz R. Modeling of CO<sub>2</sub> absorption in membrane contactors. *Separation and Purification Technology*, 2008, 59(3): 286-293.
46. Kai T, Kouketsu T, Duan S, Kazama S, Yamada K. Development of commercial-sized dendrimer composite membrane modules for CO<sub>2</sub> removal from flue gas. *Separation and Purification Technology*, 2008, 63(3): 524-530.

47. Keshavarz P, Fathikalajahi J, Ayatollahi S. Analysis of CO<sub>2</sub> separation and simulation of a partially wetted hollow fiber membrane contactor. *Journal of Hazardous Materials*, 2008, 152(3): 1237-1247.
48. Kosuri M R, Koros W J. Defect-free asymmetric hollow fiber membranes from Torlon®, a polyamide-imide polymer, for high-pressure CO<sub>2</sub> separations. *Journal of Membrane Science*, 2008, 320(1-2): 65-72.
49. Li S, Falconer J L, Noble R D. SAPO-34 membranes for CO<sub>2</sub>/CH<sub>4</sub> separations: effect of Si/Al ratio. *Microporous and Mesoporous Materials*, 2008, 110(2-3): 310-317.
50. Low B T, Xiao Y, Chung T S, Liu Y. Simultaneous occurrence of chemical grafting, cross-linking, and etching on the surface of polyimide membranes and their impact on H<sub>2</sub>/CO<sub>2</sub> separation. *Macromolecules*, 2008, 41(4): 1297-1309.
51. Modigell M, Schumacher M, Teplyakov V V, Zenkevich V B. A membrane contactor for efficient CO<sub>2</sub> removal in biohydrogen production. *Desalination*, 2008, 224(1-3): 186-190.
52. Paul S, Ghoshal A K, Mandal B. Theoretical studies on separation of CO<sub>2</sub> by single and blended aqueous alkanolamine solvents in flat sheet membrane contactor (FSMC). *Chemical Engineering Journal*, 2008, 144(3): 352-360.
53. Yan S, Fang M, Zhang W, Zhong W, Luo Z, Cen K. Comparative analysis of CO<sub>2</sub> separation from flue gas by membrane gas absorption technology and chemical absorption technology in China. *Energy Conversion and Management*, 2008, 49(11): 3188-3197.
54. Zhang H-Y, Wang R, Liang D T, Tay J H. Theoretical and experimental studies of membrane wetting in the membrane gas-liquid contacting process for CO<sub>2</sub> absorption. *Journal of Membrane Science*, 2008, 308(1-2): 162-170.
55. Zhao H-Y, Cao Y-M, Ding X-L, Zhou M-Q, Liu J-H, Yuan Q. Poly (ethylene oxide) induced cross-linking modification of Matrimid membranes for selective separation of CO<sub>2</sub>. *Journal of Membrane Science*, 2008, 320(1-2): 179-184.
56. Faiz R, Al-Marzouqi M. Mathematical modeling for the simultaneous absorption of CO<sub>2</sub> and H<sub>2</sub>S using MEA in hollow fiber membrane contactors. *Journal of Membrane Science*, 2009, 342(1-2): 269-278.
57. Ji P, Cao Y, Zhao H, Kang G, Jie X, Liu D, Liu J, Yuan Q. Preparation of hollow fiber poly (N, N-dimethylaminoethyl methacrylate)-poly (ethylene glycol methyl ether methyl acrylate)/polysulfone composite membranes for CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of membrane science*, 2009, 342(1-2): 190-197.
58. Kai T, Kazama S, Fujioka Y. Development of cesium-incorporated carbon membranes for CO<sub>2</sub> separation under humid conditions. *Journal of Membrane Science*, 2009, 342(1-2): 14-21.
59. Kosuri M R, Koros W J. Asymmetric hollow fiber membranes for separation of CO<sub>2</sub> from hydrocarbons and fluorocarbons at high-pressure conditions relevant to C<sub>2</sub>F<sub>4</sub> polymerization. *Industrial & Engineering Chemistry Research*, 2009, 48(23): 10577-10583.
60. Li Y, Rui Z, Xia C, Anderson M, Lin Y. Performance of ionic-conducting ceramic/carbonate composite material as solid oxide fuel cell electrolyte and CO<sub>2</sub> permeation membrane. *Catalysis Today*, 2009, 148(3-4): 303-309.
61. Lu J-G, Zheng Y-F, Cheng M-D. Membrane contactor for CO<sub>2</sub> absorption applying amino-acid salt solutions. *Desalination*, 2009, 249(2): 498-502.

62. Nistor C, Shishatskiy S, Popa M, Nunes S P. CO<sub>2</sub> selective membranes based on epoxy silane. *Rev. Roum. Chim*, 2009, 54: 603-610.
63. Piroonlerkgul P, Laosiripojana N, Adesina A, Assabumrungrat S. Performance of biogas-fed solid oxide fuel cell systems integrated with membrane module for CO<sub>2</sub> removal. *Chemical Engineering and Processing: Process Intensification*, 2009, 48(2): 672-682.
64. Rezvani S, Huang Y, McIlveen-Wright D, Hewitt N, Mondol J D. Comparative assessment of coal fired IGCC systems with CO<sub>2</sub> capture using physical absorption, membrane reactors and chemical looping. *Fuel*, 2009, 88(12): 2463-2472.
65. Safari M, Ghanizadeh A, Montazer-Rahmati M M. Optimization of membrane-based CO<sub>2</sub>-removal from natural gas using simple models considering both pressure and temperature effects. *International journal of greenhouse gas control*, 2009, 3(1): 3-10.
66. Sandru M, Kim T-J, Hägg M-B. High molecular fixed-site-carrier PVAm membrane for CO<sub>2</sub> capture. *Desalination*, 2009, 240(1-3): 298-300.
67. Simons K, Nijmeijer K, Wessling M. Gas-liquid membrane contactors for CO<sub>2</sub> removal. *Journal of Membrane Science*, 2009, 340(1-2): 214-220.
68. Watanabe T, Keskin S, Nair S, Sholl D S. Computational identification of a metal organic framework for high selectivity membrane-based CO<sub>2</sub>/CH<sub>4</sub> separations: Cu (hfpbb)(H<sub>2</sub> hfpbb) 0.5. *Physical Chemistry Chemical Physics*, 2009, 11(48): 11389-11394.
69. Xing R, Ho W W. Synthesis and characterization of crosslinked polyvinylalcohol/polyethyleneglycol blend membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of the Taiwan Institute of Chemical Engineers*, 2009, 40(6): 654-662.
70. Yang D, Wang Z, Wang J, Wang S. Potential of two-stage membrane system with recycle stream for CO<sub>2</sub> capture from postcombustion gas. *Energy & Fuels*, 2009, 23(10): 4755-4762.
71. Yave W, Car A, Funari S S, Nunes S P, Peinemann K-V. CO<sub>2</sub>-philic polymer membrane with extremely high separation performance. *Macromolecules*, 2009, 43(1): 326-333.
72. Ahmad F, Lau K K, Shariff A M. Modeling and Parametric Study for CO<sub>2</sub>/CH<sub>4</sub> Separation using Membrane Processes. *World Academy of Science, Engineering and Technology*, 2010.
73. Cong H, Yu B. Aminosilane cross-linked PEG/PEPEG/PPEPG membranes for CO<sub>2</sub>/N<sub>2</sub> and CO<sub>2</sub>/H<sub>2</sub> separation. *Industrial & Engineering Chemistry Research*, 2010, 49(19): 9363-9369.
74. Constantinou A, Gavrilidis A. CO<sub>2</sub> absorption in a microstructured mesh reactor. *Industrial & engineering chemistry research*, 2010, 49(3): 1041-1049.
75. El-Naas M H, Al-Marzouqi M, Marzouk S A, Abdullatif N. Evaluation of the removal of CO<sub>2</sub> using membrane contactors: membrane wettability. *Journal of Membrane Science*, 2010, 350(1-2): 410-416.
76. Franz J, Scherer V. An evaluation of CO<sub>2</sub> and H<sub>2</sub> selective polymeric membranes for CO<sub>2</sub> separation in IGCC processes. *Journal of Membrane Science*, 2010, 359(1-2): 173-183.
77. Hudiono Y C, Carlisle T K, Bara J E, Zhang Y, Gin D L, Noble R D. A three-component mixed-matrix membrane with enhanced CO<sub>2</sub> separation properties based on zeolites and ionic liquid materials. *Journal of Membrane Science*, 2010, 350(1-2): 117-123.
78. Kumar A, Yuan X, Sahu A K, Dewulf J, Ergas S J, Van Langenhove H. A hollow fiber membrane photo - bioreactor for CO<sub>2</sub> sequestration from combustion gas coupled with wastewater

treatment: a process engineering approach. *Journal of Chemical Technology & Biotechnology*, 2010, 85(3): 387-394.

79. Li S, Fan C Q. High-flux SAPO-34 membrane for CO<sub>2</sub>/N<sub>2</sub> separation. *Industrial & Engineering Chemistry Research*, 2010, 49(9): 4399-4404.
80. Li S, Carreon M A, Zhang Y, Funke H H, Noble R D, Falconer J L. Scale-up of SAPO-34 membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of Membrane Science*, 2010, 352(1-2): 7-13.
81. Li Y, Chung T-S. Molecular-level mixed matrix membranes comprising Pebax® and POSS for hydrogen purification via preferential CO<sub>2</sub> removal. *international journal of hydrogen energy*, 2010, 35(19): 10560-10568.
82. Liu Y, Hu E, Khan E A, Lai Z. Synthesis and characterization of ZIF-69 membranes and separation for CO<sub>2</sub>/CO mixture. *Journal of Membrane Science*, 2010, 353(1-2): 36-40.
83. Lu J-G, Ji Y, Zhang H, Chen M-D. CO<sub>2</sub> capture using activated amino acid salt solutions in a membrane contactor. *Separation Science and Technology*, 2010, 45(9): 1240-1251.
84. Mansourizadeh A, Ismail A. Effect of LiCl concentration in the polymer dope on the structure and performance of hydrophobic PVDF hollow fiber membranes for CO<sub>2</sub> absorption. *Chemical Engineering Journal*, 2010, 165(3): 980-988.
85. Mansourizadeh A, Ismail A, Abdullah M, Ng B. Preparation of polyvinylidene fluoride hollow fiber membranes for CO<sub>2</sub> absorption using phase-inversion promoter additives. *Journal of Membrane Science*, 2010, 355(1-2): 200-207.
86. Mansourizadeh A, Ismail A, Matsuura T. Effect of operating conditions on the physical and chemical CO<sub>2</sub> absorption through the PVDF hollow fiber membrane contactor. *Journal of Membrane Science*, 2010, 353(1-2): 192-200.
87. Marzouk S A, Al-Marzouqi M H, El-Naas M H, Abdullatif N, Ismail Z M. Removal of carbon dioxide from pressurized CO<sub>2</sub>-CH<sub>4</sub> gas mixture using hollow fiber membrane contactors. *Journal of Membrane Science*, 2010, 351(1-2): 21-27.
88. Park H B, Han S H, Jung C H, Lee Y M, Hill A J. Thermally rearranged (TR) polymer membranes for CO<sub>2</sub> separation. *Journal of Membrane Science*, 2010, 359(1-2): 11-24.
89. Reijerkerk S R, Knoef M H, Nijmeijer K, Wessling M. Poly (ethylene glycol) and poly (dimethyl siloxane): Combining their advantages into efficient CO<sub>2</sub> gas separation membranes. *Journal of membrane science*, 2010, 352(1-2): 126-135.
90. Sandru M, Haukebo S H, Hägg M-B. Composite hollow fiber membranes for CO<sub>2</sub> capture. *Journal of Membrane Science*, 2010, 346(1): 172-186.
91. Scholes C A, Smith K H, Kentish S E, Stevens G W. CO<sub>2</sub> capture from pre-combustion processes—Strategies for membrane gas separation. *International Journal of Greenhouse Gas Control*, 2010, 4(5): 739-755.
92. Simons K, Nijmeijer K, Mengers H, Brilman W, Wessling M. Highly selective amino acid salt solutions as absorption liquid for CO<sub>2</sub> capture in gas-liquid membrane contactors. *ChemSusChem*, 2010, 3(8): 939-947.
93. Tiscornia I, Kumakiri I, Bredesen R, Téllez C, Coronas J. Microporous titanosilicate ETS-10 membrane for high pressure CO<sub>2</sub> separation. *Separation and Purification Technology*, 2010, 73(1): 8-12.

94. Yave W, Car A, Wind J, Peinemann K-V. Nanometric thin film membranes manufactured on square meter scale: ultra-thin films for CO<sub>2</sub> capture. *Nanotechnology*, 2010, 21(39): 395301.
95. Yave W, Szymczyk A, Yave N, Roslaniec Z. Design, synthesis, characterization and optimization of PTT-b-PEO copolymers: A new membrane material for CO<sub>2</sub> separation. *Journal of Membrane Science*, 2010, 362(1-2): 407-416.
96. Yu X, Wang Z, Wei Z, Yuan S, Zhao J, Wang J, Wang S. Novel tertiary amino containing thin film composite membranes prepared by interfacial polymerization for CO<sub>2</sub> capture. *Journal of Membrane Science*, 2010, 362(1-2): 265-278.
97. Zhang L, Xiao Y, Chung T-S, Jiang J. Mechanistic understanding of CO<sub>2</sub>-induced plasticization of a polyimide membrane: A combination of experiment and simulation study. *Polymer*, 2010, 51(19): 4439-4447.
98. An W, Swenson P, Wu L, Waller T, Ku A, Kuznicki S M. Selective separation of hydrogen from C<sub>1</sub>/C<sub>2</sub> hydrocarbons and CO<sub>2</sub> through dense natural zeolite membranes. *Journal of membrane science*, 2011, 369(1-2): 414-419.
99. Banihashemi F, Pakizeh M, Ahmadpour A. CO<sub>2</sub> separation using PDMS/ZSM-5 zeolite composite membrane. *Separation and purification technology*, 2011, 79(3): 293-302.
100. Boributh S, Assabumrungrat S, Laosiripojana N, Jiratananon R. Effect of membrane module arrangement of gas-liquid membrane contacting process on CO<sub>2</sub> absorption performance: A modeling study. *Journal of membrane science*, 2011, 372(1-2): 75-86.
101. Chen C-C, Qiu W, Miller S J, Koros W J. Plasticization-resistant hollow fiber membranes for CO<sub>2</sub>/CH<sub>4</sub> separation based on a thermally crosslinkable polyimide. *Journal of membrane science*, 2011, 382(1-2): 212-221.
102. Chew T L, Ahmad A L, Bhatia S. Ba-SAPO-34 membrane synthesized from microwave heating and its performance for CO<sub>2</sub>/CH<sub>4</sub> gas separation. *Chemical engineering journal*, 2011, 171(3): 1053-1059.
103. Couling D J, Prakash K, Green W H. Analysis of membrane and adsorbent processes for warm syngas cleanup in integrated gasification combined-cycle power with CO<sub>2</sub> capture and sequestration. *Industrial & engineering chemistry research*, 2011, 50(19): 11313-11336.
104. Erucar I Keskin S. Screening metal-organic framework-based mixed-matrix membranes for CO<sub>2</sub>/CH<sub>4</sub> separations. *Industrial & Engineering Chemistry Research*, 2011, 50(22): 12606-12616.
105. Eslami S, Mousavi S M, Danesh S, Banazadeh H. Modeling and simulation of CO<sub>2</sub> removal from power plant flue gas by PG solution in a hollow fiber membrane contactor. *Advances in Engineering Software*, 2011, 42(8): 612-620.
106. Favre N Pierre A C. Synthesis and behaviour of hybrid polymer-silica membranes made by sol gel process with adsorbed carbonic anhydrase enzyme, in the capture of CO<sub>2</sub>. *Journal of sol-gel science and technology*, 2011, 60(2): 177.
107. Iarikov D, Hacırlıoğlu P, Oyama S. Supported room temperature ionic liquid membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Chemical Engineering Journal*, 2011, 166(1): 401-406.
108. Jin H G, Han S H, Lee Y M, Yeo Y K. Modeling and control of CO<sub>2</sub> separation process with hollow fiber membrane modules. *Korean Journal of Chemical Engineering*, 2011, 28(1): 41-48.
109. Jindaratamee P, Shimoyama Y, Ito A. Amine/glycol liquid membranes for CO<sub>2</sub> recovery from air. *Journal of membrane science*, 2011, 385: 171-176.

110. Khaisri S, deMontigny D, Tontiwachwuthikul P, Jiratananon R. CO<sub>2</sub> stripping from monoethanolamine using a membrane contactor. *Journal of Membrane Science*, 2011, 376(1-2): 110-118.
111. Khan A L, Li X, Vankelecom I F. Mixed-gas CO<sub>2</sub>/CH<sub>4</sub> and CO<sub>2</sub>/N<sub>2</sub> separation with sulfonated PEEK membranes. *Journal of membrane science*, 2011, 372(1-2): 87-96.
112. Khan A L, Li X, Vankelecom I F. SPEEK/Matrimid blend membranes for CO<sub>2</sub> separation. *Journal of membrane science*, 2011, 380(1-2): 55-62.
113. Kim D-H, Baek I-H, Hong S-U, Lee H-K. Study on immobilized liquid membrane using ionic liquid and PVDF hollow fiber as a support for CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of membrane science*, 2011, 372(1-2): 346-354.
114. Kim T-J, Uddin M W, Sandru M, Hägg M-B. The effect of contaminants on the composite membranes for CO<sub>2</sub> separation and challenges in up-scaling of the membranes. *Energy Procedia*, 2011, 4: 737-744.
115. Kumbharkar S, Liu Y, Li K. High performance polybenzimidazole based asymmetric hollow fibre membranes for H<sub>2</sub>/CO<sub>2</sub> separation. *Journal of membrane science*, 2011, 375(1-2): 231-240.
116. Kwisnek L, Heinz S, Wiggins J S, Nazarenko S. Multifunctional thiols as additives in UV-cured PEG-diacrylate membranes for CO<sub>2</sub> separation. *Journal of membrane science*, 2011, 369(1-2): 429-436.
117. Lee S H, Kim J N, Eom W H, Ko Y D, Hong S U, Baek I H. Development of water gas shift/membrane hybrid system for precombustion CO<sub>2</sub> capture in a coal gasification process. *Energy Procedia*, 2011, 4: 1139-1146.
118. Li P, Pramoda K, Chung T-S. CO<sub>2</sub> separation from flue gas using polyvinyl-(room temperature ionic liquid)-room temperature ionic liquid composite membranes. *Industrial & Engineering Chemistry Research*, 2011, 50(15): 9344-9353.
119. Li X, Remias J E, Neathery J K, Liu K. NF/RO faujasite zeolite membrane-ammonia absorption solvent hybrid system for potential post-combustion CO<sub>2</sub> capture application. *Journal of membrane science*, 2011, 366(1-2): 220-228.
120. Lotrič A, Sekavčnik M, Kunze C, Spliethoff H. Simulation of water-gas shift membrane reactor for integrated gasification combined cycle plant with CO<sub>2</sub> capture. *Strojniški vestnik-Journal of Mechanical Engineering*, 2011, 57(12): 911-926.
121. Mansourizadeh A, Ismail A. CO<sub>2</sub> stripping from water through porous PVDF hollow fiber membrane contactor. *Desalination*, 2011, 273(2-3): 386-390.
122. Mansourizadeh A, Ismail A. Preparation and characterization of porous PVDF hollow fiber membranes for CO<sub>2</sub> absorption: Effect of different non-solvent additives in the polymer dope. *International Journal of Greenhouse Gas Control*, 2011, 5(4): 640-648.
123. Martin F Z, Dijkstra J W, Boon J, Meuldijk J. A membrane reformer with permeate side combustion for CO<sub>2</sub> capture: modeling and design. *Energy Procedia*, 2011, 4: 707-714.
124. Nguyen P, Lasseguette E, Medina-Gonzalez Y, Remigy J, Roizard D, Favre E. A dense membrane contactor for intensified CO<sub>2</sub> gas/liquid absorption in post-combustion capture. *Journal of membrane science*, 2011, 377(1-2): 261-272.

125. Ostwal M, Singh R P, Dec S F, Lusk M T, Way J D. 3-Aminopropyltriethoxysilane functionalized inorganic membranes for high temperature CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of membrane science*, 2011, 369(1-2): 139-147.
126. Peters L, Hussain A, Follmann M, Melin T, Hägg M-B. CO<sub>2</sub> removal from natural gas by employing amine absorption and membrane technology—A technical and economical analysis. *Chemical Engineering Journal*, 2011, 172(2-3): 952-960.
127. Reijerkerk S R, Jordana R, Nijmeijer K, Wessling M. Highly hydrophilic, rubbery membranes for CO<sub>2</sub> capture and dehydration of flue gas. *International Journal of Greenhouse Gas Control*, 2011, 5(1): 26-36.
128. Reijerkerk S R, Wessling M, Nijmeijer K. Pushing the limits of block copolymer membranes for CO<sub>2</sub> separation. *Journal of membrane science*, 2011, 378(1-2): 479-484.
129. Sanaeepur H, Amooghin A E, Moghadassi A, Kargari A. Preparation and characterization of acrylonitrile–butadiene–styrene/poly (vinyl acetate) membrane for CO<sub>2</sub> removal. *Separation and purification technology*, 2011, 80(3): 499-508.
130. Sandström L, Sjöberg E, Hedlund J. Very high flux MFI membrane for CO<sub>2</sub> separation. *Journal of membrane science*, 2011, 380(1-2): 232-240.
131. Sohrabi M R, Marjani A, Moradi S, Davallo M, Shirazian S. Mathematical modeling and numerical simulation of CO<sub>2</sub> transport through hollow-fiber membranes. *Applied Mathematical Modelling*, 2011, 35(1): 174-188.
132. Spadaccini C M, Mukerjee E V, Letts S A, Maiti A, O'Brien K C. Ultrathin polymer membranes for high throughput CO<sub>2</sub> capture. *Energy Procedia*, 2011, 4: 731-736.
133. Tuinier M, Hamers H, van Sint Annaland M. Techno-economic evaluation of cryogenic CO<sub>2</sub> capture—A comparison with absorption and membrane technology. *International Journal of Greenhouse Gas Control*, 2011, 5(6): 1559-1565.
134. Venna S R, Carreon M A. Amino-functionalized SAPO-34 membranes for CO<sub>2</sub>/CH<sub>4</sub> and CO<sub>2</sub>/N<sub>2</sub> separation. *Langmuir*, 2011, 27(6): 2888-2894.
135. Wade J L, Lee C, West A C, Lackner K S. Composite electrolyte membranes for high temperature CO<sub>2</sub> separation. *Journal of membrane science*, 2011, 369(1-2): 20-29.
136. Xia J, Liu S, Chung T-S. Effect of end groups and grafting on the CO<sub>2</sub> separation performance of poly (ethylene glycol) based membranes. *Macromolecules*, 2011, 44(19): 7727-7736.
137. Xing R, Ho W W. Crosslinked polyvinylalcohol–polysiloxane/fumed silica mixed matrix membranes containing amines for CO<sub>2</sub>/H<sub>2</sub> separation. *Journal of Membrane Science*, 2011, 367(1-2): 91-102.
138. Ahmad F, Lau K K, Shariff A M, Murshid G. Process simulation and optimal design of membrane separation system for CO<sub>2</sub> capture from natural gas. *Computers & Chemical Engineering*, 2012, 36: 119-128.
139. Bengtson G, Neumann S, Filiz V. Optimization of PIM–membranes for Separation of CO<sub>2</sub>. *Procedia Engineering*, 2012, 44: 796-798.
140. Boributh S, Rongwong W, Assabumrungrat S, Laosiripojana N, Jiratananon R. Mathematical modeling and cascade design of hollow fiber membrane contactor for CO<sub>2</sub> absorption by monoethanolamine. *Journal of membrane science*, 2012, 401: 175-189.

141. Chabanon E, Roizard D, Favre E. Modelling Strategies of Membrane Contactor Processes for CO<sub>2</sub> POST-Combustion Capture: A Critical Reassessment. *Procedia Engineering*, 2012, 44: 343-346.
142. Constantinou A, Barrass S, Pronk F, Bril T, Wenn D, Shaw J, Gavriilidis A. CO<sub>2</sub> absorption in a high efficiency silicon nitride mesh contactor. *Chemical engineering journal*, 2012, 207: 766-771.
143. Hasan M F, Baliban R C, Elia J A, Floudas C A. Modeling, simulation, and optimization of postcombustion CO<sub>2</sub> capture for variable feed concentration and flow rate. 1. Chemical absorption and membrane processes. *Industrial & Engineering Chemistry Research*, 2012, 51(48): 15642-15664.
144. Ghasem N, Al-Marzouqi M, Zhu L. Preparation and properties of polyethersulfone hollow fiber membranes with o-xylene as an additive used in membrane contactors for CO<sub>2</sub> absorption. *Separation and purification technology*, 2012, 92: 1-10.
145. Han S H, Kwon H J, Kim K Y, Seong J G, Park C H, Kim S, Doherty C M, Thornton A W, Hill A J, Lozano A E. Tuning microcavities in thermally rearranged polymer membranes for CO<sub>2</sub> capture. *Physical chemistry chemical physics*, 2012, 14(13): 4365-4373.
146. Hwang H Y, Nam S Y, Koh H C, Ha S Y, Barbieri G, Drioli E. The effect of operating conditions on the performance of hollow fiber membrane modules for CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of Industrial and Engineering Chemistry*, 2012, 18(1): 205-211.
147. Khan A L, Klaysom C, Gahlaut A, Li X, Vankelecom I F. SPEEK and functionalized mesoporous MCM-41 mixed matrix membranes for CO<sub>2</sub> separations. *Journal of Materials Chemistry*, 2012, 22(37): 20057-20064.
148. Kim S, Lee Y M, Thermally rearranged (TR) polymer membranes with nanoengineered cavities tuned for CO<sub>2</sub> separation, in *Nanotechnology for Sustainable Development*. 2012, Springer. p. 265-275.
149. Lau C H, Paul D R, Chung T S. Molecular design of nanohybrid gas separation membranes for optimal CO<sub>2</sub> separation. *Polymer*, 2012, 53(2): 454-465.
150. Lee S H, Kim J N, Eom W H, Ryi S-K, Park J-S, Baek I H. Development of pilot WGS/multi-layer membrane for CO<sub>2</sub> capture. *Chemical engineering journal*, 2012, 207: 521-525.
151. Li H, Pieterse J, Dijkstra J, Boon J, Van Den Brink R, Jansen D. Bench-scale WGS membrane reactor for CO<sub>2</sub> capture with co-production of H<sub>2</sub>. *international journal of hydrogen energy*, 2012, 37(5): 4139-4143.
152. Li P, Paul D R, Chung T-S. High performance membranes based on ionic liquid polymers for CO<sub>2</sub> separation from the flue gas. *Green Chemistry*, 2012, 14(4): 1052-1063.
153. Lively R P, Dose M E, Xu L, Vaughn J T, Johnson J, Thompson J A, Zhang K, Lydon M E, Lee J-S, Liu L. A high-flux polyimide hollow fiber membrane to minimize footprint and energy penalty for CO<sub>2</sub> recovery from flue gas. *Journal of membrane science*, 2012, 423: 302-313.
154. Madhusoodana C, Patil M, Aminabhavi T. Ceramic Supported Composite Membranes of Hydroxy Ethyl Cellulose Loaded with AL-MCM-41 for CO<sub>2</sub> Separation. *Procedia Engineering*, 2012(44): 108-109.
155. Marzouk S A, Al-Marzouqi M H, Teramoto M, Abdullatif N, Ismail Z M. Simultaneous removal of CO<sub>2</sub> and H<sub>2</sub>S from pressurized CO<sub>2</sub>-H<sub>2</sub>S-CH<sub>4</sub> gas mixture using hollow fiber membrane contactors. *Separation and purification technology*, 2012, 86: 88-97.

156. Merkel T C, Wei X, He Z, White L S, Wijmans J, Baker R W. Selective exhaust gas recycle with membranes for CO<sub>2</sub> capture from natural gas combined cycle power plants. *Industrial & Engineering Chemistry Research*, 2012, 52(3): 1150-1159.
157. Modarresi S, Soltanieh M, Mousavi S A, Shabani I. Effect of low - frequency oxygen plasma on polysulfone membranes for CO<sub>2</sub>/CH<sub>4</sub> Separation. *Journal of Applied Polymer Science*, 2012, 124(S1): E199-E204.
158. Naim R, Ismail A, Mansourizadeh A. Effect of non-solvent additives on the structure and performance of PVDF hollow fiber membrane contactor for CO<sub>2</sub> stripping. *Journal of membrane science*, 2012, 423: 503-513.
159. Naim R, Ismail A, Mansourizadeh A. Preparation of microporous PVDF hollow fiber membrane contactors for CO<sub>2</sub> stripping from diethanolamine solution. *Journal of membrane science*, 2012, 392: 29-37.
160. Rahbari-Sisakht M, Ismail A, Matsuura T. Effect of bore fluid composition on structure and performance of asymmetric polysulfone hollow fiber membrane contactor for CO<sub>2</sub> absorption. *Separation and purification technology*, 2012, 88: 99-106.
161. Rahbari-Sisakht M, Ismail A, Rana D, Matsuura T. Effect of novel surface modifying macromolecules on morphology and performance of Polysulfone hollow fiber membrane contactor for CO<sub>2</sub> absorption. *Separation and purification technology*, 2012, 99: 61-68.
162. Rahbari-Sisakht M, Ismail A, Rana D, Matsuura T. A novel surface modified polyvinylidene fluoride hollow fiber membrane contactor for CO<sub>2</sub> absorption. *Journal of membrane science*, 2012, 415: 221-228.
163. Rongwong W, Boributh S, Assabumrungrat S, Laosiripojana N, Jiraratananon R. Simultaneous absorption of CO<sub>2</sub> and H<sub>2</sub>S from biogas by capillary membrane contactor. *Journal of membrane science*, 2012, 392: 38-47.
164. Scholes C A, Simioni M, Qader A, Stevens G W, Kentish S E. Membrane gas-solvent contactor trials of CO<sub>2</sub> absorption from syngas. *Chemical engineering journal*, 2012, 195: 188-197.
165. Shirazian S, Marjani A, Rezakazemi M. Separation of CO<sub>2</sub> by single and mixed aqueous amine solvents in membrane contactors: fluid flow and mass transfer modeling. *Engineering with Computers*, 2012, 28(2): 189-198.
166. Smart S, Vente J, Da Costa J D. High temperature H<sub>2</sub>/CO<sub>2</sub> separation using cobalt oxide silica membranes. *International Journal of Hydrogen Energy*, 2012, 37(17): 12700-12707.
167. Uddin M W Hägg M-B. Natural gas sweetening—the effect on CO<sub>2</sub>-CH<sub>4</sub> separation after exposing a facilitated transport membrane to hydrogen sulfide and higher hydrocarbons. *Journal of membrane science*, 2012, 423: 143-149.
168. Yilmaz G Keskin S. Predicting the performance of zeolite imidazolate framework/polymer mixed matrix membranes for CO<sub>2</sub>, CH<sub>4</sub>, and H<sub>2</sub> separations using molecular simulations. *Industrial & Engineering Chemistry Research*, 2012, 51(43): 14218-14228.
169. Zhang L, Hu Z, Jiang J. Metal-organic framework/polymer mixed-matrix membranes for H<sub>2</sub>/CO<sub>2</sub> separation: a fully atomistic simulation study. *The Journal of Physical Chemistry C*, 2012, 116(36): 19268-19277.
170. Zhao Y Ho W W. CO<sub>2</sub>-selective membranes containing sterically hindered amines for CO<sub>2</sub>/H<sub>2</sub> separation. *Industrial & Engineering Chemistry Research*, 2012, 52(26): 8774-8782.

171. Bae T-H, Long J R. CO<sub>2</sub>/N<sub>2</sub> separations with mixed-matrix membranes containing Mg<sub>2</sub>(dobdc) nanocrystals. *Energy & Environmental Science*, 2013, 6(12): 3565-3569.
172. Cao L, Tao K, Huang A, Kong C, Chen L. A highly permeable mixed matrix membrane containing CAU-1-NH<sub>2</sub> for H<sub>2</sub> and CO<sub>2</sub> separation. *Chemical Communications*, 2013, 49(76): 8513-8515.
173. Choi J H, Park M-J, Kim J, Ko Y, Lee S-H, Baek I. Modelling and analysis of pre-combustion CO<sub>2</sub> capture with membranes. *Korean Journal of Chemical Engineering*, 2013, 30(6): 1187-1194.
174. Duan S, Taniguchi I, Kai T, Kazama S. Development of poly (amidoamine) dendrimer/polyvinyl alcohol hybrid membranes for CO<sub>2</sub> capture at elevated pressures. *Energy Procedia*, 2013, 37: 924-931.
175. Ghadiri M, Marjani A, Shirazian S. Mathematical modeling and simulation of CO<sub>2</sub> stripping from monoethanolamine solution using nano porous membrane contactors. *International Journal of Greenhouse Gas Control*, 2013, 13: 1-8.
176. Ghasem N, Al-Marzouqi M, Rahim N A. Modeling of CO<sub>2</sub> absorption in a membrane contactor considering solvent evaporation. *Separation and Purification Technology*, 2013, 110: 1-10.
177. Hao L, Li P, Yang T, Chung T-S. Room temperature ionic liquid/ZIF-8 mixed-matrix membranes for natural gas sweetening and post-combustion CO<sub>2</sub> capture. *Journal of membrane science*, 2013, 436: 221-231.
178. Hassanlouei R N, Pelalak R, Daraei A. Wettability study in CO<sub>2</sub> capture from flue gas using nano porous membrane contactors. *International Journal of Greenhouse Gas Control*, 2013, 16: 233-240.
179. Hu T, Dong G, Li H, Chen V. Improved CO<sub>2</sub> separation performance with additives of PEG and PEG-PDMS copolymer in poly (2, 6-dimethyl-1, 4-phenylene oxide) membranes. *Journal of membrane science*, 2013, 432: 13-24.
180. Kai T, Taniguchi I, Duan S, Chowdhury F A, Saito T, Yamazaki K, Ikeda K, Ohara T, Asano S, Kazama S. Molecular gate membrane: Poly (amidoamine) dendrimer/polymer hybrid membrane modules for CO<sub>2</sub> capture. *Energy Procedia*, 2013, 37: 961-968.
181. Kanehashi S, Kishida M, Kidesaki T, Shindo R, Sato S, Miyakoshi T, Nagai K. CO<sub>2</sub> separation properties of a glassy aromatic polyimide composite membranes containing high-content 1-butyl-3-methylimidazolium bis (trifluoromethylsulfonyl) imide ionic liquid. *Journal of membrane science*, 2013, 430: 211-222.
182. Kangas J, Sandström L, Malinen I, Hedlund J, Tanskanen J. Maxwell-Stefan modeling of the separation of H<sub>2</sub> and CO<sub>2</sub> at high pressure in an MFI membrane. *Journal of membrane science*, 2013, 435: 186-206.
183. Khan A L, Klaysom C, Gahlaut A, Vankelecom I F. Polysulfone acrylate membranes containing functionalized mesoporous MCM-41 for CO<sub>2</sub> separation. *Journal of membrane science*, 2013, 436: 145-153.
184. Kim J, Abouelnasr M, Lin L-C, Smit B. Large-scale screening of zeolite structures for CO<sub>2</sub> membrane separations. *Journal of the American Chemical Society*, 2013, 135(20): 7545-7552.

185. Kim K, Ingole P G, Kim J, Lee H. Separation performance of PEBAX/PEI hollow fiber composite membrane for SO<sub>2</sub>/CO<sub>2</sub>/N<sub>2</sub> mixed gas. *Chemical Engineering Journal*, 2013, 233: 242-250.
186. Kim T-J, Vrålstad H, Sandru M, Hägg M-B. Separation performance of PVAm composite membrane for CO<sub>2</sub> capture at various pH levels. *Journal of membrane science*, 2013, 428: 218-224.
187. Koutsonikolas D E, Kaldis S P, Pantoleonos G T, Zaspalis V T, Sakellaropoulos G P. Techno-economic assessment of polymeric, ceramic and metallic membranes integration in an advanced IGCC process for H<sub>2</sub> production and CO<sub>2</sub> capture. *Trans*, 2013, 35: 715-720.
188. Lee C-B, Lee S-W, Park J-S, Lee D-W, Hwang K-R, Ryi S-K, Kim S-H. Long-term CO<sub>2</sub> capture tests of Pd-based composite membranes with module configuration. *International Journal of Hydrogen Energy*, 2013, 38(19): 7896-7903.
189. Li S, Wang Z, Zhang C, Wang M, Yuan F, Wang J, Wang S. Interfacially polymerized thin film composite membranes containing ethylene oxide groups for CO<sub>2</sub> separation. *Journal of membrane science*, 2013, 436: 121-131.
190. Lin Y F, Chen C H, Tung K L, Wei T Y, Lu S Y, Chang K S. Mesoporous Fluorocarbon - Modified Silica Aerogel Membranes Enabling Long - Term Continuous CO<sub>2</sub> Capture with Large Absorption Flux Enhancements. *ChemSusChem*, 2013, 6(3): 437-442.
191. Sainan L, Gongping L, Wang W, XIANGLI F, Wanqin J. Ceramic supported PDMS and PEGDA composite membranes for CO<sub>2</sub> separation. *Chinese Journal of Chemical Engineering*, 2013, 21(4): 348-356.
192. Mehdipour M, Karami M, Keshavarz P, Ayatollahi S. Analysis of CO<sub>2</sub> separation with aqueous potassium carbonate solution in a hollow fiber membrane contactor. *Energy & Fuels*, 2013, 27(4): 2185-2193.
193. Moghadassi A, Rajabi Z, Hosseini S, Mohammadi M. Preparation and characterization of polycarbonate-blend-raw/functionalized multi-walled carbon nano tubes mixed matrix membrane for CO<sub>2</sub> separation. *Separation Science and Technology*, 2013, 48(8): 1261-1271.
194. Mondal A Mandal B. Synthesis and characterization of crosslinked poly (vinyl alcohol)/poly (allylamine)/2-amino-2-hydroxymethyl-1, 3-propanediol/polysulfone composite membrane for CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of membrane science*, 2013, 446: 383-394.
195. Nagumo R, Iwata S, Mori H. Simulated Process Evaluation of Synthetic Natural Gas Production Based on Biomass Gasification and Potential of CO<sub>2</sub> Capture Using Membrane Separation Technology. 2013.
196. Naim R Ismail A. Effect of fiber packing density on physical CO<sub>2</sub> absorption performance in gas-liquid membrane contactor. *Separation and Purification Technology*, 2013, 115: 152-157.
197. Nasir R, Mukhtar H, Man Z, Mohshim D F. Synthesis, Characterization and Performance Study of Newly Developed Amine Polymeric Membrane (APM) for Carbon Dioxide (CO<sub>2</sub>) Removal. *World Academy of Science, Engineering and Technology, International Journal of Chemical, Molecular, Nuclear, Materials and Metallurgical Engineering*, 2013, 7(9): 670-673.
198. Peydayesh M, Asarehpour S, Mohammadi T, Bakhtiari O. Preparation and characterization of SAPO-34-Matrimid® 5218 mixed matrix membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Chemical Engineering Research and Design*, 2013, 91(7): 1335-1342.

199. Qiao Z, Wang Z, Zhang C, Yuan S, Zhu Y, Wang J, Wang S. PVAm-PIP/PS composite membrane with high performance for CO<sub>2</sub>/N<sub>2</sub> separation. *AIChE Journal*, 2013, 59(1): 215-228.
200. Rahbari-Sisakht M, Ismail A, Rana D, Matsuura T, Emadzadeh D. Effect of SMM concentration on morphology and performance of surface modified PVDF hollow fiber membrane contactor for CO<sub>2</sub> absorption. *Separation and Purification Technology*, 2013, 116: 67-72.
201. Rahman M M, Filiz V, Shishatskiy S, Abetz C, Neumann S, Bolmer S, Khan M M, Abetz V. PEBAX® with PEG functionalized POSS as nanocomposite membranes for CO<sub>2</sub> separation. *Journal of Membrane Science*, 2013, 437: 286-297.
202. Razavi S M R, Razavi S M J, Miri T, Shirazian S. CFD simulation of CO<sub>2</sub> capture from gas mixtures in nanoporous membranes by solution of 2-amino-2-methyl-1-propanol and piperazine. *International Journal of Greenhouse Gas Control*, 2013, 15: 142-149.
203. Ryi S-K, Lee C-B, Lee S-W, Park J-S. Pd-based composite membrane and its high-pressure module for pre-combustion CO<sub>2</sub> capture. *Energy*, 2013, 51: 237-242.
204. Sandru M, Kim T-J, Capala W, Huijbers M, Hägg M-B. Pilot scale testing of polymeric membranes for CO<sub>2</sub> capture from coal fired power plants. *Energy Procedia*, 2013, 37: 6473-6480.
205. Shao P, Dal-Cin M M, Guiver M D, Kumar A. Simulation of membrane-based CO<sub>2</sub> capture in a coal-fired power plant. *Journal of membrane science*, 2013, 427: 451-459.
206. Shen J-N, Yu C-C, Zeng G-N, Van der Bruggen B. Preparation of a facilitated transport membrane composed of carboxymethyl chitosan and polyethylenimine for CO<sub>2</sub>/N<sub>2</sub> separation. *International journal of molecular sciences*, 2013, 14(2): 3621-3638.
207. Stanislawski J, Holmes M, Snyder A, Tolbert S, Curran T. Advanced CO<sub>2</sub> separation technologies: coal gasification, warm-gas cleanup, and hydrogen separation membranes. *Energy Procedia*, 2013, 37: 2316-2326.
208. Tomé L C, Patinha D J, Freire C S, Rebelo L P N, Marrucho I M. CO<sub>2</sub> separation applying ionic liquid mixtures: the effect of mixing different anions on gas permeation through supported ionic liquid membranes. *Rsc Advances*, 2013, 3(30): 12220-12229.
209. Tseng H-H, Itta A K, Weng T-H, Li Y-L. SBA-15/CMS composite membrane for H<sub>2</sub> purification and CO<sub>2</sub> capture: Effect of pore size, pore volume, and loading weight on separation performance. *Microporous and Mesoporous Materials*, 2013, 180: 270-279.
210. Tziaila O, Veziri C, Papatryfon X, Beltsios K, Labropoulos A, Iliev B, Adamova G, Schubert T, Kroon M, Francisco M. Zeolite imidazolate framework-ionic liquid hybrid membranes for highly selective CO<sub>2</sub> separation. *The Journal of Physical Chemistry C*, 2013, 117(36): 18434-18440.
211. Wang L, Zhang Z, Zhao B, Zhang H, Lu X, Yang Q. Effect of long-term operation on the performance of polypropylene and polyvinylidene fluoride membrane contactors for CO<sub>2</sub> absorption. *Separation and Purification Technology*, 2013, 116: 300-306.
212. Wang M, Wang Z, Li S, Zhang C, Wang J, Wang S. A high performance antioxidative and acid resistant membrane prepared by interfacial polymerization for CO<sub>2</sub> separation from flue gas. *Energy & Environmental Science*, 2013, 6(2): 539-551.
213. Wang Z, Fang M, Pan Y, Yan S, Luo Z. Amine-based absorbents selection for CO<sub>2</sub> membrane vacuum regeneration technology by combined absorption-desorption analysis. *Chemical Engineering Science*, 2013, 93: 238-249.

214. Wang Z, Fang M, Yu H, Ma Q, Luo Z. Modeling of CO<sub>2</sub> stripping in a hollow fiber membrane contactor for CO<sub>2</sub> capture. *Energy & Fuels*, 2013, 27(11): 6887-6898.
215. Wang Z, Fang M, Yu H, Wei C-C, Luo Z. Experimental and modeling study of trace CO<sub>2</sub> Removal in a hollow-fiber membrane contactor, using CO<sub>2</sub>-loaded monoethanolamine. *Industrial & Engineering Chemistry Research*, 2013, 52(50): 18059-18070.
216. Yin X, Chu N, Yang J, Wang J, Li Z. Thin zeolite T/carbon composite membranes supported on the porous alumina tubes for CO<sub>2</sub> separation. *International Journal of Greenhouse Gas Control*, 2013, 15: 55-64.
217. Yoshimune M Haraya K. CO<sub>2</sub>/CH<sub>4</sub> mixed gas separation using carbon hollow fiber membranes. *Energy Procedia*, 2013, 37: 1109-1116.
218. Zhang K, Zou Y, Su C, Shao Z, Liu L, Wang S, Liu S. CO<sub>2</sub> and water vapor-tolerant yttria stabilized bismuth oxide (YSB) membranes with external short circuit for oxygen separation with CO<sub>2</sub> capture at intermediate temperatures. *Journal of Membrane Science*, 2013, 427: 168-175.
219. Zhu X, Chai S, Tian C, Fulvio P F, Han K S, Hagaman E W, Veith G M, Mahurin S M, Brown S, Liu H. Synthesis of Porous, Nitrogen - Doped Adsorption/Diffusion Carbonaceous Membranes for Efficient CO<sub>2</sub> Separation. *Macromolecular rapid communications*, 2013, 34(5): 452-459.
220. Ahmadpour E, Shamsabadi A A, Behbahani R M, Aghajani M, Kargari A. Study of CO<sub>2</sub> separation with PVC/Pebax composite membrane. *Journal of Natural Gas Science and Engineering*, 2014, 21: 518-523.
221. Amrei S M H H, Memardoost S, Dehkordi A M. Comprehensive modeling and CFD simulation of absorption of CO<sub>2</sub> and H<sub>2</sub>S by MEA solution in hollow fiber membrane reactors. *AIChE Journal*, 2014, 60(2): 657-672.
222. Chen H Z, Thong Z, Li P, Chung T-S. High performance composite hollow fiber membranes for CO<sub>2</sub>/H<sub>2</sub> and CO<sub>2</sub>/N<sub>2</sub> separation. *international journal of hydrogen energy*, 2014, 39(10): 5043-5053.
223. Constantinou A, Barrass S, Gavriilidis A. CO<sub>2</sub> absorption in polytetrafluoroethylene membrane microstructured contactor using aqueous solutions of amines. *Industrial & Engineering Chemistry Research*, 2014, 53(22): 9236-9242.
224. Deng L Hägg M-B. Carbon nanotube reinforced PVAm/PVA blend FSC nanocomposite membrane for CO<sub>2</sub>/CH<sub>4</sub> separation. *International Journal of Greenhouse Gas Control*, 2014, 26: 127-134.
225. Ghasem N, Al-Marsouqi M, Rahim N A. Modeling and Simulation of Membrane Contactor employed to strip CO<sub>2</sub> from Rich Solvents via COMSOL Multiphysics®.
226. He X, Kim T-J, Hägg M-B. Hybrid fixed-site-carrier membranes for CO<sub>2</sub> removal from high pressure natural gas: Membrane optimization and process condition investigation. *Journal of membrane science*, 2014, 470: 266-274.
227. Huang A, Chen Y, Liu Q, Wang N, Jiang J, Caro J. Synthesis of highly hydrophobic and permselective metal-organic framework Zn (BDC)(TED) 0.5 membranes for H<sub>2</sub>/CO<sub>2</sub> separation. *Journal of Membrane Science*, 2014, 454: 126-132.
228. Hussain A, Nasir H, Ahsan M. Process Design Analyses of CO<sub>2</sub> Capture from Natural Gas by Polymer Membrane. *Journal of The Chemical Society of Pakistan*, 2014, 36(3).

229. Junaidi M, Khoo C, Leo C, Ahmad A. The effects of solvents on the modification of SAPO-34 zeolite using 3-aminopropyl trimethoxy silane for the preparation of asymmetric polysulfone mixed matrix membrane in the application of CO<sub>2</sub> separation. *Microporous and Mesoporous Materials*, 2014, 192: 52-59.
230. Jusoh N, Lau K, Shariff A M, Yeong Y. Capture of bulk CO<sub>2</sub> from methane with the presence of heavy hydrocarbon using membrane process. *International Journal of Greenhouse Gas Control*, 2014, 22: 213-222.
231. Kimball E, Al-Azki A, Gomez A, Goetheer E, Booth N, Adams D, Ferre D. Hollow fiber membrane contactors for CO<sub>2</sub> capture: modeling and up-scaling to CO<sub>2</sub> capture for an 800 MWe coal power station. *Oil & Gas Science and Technology–Revue d’IFP Energies nouvelles*, 2014, 69(6): 1047-1058.
232. Kosinov N, Auffret C, Gücüyener C, Szyja B M, Gascon J, Kapteijn F, Hensen E J. High flux high-silica SSZ-13 membrane for CO<sub>2</sub> separation. *Journal of Materials Chemistry A*, 2014, 2(32): 13083-13092.
233. Kundu P K, Chakma A, Feng X. Effectiveness of membranes and hybrid membrane processes in comparison with absorption using amines for post-combustion CO<sub>2</sub> capture. *International Journal of Greenhouse Gas Control*, 2014, 28: 248-256.
234. Li S, Wang Z, He W, Zhang C, Wu H, Wang J, Wang S. Effects of minor SO<sub>2</sub> on the transport properties of fixed carrier membranes for CO<sub>2</sub> capture. *Industrial & Engineering Chemistry Research*, 2014, 53(18): 7758-7767.
235. Lin H, He Z, Sun Z, Vu J, Ng A, Mohammed M, Knief J, Merkel T C, Wu T, Lambrecht R C. CO<sub>2</sub>-selective membranes for hydrogen production and CO<sub>2</sub> capture–Part I: Membrane development. *Journal of membrane science*, 2014, 457: 149-161.
236. Lin Y-F, Ko C-C, Chen C-H, Tung K-L, Chang K-S, Chung T-W. Sol–gel preparation of polymethylsilsesquioxane aerogel membranes for CO<sub>2</sub> absorption fluxes in membrane contactors. *Applied energy*, 2014, 129: 25-31.
237. Lindqvist K, Roussanaly S, Anantharaman R. Multi-stage membrane processes for CO<sub>2</sub> capture from cement industry. *Energy Procedia*, 2014, 63: 6476-6483.
238. Lu J-G, Lu C-T, Chen Y, Gao L, Zhao X, Zhang H, Xu Z-W. CO<sub>2</sub> capture by membrane absorption coupling process: application of ionic liquids. *Applied energy*, 2014, 115: 573-581.
239. Ma C, Koros W J. Effects of hydrocarbon and water impurities on CO<sub>2</sub>/CH<sub>4</sub> separation performance of ester-crosslinked hollow fiber membranes. *Journal of membrane science*, 2014, 451: 1-9.
240. Ma Z, Qiao Z, Wang Z, Cao X, He Y, Wang J, Wang S. CO<sub>2</sub> separation enhancement of the membrane by modifying the polymer with a small molecule containing amine and ester groups. *Rsc Advances*, 2014, 4(41): 21313-21317.
241. Maghsoudi H, Soltanieh M. Simultaneous separation of H<sub>2</sub>S and CO<sub>2</sub> from CH<sub>4</sub> by a high silica CHA-type zeolite membrane. *Journal of membrane science*, 2014, 470: 159-165.
242. Makhouloufi C, Lasseguette E, Remigy J C, Belaisaoui B, Roizard D, Favre E. Ammonia based CO<sub>2</sub> capture process using hollow fiber membrane contactors. *Journal of membrane science*, 2014, 455: 236-246.

243. Mansourizadeh A, Aslmahdavi Z, Ismail A, Matsuura T. Blend polyvinylidene fluoride/surface modifying macromolecule hollow fiber membrane contactors for CO<sub>2</sub> absorption. *International Journal of Greenhouse Gas Control*, 2014, 26: 83-92.
244. Mansourizadeh A Pouranfard A-R. Microporous polyvinylidene fluoride hollow fiber membrane contactors for CO<sub>2</sub> stripping: effect of PEG-400 in spinning dope. *Chemical Engineering Research and Design*, 2014, 92(1): 181-190.
245. Masoumi S, Keshavarz P, Rastgoo Z. Theoretical investigation on CO<sub>2</sub> absorption into DEAB solution using hollow fiber membrane contactors. *Journal of Natural Gas Science and Engineering*, 2014, 18: 23-30.
246. Mohshim D F, Mukhtar H, Man Z. The effect of incorporating ionic liquid into polyethersulfone-SAPO34 based mixed matrix membrane on CO<sub>2</sub> gas separation performance. *Separation and Purification Technology*, 2014, 135: 252-258.
247. Mondal A Mandal B. CO<sub>2</sub> separation using thermally stable crosslinked poly (vinyl alcohol) membrane blended with polyvinylpyrrolidone/polyethyleneimine/tetraethylenepentamine. *Journal of membrane science*, 2014, 460: 126-138.
248. Mondal A Mandal B. Novel CO<sub>2</sub>-selective cross-linked poly (vinyl alcohol)/polyvinylpyrrolidone blend membrane containing amine carrier for CO<sub>2</sub>-N<sub>2</sub> separation: synthesis, characterization, and gas permeation study. *Industrial & Engineering Chemistry Research*, 2014, 53(51): 19736-19746.
249. Nabian N, Ghoreyshi A, Rahimpour A, Shakeri M. Effect of Polymer Concentration on the Structure and Performance of Polysulfone Flat Membrane for CO<sub>2</sub> Absorption in Membrane Contactor. *Iranian Journal of Chemical Engineering*, 2014, 11(2): 79.
250. Nafisi V Hägg M-B. Development of dual layer of ZIF-8/PEBAX-2533 mixed matrix membrane for CO<sub>2</sub> capture. *Journal of Membrane Science*, 2014, 459: 244-255.
251. Patel R, Kim S J, Roh D K, Kim J H. Synthesis of amphiphilic PCZ-r-PEG nanostructural copolymers and their use in CO<sub>2</sub>/N<sub>2</sub> separation membranes. *Chemical Engineering Journal*, 2014, 254: 46-53.
252. Pedram M Z, Omidkhan M, Amooghin A E. Synthesis and characterization of diethanolamine-impregnated cross-linked polyvinylalcohol/glutaraldehyde membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of Industrial and Engineering Chemistry*, 2014, 20(1): 74-82.
253. Rabiee H, Soltanieh M, Mousavi S A, Ghadimi A. Improvement in CO<sub>2</sub>/H<sub>2</sub> separation by fabrication of poly (ether-b-amide6)/glycerol triacetate gel membranes. *Journal of Membrane Science*, 2014, 469: 43-58.
254. Rahbari-Sisakht M, Rana D, Matsuura T, Emadzadeh D, Padaki M, Ismail A. Study on CO<sub>2</sub> stripping from water through novel surface modified PVDF hollow fiber membrane contactor. *Chemical Engineering Journal*, 2014, 246: 306-310.
255. Rahim N A, Ghasem N, Al-Marzouqi M. Stripping of CO<sub>2</sub> from different aqueous solvents using PVDF hollow fiber membrane contacting process. *Journal of Natural Gas Science and Engineering*, 2014, 21: 886-893.
256. Rezaei M, Ismail A, Hashemifard S, Bakeri G, Matsuura T. Experimental study on the performance and long-term stability of PVDF/montmorillonite hollow fiber mixed matrix membranes for CO<sub>2</sub> separation process. *International Journal of Greenhouse Gas Control*, 2014, 26: 147-157.

257. Rodenas T, van Dalen M, García - Pérez E, Serra - Crespo P, Zornoza B, Kapteijn F, Gascon J. Visualizing MOF mixed matrix membranes at the nanoscale: towards structure - performance relationships in CO<sub>2</sub>/CH<sub>4</sub> separation over NH<sub>2</sub> - MIL - 53 (Al)@ PI. *Advanced Functional Materials*, 2014, 24(2): 249-256.
258. Rodenas T, van Dalen M, Serra-Crespo P, Kapteijn F, Gascon J. Mixed matrix membranes based on NH<sub>2</sub>-functionalized MIL-type MOFs: Influence of structural and operational parameters on the CO<sub>2</sub>/CH<sub>4</sub> separation performance. *Microporous and Mesoporous Materials*, 2014, 192: 35-42.
259. Roh D K, Kim S J, Chi W S, Kim J K, Kim J H. Dual-functionalized mesoporous TiO<sub>2</sub> hollow nanospheres for improved CO<sub>2</sub> separation membranes. *Chemical Communications*, 2014, 50(43): 5717-5720.
260. Ryi S-K, Lee S-W, Park J-W, Oh D-K, Park J-S, Kim S S. Combined steam and CO<sub>2</sub> reforming of methane using catalytic nickel membrane for gas to liquid (GTL) process. *Catalysis Today*, 2014, 236: 49-56.
261. Salih A A, Yi C, Peng H, Yang B, Yin L, Wang W. Interfacially polymerized polyetheramine thin film composite membranes with PDMS inter-layer for CO<sub>2</sub> separation. *Journal of membrane science*, 2014, 472: 110-118.
262. Scholes C A, Ho M T, Aguiar A A, Wiley D E, Stevens G W, Kentish S E. Membrane gas separation processes for CO<sub>2</sub> capture from cement kiln flue gas. *International Journal of Greenhouse Gas Control*, 2014, 24: 78-86.
263. Scholes C A, Ribeiro C P, Kentish S E, Freeman B D. Thermal rearranged poly (benzoxazole)/polyimide blended membranes for CO<sub>2</sub> separation. *Separation and Purification Technology*, 2014, 124: 134-140.
264. Shi H. Synthesis of SAPO-34 zeolite membranes with the aid of crystal growth inhibitors for CO<sub>2</sub>-CH<sub>4</sub> separation. *New Journal of Chemistry*, 2014, 38(11): 5276-5278.
265. Taniguchi I, Fujikawa S. CO<sub>2</sub> separation with nano-thick polymeric membrane for pre-combustion. *Energy Procedia*, 2014, 63: 235-242.
266. Thompson J A, Vaughn J T, Brunelli N A, Koros W J, Jones C W, Nair S. Mixed-linker zeolitic imidazolate framework mixed-matrix membranes for aggressive CO<sub>2</sub> separation from natural gas. *Microporous and Mesoporous Materials*, 2014, 192: 43-51.
267. Tseng H-H, Chang S-H, Wey M-Y. A carbon gutter layer-modified  $\alpha$ -Al<sub>2</sub>O<sub>3</sub> substrate for PPO membrane fabrication and CO<sub>2</sub> separation. *Journal of Membrane Science*, 2014, 454: 51-61.
268. Wang L, Li Y, Li S, Ji P, Jiang C. Preparation of composite poly (ether block amide) membrane for CO<sub>2</sub> capture. *Journal of Energy Chemistry*, 2014, 23(6): 717-725.
269. Wang S, Liu Y, Huang S, Wu H, Li Y, Tian Z, Jiang Z. Pebax-PEG-MWCNT hybrid membranes with enhanced CO<sub>2</sub> capture properties. *Journal of membrane science*, 2014, 460: 62-70.
270. Wang Z, Fang M, Ma Q, Zhao Z, Wang T, Luo Z. Membrane stripping technology for CO<sub>2</sub> desorption from CO<sub>2</sub>-rich absorbents with low energy consumption. *Energy Procedia*, 2014, 63: 765-772.
271. Wickramanayake S, Hopkinson D, Myers C, Hong L, Feng J, Seol Y, Plasynski D, Zeh M, Luebke D. Mechanically robust hollow fiber supported ionic liquid membranes for CO<sub>2</sub> separation applications. *Journal of membrane science*, 2014, 470: 52-59.

272. Wu D, Maurin G, Yang Q, Serre C, Jobic H, Zhong C. Computational exploration of a Zr-carboxylate based metal-organic framework as a membrane material for CO<sub>2</sub> capture. *Journal of Materials Chemistry A*, 2014, 2(6): 1657-1661.
273. Wu T, Wang B, Lu Z, Zhou R, Chen X. Alumina-supported AlPO-18 membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of membrane science*, 2014, 471: 338-346.
274. Xin Q, Wu H, Jiang Z, Li Y, Wang S, Li Q, Li X, Lu X, Cao X, Yang J. SPEEK/amine-functionalized TiO<sub>2</sub> submicrospheres mixed matrix membranes for CO<sub>2</sub> separation. *Journal of membrane science*, 2014, 467: 23-35.
275. Xiong L, Gu S, Jensen K O, Yan Y S. Facilitated Transport in Hydroxide - Exchange Membranes for Post - Combustion CO<sub>2</sub> Separation. *ChemSusChem*, 2014, 7(1): 114-116.
276. Yan S, He Q, Zhao S, Wang Y, Ai P. Biogas upgrading by CO<sub>2</sub> removal with a highly selective natural amino acid salt in gas-liquid membrane contactor. *Chemical Engineering and Processing: Process Intensification*, 2014, 85: 125-135.
277. Yan Y, Zhang Z, Zhang L, Wang J, Li J, Ju S. Modeling of CO<sub>2</sub> separation from flue gas by methyl-diethanolamine and 2-(1-piperazinyl)-ethylamine in membrane contactors: effect of gas and liquid parameters. *Journal of Energy Engineering*, 2014, 141(4): 04014034.
278. Yin H, Wang J, Xie Z, Yang J, Bai J, Lu J, Zhang Y, Yin D, Lin J Y. A highly permeable and selective amino-functionalized MOF CAU-1 membrane for CO<sub>2</sub>-N<sub>2</sub> separation. *Chemical Communications*, 2014, 50(28): 3699-3701.
279. Zaidiza D A, Billaud J, Belaisaoui B, Rode S, Roizard D, Favre E. Modeling of CO<sub>2</sub> post-combustion capture using membrane contactors, comparison between one-and two-dimensional approaches. *Journal of membrane science*, 2014, 455: 64-74.
280. Zhang Y Wang R. Novel method for incorporating hydrophobic silica nanoparticles on polyetherimide hollow fiber membranes for CO<sub>2</sub> absorption in a gas-liquid membrane contactor. *Journal of membrane science*, 2014, 452: 379-389.
281. Zhang L, Gong Y, Brinkman K S, Wei T, Wang S, Huang K. Flux of silver-carbonate membranes for post-combustion CO<sub>2</sub> capture: The effects of membrane thickness, gas concentration and time. *Journal of membrane science*, 2014, 455: 162-167.
282. Zhang L, Gong Y, Yaggie J, Wang S, Romito K, Huang K. Surface modified silver-carbonate mixed conducting membranes for high flux CO<sub>2</sub> separation with enhanced stability. *Journal of membrane science*, 2014, 453: 36-41.
283. Zhang X, Seames W S, Tande B M. Recovery of CO<sub>2</sub> from Monoethanolamine using a Membrane Contactor. *Separation Science and Technology*, 2014, 49(1): 1-11.
284. Zhang Z, Yan Y, Zhang L, Chen Y, Ju S. CFD investigation of CO<sub>2</sub> capture by methyl-diethanolamine and 2-(1-piperazinyl)-ethylamine in membranes: Part B. Effect of membrane properties. *Journal of Natural Gas Science and Engineering*, 2014, 19: 311-316.
285. Zhang Z, Yan Y, Zhang L, Ju S. Numerical simulation and analysis of CO<sub>2</sub> removal in a polypropylene hollow fiber membrane contactor. *International Journal of Chemical Engineering*, 2014, 2014.
286. Zhao D, Ren J, Li H, Hua K, Deng M. Poly (amide-6-b-ethylene oxide)/SAPO-34 mixed matrix membrane for CO<sub>2</sub> separation. *Journal of Energy Chemistry*, 2014, 23(2): 227-234.

287. Zhao Y, Jung B T, Ansaloni L, Ho W W. Multiwalled carbon nanotube mixed matrix membranes containing amines for high pressure CO<sub>2</sub>/H<sub>2</sub> separation. *Journal of membrane science*, 2014, 459: 233-243.
288. Zhou J, Tran M-M, Haldeman A T, Jin J, Wagener E H, Husson S M. Perfluorocyclobutyl polymer thin-film composite membranes for CO<sub>2</sub> separations. *Journal of membrane science*, 2014, 450: 478-486.
289. Zhou M, Korelskiy D, Ye P, Grahn M, Hedlund J. A uniformly oriented MFI membrane for improved CO<sub>2</sub> separation. *Angewandte Chemie International Edition*, 2014, 53(13): 3492-3495.
290. Azizi M Mousavi S A. CO<sub>2</sub>/H<sub>2</sub> separation using a highly permeable polyurethane membrane: Molecular dynamics simulation. *Journal of Molecular Structure*, 2015, 1100: 401-414.
291. Carapellucci R, Giordano L, Vaccarelli M. Study of a natural gas combined cycle with multi-stage membrane systems for CO<sub>2</sub> post-combustion capture. *Energy Procedia*, 2015, 81: 412-421.
292. Dong L, Sun Y, Zhang C, Han D, Bai Y, Chen M. Efficient CO<sub>2</sub> capture by metallo-supramolecular polymers as fillers to fabricate a polymeric blend membrane. *RSC Advances*, 2015, 5(83): 67658-67661.
293. Farjami M, Moghadassi A, Vatanpour V. Modeling and simulation of CO<sub>2</sub> removal in a polyvinylidene fluoride hollow fiber membrane contactor with computational fluid dynamics. *Chemical Engineering and Processing: Process Intensification*, 2015, 98: 41-51.
294. Fazaeli R, Razavi S M R, Najafabadi M S, Torkaman R, Hemmati A. Computational simulation of CO<sub>2</sub> removal from gas mixtures by chemical absorbents in porous membranes. *RSC Advances*, 2015, 5(46): 36787-36797.
295. Gilassi S Rahmanian N. Mathematical modelling and numerical simulation of CO<sub>2</sub>/CH<sub>4</sub> separation in a polymeric membrane. *Applied Mathematical Modelling*, 2015, 39(21): 6599-6611.
296. Goyal N, Suman S, Gupta S. Mathematical modeling of CO<sub>2</sub> separation from gaseous-mixture using a Hollow-Fiber Membrane Module: Physical mechanism and influence of partial-wetting. *Journal of membrane science*, 2015, 474: 64-82.
297. Hwang S, Chi W S, Lee S J, Im S H, Kim J H, Kim J. Hollow ZIF-8 nanoparticles improve the permeability of mixed matrix membranes for CO<sub>2</sub>/CH<sub>4</sub> gas separation. *Journal of membrane science*, 2015, 480: 11-19.
298. Kammakakam I, Nam S, Kim T-H. Ionic group-mediated crosslinked polyimide membranes for enhanced CO<sub>2</sub> separation. *RSC Advances*, 2015, 5(86): 69907-69914.
299. Khalilinejad I, Sanaeepur H, Kargari A. Preparation of poly (ether-6-block amide)/PVC thin film composite membrane for CO<sub>2</sub> separation: effect of top layer thickness and operating parameters. *Journal of Membrane Science and Research*, 2015, 1(3): 124-129.
300. Kim S J, Jeon H, Kim D J, Kim J H. High - performance Polymer Membranes with Multi - functional Amphiphilic Micelles for CO<sub>2</sub> Capture. *ChemSusChem*, 2015, 8(22): 3783-3792.
301. Konruang S, Sirijarukul S, Wanichapichart P, Yu L, Chittrakarn T. Ultraviolet - ray treatment of polysulfone membranes on the O<sub>2</sub>/N<sub>2</sub> and CO<sub>2</sub>/CH<sub>4</sub> separation performance. *Journal of Applied Polymer Science*, 2015, 132(25).
302. Lee H J, Magnone E, Park J H. Preparation, characterization and laboratory-scale application of modified hydrophobic aluminum oxide hollow fiber membrane for CO<sub>2</sub> capture using H<sub>2</sub>O as low-cost absorbent. *Journal of membrane science*, 2015, 494: 143-153.

303. Lee S, Choi J-W, Lee S-H. Separation of greenhouse gases (SF<sub>6</sub>, CF<sub>4</sub> and CO<sub>2</sub>) in an industrial flue gas using pilot-scale membrane. *Separation and Purification Technology*, 2015, 148: 15-24.
304. Li P, Wang Z, Li W, Liu Y, Wang J, Wang S. High-performance multilayer composite membranes with mussel-inspired polydopamine as a versatile molecular bridge for CO<sub>2</sub> separation. *ACS applied materials & interfaces*, 2015, 7(28): 15481-15493.
305. Li P, Wang Z, Liu Y, Zhao S, Wang J, Wang S. A synergistic strategy via the combination of multiple functional groups into membranes towards superior CO<sub>2</sub> separation performances. *Journal of membrane science*, 2015, 476: 243-255.
306. Li X, Cheng Y, Zhang H, Wang S, Jiang Z, Guo R, Wu H. Efficient CO<sub>2</sub> capture by functionalized graphene oxide nanosheets as fillers to fabricate multi-permselective mixed matrix membranes. *ACS applied materials & interfaces*, 2015, 7(9): 5528-5537.
307. Li X, Jiang Z, Wu Y, Zhang H, Cheng Y, Guo R, Wu H. High-performance composite membranes incorporated with carboxylic acid nanogels for CO<sub>2</sub> separation. *Journal of Membrane Science*, 2015, 495: 72-80.
308. Li X, Ma L, Zhang H, Wang S, Jiang Z, Guo R, Wu H, Cao X, Yang J, Wang B. Synergistic effect of combining carbon nanotubes and graphene oxide in mixed matrix membranes for efficient CO<sub>2</sub> separation. *Journal of membrane science*, 2015, 479: 1-10.
309. Li Y, Li X, Wu H, Xin Q, Wang S, Liu Y, Tian Z, Zhou T, Jiang Z, Tian H. Anionic surfactant-doped Pebax membrane with optimal free volume characteristics for efficient CO<sub>2</sub> separation. *Journal of Membrane Science*, 2015, 493: 460-469.
310. Li Y, Xin Q, Wang S, Tian Z, Wu H, Liu Y, Jiang Z. Trapping bound water within a polymer electrolyte membrane of calcium phosphotungstate for efficient CO<sub>2</sub> capture. *Chemical Communications*, 2015, 51(10): 1901-1904.
311. Liao J, Wang Z, Gao C, Wang M, Yan K, Xie X, Zhao S, Wang J, Wang S. A high performance PVAm-HT membrane containing high-speed facilitated transport channels for CO<sub>2</sub> separation. *Journal of Materials Chemistry A*, 2015, 3(32): 16746-16761.
312. Lin R, Ge L, Liu S, Rudolph V, Zhu Z. Mixed-matrix membranes with metal-organic framework-decorated CNT fillers for efficient CO<sub>2</sub> separation. *ACS applied materials & interfaces*, 2015, 7(27): 14750-14757.
313. Lin Y-F, Chang J-M, Ye Q, Tung K-L. Hydrophobic fluorocarbon-modified silica aerogel tubular membranes with excellent CO<sub>2</sub> recovery ability in membrane contactors. *Applied energy*, 2015, 154: 21-25.
314. Lock S S M, Lau K K, Ahmad F, Shariff A. Modeling, simulation and economic analysis of CO<sub>2</sub> capture from natural gas using cocurrent, countercurrent and radial crossflow hollow fiber membrane. *International Journal of Greenhouse Gas Control*, 2015, 36: 114-134.
315. Loloie M, Omidkhah M, Moghadassi A, Amooghin A E. Preparation and characterization of Matrimid® 5218 based binary and ternary mixed matrix membranes for CO<sub>2</sub> separation. *International Journal of Greenhouse Gas Control*, 2015, 39: 225-235.
316. Mahmoudi A, Asghari M, Zargar V. CO<sub>2</sub>/CH<sub>4</sub> separation through a novel commercializable three-phase PEBA/PEG/NaX nanocomposite membrane. *Journal of Industrial and Engineering Chemistry*, 2015, 23: 238-242.

317. Mondal A, Barooah M, Mandal B. Effect of single and blended amine carriers on CO<sub>2</sub> separation from CO<sub>2</sub>/N<sub>2</sub> mixtures using crosslinked thin-film poly (vinyl alcohol) composite membrane. *International Journal of Greenhouse Gas Control*, 2015, 39: 27-38.
318. Mulukutla T, Chau J, Singh D, Obuskovic G, Sirkar K K. Novel membrane contactor for CO<sub>2</sub> removal from flue gas by temperature swing absorption. *Journal of Membrane Science*, 2015, 493: 321-328.
319. Nabian N, Ghoreyshi A A, Rahimpour A, Shakeri M. Performance evaluation and mass transfer study of CO<sub>2</sub> absorption in flat sheet membrane contactor using novel porous polysulfone membrane. *Korean Journal of Chemical Engineering*, 2015, 32(11): 2204-2211.
320. Nasir R, Mukhtar H, Man Z, Shaharun M S, Bakar M A. Development and Performance Prediction of Polyethersulfone-Carbon Molecular Sieve Mixed Matrix Membrane for CO<sub>2</sub>/CH<sub>4</sub> Separation. *Chemical Engineering Transactions*, 2015, 45: 1417-1422.
321. Nasir R, Mukhtar H, Man Z, Shaharun M S, Bakar M Z A. Effect of fixed carbon molecular sieve (CMS) loading and various di-ethanolamine (DEA) concentrations on the performance of a mixed matrix membrane for CO<sub>2</sub>/CH<sub>4</sub> separation. *RSC Advances*, 2015, 5(75): 60814-60822.
322. Nordin N A H M, Racha S M, Matsuura T, Misdan N, Sani N A A, Ismail A F, Mustafa A. Facile modification of ZIF-8 mixed matrix membrane for CO<sub>2</sub>/CH<sub>4</sub> separation: synthesis and preparation. *RSC Advances*, 2015, 5(54): 43110-43120.
323. Nwogu N C, Kajama M N, Osueke G, Gobina E. High performance valuation of CO<sub>2</sub> gas separation ceramic membrane system. 2015.
324. Park C H, Lee J H, Jung J P, Jung B, Kim J H. A highly selective PEGBEM-g-POEM comb copolymer membrane for CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of membrane science*, 2015, 492: 452-460.
325. Park S, Lee A S, Do Y S, Hwang S S, Lee Y M, Lee J-H, Lee J S. Rational molecular design of PEOlated ladder-structured polysilsesquioxane membranes for high performance CO<sub>2</sub> removal. *Chemical Communications*, 2015, 51(83): 15308-15311.
326. Qiao Z, Wang Z, Yuan S, Wang J, Wang S. Preparation and characterization of small molecular amine modified PVAm membranes for CO<sub>2</sub>/H<sub>2</sub> separation. *Journal of Membrane Science*, 2015, 475: 290-302.
327. Rabiee H, Alsadat S M, Soltanieh M, Mousavi S A, Ghadimi A. Gas permeation and sorption properties of poly (amide-12-b-ethyleneoxide)(Pebax1074)/SAPO-34 mixed matrix membrane for CO<sub>2</sub>/CH<sub>4</sub> and CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of Industrial and Engineering Chemistry*, 2015, 27: 223-239.
328. Rahim N A, Ghasem N, Al-Marzouqi M. Absorption of CO<sub>2</sub> from natural gas using different amino acid salt solutions and regeneration using hollow fiber membrane contactors. *Journal of Natural Gas Science and Engineering*, 2015, 26: 108-117.
329. Rezaei M, Ismail A, Bakeri G, Hashemifard S, Matsuura T. Effect of general montmorillonite and Cloisite 15A on structural parameters and performance of mixed matrix membranes contactor for CO<sub>2</sub> absorption. *Chemical engineering journal*, 2015, 260: 875-885.
330. Sadoogh M, Mansourizadeh A, Mohammadinik H. An experimental study on the stability of PVDF hollow fiber membrane contactors for CO<sub>2</sub> absorption with alkanolamine solutions. *RSC Advances*, 2015, 5(105): 86031-86040.

331. Scofield J M, Gurr P A, Kim J, Fu Q, Halim A, Kentish S E, Qiao G G. High - performance thin film composite membranes with well - defined poly (dimethylsiloxane) - b - poly (ethylene glycol) copolymer additives for CO<sub>2</sub> separation. *Journal of Polymer Science Part A: Polymer Chemistry*, 2015, 53(12): 1500-1511.
332. Seoane B, Coronas J, Gascon I, Benavides M E, Karvan O, Caro J, Kapteijn F, Gascon J. Metal-organic framework based mixed matrix membranes: a solution for highly efficient CO<sub>2</sub> capture? *Chemical Society Reviews*, 2015, 44(8): 2421-2454.
333. Shen J, Liu G, Huang K, Jin W, Lee K R, Xu N. Membranes with fast and selective gas - transport channels of laminar graphene oxide for efficient CO<sub>2</sub> capture. *Angewandte Chemie*, 2015, 127(2): 588-592.
334. Shin D-Y, Hwang K-R, Park J-S, Park M-J. Computational fluid dynamics modeling and analysis of Pd-based membrane module for CO<sub>2</sub> capture from H<sub>2</sub>/CO<sub>2</sub> binary gas mixture. *Korean Journal of Chemical Engineering*, 2015, 32(7): 1414-1421.
335. Skorek-Osikowska A, Bartela Ł, Kotowicz J. Thermodynamic and economic evaluation of a CO<sub>2</sub> membrane separation unit integrated into a supercritical coal-fired heat and power plant. *Journal of Power Technologies*, 2015, 95(3): 201-210.
336. Sorribas S, Comesaña-Gándara B, Lozano A E, Zornoza B, Téllez C, Coronas J. Insight into ETS-10 synthesis for the preparation of mixed matrix membranes for CO<sub>2</sub>/CH<sub>4</sub> gas separation. *RSC Advances*, 2015, 5(124): 102392-102398.
337. Sun C, Wen B, Bai B. Application of nanoporous graphene membranes in natural gas processing: Molecular simulations of CH<sub>4</sub>/CO<sub>2</sub>, CH<sub>4</sub>/H<sub>2</sub>S and CH<sub>4</sub>/N<sub>2</sub> separation. *Chemical Engineering Science*, 2015, 138: 616-621.
338. Taniguchi I, Kai T, Duan S, Kazama S, Jinnai H. A compatible crosslinker for enhancement of CO<sub>2</sub> capture of poly (amidoamine) dendrimer-containing polymeric membranes. *Journal of Membrane Science*, 2015, 475: 175-183.
339. Tong J, Zhang L, Fang J, Han M, Huang K. Electrochemical capture of CO<sub>2</sub> from natural gas using a high-temperature ceramic-carbonate membrane. *Journal of The Electrochemical Society*, 2015, 162(4): E43-E46.
340. Vakharia V, Ramasubramanian K, Ho W W. An experimental and modeling study of CO<sub>2</sub>-selective membranes for IGCC syngas purification. *Journal of membrane science*, 2015, 488: 56-66.
341. Wang B, Sun C, Li Y, Zhao L, Ho W W, Dutta P K. Rapid synthesis of faujasite/polyethersulfone composite membrane and application for CO<sub>2</sub>/N<sub>2</sub> separation. *Microporous and Mesoporous Materials*, 2015, 208: 72-82.
342. Wang N, Mundstock A, Liu Y, Huang A, Caro J. Amine-modified Mg-MOF-74/CPO-27-Mg membrane with enhanced H<sub>2</sub>/CO<sub>2</sub> separation. *Chemical Engineering Science*, 2015, 124: 27-36.
343. Wang S, Tian Z, Feng J, Wu H, Li Y, Liu Y, Li X, Xin Q, Jiang Z. Enhanced CO<sub>2</sub> separation properties by incorporating poly (ethylene glycol)-containing polymeric submicrospheres into polyimide membrane. *Journal of membrane science*, 2015, 473: 310-317.
344. Xin Q, Gao Y, Wu X, Li C, Liu T, Shi Y, Li Y, Jiang Z, Wu H, Cao X. Incorporating one-dimensional aminated titania nanotubes into sulfonated poly (ether ether ketone) membrane to construct CO<sub>2</sub>-facilitated transport pathways for enhanced CO<sub>2</sub> separation. *Journal of membrane science*, 2015, 488: 13-29.

345. Xin Q, Li Z, Li C, Wang S, Jiang Z, Wu H, Zhang Y, Yang J, Cao X. Enhancing the CO<sub>2</sub> separation performance of composite membranes by the incorporation of amino acid-functionalized graphene oxide. *Journal of Materials Chemistry A*, 2015, 3(12): 6629-6641.
346. Xing W, Peters T, Fontaine M-L, Evans A, Henriksen P P, Norby T, Bredesen R. Steam-promoted CO<sub>2</sub> flux in dual-phase CO<sub>2</sub> separation membranes. *Journal of membrane science*, 2015, 482: 115-119.
347. Zaidiza D A, Belaisaoui B, Rode S, Neveux T, Makhloufi C, Castel C, Roizard D, Favre E. Adiabatic modelling of CO<sub>2</sub> capture by amine solvents using membrane contactors. *Journal of Membrane Science*, 2015, 493: 106-119.
348. Zhang L, Qu R, Sha Y, Wang X, Yang L. Membrane gas absorption for CO<sub>2</sub> capture from flue gas containing fine particles and gaseous contaminants. *International Journal of Greenhouse Gas Control*, 2015, 33: 10-17.
349. Zhang L, Qu Z-Y, Yan Y-F, Ju S-X, Zhang Z-E. Numerical investigation of the effects of polypropylene hollow fibre membrane structure on the performance of CO<sub>2</sub> removal from flue gas. *RSC Advances*, 2015, 5(1): 424-433.
350. Zheng Y, Hu N, Wang H, Bu N, Zhang F, Zhou R. Preparation of steam-stable high-silica CHA (SSZ-13) membranes for CO<sub>2</sub>/CH<sub>4</sub> and C<sub>2</sub>H<sub>4</sub>/C<sub>2</sub>H<sub>6</sub> separation. *Journal of Membrane Science*, 2015, 475: 303-310.
351. Zhou R, Wang H, Wang B, Chen X, Li S, Yu M. Defect-patching of zeolite membranes by surface modification using siloxane polymers for CO<sub>2</sub> separation. *Industrial & Engineering Chemistry Research*, 2015, 54(30): 7516-7523.
352. Zhou T, Luo L, Hu S, Wang S, Zhang R, Wu H, Jiang Z, Wang B, Yang J. Janus composite nanoparticle-incorporated mixed matrix membranes for CO<sub>2</sub> separation. *Journal of Membrane Science*, 2015, 489: 1-10.
353. Adewole J K, Ahmad A L. Process modeling and optimization studies of high pressure membrane separation of CO<sub>2</sub> from natural gas. *Korean Journal of Chemical Engineering*, 2016, 33(10): 2998-3010.
354. Amooghini A E, Omidkhah M, Sanaeepur H, Kargari A. Preparation and characterization of Ag<sup>+</sup> ion-exchanged zeolite-Matrimid® 5218 mixed matrix membrane for CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of Energy Chemistry*, 2016, 25(3): 450-462.
355. Arias A M, Mussati M C, Mores P L, Scenna N J, Caballero J A, Mussati S F. Optimization of multi-stage membrane systems for CO<sub>2</sub> capture from flue gas. *International Journal of Greenhouse Gas Control*, 2016, 53: 371-390.
356. Hosseinzadeh Beiragh H, Omidkhah M, Abedini R, Khosravi T, Pakseresht S. Synthesis and characterization of poly (ether - block - amide) mixed matrix membranes incorporated by nanoporous ZSM - 5 particles for CO<sub>2</sub>/CH<sub>4</sub> separation. *Asia - Pacific Journal of Chemical Engineering*, 2016, 11(4): 522-532.
357. Chang J, Kang S W. CO<sub>2</sub> separation through poly (vinylidene fluoride-co-hexafluoropropylene) membrane by selective ion channel formed by tetrafluoroboric acid. *Chemical Engineering Journal*, 2016, 306: 1189-1192.
358. Chen Y, Ho W W. High-molecular-weight polyvinylamine/piperazine glycinate membranes for CO<sub>2</sub> capture from flue gas. *Journal of Membrane Science*, 2016, 514: 376-384.

359. Dai Z, Bai L, Hval K N, Zhang X, Zhang S, Deng L. Pebax®/TSIL blend thin film composite membranes for CO<sub>2</sub> separation. *Science China Chemistry*, 2016, 59(5): 538-546.
360. Dong G, Hou J, Wang J, Zhang Y, Chen V, Liu J. Enhanced CO<sub>2</sub>/N<sub>2</sub> separation by porous reduced graphene oxide/Pebax mixed matrix membranes. *Journal of Membrane Science*, 2016, 520: 860-868.
361. Dong G, Zhang Y, Hou J, Shen J, Chen V. Graphene oxide nanosheets based novel facilitated transport membranes for efficient CO<sub>2</sub> capture. *Industrial & Engineering Chemistry Research*, 2016, 55(18): 5403-5414.
362. Dong L, Zhang C, Bai Y, Shi D, Li X, Zhang H, Chen M. High-performance PEBA2533-functional MMT mixed matrix membrane containing high-speed facilitated transport channels for CO<sub>2</sub>/N<sub>2</sub> separation. *ACS Sustainable Chemistry & Engineering*, 2016, 4(6): 3486-3496.
363. Dong X, Liu Q, Huang A. Highly permselective MIL - 68 (Al)/matrimid mixed matrix membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of Applied Polymer Science*, 2016, 133(22).
364. Ebrahimi S, Mollaiy-Berneti S, Asadi H, Peydayesh M, Akhlaghian F, Mohammadi T. PVA/PES-amine-functional graphene oxide mixed matrix membranes for CO<sub>2</sub>/CH<sub>4</sub> separation: Experimental and modeling. *Chemical Engineering Research and Design*, 2016, 109: 647-656.
365. Fan T, Xie W, Ji X, Liu C, Feng X, Lu X. CO<sub>2</sub>/N<sub>2</sub> separation using supported ionic liquid membranes with green and cost-effective [Choline][Pro]/PEG200 mixtures. *Chinese journal of chemical engineering*, 2016, 24(11): 1513-1521.
366. Gilassi S, Rahmanian N. CFD modelling of a hollow fibre membrane for CO<sub>2</sub> removal by aqueous amine solutions of MEA, DEA and MDEA. *International Journal of Chemical Reactor Engineering*, 2016, 14(1): 53-61.
367. Jeon H, Kim D J, Park M S, Ryu D Y, Kim J H. Amphiphilic graft copolymer nanospheres: From colloidal self-assembly to CO<sub>2</sub> capture membranes. *ACS applied materials & interfaces*, 2016, 8(14): 9454-9461.
368. Jomekian A, Behbahani R M, Mohammadi T, Kargari A. CO<sub>2</sub>/CH<sub>4</sub> separation by high performance co-casted ZIF-8/Pebax 1657/PES mixed matrix membrane. *Journal of Natural Gas Science and Engineering*, 2016, 31: 562-574.
369. Jusoh N, Lau K K, Yeong Y F, Shariff A M. Bulk CO<sub>2</sub>/CH<sub>4</sub> Separation for Offshore Operating Conditions using Membrane Process. *Sains Malaysiana*, 2016, 45(11): 1707-1714.
370. Kang Z, Peng Y, Qian Y, Yuan D, Addicoat M A, Heine T, Hu Z, Tee L, Guo Z, Zhao D. Mixed matrix membranes (MMMs) comprising exfoliated 2D covalent organic frameworks (COFs) for efficient CO<sub>2</sub> separation. *Chemistry of Materials*, 2016, 28(5): 1277-1285.
371. Karamouz F, Maghsoudi H, Yegani R. Synthesis and characterization of high permeable PEBA membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of Natural Gas Science and Engineering*, 2016, 35: 980-985.
372. Karimi S, Korelskiy D, Mortazavi Y, Khodadadi A A, Sardari K, Esmaeili M, Antzutkin O N, Shah F U, Hedlund J. High flux acetate functionalized silica membranes based on in-situ co-condensation for CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of Membrane Science*, 2016, 520: 574-582.
373. Kertik A, Khan A L, Vankelecom I F. Mixed matrix membranes prepared from non-dried MOFs for CO<sub>2</sub>/CH<sub>4</sub> separations. *RSC Advances*, 2016, 6(115): 114505-114512.

374. Kim J, Choi J, Soo Kang Y, Won J. Matrix effect of mixed - matrix membrane containing CO<sub>2</sub> - selective MOF s. *Journal of Applied Polymer Science*, 2016, 133(1).
375. Kim J, Fu Q, Scofield J M, Kentish S E, Qiao G G. Ultra-thin film composite mixed matrix membranes incorporating iron (III)-dopamine nanoparticles for CO<sub>2</sub> separation. *Nanoscale*, 2016, 8(15): 8312-8323.
376. Kim J, Fu Q, Xie K, Scofield J M, Kentish S E, Qiao G G. CO<sub>2</sub> separation using surface-functionalized SiO<sub>2</sub> nanoparticles incorporated ultra-thin film composite mixed matrix membranes for post-combustion carbon capture. *Journal of membrane science*, 2016, 515: 54-62.
377. Kim S-J, Park A, Nam S-E, Park Y-I, Lee P S. Practical designs of membrane contactors and their performances in CO<sub>2</sub>/CH<sub>4</sub> separation. *Chemical Engineering Science*, 2016, 155: 239-247.
378. Kim S J, Chi W S, Jeon H, Kim J H, Patel R. Spontaneously self-assembled dual-layer mixed matrix membranes containing mass-produced mesoporous TiO<sub>2</sub> for CO<sub>2</sub> capture. *Journal of Membrane Science*, 2016, 508: 62-72.
379. Koolivand H, Sharif A, Chehrazi E, Kashani M R, Paran S M R. Mixed-matrix membranes comprising graphene-oxide nanosheets for CO<sub>2</sub>/CH<sub>4</sub> separation: A comparison between glassy and rubbery polymer matrices. *Polymer Science, Series A*, 2016, 58(5): 801-809.
380. Korelskiy D, Grahn M, Ye P, Zhou M, Hedlund J. A study of CO<sub>2</sub>/CO separation by sub-micron b-oriented MFI membranes. 2016.
381. Krea M, Roizard D, Favre E. Copoly (alkyl ether imide) membranes as promising candidates for CO<sub>2</sub> capture applications. *Separation and Purification Technology*, 2016, 161: 53-60.
382. Lee J H, Jung J P, Jang E, Lee K B, Hwang Y J, Min B K, Kim J H. PEDOT-PSS embedded comb copolymer membranes with improved CO<sub>2</sub> capture. *Journal of Membrane Science*, 2016, 518: 21-30.
383. Li W, Zheng X, Dong Z, Li C, Wang W, Yan Y, Zhang J. Molecular dynamics simulations of CO<sub>2</sub>/N<sub>2</sub> separation through two-dimensional graphene oxide membranes. *The Journal of Physical Chemistry C*, 2016, 120(45): 26061-26066.
384. Li W, Zhang Y, Su P, Xu Z, Zhang G, Shen C, Meng Q. Metal-organic framework channelled graphene composite membranes for H<sub>2</sub>/CO<sub>2</sub> separation. *Journal of Materials Chemistry A*, 2016, 4(48): 18747-18752.
385. Liao J, Wang Z, Wang M, Gao C, Zhao S, Wang J, Wang S. Adjusting carrier microenvironment in CO<sub>2</sub> separation fixed carrier membrane. *Journal of membrane science*, 2016, 511: 9-19.
386. Lin Y-F Kuo J-W. Mesoporous bis (trimethoxysilyl) hexane (BTMSH)/tetraethyl orthosilicate (TEOS)-based hybrid silica aerogel membranes for CO<sub>2</sub> capture. *Chemical Engineering Journal*, 2016, 300: 29-35.
387. Liu Z, Liu C, Li L, Qin W, Xu A. CO<sub>2</sub> separation by supported ionic liquid membranes and prediction of separation performance. *International Journal of Greenhouse Gas Control*, 2016, 53: 79-84.
388. Lu S-C, Khan A L, Vankelecom I F. Polysulfone-ionic liquid based membranes for CO<sub>2</sub>/N<sub>2</sub> separation with tunable porous surface features. *Journal of Membrane Science*, 2016, 518: 10-20.
389. Moradi M R, Chenar M P, Noie S H. Using PDMS coated TFC-RO membranes for CO<sub>2</sub>/N<sub>2</sub> gas separation: Experimental study, modeling and optimization. *Polymer Testing*, 2016, 56: 287-298.

390. Mosleh S, Mozdianfard M, Hemmati M, Khanbabaei G. Synthesis and characterization of rubbery/glassy blend membranes for CO<sub>2</sub>/CH<sub>4</sub> gas separation. *Journal of Polymer Research*, 2016, 23(6): 120.
391. Mubashir M, Yeong Y F, Lau K K. Ultrasonic-assisted secondary growth of deca-dodecyl 3 rhombohedral (DD3R) membrane and its process optimization studies in CO<sub>2</sub>/CH<sub>4</sub> separation using response surface methodology. *Journal of Natural Gas Science and Engineering*, 2016, 30: 50-63.
392. Nematollahi M H, Dehaghani A H S, Abedini R. CO<sub>2</sub>/CH<sub>4</sub> separation with poly (4-methyl-1-pentene)(TPX) based mixed matrix membrane filled with Al<sub>2</sub>O<sub>3</sub> nanoparticles. *Korean Journal of Chemical Engineering*, 2016, 33(2): 657-665.
393. Nematollahi M H, Dehaghani A H S, Pirouzfard V, Akhondi E. Mixed matrix membranes comprising PMP polymer with dispersed alumina nanoparticle fillers to separate CO<sub>2</sub>/N<sub>2</sub>. *Macromolecular Research*, 2016, 24(9): 782-792.
394. Nguyen T H, Gong H, Lee S S, Bae T H. Amine - Appended Hierarchical Ca - A Zeolite for Enhancing CO<sub>2</sub>/CH<sub>4</sub> Selectivity of Mixed - Matrix Membranes. *ChemPhysChem*, 2016, 17(20): 3165-3169.
395. Otani A, Zhang Y, Matsuki T, Kamio E, Matsuyama H, Maginn E J. Molecular design of high CO<sub>2</sub> reactivity and low viscosity ionic liquids for CO<sub>2</sub> separative facilitated transport membranes. *Industrial & Engineering Chemistry Research*, 2016, 55(10): 2821-2830.
396. Pohlmann J, Bram M, Wilkner K, Brinkmann T. Pilot scale separation of CO<sub>2</sub> from power plant flue gases by membrane technology. *International Journal of Greenhouse Gas Control*, 2016, 53: 56-64.
397. Qin Y, Lv J, Fu X, Guo R, Li X, Zhang J, Wei Z. High-performance SPEEK/amino acid salt membranes for CO<sub>2</sub> separation. *RSC Advances*, 2016, 6(3): 2252-2258.
398. Rafiq S, Deng L, Hägg M B. Role of facilitated transport membranes and composite membranes for efficient CO<sub>2</sub> capture—a review. *ChemBioEng Reviews*, 2016, 3(2): 68-85.
399. Rahmani M, Kazemi A, Talebnia F. Matrimid mixed matrix membranes for enhanced CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of Polymer Engineering*, 2016, 36(5): 499-511.
400. Razavi S M R, Shirazian S, Nazemian M. Numerical simulation of CO<sub>2</sub> separation from gas mixtures in membrane modules: Effect of chemical absorbent. *Arabian Journal of Chemistry*, 2016, 9(1): 62-71.
401. Rui Z, James J B, Kasik A, Lin Y. Metal - organic framework membrane process for high purity CO<sub>2</sub> production. *AIChE Journal*, 2016, 62(11): 3836-3841.
402. Saedi S, Seidi F, Moradi F, Xiang X. Preparation and characterization of an amino - cellulose (AC) derivative for development of thin - film composite membrane for CO<sub>2</sub>/CH<sub>4</sub> separation. *Starch - Stärke*, 2016, 68(7-8): 651-661.
403. Saeed M, Deng L. Carbon nanotube enhanced PVA-mimic enzyme membrane for post-combustion CO<sub>2</sub> capture. *International Journal of Greenhouse Gas Control*, 2016, 53: 254-262.
404. Sanaeepur H, Kargari A, Nasernejad B, Amooghini A E, Omidkhan M. A novel CO<sub>2</sub>-exchanged zeolite Y/cellulose acetate mixed matrix membrane for CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of the Taiwan Institute of Chemical Engineers*, 2016, 60: 403-413.

405. Sánchez-Laínez J, Zornoza B, Friebe S, Caro J, Cao S, Sabetghadam A, Seoane B, Gascon J, Kapteijn F, Le Guillouzer C. Influence of ZIF-8 particle size in the performance of polybenzimidazole mixed matrix membranes for pre-combustion CO<sub>2</sub> capture and its validation through interlaboratory test. *Journal of membrane science*, 2016, 515: 45-53.
406. Sánchez-Laínez J, Zornoza B, Téllez C, Coronas J. On the chemical filler–polymer interaction of nano-and micro-sized ZIF-11 in PBI mixed matrix membranes and their application for H<sub>2</sub>/CO<sub>2</sub> separation. *Journal of Materials Chemistry A*, 2016, 4(37): 14334-14341.
407. Scofield J M, Gurr P A, Kim J, Fu Q, Kentish S E, Qiao G G. Development of novel fluorinated additives for high performance CO<sub>2</sub> separation thin-film composite membranes. *Journal of Membrane Science*, 2016, 499: 191-200.
408. Shen J, Liu G, Huang K, Li Q, Guan K, Li Y, Jin W. UiO-66-polyether block amide mixed matrix membranes for CO<sub>2</sub> separation. *Journal of membrane science*, 2016, 513: 155-165.
409. Shen J, Zhang M, Liu G, Guan K, Jin W. Size effects of graphene oxide on mixed matrix membranes for CO<sub>2</sub> separation. *AIChE Journal*, 2016, 62(8): 2843-2852.
410. Shen Y, Wang H, Zhang X, Zhang Y. MoS<sub>2</sub> nanosheets functionalized composite mixed matrix membrane for enhanced CO<sub>2</sub> capture via surface drop-coating method. *ACS applied materials & interfaces*, 2016, 8(35): 23371-23378.
411. Solimando X, Lherbier C, Babin J, Arnal - Herault C, Romero E, Acherar S, Jamart - Gregoire B, Barth D, Roizard D, Jonquieres A. Pseudopeptide bioconjugate additives for CO<sub>2</sub> separation membranes. *Polymer International*, 2016, 65(12): 1464-1473.
412. Sumer Z, Keskin S. Computational Screening of MOF-Based Mixed Matrix Membranes for CO<sub>2</sub>/N<sub>2</sub> Separations. *Journal of Nanomaterials*, 2016, 2016.
413. Sun C, Srivastava D J, Grandinetti P J, Dutta P K. Synthesis of chabazite/polymer composite membrane for CO<sub>2</sub>/N<sub>2</sub> separation. *Microporous and Mesoporous Materials*, 2016, 230: 208-216.
414. Waheed N, Mushtaq A, Tabassum S, Gilani M A, Ilyas A, Ashraf F, Jamal Y, Bilad M R, Khan A U, Khan A L. Mixed matrix membranes based on polysulfone and rice husk extracted silica for CO<sub>2</sub> separation. *Separation and Purification Technology*, 2016, 170: 122-129.
415. Wang S, Li X, Wu H, Tian Z, Xin Q, He G, Peng D, Chen S, Yin Y, Jiang Z. Advances in high permeability polymer-based membrane materials for CO<sub>2</sub> separations. *Energy & Environmental Science*, 2016, 9(6): 1863-1890.
416. Wang Y, Yang Q, Li J, Yang J, Zhong C. Exploration of nanoporous graphene membranes for the separation of N<sub>2</sub> from CO<sub>2</sub>: a multi-scale computational study. *Physical Chemistry Chemical Physics*, 2016, 18(12): 8352-8358.
417. Wong K, Goh P, Ismail A. Thin film nanocomposite: the next generation selective membrane for CO<sub>2</sub> removal. *Journal of Materials Chemistry A*, 2016, 4(41): 15726-15748.
418. Woo K T, Dong G, Lee J, Kim J S, Do Y S, Lee W H, Lee H S, Lee Y M. Ternary mixed-gas separation for flue gas CO<sub>2</sub> capture using high performance thermally rearranged (TR) hollow fiber membranes. *Journal of Membrane Science*, 2016, 510: 472-480.
419. Wu D, Zhao L, Vakharia V K, Salim W, Ho W W. Synthesis and characterization of nanoporous polyethersulfone membrane as support for composite membrane in CO<sub>2</sub> separation: From lab to pilot scale. *Journal of Membrane Science*, 2016, 510: 58-71.

420. Xiang L, Pan Y, Zeng G, Jiang J, Chen J, Wang C. Preparation of poly (ether-block-amide)/attapulgite mixed matrix membranes for CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of Membrane Science*, 2016, 500: 66-75.
421. Xin Q, Zhang Y, Huo T, Ye H, Ding X, Lin L, Zhang Y, Wu H, Jiang Z. Mixed matrix membranes fabricated by a facile in situ biomimetic mineralization approach for efficient CO<sub>2</sub> separation. *Journal of Membrane Science*, 2016, 508: 84-93.
422. Xin Q, Zhang Y, Shi Y, Ye H, Lin L, Ding X, Zhang Y, Wu H, Jiang Z. Tuning the performance of CO<sub>2</sub> separation membranes by incorporating multifunctional modified silica microspheres into polymer matrix. *Journal of Membrane Science*, 2016, 514: 73-85.
423. Zaidiza D A, Wilson S G, Belaisaoui B, Rode S, Castel C, Roizard D, Favre E. Rigorous modelling of adiabatic multicomponent CO<sub>2</sub> post-combustion capture using hollow fibre membrane contactors. *Chemical Engineering Science*, 2016, 145: 45-58.
424. Zhang H, Guo R, Hou J, Wei Z, Li X. Mixed-matrix membranes containing carbon nanotubes composite with hydrogel for efficient CO<sub>2</sub> separation. *ACS applied materials & interfaces*, 2016, 8(42): 29044-29051.
425. Zhang L, Li J, Zhou L, Liu R, Wang X, Yang L. Fouling of Impurities in Desulfurized Flue Gas on Hollow Fiber Membrane Absorption for CO<sub>2</sub> Capture. *Industrial & Engineering Chemistry Research*, 2016, 55(29): 8002-8010.
426. Zhang P, Tong J, Jee Y, Huang K. Stabilizing a high-temperature electrochemical silver-carbonate CO<sub>2</sub> capture membrane by atomic layer deposition of a ZrO<sub>2</sub> overcoat. *Chemical Communications*, 2016, 52(63): 9817-9820.
427. Zhong S, Bu N, Zhou R, Jin W, Yu M, Li S. Aluminophosphate-17 and silicoaluminophosphate-17 membranes for CO<sub>2</sub> separations. *Journal of Membrane Science*, 2016, 520: 507-514.
428. Alavi S A, Kargari A, Sanaeepur H, Karimi M. Preparation and characterization of PDMS/zeolite 4A/PAN mixed matrix thin film composite membrane for CO<sub>2</sub>/N<sub>2</sub> and CO<sub>2</sub>/CH<sub>4</sub> separations. *Research on Chemical Intermediates*, 2017, 43(5): 2959-2984.
429. Azizi N, Arzani M, Mahdavi H R, Mohammadi T. Synthesis and characterization of poly (ether-block-amide) copolymers/multi-walled carbon nanotube nanocomposite membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Korean Journal of Chemical Engineering*, 2017, 34(9): 2459-2470.
430. Azizi N, Mohammadi T, Behbahani R M. Synthesis of a new nanocomposite membrane (PEBAX-1074/PEG-400/TiO<sub>2</sub>) in order to separate CO<sub>2</sub> from CH<sub>4</sub>. *Journal of Natural Gas Science and Engineering*, 2017, 37: 39-51.
431. Azizi N, Mohammadi T, Behbahani R M. Synthesis of a PEBAX-1074/ZnO nanocomposite membrane with improved CO<sub>2</sub> separation performance. *Journal of energy chemistry*, 2017, 26(3): 454-465.
432. Benito J, Sánchez - Laínez J, Zornoza B, Martín S, Carta M, Malpass - Evans R, Téllez C, McKeown N B, Coronas J, Gascón I. Ultrathin composite polymeric membranes for CO<sub>2</sub>/N<sub>2</sub> separation with minimum thickness and high CO<sub>2</sub> permeance. *ChemSusChem*, 2017, 10(20): 4014-4017.
433. Brunetti A, Cersosimo M, Kim J S, Dong G, Fontananova E, Lee Y M, Drioli E, Barbieri G. Thermally rearranged mixed matrix membranes for CO<sub>2</sub> separation: An aging study. *International Journal of Greenhouse Gas Control*, 2017, 61: 16-26.

434. Cheng Y, Wang X, Jia C, Wang Y, Zhai L, Wang Q, Zhao D. Ultrathin mixed matrix membranes containing two-dimensional metal-organic framework nanosheets for efficient CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of Membrane Science*, 2017, 539: 213-223.
435. Galaleldin S, Mannan H, Mukhtar H. Development and characterization of polyethersulfone/TiO<sub>2</sub> mixed matrix membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. in *AIP Conference Proceedings*. 2017. AIP Publishing.
436. Ghadiri M, Marjani A, Shirazian S. Development of a mechanistic model for prediction of CO<sub>2</sub> capture from gas mixtures by amine solutions in porous membranes. *Environmental Science and Pollution Research*, 2017, 24(16): 14508-14515.
437. Hosseini S, Mansourizadeh A. Preparation of porous hydrophobic poly (vinylidene fluoride-co-hexafluoropropylene) hollow fiber membrane contactors for CO<sub>2</sub> stripping. *Journal of the Taiwan Institute of Chemical Engineers*, 2017, 76: 156-166.
438. Hu L, Cheng J, Li Y, Liu J, Zhang L, Zhou J, Cen K. Composites of ionic liquid and amine-modified SAPO 34 improve CO<sub>2</sub> separation of CO<sub>2</sub>-selective polymer membranes. *Applied Surface Science*, 2017, 410: 249-258.
439. Hussain A. Three Stage Membrane Process for CO<sub>2</sub> Capture from Natural Gas. *A A*, 50: 1.
440. Ilyas A, Muhammad N, Gilani M A, Ayub K, Vankelecom I F, Khan A L. Supported protic ionic liquid membrane based on 3-(trimethoxysilyl) propan-1-aminium acetate for the highly selective separation of CO<sub>2</sub>. *Journal of Membrane Science*, 2017, 543: 301-309.
441. Isfahani A P, Sadeghi M, Wakimoto K, Gibbons A H, Bagheri R, Sivaniah E, Ghalei B. Enhancement of CO<sub>2</sub> capture by polyethylene glycol-based polyurethane membranes. *Journal of Membrane Science*, 2017, 542: 143-149.
442. Jin P, Huang C, Shen Y, Zhan X, Hu X, Wang L, Wang L. Simultaneous Separation of H<sub>2</sub>S and CO<sub>2</sub> from Biogas by Gas-Liquid Membrane Contactor Using Single and Mixed Absorbents. *Energy & Fuels*, 2017, 31(10): 11117-11126.
443. Jo E-S, An X, Ingole P G, Choi W-K, Park Y-S, Lee H-K. CO<sub>2</sub>/CH<sub>4</sub> separation using inside coated thin film composite hollow fiber membranes prepared by interfacial polymerization. *Chinese Journal of Chemical Engineering*, 2017, 25(3): 278-287.
444. Jung J P, Park C H, Lee J H, Bae Y-S, Kim J H. Room-temperature, one-pot process for CO<sub>2</sub> capture membranes based on PEMA-g-PPG graft copolymer. *Chemical Engineering Journal*, 2017, 313: 1615-1622.
445. Jusoh N, Yeong Y F, Lau K K, Shariff A M. Transport properties of mixed matrix membranes encompassing zeolitic imidazolate framework 8 (ZIF-8) nanofiller and 6FDA-durene polymer: Optimization of process variables for the separation of CO<sub>2</sub> from CH<sub>4</sub>. *Journal of cleaner production*, 2017, 149: 80-95.
446. Kang G, Chan Z P, Saleh S B M, Cao Y. Removal of high concentration CO<sub>2</sub> from natural gas using high pressure membrane contactors. *International Journal of Greenhouse Gas Control*, 2017, 60: 1-9.
447. Karousos D S, Labropoulos A I, Sapalidis A, Kanellopoulos N K, Iliev B, Schubert T J, Romanos G E. Nanoporous ceramic supported ionic liquid membranes for CO<sub>2</sub> and SO<sub>2</sub> removal from flue gas. *Chemical Engineering Journal*, 2017, 313: 777-790.

448. Karunakaran M, Villalobos L F, Kumar M, Shevate R, Akhtar F H, Peinemann K-V. Graphene oxide doped ionic liquid ultrathin composite membranes for efficient CO<sub>2</sub> capture. *Journal of Materials Chemistry A*, 2017, 5(2): 649-656.
449. Kgaphola K, Sigalas I, Daramola M O. Synthesis and characterization of nanocomposite SAPO - 34/ceramic membrane for post - combustion CO<sub>2</sub> capture. *Asia - Pacific Journal of Chemical Engineering*, 2017, 12(6): 894-904.
450. Khakpay A, Rahmani F, Nouranian S, Scovazzo P. Molecular insights on the CH<sub>4</sub>/CO<sub>2</sub> separation in nanoporous graphene and graphene oxide separation platforms: Adsorbents versus membranes. *The Journal of Physical Chemistry C*, 2017, 121(22): 12308-12320.
451. Khalilinejad I, Kargari A, Sanaeepur H. Preparation and characterization of (Pebax 1657+ silica nanoparticle)/PVC mixed matrix composite membrane for CO<sub>2</sub>/N<sub>2</sub> separation. *Chemical Papers*, 2017, 71(4): 803-818.
452. Khosravi T, Omidkhah M, Kaliaguine S, Rodrigue D. Amine - functionalized CuBTC/poly (ether - b - amide - 6)(Pebax® MH 1657) mixed matrix membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *The Canadian Journal of Chemical Engineering*, 2017, 95(10): 2024-2033.
453. Kim N U, Park B J, Choi Y, Lee K B, Kim J H. High-performance self-cross-linked PGP-POEM comb copolymer membranes for CO<sub>2</sub> capture. *Macromolecules*, 2017, 50(22): 8938-8947.
454. Kim S H, Kim J-K, Yeo J-g, Yeo Y-K. Comparative feasibility study of CO<sub>2</sub> capture in hollowfiber membrane processes based on process models and heat exchanger analysis. *Chemical Engineering Research and Design*, 2017, 117: 659-669.
455. Kline G K, Weidman J R, Zhang Q, Guo R. Studies of the synergistic effects of crosslink density and crosslink inhomogeneity on crosslinked PEO membranes for CO<sub>2</sub>-selective separations. *Journal of Membrane Science*, 2017, 544: 25-34.
456. Lai L S, Yeong Y F, Lau K K, Shariff A M. Single and Binary CO<sub>2</sub>/CH<sub>4</sub> Separation of a Zeolitic Imidazolate Framework - 8 Membrane. *Chemical Engineering & Technology*, 2017, 40(6): 1031-1042.
457. Lai L S, Yeong Y F, Lau K K, Shariff A M. Synthesis of zeolitic imidazolate frameworks (ZIF) - 8 membrane and its process optimization study in separation of CO<sub>2</sub> from natural gas. *Journal of Chemical Technology & Biotechnology*, 2017, 92(2): 420-431.
458. Lee H, Park S C, Roh J S, Moon G H, Shin J E, Kang Y S, Park H B. Metal-organic frameworks grown on a porous planar template with an exceptionally high surface area: promising nanofiller platforms for CO<sub>2</sub> separation. *Journal of Materials Chemistry A*, 2017, 5(43): 22500-22505.
459. Lee J H, Lee J, Jo H J, Seong J G, Kim J S, Lee W H, Moon J, Lee D, Oh W J, Yeo J-g. Wet CO<sub>2</sub>/N<sub>2</sub> permeation through a crosslinked thermally rearranged poly (benzoxazole-co-imide)(XTR-PBOI) hollow fiber membrane module for CO<sub>2</sub> capture. *Journal of Membrane Science*, 2017, 539: 412-420.
460. Lee S, Binns M, Lee J H, Moon J-H, Yeo J-g, Yeo Y-K, Lee Y M, Kim J-K. Membrane separation process for CO<sub>2</sub> capture from mixed gases using TR and XTR hollow fiber membranes: process modeling and experiments. *Journal of Membrane Science*, 2017, 541: 224-234.

461. Li H, Ding X, Zhang Y, Liu J. Porous graphene nanosheets functionalized thin film nanocomposite membrane prepared by interfacial polymerization for CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of Membrane Science*, 2017, 543: 58-68.
462. Li S, Pyrzyński T J, Klinghoffer N B, Tamale T, Zhong Y, Aderhold J L, Zhou S J, Meyer H S, Ding Y, Bikson B. Scale-up of PEEK hollow fiber membrane contactor for post-combustion CO<sub>2</sub> capture. *Journal of Membrane Science*, 2017, 527: 92-101.
463. Liu B, Zhou R, Bu N, Wang Q, Zhong S, Wang B, Hidetoshi K. Room-temperature ionic liquids modified zeolite SSZ-13 membranes for CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of membrane science*, 2017, 524: 12-19.
464. Liu Y, Li X, Qin Y, Guo R, Zhang J. Pebax–polydopamine microsphere mixed - matrix membranes for efficient CO<sub>2</sub> separation. *Journal of Applied Polymer Science*, 2017, 134(10).
465. Lu J-G, Ge H, Chen Y, Ren R-T, Xu Y, Zhao Y-X, Zhao X, Qian H. CO<sub>2</sub> capture using a functional protic ionic liquid by membrane absorption. *Journal of the Energy Institute*, 2017, 90(6): 933-940.
466. Mahdavi H R, Azizi N, Mohammadi T. Performance evaluation of a synthesized and characterized Pebax1657/PEG1000/γ-Al<sub>2</sub>O<sub>3</sub> membrane for CO<sub>2</sub>/CH<sub>4</sub> separation using response surface methodology. *Journal of Polymer Research*, 2017, 24(5): 67.
467. Mannan H, Mohshim D, Mukhtar H, Murugesan T, Man Z, Bustam M. Synthesis, characterization, and CO<sub>2</sub> separation performance of polyether sulfone/[EMIM][Tf<sub>2</sub>N] ionic liquid-polymeric membranes (ILPMs). *Journal of industrial and engineering chemistry*, 2017, 54: 98-106.
468. Marti A M, Wickramanayake W, Dahe G, Sekizkardes A, Bank T L, Hopkinson D P, Venna S R. Continuous flow processing of ZIF-8 membranes on polymeric porous hollow fiber supports for CO<sub>2</sub> capture. *ACS applied materials & interfaces*, 2017, 9(7): 5678-5682.
469. Martín-Gil V, López A, Hrabanek P, Mallada R, Vankelecom I, Fila V. Study of different titanosilicate (TS-1 and ETS-10) as fillers for Mixed Matrix Membranes for CO<sub>2</sub>/CH<sub>4</sub> gas separation applications. *Journal of Membrane Science*, 2017, 523: 24-35.
470. Mirfendereski M Mohammadi T. Investigation of H<sub>2</sub>S and CO<sub>2</sub> Removal from Gas Streams Using Hollow Fiber Membrane Gas–liquid Contactors. *Chemical and biochemical engineering quarterly*, 2017, 31(2): 139-144.
471. Monteiro B, Nabais A R, Almeida Paz F A, Cabrita L, Branco L C, Marrucho I M, Neves L A, Pereira C C. Membranes with a low loading of Metal–Organic Framework - Supported Ionic Liquids for CO<sub>2</sub>/N<sub>2</sub> separation in CO<sub>2</sub> capture. *Energy Technology*, 2017, 5(12): 2158-2162.
472. Morris C G, Jacques N M, Godfrey H G, Mitra T, Fritsch D, Lu Z, Murray C A, Potter J, Cobb T M, Yuan F. Stepwise observation and quantification and mixed matrix membrane separation of CO<sub>2</sub> within a hydroxy-decorated porous host. *Chemical science*, 2017, 8(4): 3239-3248.
473. Nordin N A H M, Ismail A F, Misdan N, Nazri N A M. Modified ZIF-8 mixed matrix membrane for CO<sub>2</sub>/CH<sub>4</sub> separation. in *AIP Conference Proceedings*. 2017. AIP Publishing.
474. Park C H, Lee J H, Jang E, Lee K B, Kim J H. MgCO<sub>3</sub>-crystal-containing mixed matrix membranes with enhanced CO<sub>2</sub> permselectivity. *Chemical Engineering Journal*, 2017, 307: 503-512.
475. Peng D, Wang S, Tian Z, Wu X, Wu Y, Wu H, Xin Q, Chen J, Cao X, Jiang Z. Facilitated transport membranes by incorporating graphene nanosheets with high zinc ion loading for enhanced CO<sub>2</sub> separation. *Journal of Membrane Science*, 2017, 522: 351-362.

476. Prasad B, Mandal B. CO<sub>2</sub> separation performance by chitosan/tetraethylenepentamine/poly(ether sulfone) composite membrane. *Journal of Applied Polymer Science*, 2017, 134(34): 45206.
477. Qu Y, Li F, Zhao M. Theoretical design of highly efficient CO<sub>2</sub>/N<sub>2</sub> separation membranes based on electric quadrupole distinction. *The Journal of Physical Chemistry C*, 2017, 121(33): 17925-17931.
478. Quan S, Li S W, Xiao Y C, Shao L. CO<sub>2</sub>-selective mixed matrix membranes (MMMs) containing graphene oxide (GO) for enhancing sustainable CO<sub>2</sub> capture. *International Journal of Greenhouse Gas Control*, 2017, 56: 22-29.
479. Rahmawati Y, Nurkhamidah S, Susianto, Listiyana N I, Putricahyani W. Application of dual membrane contactor for simultaneous CO<sub>2</sub> removal using continuous diethanolamine (DEA). in *AIP Conference Proceedings*. 2017. AIP Publishing.
480. Ranjbaran F, Kamio E, Matsuyama H. Ion gel membrane with tunable inorganic/organic composite network for CO<sub>2</sub> separation. *Industrial & Engineering Chemistry Research*, 2017, 56(44): 12763-12772.
481. Ur Rehman R, Rafiq S, Muhammad N, Khan A L, Ur Rehman A, TingTing L, Saeed M, Jamil F, Ghauri M, Gu X. Development of ethanalamine - based ionic liquid membranes for efficient CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of Applied Polymer Science*, 2017, 134(44): 45395.
482. Ricci E, Minelli M, De Angelis M G. A multiscale approach to predict the mixed gas separation performance of glassy polymeric membranes for CO<sub>2</sub> capture: The case of CO<sub>2</sub>/CH<sub>4</sub> mixture in Matrimid®. *Journal of membrane science*, 2017, 539: 88-100.
483. Rudaini I A, Naim R, Abdullah S, Mokhtar N M, Jaafar J. PVDF-Cloisite Hollow Fiber Membrane For CO<sub>2</sub> Absorption Via Membrane Contactor. *Jurnal Teknologi*, 2017, 79(1-2).
484. Saidi M. Kinetic study and process model development of CO<sub>2</sub> absorption using hollow fiber membrane contactor with promoted hot potassium carbonate. *Journal of environmental chemical engineering*, 2017, 5(5): 4415-4430.
485. Saidi M. Mathematical modeling of CO<sub>2</sub> absorption into novel reactive DEAB solution in hollow fiber membrane contactors; kinetic and mass transfer investigation. *Journal of membrane science*, 2017, 524: 186-196.
486. Selyanchyn R, Fujikawa S. Membrane thinning for efficient CO<sub>2</sub> capture. *Science and Technology of advanced Materials*, 2017, 18(1): 816-827.
487. Shafie S N A, Man Z, Idris A. Development of polycarbonate-silica matrix membrane for CO<sub>2</sub>/CH<sub>4</sub> separation. in *AIP Conference Proceedings*. 2017. AIP Publishing.
488. Shamsabadi A A, Seidi F, Salehi E, Nozari M, Rahimpour A, Soroush M. Efficient CO<sub>2</sub>-removal using novel mixed-matrix membranes with modified TiO<sub>2</sub> nanoparticles. *Journal of Materials Chemistry A*, 2017, 5(8): 4011-4025.
489. Shin H, Chi W S, Bae S, Kim J H, Kim J. High-performance thin PVC-POEM/ZIF-8 mixed matrix membranes on alumina supports for CO<sub>2</sub>/CH<sub>4</sub> separation. *Journal of industrial and engineering chemistry*, 2017, 53: 127-133.
490. Song C, Liu Q, Ji N, Deng S, Zhao J, Li Y, Kitamura Y. Reducing the energy consumption of membrane-cryogenic hybrid CO<sub>2</sub> capture by process optimization. *Energy*, 2017, 124: 29-39.

491. Song Z, Qiu F, Zaia E W, Wang Z, Kunz M, Guo J, Brady M, Mi B, Urban J J. Dual-channel, molecular-sieving core/shell ZIF@ MOF architectures as engineered fillers in hybrid membranes for highly selective CO<sub>2</sub> separation. *Nano letters*, 2017, 17(11): 6752-6758.
492. Taniguchi I, Kinugasa K, Toyoda M, Minezaki K. Effect of amine structure on CO<sub>2</sub> capture by polymeric membranes. *Science and Technology of advanced MaTerialS*, 2017, 18(1): 950-958.
493. Taniguchi I, Wada N, Kinugasa K, Higa M. CO<sub>2</sub> capture by polymeric membranes composed of hyper-branched polymers with dense poly (oxyethylene) comb and poly (amidoamine). *Open Physics*, 2017, 15(1): 662-670.
494. Tong Z, Ho W W. New sterically hindered polyvinylamine membranes for CO<sub>2</sub> separation and capture. *Journal of Membrane Science*, 2017, 543: 202-211.
495. Tseng H-H, Chuang H-W, Zhuang G-L, Lai W-H, Wey M-Y. Structure-controlled mesoporous SBA-15-derived mixed matrix membranes for H<sub>2</sub> purification and CO<sub>2</sub> capture. *International Journal of Hydrogen Energy*, 2017, 42(16): 11379-11391.
496. Turi D, Ho M, Ferrari M, Chiesa P, Wiley D, Romano M C. CO<sub>2</sub> capture from natural gas combined cycles by CO<sub>2</sub> selective membranes. *International Journal of Greenhouse Gas Control*, 2017, 61: 168-183.
497. Usman M, Dai Z, Hillestad M, Deng L. Mathematical modeling and validation of CO<sub>2</sub> mass transfer in a membrane contactor using ionic liquids for pre-combustion CO<sub>2</sub> capture. *Chemical Engineering Research and Design*, 2017, 123: 377-387.
498. Wang P, Li W, Du C, Zheng X, Sun X, Yan Y, Zhang J. CO<sub>2</sub>/N<sub>2</sub> separation via multilayer nanoslit graphene oxide membranes: Molecular dynamics simulation study. *Computational Materials Science*, 2017, 140: 284-289.
499. Wang S, Xie Y, He G, Xin Q, Zhang J, Yang L, Li Y, Wu H, Zhang Y, Guiver M D. Graphene oxide membranes with heterogeneous nanodomains for efficient CO<sub>2</sub> separations. *Angewandte Chemie International Edition*, 2017, 56(45): 14246-14251.
500. Wang Z, Ren H, Zhang S, Zhang F, Jin J. Polymers of intrinsic microporosity/metal-organic framework hybrid membranes with improved interfacial interaction for high-performance CO<sub>2</sub> separation. *Journal of Materials Chemistry A*, 2017, 5(22): 10968-10977.
501. Xiang L, Sheng L, Wang C, Zhang L, Pan Y, Li Y. Amino - Functionalized ZIF - 7 Nanocrystals: Improved Intrinsic Separation Ability and Interfacial Compatibility in Mixed - Matrix Membranes for CO<sub>2</sub>/CH<sub>4</sub> Separation. *Advanced Materials*, 2017, 29(32): 1606999.
502. Yoon K W, Kim H, Kang Y S, Kang S W. 1-Butyl-3-methylimidazolium tetrafluoroborate/zinc oxide composite membrane for high CO<sub>2</sub> separation performance. *Chemical Engineering Journal*, 2017, 320: 50-54.
503. Zainab G, Iqbal N, Babar A A, Huang C, Wang X, Yu J, Ding B. Free-standing, spider-web-like polyamide/carbon nanotube composite nanofibrous membrane impregnated with polyethyleneimine for CO<sub>2</sub> capture. *Composites Communications*, 2017, 6: 41-47.
504. Zhang C, Zhang W, Gao H, Bai Y, Sun Y, Chen Y. Synthesis and gas transport properties of poly (ionic liquid) based semi-interpenetrating polymer network membranes for CO<sub>2</sub>/N<sub>2</sub> separation. *Journal of Membrane Science*, 2017, 528: 72-81.

505. Zhang L, Wang X, Yu R, Li J, Hu B, Yang L. Hollow fiber membrane separation process in the presence of gaseous and particle impurities for post-combustion CO<sub>2</sub> capture. *International journal of green energy*, 2017, 14(1): 15-23.
506. Zhang X-M, Tu Z-H, Li H, Li L, Wu Y-T, Hu X-B. Supported protic-ionic-liquid membranes with facilitated transport mechanism for the selective separation of CO<sub>2</sub>. *Journal of Membrane Science*, 2017, 527: 60-67.
507. Zhang Y, Wang H, Zhang Y, Ding X, Liu J. Thin film composite membranes functionalized with montmorillonite and hydrotalcite nanosheets for CO<sub>2</sub>/N<sub>2</sub> separation. *Separation and Purification Technology*, 2017, 189: 128-137.
508. Zhao D, Ren J, Wang Y, Qiu Y, Li H, Hua K, Li X, Ji J, Deng M. High CO<sub>2</sub> separation performance of Pebax®/CNTs/GTA mixed matrix membranes. *Journal of membrane science*, 2017, 521: 104-113.
509. Zhao L, Sang P, Guo S, Liu X, Li J, Zhu H, Guo W. Promising monolayer membranes for CO<sub>2</sub>/N<sub>2</sub>/CH<sub>4</sub> separation: graphdiynes modified respectively with hydrogen, fluorine, and oxygen atoms. *Applied Surface Science*, 2017, 405: 455-464.
510. Zhou F, Tien H N, Xu W L, Chen J-T, Liu Q, Hicks E, Fathizadeh M, Li S, Yu M. Ultrathin graphene oxide-based hollow fiber membranes with brush-like CO<sub>2</sub>-philic agent for highly efficient CO<sub>2</sub> capture. *Nature communications*, 2017, 8(1): 2107.
511. Zhu L, Swihart M T, Lin H. Tightening polybenzimidazole (PBI) nanostructure via chemical cross-linking for membrane H<sub>2</sub>/CO<sub>2</sub> separation. *Journal of Materials Chemistry A*, 2017, 5(37): 19914-19923.
512. Ahmad M Z, Navarro M, Lhotka M, Zornoza B, Téllez C, Fila V, Coronas J. Enhancement of CO<sub>2</sub>/CH<sub>4</sub> separation performances of 6FDA-based co-polyimides mixed matrix membranes embedded with UiO-66 nanoparticles. *Separation and Purification Technology*, 2018, 192: 465-474.
513. Baghban A Azar A A. ANFIS modeling of CO<sub>2</sub> separation from natural gas using hollow fiber polymeric membrane. *Energy sources, part a: Recovery, utilization, and environmental effects*, 2018, 40(2): 193-199.
514. Constantinou A, Barrass S, Gavriilidis A. CO<sub>2</sub> absorption in flat membrane microstructured contactors of different wettability using aqueous solution of NaOH. *Green Processing and Synthesis*, 2018, 7(6): 471-476.
515. Fu X, Li X, Guo R, Zhang J, Cao X. Block copolymer membranes based on polyetheramine and methyl-containing polyisophthalamides designed for efficient CO<sub>2</sub> separation. *High Performance Polymers*, 2018, 30(9): 1064-1074.
516. Hu L, Cheng J, Li Y, Liu J, Zhou J, Cen K. Optimization of coating solution viscosity of hollow fiber - supported polydimethylsiloxane membrane for CO<sub>2</sub>/H<sub>2</sub> separation. *Journal of Applied Polymer Science*, 2018, 135(5): 45765.
517. Hu L, Cheng J, Li Y, Liu J, Zhou J, Cen K. In-situ grafting to improve polarity of polyacrylonitrile hollow fiber-supported polydimethylsiloxane membranes for CO<sub>2</sub> separation. *Journal of colloid and interface science*, 2018, 510: 12-19.
518. Ko D. Development of a dynamic simulation model of a hollow fiber membrane module to sequester CO<sub>2</sub> from coalbed methane. *Journal of Membrane Science*, 2018, 546: 258-269.

519. Ovalle-Encinia O, Pfeiffer H, Ortiz-Landeros J. Ce<sub>0.85</sub>Sm<sub>0.15</sub>O<sub>2</sub>-Sm<sub>0.6</sub>Sr<sub>0.4</sub>Al<sub>0.3</sub>Fe<sub>0.7</sub>O<sub>3</sub> composite for the preparation of dense ceramic-carbonate membranes for CO<sub>2</sub> separation. *Journal of Membrane Science*, 2018, 547: 11-18.
520. Pang H, Gong H, Du M, Shen Q, Chen Z. Effect of non-solvent additive concentration on CO<sub>2</sub> absorption performance of polyvinylidene fluoride hollow fiber membrane contactor. *Separation and Purification Technology*, 2018, 191: 38-47.
521. Ramli N A, Hashim N A, Aroua M K. Prediction of CO<sub>2</sub>/O<sub>2</sub> absorption selectivity using supported ionic liquid membranes (SILMs) for gas-liquid membrane contactor. *Chemical Engineering Communications*, 2018, 205(3): 295-310.
522. Russo G, Prpich G, Anthony E J, Montagnaro F, Jurado N, Di Lorenzo G, Darabkhani H G. Selective-exhaust gas recirculation for CO<sub>2</sub> capture using membrane technology. *Journal of Membrane Science*, 2018, 549: 649-659.
523. Wang F, Kang G, Liu D, Li M, Cao Y. Enhancing CO<sub>2</sub> absorption efficiency using a novel PTFE hollow fiber membrane contactor at elevated pressure. *AIChE Journal*, 2018, 64(6): 2135-2145.
524. Yu L, Kanezashi M, Nagasawa H, Moriyama N, Tsuru T, Ito K. Enhanced CO<sub>2</sub> separation performance for tertiary amine - silica membranes via thermally induced local liberation of CH<sub>3</sub>Cl. *AIChE Journal*, 2018, 64(5): 1528-1539.
525. Zhang N, Peng D, Wu H, Ren Y, Yang L, Wu X, Wu Y, Qu Z, Jiang Z, Cao X. Significantly enhanced CO<sub>2</sub> capture properties by synergy of zinc ion and sulfonate in Pebax-pitch hybrid membranes. *Journal of Membrane Science*, 2018, 549: 670-679.