

Influences of spinel type and polymeric surfactants on the size evolution of colloidal magnetic nanocrystals (MFe₂O₄, M = Fe, Mn)

Tahereh ROHANI BASTAMI(✉)^{1,2}, Mohammad H. ENTEZARI², Chiwai KWONG(✉)³, Shizhang QIAO(✉)³

1 Department of Chemical Engineering, Quchan University of Advanced Technology, Quchan 94771, Iran

2 Department of Chemistry, Ferdowsi University of Mashhad, Mashhad 91775, Iran

3 School of Chemical Engineering, The University of Adelaide, SA 5005, Australia

E-mail: tahereh.rohani@gmail.com, philip.kwong@adelaide.edu.au, s.qiao@adelaide.edu.au

Table S1 Results and analysis based on TGA curves for manganese ferrite nanoparticles

Sample condition	Second weight loss /%	Third weight loss /%	Total mass loss + uncoordinated surfactants + adsorbed water /%
12 h, 160 °C, PVP	32	6	50
12 h, 180 °C, PVP	6.5	2	11
72 h, 180 °C, PVP	7.5	1.5	12
12 h, 160 °C, PEG ₃₀₀	12	1.7	16
12 h, 180 °C, PEG ₃₀₀	8.5	1.5	12

Table S2 Results and analysis based on TGA curves for magnetite nanoparticles

Sample condition	Second weight loss /%	Third weight loss /%	Total mass loss + uncoordinated surfactants + adsorbed water /%
12 h, 160 °C, PVP	9	1.5	12.5
12 h, 180 °C, PVP	5	2	8
2 h, 180 °C, PVP	3	2	5.5
12 h, 160 °C, PEG ₃₀₀	10	1.3	13
12 h, 180 °C, PEG ₃₀₀	6	1.2	8

Table S3 Infrared transmission frequencies of pure surfactants and final product, manganese ferrite nanoparticles (12 h, 160 °C)

Observed bands /cm ⁻¹			Observed bands /cm ⁻¹		
PEG ₃₀₀	Product	Band assignment	PVP	Product	Band assignment
3380	3360	$\nu_{\text{O-H}}$ and adsorbed water	3430-3530	3320-3430	adsorbed water
2890	2870, 2930	$(\nu_{\text{S}}, \nu_{\text{as}})_{\text{CH}_2}$	2890, 2960	2870, 2950	$(\nu_{\text{S}}, \nu_{\text{as}})_{\text{CH}_2}$
	1570	$\delta_{\text{H}_2\text{O}}$		1570	$\delta_{\text{H}_2\text{O}}$
1470, 1360	1430	CH ₂ bending	1670	1570-1670 ^{sh}	$\nu_{\text{C=O}}$
1100	1080	$\nu_{\text{C-O}}$	1430	1430	CH ₂ bending
940	895	ρ_{CH_2} out of plane bending	1280	1220	$\nu_{\text{C-N}}$
	540	$\nu_{\text{M-O}}$		1050-1080	$\nu_{\text{C-O}}$
				879	CH ₂ out of plane bending
				550	$\nu_{\text{M-O}}$

sh = shoulder, ν_{as} = asymmetric stretching, ν_{S} = symmetric stretching, δ = scissoring, ρ = rocking

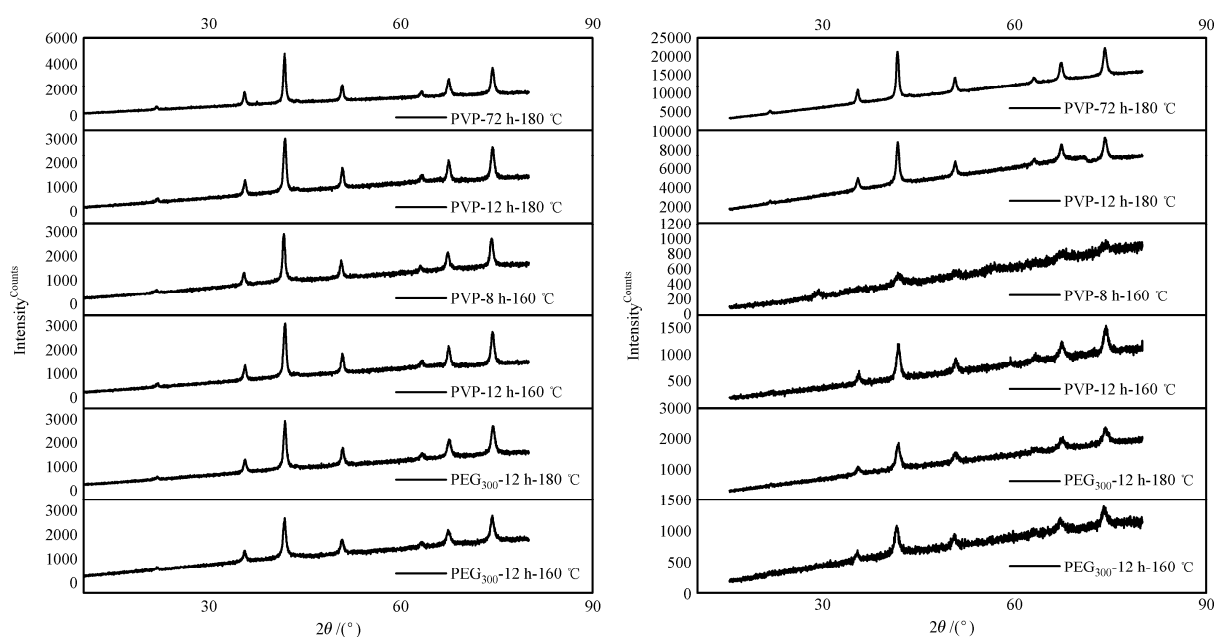
Table S4 Infrared transmission frequencies of surfactant and final product, magnetite nanoparticles (12 h, 160 °C)

Observed bands /cm ⁻¹			Observed bands /cm ⁻¹		
PEG ₃₀₀ 3380	Product 3360	Band assignment ν _{O-H} and adsorbed water	PVP 3430-3530	Product	Band assignment adsorbed water
2890	2870, 2920 1590	(ν _s , ν _{as}) _{CH2} δ _{H2O}	2890, 2960	1590	(ν _s , ν _{as}) _{CH2} δ _{H2O}
1470, 1360	1430	CH ₂ bending	1670	1590-1650 ^{sh}	C=O stretch
1100	1050-1080	ν _{C-O}	1430	1430	CH ₂ bending
940	895	ρ _{CH2} out of plane bending	1280	1050-1100	ν _{C-N}
	555	ν _{M-O}		895	ν _{C-O} CH ₂ out of plane bending
				555	M-O stretching

sh = shoulder, ν_{as} = asymmetric stretching, ν_s = symmetric stretching, δ = scissoring, ρ = rocking

Table S5 Magnetization saturation for magnetic ferrite nanoparticles

Sample	M _s / (emu · g · particle ⁻¹), T = 300 K
MnFe ₂ O ₄ (12 h, 200 °C)	70
MnFe ₂ O ₄ (12 h, 180 °C)	62
Fe ₃ O ₄ (12 h, 180 °C)	81.81
MnFe ₂ O ₄ (12 h, 200 °C) [21]	53.2
Fe ₃ O ₄ (12 h, 200 °C) [21]	81.9

**Fig. S1** XRD patterns of (a) Fe₃O₄, and (b) MnFe₂O₄ nanoparticles

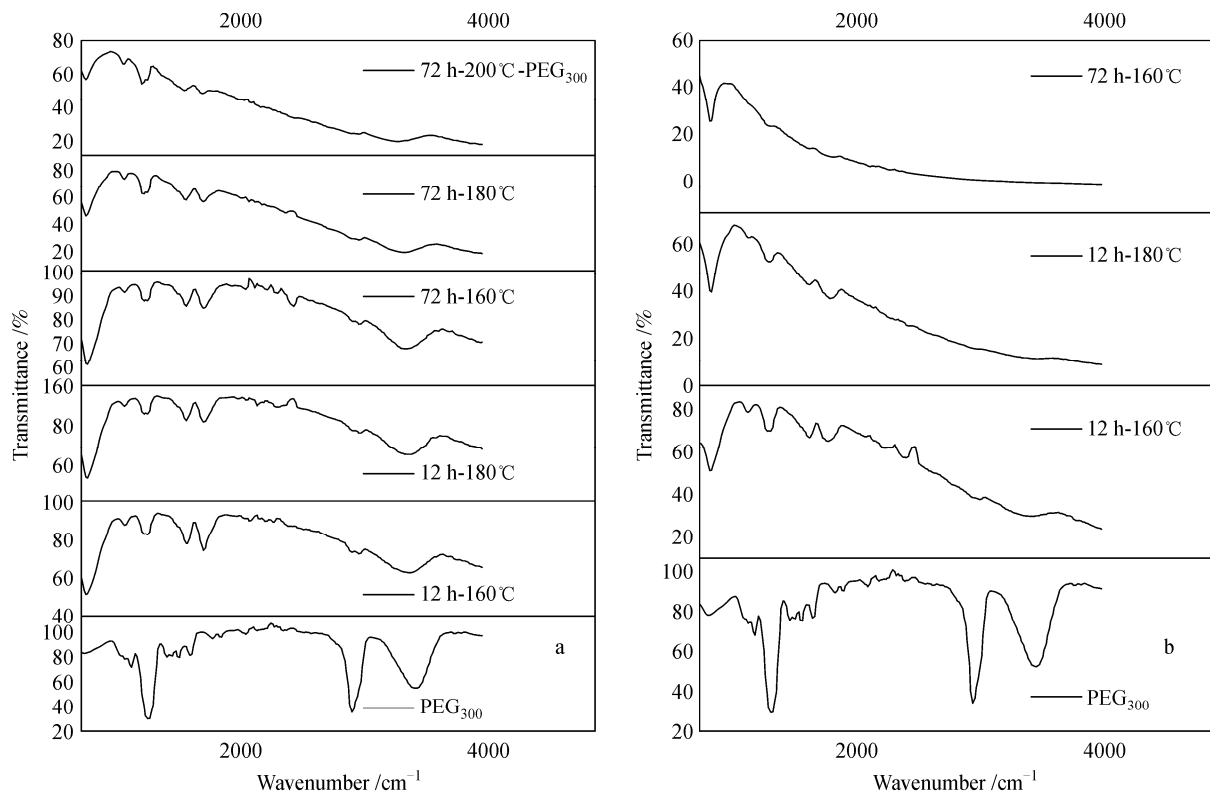


Fig. S2 FTIR spectra of (a) MnFe_2O_4 , and (b) Fe_3O_4 nanoparticles coated with PEG_{300}

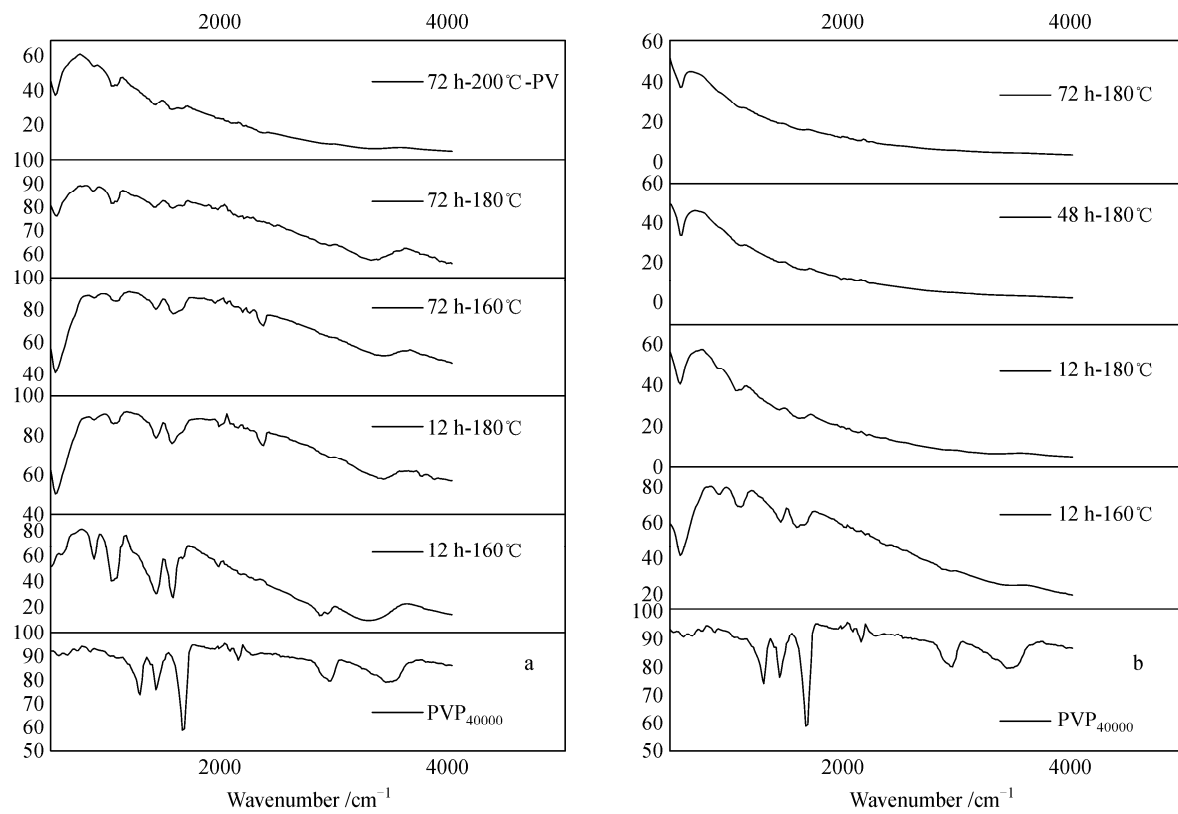


Fig. S3 FTIR spectra of (a) MnFe_2O_4 and (b) Fe_3O_4 nanoparticles coated with PVP