



Original Research

Costa Rican Crisp Products: Incidence of Acrylamide and Evaluation of Their Respective Raw Materials

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Abstract

Background: The study's main objective was to quantify the acrylamide (ACR) content in potato, plantain, and cassava crisps—products widely consumed in Costa Rica. ACR, a possible carcinogen according to various global organizations, is generated during the Maillard reaction when foods rich in asparagine (ASN) and reducing sugars are subjected to temperatures above 120 °C. **Methods:** Using GC-MS analysis on n = 54 samples (24 potatoes, 18 plantains, and 12 cassavas), it was determined that ACR levels were within the ranges established by international organizations such as Codex Alimentarius (CODEX) and the Food and Drug Administration (FDA). The reducing sugar and ASN content of the raw materials was quantified to correlate them with the ACR in the final product. **Results:** One potato product was identified with an ACR concentration that significantly exceeded the 750 µg kg⁻¹ limit stipulated in Recommendation (EU) 2019/1888. For plantain and cassava chips, which currently have no specific regulations, the results showed ACR content to be significantly lower compared to potato crisps. The findings demonstrated a significant correlation between the initial asparagine content and ACR formation in potato crisps, a relationship not observed with reducing sugars. In contrast, no direct correlation was found between precursors and ACR in plantain chips. The analysis also revealed that, in addition to asparagine concentration, the crisps' surface-to-volume ratio is a crucial physical parameter for minimizing ACR formation. **Conclusion:** The data obtained on daily ACR intake will serve as a valuable input for future risk studies in the Costa Rican population, suggesting that plantain and cassava chips are a safer alternative due to their lower ACR content.

Keywords: Costa Rica; snack foods; acrylamide; asparagine

1. Introduction

Plant-based crisps are revolutionizing the snack industry by providing consumers with a healthier and more sustainable alternative to traditional potato chips [1,2]. As a relevant food commodity, snack consumption has been estimated worldwide, including in the United States [3], Iran [4], Germany, and the United Kingdom [5]. It is an industry projected to grow. For example, in Costa Rica alone, the snack food market is expected to grow at an annual rate of 4.32% [6]. It is expected to represent a total revenue of up to USD 275.15 million in 2025, from which salty snacks made from potatoes, plantains, and cassava account for 9% of the trade items [7]. However, despite their popularity, findings suggest that fried snack overeating is linked to an increase in the number of diseases such as cancer, heart disease, obesity, diabetes, and high blood pressure [8]. Additionally, high temperatures increase the risk of nutrient loss, oil oxidation, and acrylamide (ACR) formation during the frying process [9,10], a process primarily used to make crispy snacks.

Heat treatments, such as drying or frying, are often used in the food industry to process and preserve plant-based foods [11]. Frying enables physicochemical changes, such as starch gelatinization (resulting in larger starch granules), crust formation (due to the drying of the fried product), and the formation of taste and aroma components [8]. When heated above 120 °C, plant-based foods are particularly susceptible to the Maillard Reaction [12]. They are likely to produce a significant amount of ACR due to the presence of reducing sugars and the amino acid asparagine (ASN) [13,14]. In addition, ACR formation is susceptible to frying temperature, time, the size of the contact surface with oil, and the interaction between raw materials and oil. As a result, there has been significant concern about the ACR incidence in fried foods due to the mounting evidence of its adverse health effects.

The International Agency for Research on Cancer [15] classified ACR as a probable human carcinogen (Group 2A). It is an already familiar hazardous compound with reported carcinogenicity, genotoxicity, neurotoxicity, and reproductive toxicity in animal studies [16]. In addition,



several recent studies have focused on the levels of ACR in vegetable crisps [17–20]. Additionally, several studies have examined the impact of various cooking parameters on ACR formation in fried foods [21]. Although researchers have explored strategies to mitigate ACR formation during high-temperature cooking processes [16,22,23], few studies are comprehensive and seldom include non-traditional food ingredients used in snack production, as well as their impact on ACR levels in the final product [23,24]. This research provides an overview of the current ACR content in three fried snacks popular in the Costa Rican industry, and, especially, how the chemical nature of the different plant-based raw materials used in the preparation of such foods directly determines ACR levels in the crisp products.

2. Materials and Methods

2.1 Industrial Crisp Samples

The analysis of the incidence of ACR content in these three fried crisps produced in Costa Rica was carried out with the collaboration of three companies (coded as A, B, and C) specialized in the production of fried potato (companies A, B, and C), plantain (companies A and B), and cassava crisps (company A). Each company provided three different batches (1 kg per batch) of raw materials and their corresponding finished products, along with information on the frying process. The crop varieties used by the companies were the same: for potatoes (*Solanum tuberosum* L.), the Floresta and Única varieties; for plantains (*Musa paradisiaca* L.), the Curare variety; and for cassava (*Manihot esculenta* Crantz), the Valencia and Señorita varieties. The samples were delivered to the laboratory frozen in coolers and were kept at $-20\text{ }^{\circ}\text{C}$ until processing.

For the frying treatment, all three companies used the immersion method in palm olein. The frying temperatures vary depending on the vegetable type and shape: for corrugated potato crisps, $165\text{ }^{\circ}\text{C}$; for sliced potatoes, $170\text{ }^{\circ}\text{C}$; and for potato sticks, $180\text{ }^{\circ}\text{C}$ for 4–5 minutes. For plantains (“patacón”, unripe and ripe slices), a pre-frying step was performed for 1–1.5 minutes at $160\text{--}170\text{ }^{\circ}\text{C}$, followed by a final frying step at $175\text{--}185\text{ }^{\circ}\text{C}$ for 3.5–4.0 minutes. In the case of cassava (“patacón” and slice), the frying lasted 2–3 minutes at $170\text{--}180\text{ }^{\circ}\text{C}$. Following the frying process, the products were cooled on conveyor belts with air circulation before packaging. Samples from each batch were delivered to the laboratory in their respective packaging and are stored at $25\text{ }^{\circ}\text{C}$ until processing.

2.2 Commercial Crisps Samples

The analysis of ACR content in commercially available crisps was conducted on a total of 54 crisp-type products sampled from various local markets in the Central Metropolitan Area. The samples were classified based on their matrix: potato ($n = 24$), plantain ($n = 18$), and cassava ($n = 12$). Each lot was subjected to three replicates, which were then homogenized using a food mill (Knife Mill

Grindomix, Retsch GM 200, Haan, Germany) to create a composite sample. This composite sample was placed in vacuum-sealed, high-density polyethylene bags and stored at $-20\text{ }^{\circ}\text{C}$ until the ACR analysis could be performed.

2.3 Reagents

Analytical standards as: ACR (catalog number 23701), ACR-2,3,3- d_3 (catalog number 636568, 98 atom % D, $\text{D}_2\text{C}=\text{CDCONH}_2$, used as internal standard), Fructose (catalog number PHR1002), dextrose (catalog number PHR1000), sucrose (catalog number PHR1001-16), ASN (catalog number PHR1001-16) were purchased from Millipore Sigma (Saint Louis, MO, USA). Sodium dihydrogen phosphate monohydrate (NaH_2PO_4 , catalog number 1.06346) *o*-phthalaldehyde (OPA, $\geq 99\%$, HPLC catalog number 79760), mercaptoethanol ($\geq 99\%$ catalog number M6250), boric acid (catalog number 1.93409), sodium chloride (ACS reagent, $\geq 99.0\%$, catalog number S9888) and sodium sulfate (anhydrous, ACS reagent, powder, catalog number, 238597), Hexane (gradient grade, catalog number 110-54-3), acetonitrile (ACN, gradient grade, $\geq 99.9\%$, catalog number 439134), methanol (MeOH, $\geq 99.9\%$, suitable, HPLC catalog number 34860) were purchased from Millipore Sigma Ultrapure water (type I, $0.055\text{ }\mu\text{S cm}^{-1}$ at $25\text{ }^{\circ}\text{C}$, $5\text{ }\mu\text{g L}^{-1}$ total organic carbon was obtained using an A10 Milli-Q® Advantage system and an Elix® Advantage 10 system (Merck KGaA, Darmstadt, Germany). A quality control material of known ACR concentration for vegetable Crisps was used to assess method accuracy (product code FCCP3-PRO37QC, purchased from the ISO/IEC 17043:2010 accredited FAPAS®, Sand Hutton, York, United Kingdom).

2.4 Chemical Analysis of Raw Material

The raw samples (potato, plantain, and cassava) were homogenized in a food mill (Retsch GM 200, Germany). A portion of the fresh sample was immediately analyzed for water activity (a_w) using AQUA LAB 4TE (Decagon Devices, Pullman, WA, USA). Moisture content was determined according to AOAC 964.22 [25], and pH was determined according to AOAC 981.12 [25]. The remaining portion was frozen ($-20\text{ }^{\circ}\text{C}$) and then lyophilized (FreeZone 6, Labconco, Kansas, MO, USA).

The physicochemical analysis performed on freeze-dried material matter was total starch according to AOAC 996.11 [25] using the Total Starch Assay Kit purchased from Megazyme International Ireland Ltd. (Wicklow, Ireland), moisture content was determined using TGA701 (LECO, Saint Joseph, MI, USA) at $100\text{ }^{\circ}\text{C}$, and the sugar profile and free ASN content.

2.5 Chemical Analysis of Crisps

Each company sent ca. 250 g of crisps (potato, plantain, and cassava). Three replicates were taken from each batch and homogenized in a food mill (Retsch GM 200,

Germany) to obtain a composite sample. The sample was placed in vacuum-packed high-density polyethylene bags and stored at $-20\text{ }^{\circ}\text{C}$ until further analysis (including aw, moisture, fat content, sugar profile, free ASN, and ACR content). Physicochemical analyses were performed: a_w using AQUA LAB 4TE (Decagon Devices, WA, USA). Moisture content was determined using TGA701 (LECO, Saint Joseph, MI, USA) at $100\text{ }^{\circ}\text{C}$, and fat content was determined according to AOAC 920.85 [25].

2.6 Sugar Profile

A sugar profile was analyzed according to Sullivan and Carpenter [26] with the following modifications: a 3.0 g subsample of freeze-dried material was extracted with 30 g type I water. The mixture was stirred for 20 min, centrifuged (Sorvall™, ST16R, catalog number 75007203, ThermoFisher Scientific, Waltham, MA, USA) for 10 min at 4000 rpm. A 2 mL aliquot was filtered through a syringe filter (catalog number 1751AQ, Minisart®, Polytetrafluorethylene (PTFE), Pore Size $0.2\text{ }\mu\text{m}$, Göttingen, Germany) and finally transferred to a HPLC vial (2 mL, Type 1 borosilicate amber glass, PTFE/silicone screw cap and septa, catalog number 5182-0716, Santa Clara, CA, USA).

For the crisps, five grams were extracted with 30–50 mL hexane on an orbital shaker (KS 130 Basic, IKA™, Staufen, Germany) for at least 10 min. The sample was centrifuged for 10 min at 4000 rpm and the organic phase was discarded. The process was repeated once more. Hexane remnant was evaporated in a convection oven at $60\text{ }^{\circ}\text{C}$ until the solvent odor was no longer perceptible. To the fat-free residue, 30 g type I water was added and shaken for 5 min at 300 rpm. Later, the mixture was centrifuged for 10 min at 4000 rpm. A 2 mL aliquot of the supernatant was filtered through a $0.2\text{ }\mu\text{m}$ pore size syringe filter into an HPLC vial.

The HPLC analysis was performed with a Shimadzu HPLC system (Shimadzu, Kyoto, Japan) equipped with a refractive index detector (RID-10A), column compartment (CTO-20A), autosampler (SIL-20A HT), and a quaternary pump (LC-20AT). An amino column (Zorbax Carbohydrate $5\text{ }\mu\text{m}$, $150\text{ mm} \times 4.6\text{ mm}$, part number 883952-708, Agilent Technologies) was used with a flow of 1.2 mL min^{-1} at $30\text{ }^{\circ}\text{C}$. The eluent was a mixture of acetonitrile (ACN) and H_2O (75:25) and was maintained in an isocratic condition for 20 min. The analysis was performed in duplicate. Sugars (sucrose, fructose, and glucose) were identified by comparing their retention times with those of standards and quantified using external calibration curves with a linearity range of 0.1–1.0 g sugar/100 mL mobile phase. The correlation obtained was ($r^2 = 0.9996$). The concentration of each sugar was reported as grams of sugar per 100 g of fresh sample. The chromatographic analysis was performed using LabSolutions Lite (version 5.93, Shimadzu Corporation, Kyoto, Japan).

2.7 Free ASN Content Method

The method was based on Žilić and coworkers [27] with the following modifications: Two grams of the previously homogenized crisps or a subsample of freeze-dried material was weighed in centrifuge tubes (50 mL, CLS430829, polypropylene, Corning®, Corning, NY, USA), 20 mL of a phosphate buffer (NaH_2PO_4 , 40 mmol L^{-1} , $\text{pH} = 7.8$) was added and was stirred for 20 minutes at room temperature in an orbital shaker. Then the mixture was centrifuged at 4500 rpm for 10 minutes. A 5 mL sample of the supernatant was filtered through a $0.20\text{ }\mu\text{m}$ PTFE micropore, $10\text{ }\mu\text{L}$ was taken into a 2 mL vial, $100\text{ }\mu\text{L}$ of *o*-phthalaldehyde reagent (prepared as follows: 10 mg OPA, $20\text{ }\mu\text{L}$ of mercaptoethanol in 10 mL of borate buffer 50 mmol L^{-1} , $\text{pH} = 10$) was added, along with $130\text{ }\mu\text{L}$ of borate buffer (50 mmol L^{-1} , $\text{pH} = 10$) and $260\text{ }\mu\text{L}$ of water type I.

The analytical determination for free ASN was performed using HPLC system (Agilent 1260 infinity, Agilent Technologies) equipped with a quaternary pump (61311C), a column compartment (kept at $25\text{ }^{\circ}\text{C}$ during analysis, G1316A), a fluorescence detector (G1321C) and an automatic liquid sampler module (injection volume set at one μL , ALS, G7129A) prepared with a reverse phase chromatographic column (Zorbax Eclipse AAA column $75\text{ mm} \times 4.6\text{ mm} \times 3.5\text{ }\mu\text{m}$, Part number AG966400-902, Agilent Technologies), at a flow rate of 2 mL min^{-1} at room temperature, using mobile phase A (NaH_2PO_4 40 mmol L^{-1} , $\text{pH} = 7.8$) and mobile phase B (45:45:10 MeOH:ACN: H_2O) with a gradient of 0 min 0% B, 1 min 0% B, 9.8 min 57% B, 10 min 100% B, 12 min 100% B, 12.5 min 0% B, 14 min 0% B, 16 min 0% B, $1\text{ }\mu\text{L}$ of the sample was injected, using wavelengths of 340 nm (excitation) and 450 nm (emission). The ASN content was quantified using a calibration curve (in a range of 0.125 to $2.0\text{ }\mu\text{g mL}^{-1}$, prepared in phosphate-buffered media). The chromatographic analysis was performed using OpenLab ChemStation (version 3.7.1, Agilent Technologies).

2.8 ACR Content Method

Two grams of the previously homogenized crisp samples were weighed in a centrifuge tube (50 mL, polypropylene, conical bottom, Corning®). Afterward, 10 mL of hexane was added to the mixture, which was then shaken for 5 min using a vortex (SI™ Vortex-Genie™ 2, Scientific Industries Inc., Bohemia, NY, USA). The mixture was centrifuged for 10 min at 4000 rpm, and the final hexane supernatant was discarded. This same defatting process was performed once more. To the residue, 10 mL of distilled water and $100\text{ }\mu\text{L}$ of the internal standard solution were added. A 10 mL sample of acetonitrile was added, and the solution was stirred for 30 min in an orbital shaker (KS 130 Basic, IKA™, Staufen, Germany). 4 g of anhydrous Na_2SO_4 and 0.5 g of NaCl were added, and the solution was stirred for 10 more min. The solution was centrifuged

at 4 °C for 8 min at 5500 rpm, 1.00 mL of the upper liquid phase was taken and filtered through a 0.20 µm micropore (regenerated cellulose, catalog 18407, Sartorius, Gotinga, Germany) into a GC/MS-ready vial (2 mL vials and 11 mm aluminum crimp vial caps, red silicone/clear PTFE seal, catalog number 5190-9045, Agilent Technologies)

The ACR analysis was performed using an Agilent 7820A gas chromatograph (Agilent Technologies) coupled with a 5977B mass spectrometry detector and equipped with a 30 m × 250 µm × 0.25 µm chromatographic column (catalog 122-7032UI, J&W DB-WAX Ultra Inert, Agilent Technologies). The injection volume was set at five µL in splitless mode at 15 mL min⁻¹, 150 °C, and a helium flow of 1 mL min⁻¹ (220 cubic ft, ultra-pure grade, GasPro, San Antonio, Alajuela, Costa Rica). The temperature ramp used to separate ACR was as follows: 60 °C for 4 min, 70 °C min⁻¹ to 160 °C for 4 min, and 10 °C min⁻¹ to 190 °C. This temperature was sustained for an additional 4 min, for a total of 17 min of analysis. Detection parameters were set as follows: ionization was mediated by an electron impact system at 70 eV, and the quadrupole temperature was set at 280 °C; selected ions of 71 m/z and 55 m/z were monitored using simultaneous ion monitoring (SIM) mode. The analyte signal was recorded under these conditions at 10.62 ± 0.22 min. The internal standard was assessed in all samples and standards at a fixed final concentration of 1.5 µg mL⁻¹. Selected ions of 74 m/z and 58 m/z were monitored using SIM mode. Nine-point calibration curves were constructed over the range of 0.1 to 100 µg mL⁻¹ in ACN for each measurement and used to interpolate sample data. The chromatographic analysis was performed using Mass Hunter Workstation (version 10.0, Agilent Technologies).

2.9 Statistical Analysis

The Kruskal-Wallis test was used to assess significant differences between types of crisp within each category. ANOVA post hoc Dunnett's test was performed to compare potato crisps ACR values to the benchmark value of 750 µg kg⁻¹. Spearman's rank order test was used to evaluate the association between ACR (µg kg⁻¹ dry basis) and their respective precursors (ASN and reducing sugars on a dry basis) found in the raw materials when compared to the crisp products; in this case, the pair of variables with positive correlation coefficients and *p* values below 0.05 tend to increase together. All statistical tests were performed using a threshold of $\alpha = 0.05$ and IBM SPSS Statistics 30.0 (IBM Inc., Armonk, NY, USA).

2.10 Estimation of Acrylamide Dietary Exposure (ADE)

The Latin American Nutrition and Health Study (ELANS) [28] provided food consumption data for the Costa Rican population. The dataset comprised *n* = 798 individuals aged 15 to 65 years, consisting of *n* = 404 women and *n* = 394 men. For analysis, the data were stratified by gender and then by age group. ADE from crisp product con-

sumption in Costa Rica was estimated using the following equation,

$$ADE = (RI \cdot C)/bw$$

with results expressed in micrograms per kilogram of body weight per day (µg kg⁻¹ bw day⁻¹) for each person. RI represents the daily intake (g day⁻¹) of crisp products, averaged over a two-day consumption period. C means the average ACR concentration in each crisp product (µg kg⁻¹). Bw is the individual's body weight (kg). The average dietary exposure to ACR was calculated for each gender and age group.

3. Results and Discussion

3.1 Vegetable Crisps: Raw Material Description and Impact on ACR Formation

The three raw vegetable samples exhibited an optimal pH range of 4–6. Following the frying process, the final product exhibited an ideal water activity (*a_w*) of below 0.3. The precursor content within the potato varieties utilized by the Costa Rican industry is consistent with findings reported by several authors [29–31]. These studies showed that reducing sugar content is considerably lower than asparagine content. Asparagine, specifically, is recognized as the primary precursor for the formation of induced ACR in fried potato products, and its formation can be mitigated by reducing the initial ASN concentration [32]. A positive correlation was observed between the formation of ASN and ACR after frying the raw potato samples. Significant relationships were found between precursor ASN and ACR content in potato products. A Spearman's rho of 0.829 (*p* < 0.05, *n* = 6) was determined and illustrated in Fig. 1A.

Bassama and coworkers [33] reported undetectable levels of up to 321.5 mg ASN/100 g (dry basis) in a wide variety of plantain cultivars, noting that this precursor decreases with increasing postharvest time. In the case of the plantain chips evaluated, the asparagine content of the Curare variety was very similar to that of the Harton variety [32]. There are no significant differences in the ASN values between potatoes and plantains (*p* > 0.05). However, a significant relationship was observed between precursor ASN and ACR content in potato products, but not in plantain crisps. A Spearman's rho of 0.600 (*p* > 0.050, *n* = 4) was determined and illustrated in Fig. 1B.

Cassava slices presented significantly lower ASN content (*p* < 0.05) than potatoes and plantains, which is reflected in the considerably lower ACR content (*p* < 0.05) compared to plantains and potatoes. This, together with the results observed in potatoes, reinforces the hypothesis that nitrogenous compounds are the primary drivers of ACR formation [34,35].

No significant relationship between fructose or glucose content and ACR in any of the vegetable crisps evaluated (Tables 1,2,3 and Fig. 1A,B, *p* > 0.05), even if they

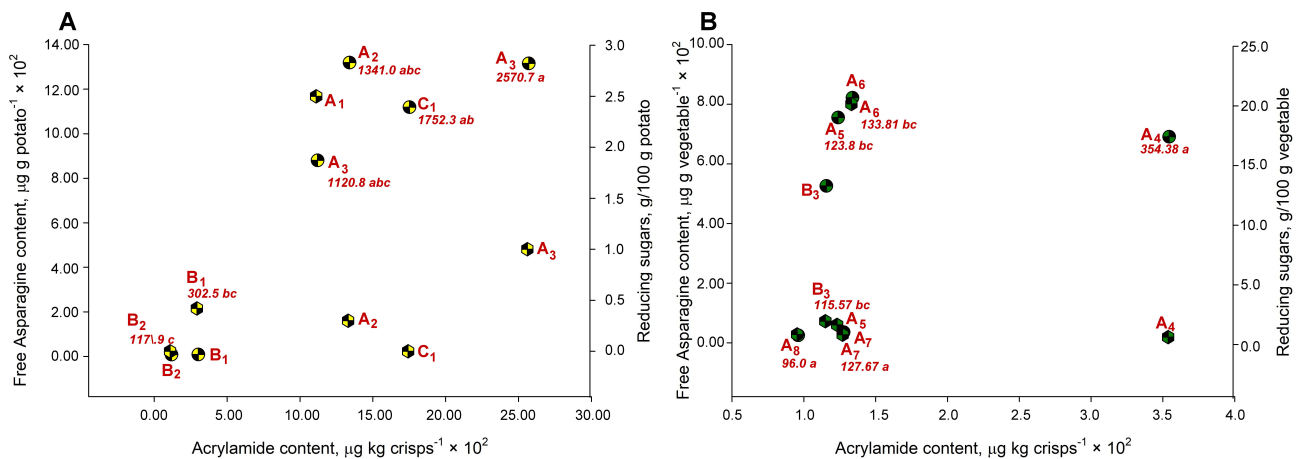


Fig. 1. Multiple-axis chart showing the spatial distribution among three simultaneous variables, acrylamide (ACR) and two different precursors in fried crisps. Correlation between the acrylamide (ACR) content in fried chips (A: potato; B: plantain and cassava) and the content of precursors asparagine (asparagine (ASN), hourglass circles) and reducing sugars (Σ fructose and glucose, hourglass hexagons) in the corresponding raw plant materials. Different lowercase letters represent significant differences at $p < 0.05$. The analyzed samples of both chips and raw materials are produced in Costa Rica by three companies coded as A, B and C.

Table 1. Characterization of the raw material and crisps of each of the types of potato-based snacks produced by three different Costa Rican companies.

Parameter	Material	Potato crisp presentation					
		Corrugated		Slice			Stick
		A ₁	B ₁	A ₂	B ₂	C ₁	A ₃
pH	Raw	5.80 ± 0.16	5.81 ± 0.06	5.59 ± 0.26	5.83 ± 0.06	5.86 ± 0.06	5.88 ± 0.19
Starch (g 100 g ⁻¹)	Raw	11.0 ± 3.7	10.1 ± 1.7	10.88 ± 0.95	11.2 ± 1.4	13.8 ± 5.4	10.9 ± 1.6
Asparagine (µg g ⁻¹)	Raw	153 ± 47	1.57 ± 0.29	217 ± 129	1.75 ± 0.32	225 ± 79	224 ± 33
Moisture (g 100 g ⁻¹)	Raw	82.3 ± 2.9	81.7 ± 1.4	84.7 ± 2.1	81.67 ± 0.84	80.67 ± 5.8	83.7 ± 1.5
	Crisps	2.45 ± 0.34	1.27 ± 0.18	2.98 ± 0.69	1.92 ± 0.68	2.05 ± 0.50	3.1 ± 1.6
a _w	Raw	0.9960 ± 0.0011	0.9943 ± 0.0006	0.9982 ± 0.0007	0.9948 ± 0.0019	0.9967 ± 0.0004	0.9955 ± 0.0010
	Crisps	0.239 ± 0.040	0.0847 ± 0.020	0.241 ± 0.065	0.139 ± 0.050	0.074 ± 0.016	0.23 ± 0.15
Glucose (g 100 g ⁻¹)	Raw	(ND < 0.05)–0.37	ND < 0.09	(ND < 0.05)–0.14	ND < 0.09	0.155 ± 0.048	(ND < 0.06)–0.24
	Crisps	(ND < 0.25)–1.55	ND < 0.98	ND < 0.21	ND < 0.75	(ND < 0.08)–1.55	ND < 0.18
Fructose (g 100 g ⁻¹)	Raw	(ND < 0.06)–0.52	0.079 ± 0.042	ND < 0.05	ND < 0.03	0.040 ± 0.013	(ND < 0.05)–0.10
	Crisps	(ND < 0.20)	ND < 0.33	ND < 0.25	ND < 0.30	ND < 0.07	ND < 0.25
Sucrose (g 100 g ⁻¹)	Raw	(ND < 0.02)–0.40	0.41 ± 0.15	(ND < 0.04)–0.23	0.31 ± 0.13	0.036 ± 0.011	(ND < 0.02)–0.36
	Crisps	(NC < 0.61)–1.81	ND < 1.0	(ND < 0.20)–1.29	ND < 1.46	(NC < 0.22)–0.38	ND < 0.23
Fat (g 100 g ⁻¹)	Crisps	36.1 ± 3.6	41.0 ± 2.3	29.5 ± 7.5	30.8 ± 6.4	27.4 ± 1.1	34.83 ± 0.95
Acrylamide (µg kg ⁻¹)	Crisps	1093 ± 157	298 ± 258	1301 ± 607	116 ± 71	1716 ± 1574	2491 ± 1536

ND, not detectable; NC, not quantifiable. Three different Costa Rican companies coded as A, B and C.

had been depleted during the frying process (reducing sugars as substrate for the Maillard reaction and sucrose because of its hydrolysis) [36]. Ripe plantain chips showed significantly higher concentrations ($p < 0.05$) of reducing sugars than the other samples; unripe plantain presented a content significantly similar to that of potatoes and cassava ($p > 0.05$). Despite this, potatoes presented significantly higher ACR contents ($p < 0.05$), supporting the thesis that the relationship between reducing sugars and ACR formation is complex [36]. It is possible to explain that the low ACR content in the plantain crisps samples may be because

the moisture content of the raw material is lower than that of the potatoes, which allows for low heat and mass transfer during frying [37], in addition to the fact that the plantains undergo pre-frying that can decrease both the a_w and the moisture content before the final frying process [33].

The surface-to-volume ratio (SVR) seems to play a crucial role in ACR formation. The potato product with the highest ACR values ($p < 0.05$) was the extruded potato sticks (Table 1 A₃, Fig. 2A SPH), which have a high SVR (approximately 10.4 mm⁻¹) compared to the smooth and

Table 2. Characterization of the raw material and crisps of each plantain-based snack produced by two different Costa Rican companies.

Parameter	Material	Plantain crisp presentation			
		“Patacón”		Unripe slice plantain	Ripe Slice plantain
		A ₄	B ₃	A ₅	A ₆
pH	Raw	5.63 ± 0.81	6.05 ± 0.42	5.23 ± 0.45	4.78 ± 0.05
Starch (g 100 g ⁻¹)	Raw	20.1 ± 6.6	23.8 ± 2.3	22.5 ± 6.1	19.7 ± 7.1
Asparagine (µg g ⁻¹)	Raw	268 ± 97	216 ± 45	298 ± 104	323 ± 28
Moisture (g 100 g ⁻¹)	Raw	62.7 ± 1.4	60.4 ± 1.2	61.2 ± 1.1	62.1 ± 2.4
	Crisps	3.21 ± 1.07	1.36 ± 0.62	3.88 ± 0.62	4.02 ± 0.3
a _w	Raw	0.9937 ± 0.0012	0.9931 ± 0.0011	0.9932 ± 0.0019	0.9761 ± 0.0016
	Crisps	0.199 ± 0.1039	0.0775 ± 0.0265	0.2406 ± 0.0833	0.2713 ± 0.0503
Glucose (g 100 g ⁻¹)	Raw	0.157 ± 0.060	0.57 ± 0.11	0.38 ± 0.18	4.09 ± 0.69
	Crisps	ND <0.21	ND <0.28	ND <0.26	(ND <0.27)–1.24
Fructose (g 100 g ⁻¹)	Raw	0.39 ± 0.32	0.229 ± 0.029	0.51 ± 0.18	3.77 ± 0.41
	Crisps	ND <0.18	ND <0.24	ND <0.23	(ND <0.27)–1.53
Sucrose (g 100 g ⁻¹)	Raw	0.71 ± 0.62	0.65 ± 0.11	0.79 ± 0.48	0.93 ± 0.76
	Crisps	NC <0.58	ND <1.54	(ND <0.20)–2.26	9.61 ± 5.2
Fat (g 100 g ⁻¹)	Crisps	36.03 ± 1.12	26.4 ± 2.5	19.2 ± 1.77	24.5 ± 9.97
Acrylamide (µg kg ⁻¹)	Crisps	343 ± 46	114 ± 63	120 ± 63	128 ± 22

ND, not detectable; NC, not quantifiable. Three different Costa Rican companies coded as A, B and C.

Table 3. Characterization of the raw material and crisps for each type of cassava-based snacks produced by a Costa Rican company.

Parameter	Material	Cassava crisp presentation	
		“Patacón”	Slice
		A ₇	A ₈
pH	Raw	5.49 ± 1.15	5.32 ± 0.59
Starch (g 100 g ⁻¹)	Raw	33.9 ± 5.0	33.7 ± 6.0
Asparagine (µg g ⁻¹)	Raw	13.9 ± 8.1	9.93 ± 0.46
Moisture (g 100 g ⁻¹)	Raw	60.8 ± 2.9	60.6 ± 6.9
	Crisps	3.66 ± 0.85	2.08 ± 0.70
a _w	Raw	0.9945 ± 0.0010	0.9943 ± 0.0022
	Crisps	0.2287 ± 0.097	0.0848 ± 0.0361
Glucose (g 100 g ⁻¹)	Raw	0.37 ± 0.21	0.23 ± 0.06
	Crisps	ND <0.23	ND <0.28
Fructose (g 100 g ⁻¹)	Raw	0.48 ± 0.09	0.51 ± 0.23
	Crisps	ND <0.20	ND <0.24
Sucrose (g 100 g ⁻¹)	Raw	1.72 ± 0.25	1.67 ± 0.52
	Crisps	1.33 ± 0.30	(NC <0.67)–1.81
Fat (g 100 g ⁻¹)	Crisps	27.63 ± 4.81	21.8 ± 4.39
Acrylamide (µg kg ⁻¹)	Crisps	123 ± 47	94 ± 46

ND, not detectable; NC, not quantifiable.

wavy sliced samples (approximately 2.72 mm⁻¹). This finding aligns with established principles of heat transfer and chemical kinetics during the frying process. It has been demonstrated that with a larger exposed area, heat transfer is increased, leading to greater water loss and, consequently, higher ACR formation [38]. Additionally, a larger exposed area allows for the absorption of a higher fat con-

tent [39]. Supporting this, Gökmen and Palazoğlu [40] observed that changing the slice geometry from strip to cube (resulting in a 1.4-fold increase in SRV) led to a 2.4-fold increase in ACR concentration. They attributed this to the higher SVR, which exposes more precursors to elevated surface temperatures, thereby promoting greater ACR formation.

In the case of plantain crisps, it was observed that product “patacón” (Table 2 A₄, Fig. 2B STE), shaped (larger surface area and volume), presented a higher amount of fat, and it’s the product with the highest ACR values ($p < 0.05$) compared with other plantain products. This same behavior occurs with the cassava’s “patacón” (Table 3 A₇, Fig. 2C SYB), whose content is significantly higher than that of the other cassava products ($p < 0.05$), yielding a value almost double. Studies show that ACR formation occurs mainly at or near the surface, as the maximum temperature during frying is reached much faster on the surface than in the interior, accompanied by rapid moisture loss and oil absorption. Consequently, reducing SRV decreases the ACR content [41].

3.2 Levels of ACR Found in Snack Foods

It was observed (Fig. 2A) that the maximum ACR values for potato crisps are ten-fold higher ($p < 0.05$) than those of their plantain and cassava counterparts. In 75% of the potato samples analyzed, the average ACR content did not exceed the 750-µg kg⁻¹ legal threshold set for these foods. For example, among five sliced potato crisp samples (SPA, SPB, SPC, SPE, SPF), no significant variance was found ($p > 0.05$); therefore, the ACR overall mean value lies at 658 ± 218 µg kg⁻¹, and none of these sam-

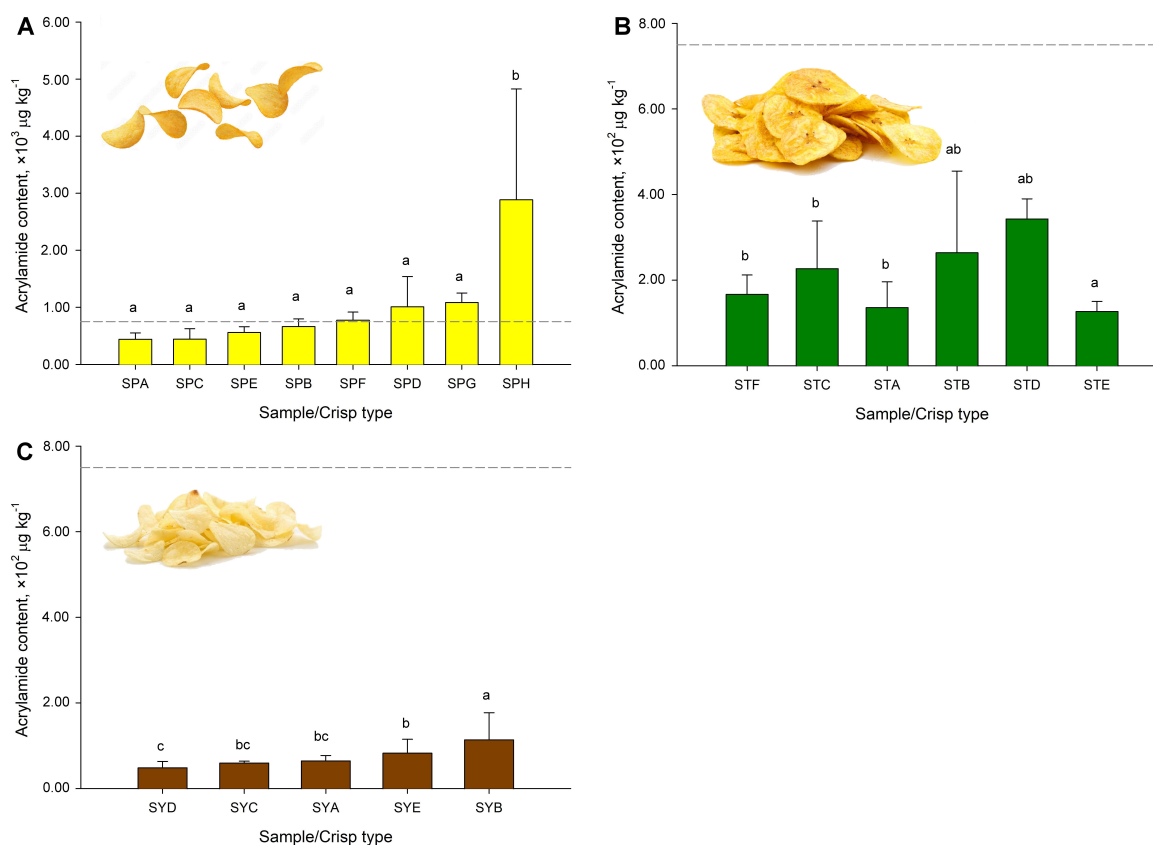


Fig. 2. Acrylamide (ACR) formation during the frying process of different snacks, each colored bar represents a distinct matrix or sample presentation. Acrylamide (ACR) content in (A) sliced potato crisps (SPA, SPB, SPC, SPE, SPF), potato corrugated crisps (SPG) and stick-cut potato crisps (SPH), (B) unripe-sliced plantain crisps (STA, STB, STC, STD), unripe plantain “patacón” (STE), ripe-sliced plantain crisps (STF), (C) sliced cassava crisps (SYA, SYC, SYD, SYE), cassava “patacón” (SYB), commercially available in Costa Rica with respect to the benchmark of $750 \mu\text{g kg}^{-1}$. Dissimilar lowercase letters (a, b, c) represent significant differences at $p < 0.05$ between samples within each product type. Dissimilar lowercase letters represent significant differences at $p < 0.05$ between samples within each product type.

ples differ from the aforementioned legal threshold ($p > 0.05$). The SPD and SPF samples showed an average ACR level exceeding the EU limit; however, this finding was not statistically significant ($p > 0.05$). Similarly, the SPG sample, which was a corrugated potato crisp, showed an average ACR level above the EU limit; however, this difference was not statistically significant ($p > 0.05$). Only the stick-cut potato crisps (SPH) sample was statistically above the permitted level ($p < 0.05$), a result that aligns with previous findings, which show that this type of cut favors the formation of ACR during the frying process.

The ACR content in potato crisps produced in Costa Rica is similar to, or even lower than, that found in other locations. According to Mesias and coworkers [21] although the average ACR content found in 2019 in potato snacks marketed in Spain ($664 \mu\text{g kg}^{-1}$, range $89\text{--}1930 \mu\text{g kg}^{-1}$) has decreased over the years (55.3% lower than in 2004 and 10.3% lower than in 2008), 27% of the samples still exhibited concentrations above the reference level established in the Regulation ($750 \mu\text{g kg}^{-1}$), suggesting that ef-

forts to reduce ACR formation in this industry should be continued. Data from Ethiopia showed maximum values of ACR reported in potato crisps as high as $3515 \mu\text{g kg}^{-1}$ [42], Pakistan’s products reported $2649.8 \pm 1.1 \mu\text{g kg}^{-1}$ as maximum [43], and a study in Bangladesh showed that only 40% of local potato crisps exceeded the reference level and ranged from $461 \mu\text{g kg}^{-1}$ to $2129 \mu\text{g kg}^{-1}$ [44]. ACR levels also varied widely ($166.7\text{--}1101.8 \text{ mg kg}^{-1}$) in 14 original-cut potato chip brands in China’s markets [45].

Unlike potato crisps, the “alternative” tubers used in this experiment (plantain and cassava) showed relatively the lowest ACR production (Fig. 2B,C). The unripe plantain-based products (STA, STB, STC, STD) had ACR levels ranging from $137 \mu\text{g kg}^{-1}$ to $265 \mu\text{g kg}^{-1}$, which are comparable to previous findings in Malaysian ($28\text{--}243 \mu\text{g ACR kg}^{-1}$) [46] and Colombian products ($130.4 \mu\text{g ACR kg}^{-1}$) [47]. In ripe plantain, the average ACR level was $128 \pm 22 \mu\text{g kg}^{-1}$. This value is considerably lower than those reported by other researchers [48,49], who found values of $894.40 \pm 56.35 \mu\text{g ACR kg}^{-1}$ and a maximum of

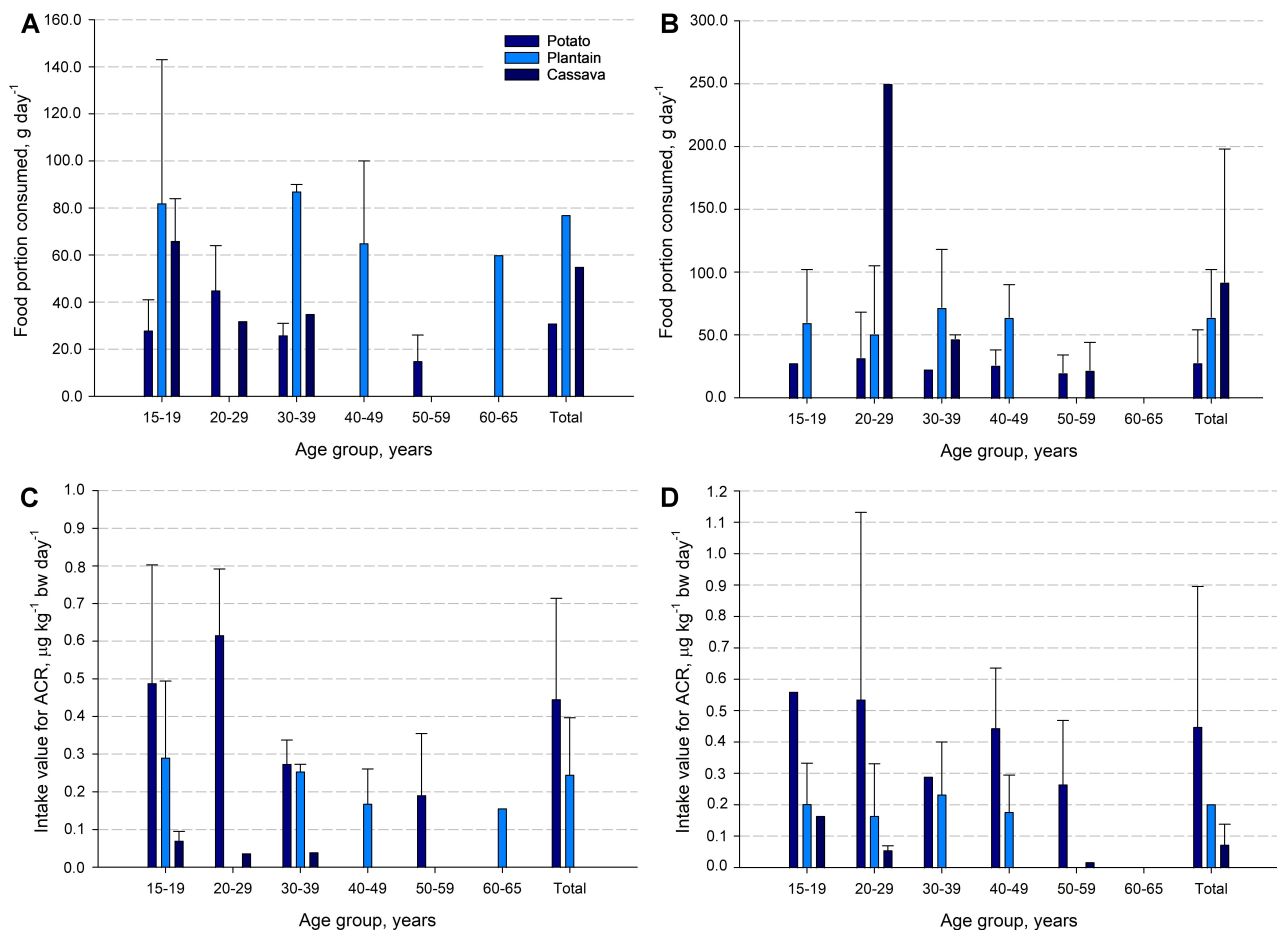


Fig. 3. General population consumption practices in Costa Rica regarding snack foods. Crisps product consumption by gender and age (A: men and B: women) according to the study Latin American Nutrition and Health Study (ELANS) [28], and estimated acrylamide exposure value ($\text{ng kg}^{-1} \text{bw day}^{-1}$) by crisps products consumption trends among the local population for Costa Rican (C: men and D: women).

1574 $\mu\text{g ACR kg}^{-1}$, respectively. However, it is consistent with the level documented by Barón Cortés and coworkers [47]. This discrepancy highlights the critical role of the ripeness stage in ACR formation in plantain, a factor previously noted by Shamla and Nisha [50].

The average ACR content in the cassava products processed in this study was 75.2 $\mu\text{g/kg}$, a value similar to the 59.42 $\mu\text{g/kg}$ reported by Díaz-Ávila and coworkers [48]. A survey by González-Cuello and coworkers [51] found a significantly higher maximum ACR content of 101.10 mg/kg when cassava chips were fried at 180 °C for 5 minutes. These types of products are a good alternative for consumers who prefer fried chips and have a low ACR content [52].

3.3 ACR Exposure in the Costa Rican Population Through Snacks

The synthesis of ACR in foods is dependent on a myriad of factors (including cooking temperature, time, pH, surface/area ratio, moisture content, cooking method, use of fertilizer, harvesting, cultivar, and storage conditions, to

name a few), which may or may not be easy to manipulate or even assess. Hence, ACR behavior is complex, and even toxin bioavailability comes into play when considering exposure [18]. Interestingly, recent findings support a deleterious impact of ACR within the gastrointestinal tract [53].

As reported in a consumption survey conducted by the Latin American Nutrition and Health Study (ELANS) [28], among $n = 798$ individuals aged 15 to 65 years. Only 19.55% of this population consumes fried crisps (Fig. 3A,B, Ref. [28]). The results revealed that corn-based crisps are the most consumed (61.53%), followed by potato-based crisps (19.23%), then plantain-based crisps (12.82%), and finally cassava-based crisps (6.41%). It was observed that the portion of crisps varies considerably by gender and age (Fig. 3A,B). In the case of potato crisps, both men and women consume a daily portion of 30 g, whereas for plantains, the average daily intake increases to 70 g per day (Fig. 3A,B). However, men consume a larger portion (77 g per day) than women (64 g per day). Unlike cassava crisps,

women consume more (92 g per day) than men (55 g per day; Fig. 3A,B).

The exposure value of ACR ($\text{ng kg}^{-1} \text{ bw day}^{-1}$) in the population consuming the products under study was obtained taking into account gender and age, and the results are presented in Fig. 3C,D. An average of 447 ± 378 , 224 ± 138 , and 65 ± 41 was obtained for potato, plantain, and cassava, respectively. These results are very similar to those obtained in other studies, for example, in Lebanese potato/corn crisps, the average daily intake of the population was $453 \pm 251 \text{ ng kg}^{-1} \text{ bw day}^{-1}$ [54]. In an Algerian risk assessment analysis, the major contributors to ACR exposure were potato fries, with an estimated exposure of 0.2 to 0.4 $\mu\text{g kg}^{-1} \text{ bw day}^{-1}$ [12]. Within the Italian market, the values in crisps were 0.172 and 0.901 $\mu\text{g kg}^{-1} \text{ bw day}^{-1}$ [55]. The tolerable daily intake (TDI) for neurotoxicity from ACR was estimated as 40 $\mu\text{g kg}^{-1} \text{ bw}$; TDIs for cancer were estimated as 2.6 $\mu\text{g kg}^{-1} \text{ day ca. } 200 \text{ ng kg}^{-1} \text{ bw day}^{-1}$ [56,57], which indicates that there may be a risk to the population that consumes these crisps daily since the values are very close to this tolerable value and are even higher.

4. Conclusion

This preliminary investigation analyzed the chemical safety of fried snacks on the Costa Rican market. Our findings indicate that although potato chips produced the highest quantities of ACR upon frying, only one of the products examined statistically surpassed the EU-established maximum levels. This was likely attributable to its high surface-to-volume ratio. These results underscore the importance of advising manufacturers to implement mitigation strategies, such as controlling the physical properties of the raw ingredients, to reduce ACR formation and guarantee product safety. Moreover, plantain and cassava present promising alternatives for fried snacks. Their inherent asparagine content and processing methods result in significantly lower ACR levels compared to those found in potato-based products. Despite this, further research is warranted to optimize production conditions. The estimated daily ACR intake from these snacks alone suggests a potential health risk for the Costa Rican population, given their consumption habits and typical portion sizes.

Availability of Data and Materials

The datasets generated and analyzed during the present study are available upon reasonable request. The data is not publicly available due to privacy and ethical restrictions.

Author Contributions

Conceptualization, CCH and MQV; Data curation, MQV, CCH, and FGC; Formal analysis, MQV, CCH, FGC, and GA; Funding acquisition, CCH and GA; Investiga-

tion, MQV, CCH, FGC, and GA; Methodology, CCH, FGC, and GA; Resources, CCH and GA; Supervision, CCH, FGC, and GA; Writing—original draft, CCH and FGC. All authors contributed to editorial changes in the manuscript. All authors read and approved the final manuscript. All authors have participated sufficiently in the work and agreed to be accountable for all aspects of the work.

Ethics Approval and Consent to Participate

Not applicable.

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Conflict of Interest

The authors declare no conflict of interest.

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